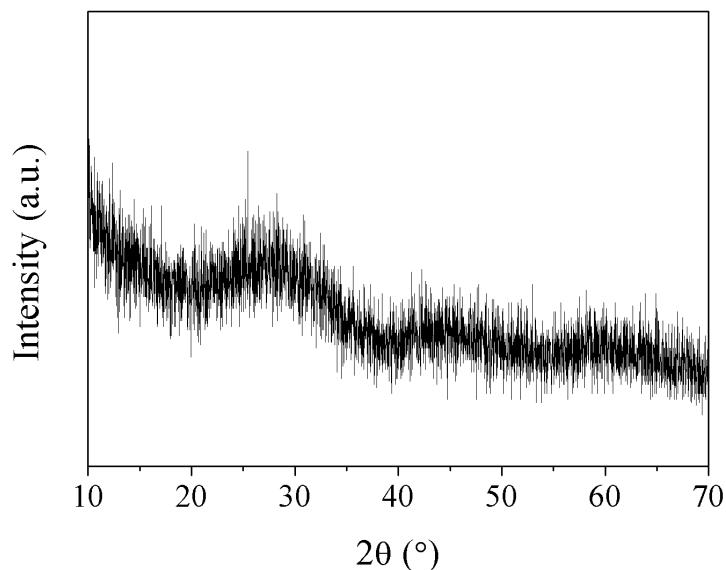


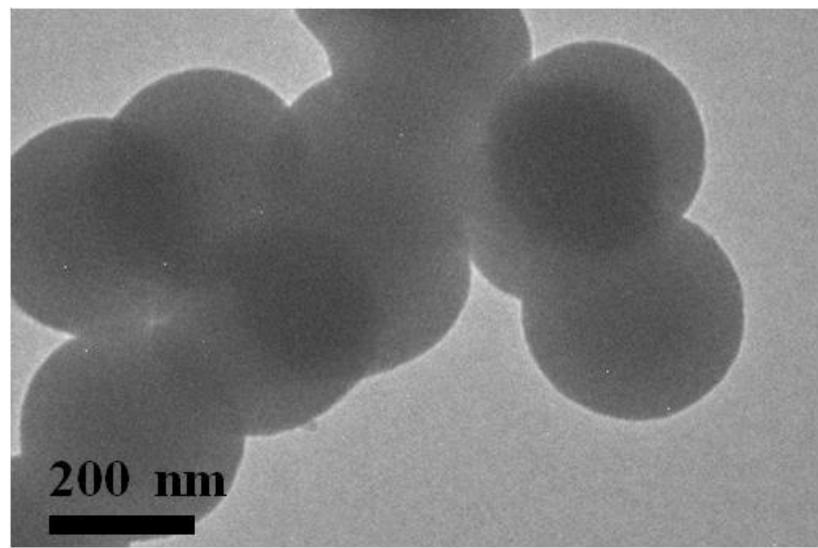
## **Supporting Information**

### **Hollow Titanium Dioxide Spheres as Anode Material for Lithium Ion Battery with Largely Improved Rate Stability and Cycle Performance by Suppressing the Formation of Solid Electrolyte Interface Layer**

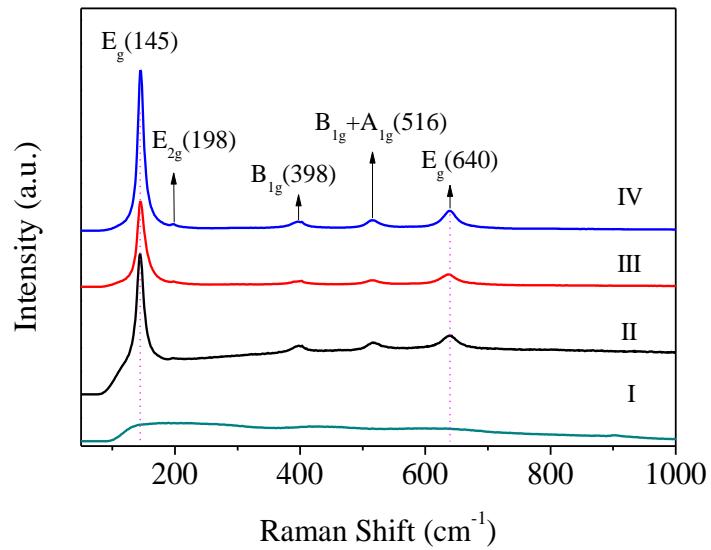
Cuiping Han,<sup>a,b,c</sup> Di Yang,<sup>a,d</sup> Yingkui Yang,<sup>a,e</sup> Beibei Jiang,<sup>a</sup> Yanjie He,<sup>a</sup> Mengye Wang,<sup>a</sup>  
Ah-Young Song,<sup>a</sup> Yan-Bing He,<sup>b</sup> Baohua Li,<sup>b\*</sup> and Zhiqun Lin<sup>a\*</sup>



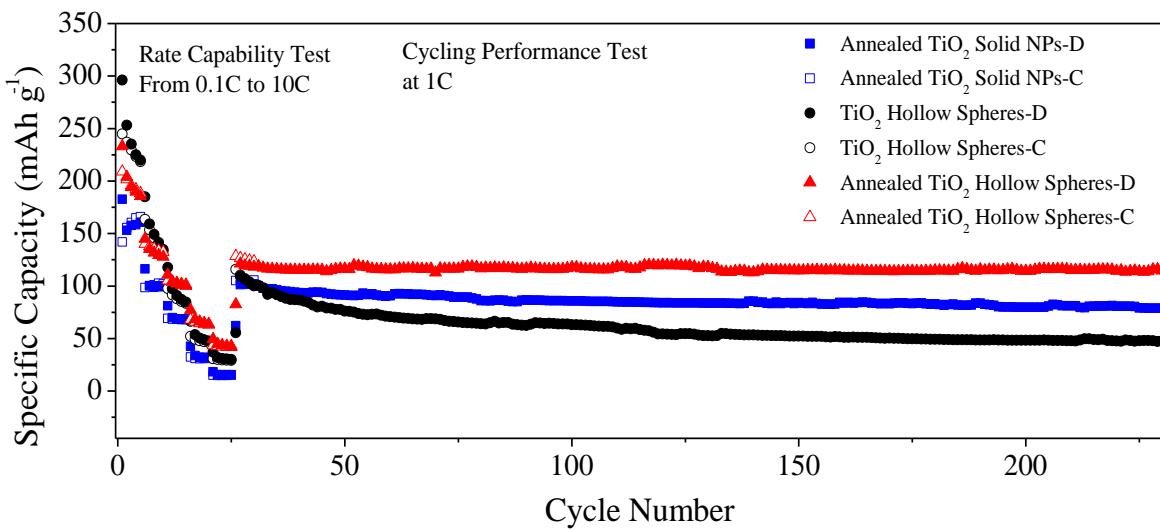
**Figure S1.** XRD patterns of as-prepared colloidal TiO<sub>2</sub> nanoparticles.



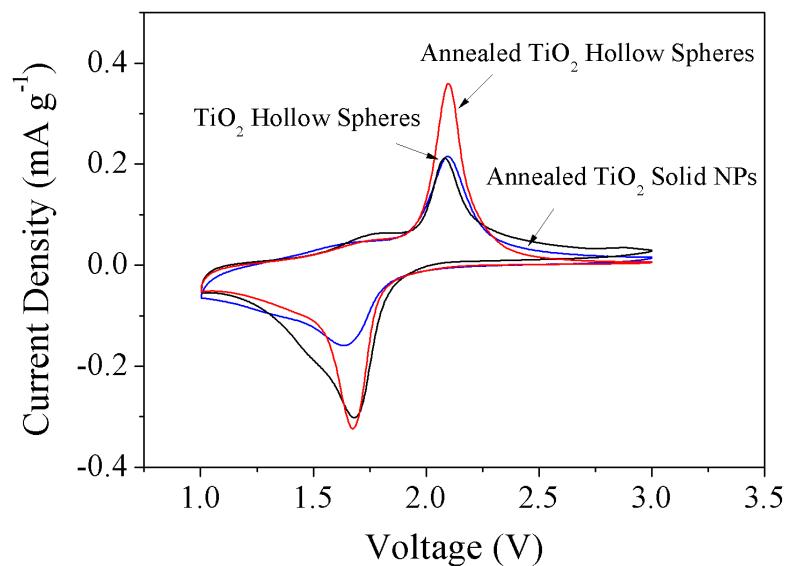
**Figure S2.** TEM image of as-prepared colloidal  $\text{TiO}_2$  nanoparticles.



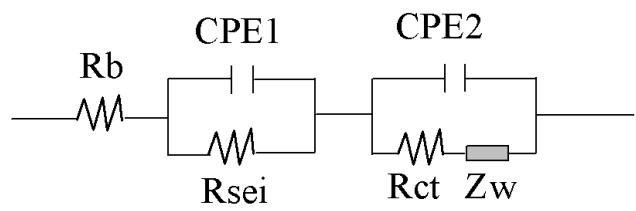
**Figure S3.** Raman spectra of (I) as-prepared colloidal  $\text{TiO}_2$  nanoparticles, (II) annealed  $\text{TiO}_2$  solid nanoparticles, (III)  $\text{TiO}_2$  hollow spheres after hydrothermal treatment, and (IV) annealed  $\text{TiO}_2$  hollow spheres.



**Figure S4.** Rate capability and cycling performance tests as demonstrated in Figure 6. Note that the figure is divided into Figure 6a and 6b for a better comparison.



**Figure S5.** CV characteristics of annealed  $\text{TiO}_2$  solid nanoparticles,  $\text{TiO}_2$  hollow spheres, and annealed  $\text{TiO}_2$  hollow spheres at a scanning rate of  $0.2\text{mVs}^{-1}$ .



**Figure S6.** The equivalent circuit used for the EIS simulation in Figure 6.