

Robust α -Fe₂O₃ nanorod arrays with optimized interstices as high-performance 3D anodes for high-rate lithium ion batteries

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Electronic supplementary information

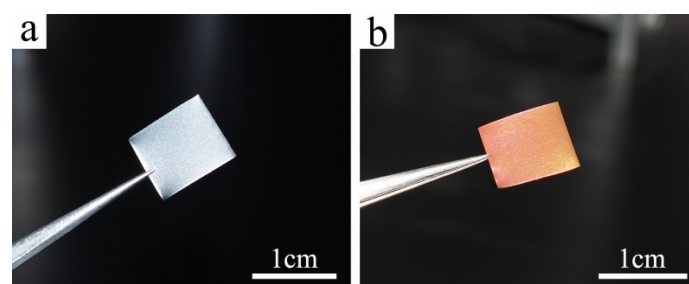


Fig. S1. Optical photographs of Ti foil: (a) bare, (b) after growth of α -Fe₂O₃ nanorod arrays.

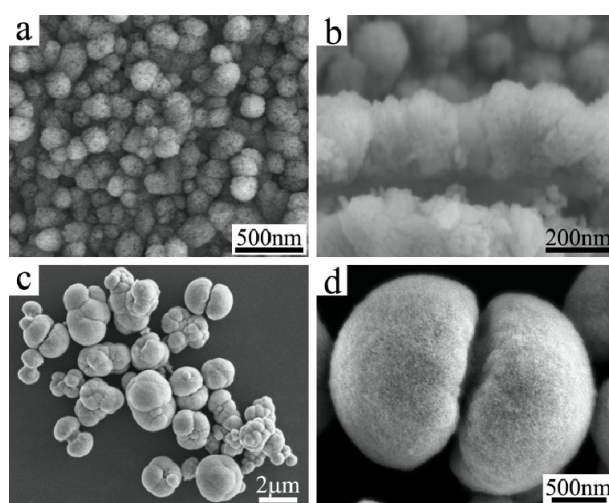


Fig. S2. SEM images of α -Fe₂O₃ products obtained with different concentrations of $\text{Fe}_2(\text{SO}_4)_3$: (a, b) 2.5 mM, (c, d) 15 mM.

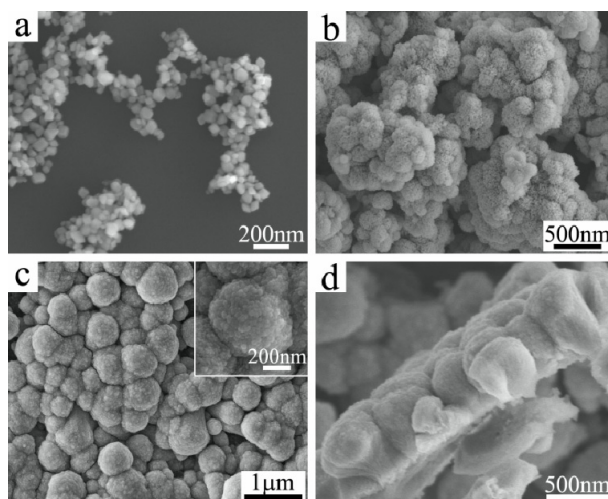


Fig. S3. SEM images of α -Fe₂O₃ products obtained with different concentrations of HAc: (a) 0 M, (b) 2.1 M, (c, d) 6.3 M.

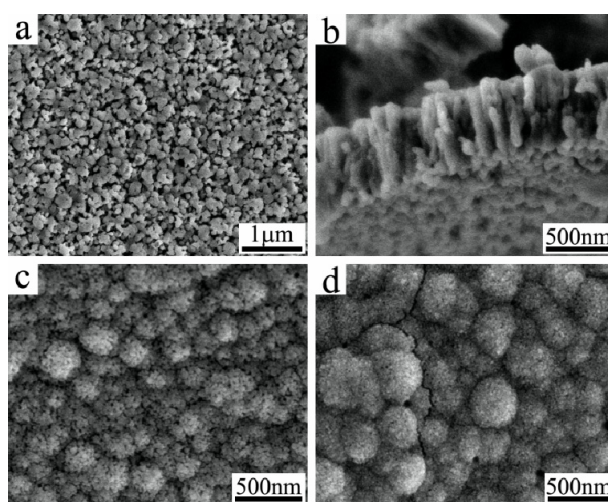


Fig. S4. SEM images of α -Fe₂O₃ nanorod arrays obtained with different concentrations of Na₂SO₄: (a, b) 0 M, (c) 0.3 M, and (d) 0.9 M.

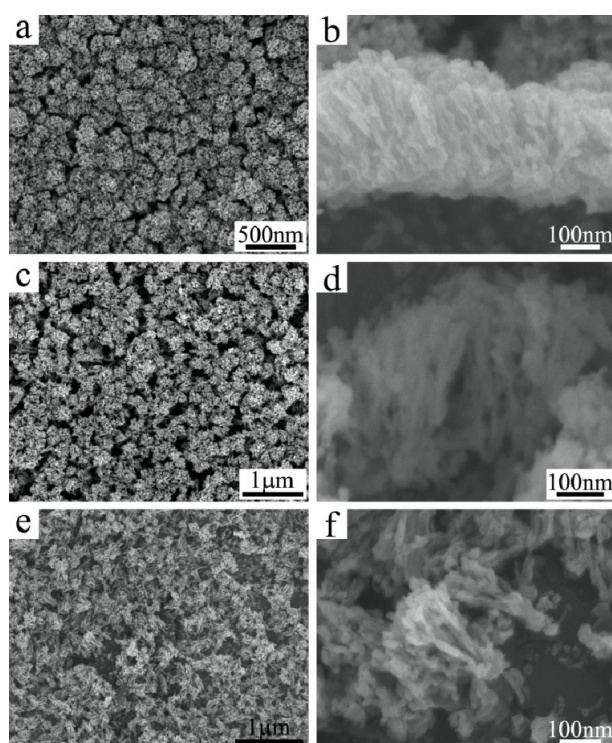


Fig. S5. SEM images of α -Fe₂O₃ nanorod arrays after etching by oxalic acid for different times: (a, b) 1 h, (c, d) 1.5 h, (e, f) 2 h.

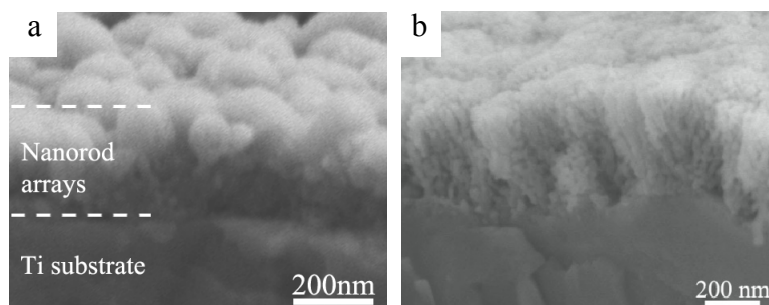


Fig. S6. Cross-sectional SEM images of α -Fe₂O₃ nanorod arrays on Ti foil before etching without calcination (a) and after etching for 1.5 h and calcination at 450 °C for 30 min (b).

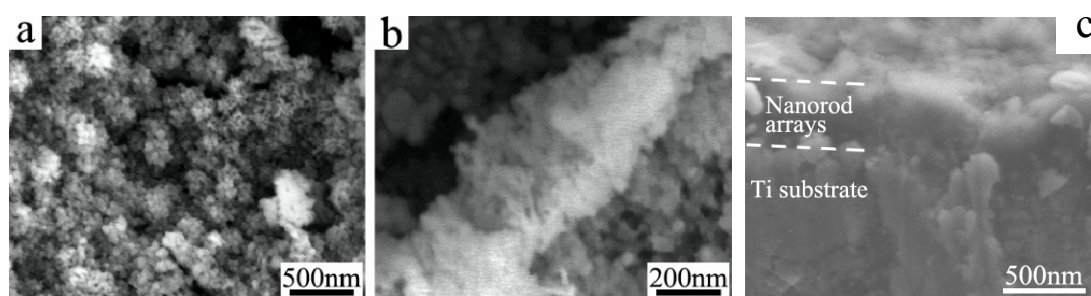


Fig. S7. (a, b) SEM images of Fe₂O₃-NA-1.5 after 500 cycles at 5 °C and (c) cross-sectional SEM image of Fe₂O₃-NA-1.5 on Ti foil after 100 cycles at 5 °C.

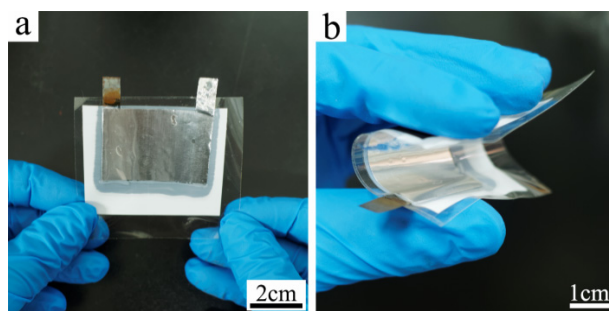


Fig. S8. Optical photographs of flexible LIB full battery.

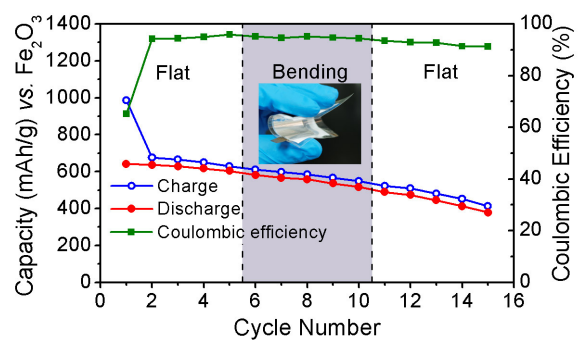


Fig. S9. Cycling performance of a flexible lithium ion battery at 5 C at varied bending states.