Electronic Supplementary Information

Size-exclusive and coordination-induced selective dyes adsorption in a nanotubular metal-organic framework

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BUT-51				
Formula	$C_{32}H_{18}Co_2N_3NaO_{12}\\$			
Fw	777.34			
temp/K	173(2)			
cryst syst	tetragonal			
space group	$P4_{2}/m$			
<i>a</i> (Å)	14.8967(2)			
<i>c</i> (Å)	10.0502(3)			
$V(Å^3)$	2230.26(8)			
Ζ	2			
$D_c (\mathrm{mg}\cdot\mathrm{m}^{-3})$	1.158			
$\mu/(\mathrm{mm}^{-1})$	6.367			
F (000)	784			
rflns collected	4616			
unique rflns	2308			
$R_{\rm int}$	0.0275			
GOF on F^2	1.060			
R_1^a (I>2sigmaI)	0.0632			
wR_2 a	0.1874			

Table S	1. Crystal Data and Structure Refinement Details of BUT-51	

^a R₁ = $||F_0| - |F_c|| / |F_0|$. $wR_2 = [w(|F_0^2| - |F_c^2|)^2 / w|F_0^2|^2]^{1/2}$.

BUT-51					
Co(1)–O(4)	2.036(5)	Co(1)–N(1) ^{#2}	2.003(7)		
Co(1)-O(1)	2.065(3)	Co(1)–O(3)	2.379(5)		
Na(1)–O(1)	2.445(3)	Na(1)–O(1) ^{#1}	2.114(9)		
Na(1)–O(3)	2.445(3)				
O(4)-Co(1)-O(1)	123.33(13)	O(1)#1–Co(1)–O(1)	89.50(17)		
O(4)-Co(1)-N(1) ^{#2}	112.3(2)	O(1)-Co(1)-N(1) ^{#2}	101.97(13)		
O(4)-Co(1)-O(3)	59.3(2)	O(1)-Co(1)-O(3)	83.89(13)		
N(1) ^{#2} -Co(1)-O(3)	171.65(19)	O(1)#3–Na(1)–O(1)	180.0(2)		
$O(1)^{#3}$ -Na(1)-O(1) ^{#1}	107.04(15)	O(1)-Na(1)-O(1) ^{#1}	72.96(15)		
O(1)-Na(1)-O(3)	74.55(12)	O(1) ^{#3} –Na(1)–O(3)	105.45(12)		

Table S2. Selected bond distances (Å) and angles (°) in BUT-51.

Symmetry codes #1: x, y, -z; #2: -y, x, z - 1/2; #3, -x + 1, -y, -z



13.1×5.8×4.5 Å

OH





Fig. S2. TGA curves of as-synthesized BUT-51 and AH@BUT-51.



Fig. S3. UV-vis spectra of acetone solutions of SY7 and AY1 in the presence of BUT-51 monitored with time.



Fig. S4. The dyes release of AH@BUT-51 in the saturated acetone solution of NH₄Cl.



Fig. S5. Solid-state UV-Vis spectra of BUT-51 and AH@BUT-51.