

Supporting information for:
Ammonium-based Protic Ionic Liquid Doped
Nafion Membrane as Anhydrous Fuel Cell
Electrolyte

Anurag Prakash Sunda^{*,†,‡}

*Chemistry and Physics of Materials Unit, Jawaharlal Nehru Centre for Advanced Scientific
Research, Bangalore 560064, INDIA, and Materials Research Laboratory, Department of
Chemistry, Pandit Deendayal Upadhyay Shekhawati University, Sikar 332001, Rajasthan, INDIA.*

E-mail: anurag.sunda@gmail.com

*To whom correspondence should be addressed

[†]Chemistry and Physics of Materials Unit, Jawaharlal Nehru Centre for Advanced Scientific Research, Bangalore 560064, INDIA

[‡]Materials Research Laboratory, Department of Chemistry, Pandit Deendayal Upadhyay Shekhawati University, Sikar 332001, Rajasthan, INDIA.

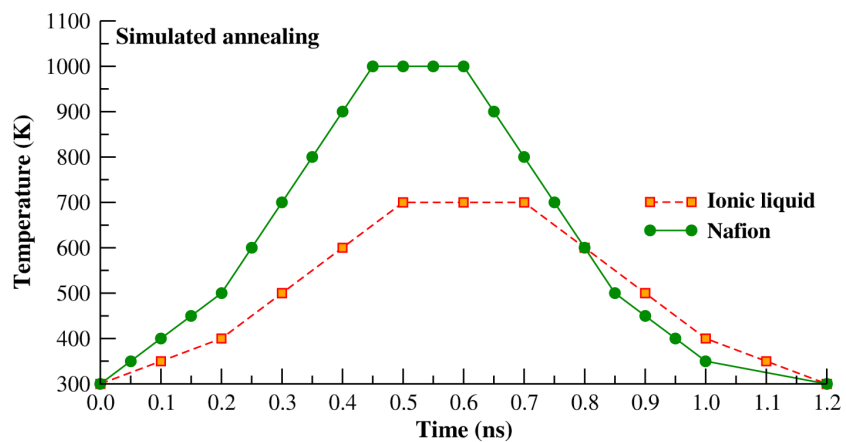


Figure S1: Variation in temperature with time for simulated annealing.

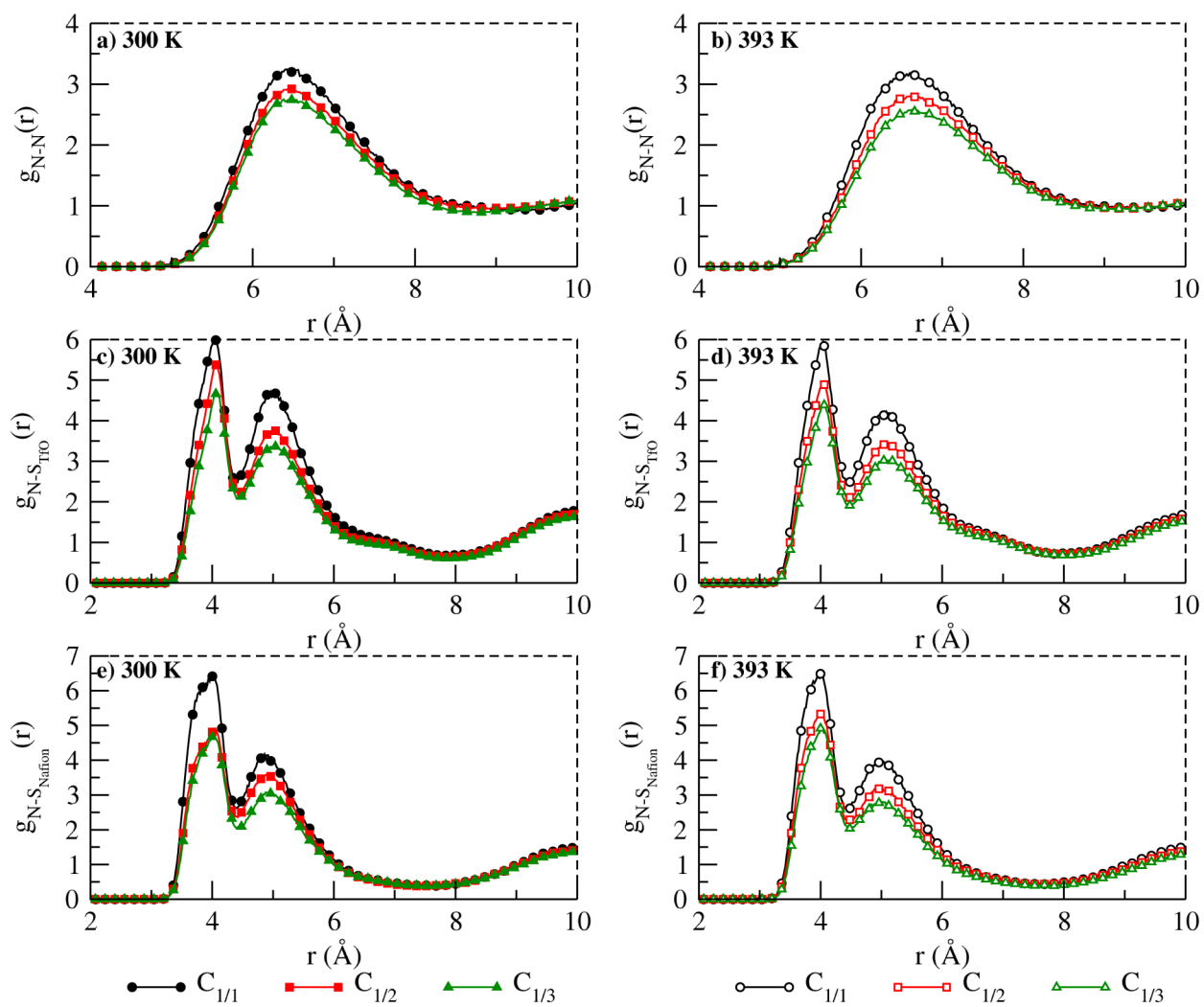


Figure S2: RDFs for ammonium cation interactions at 300 K and 393 K in (a, b) N-N, (c, d) N-S_{TiO}, and (e, f) N-S_{Nafion} respectively.

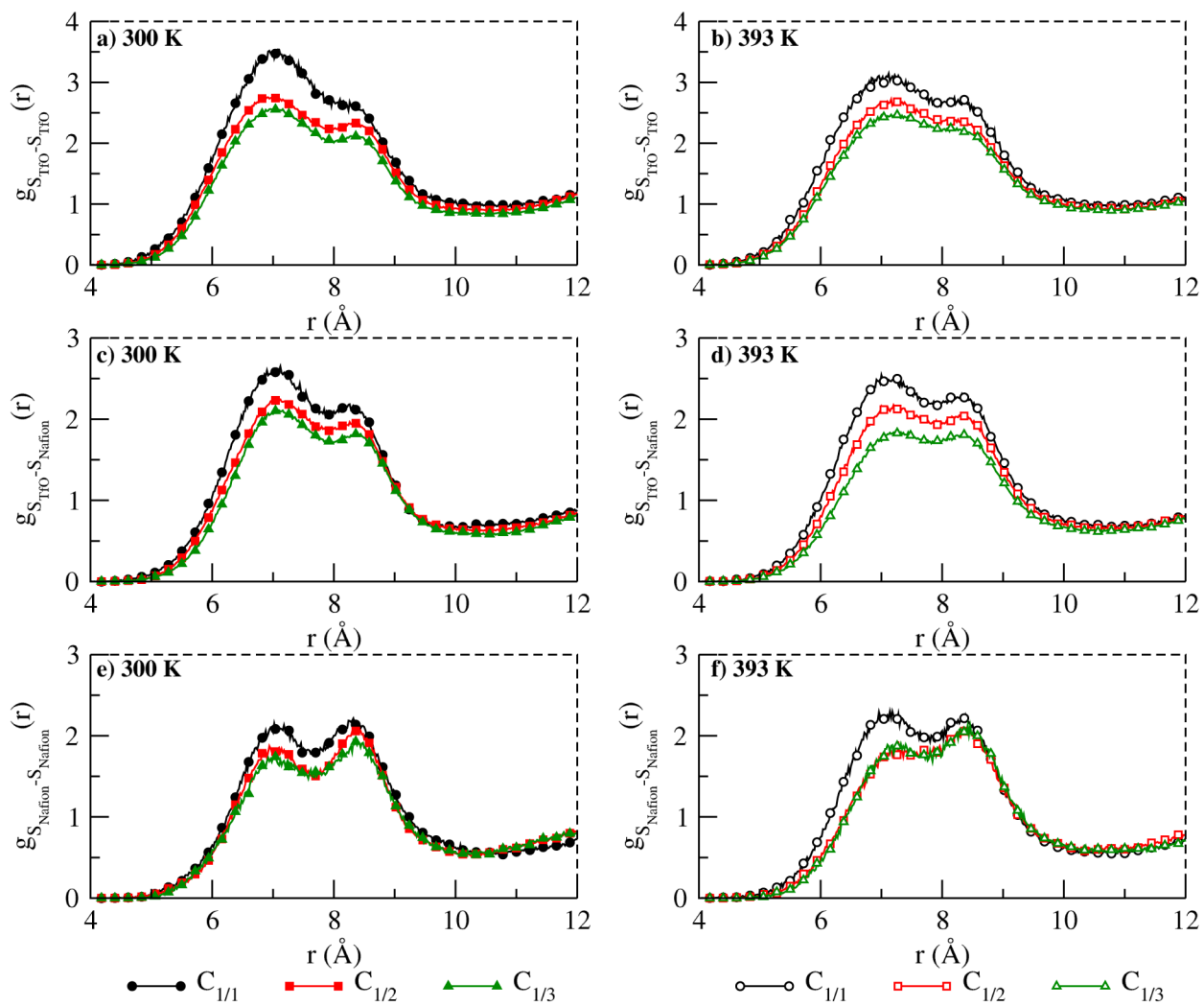


Figure S3: RDFs for sulfonate-sulfonate (S-S) interactions at 300 K and 393 K in (a, b) S_{TFO}-S_{TFO}, (c, d) S_{TFO}-S_{Nafion}, and (e, f) S_{Nafion}-S_{Nafion} respectively.

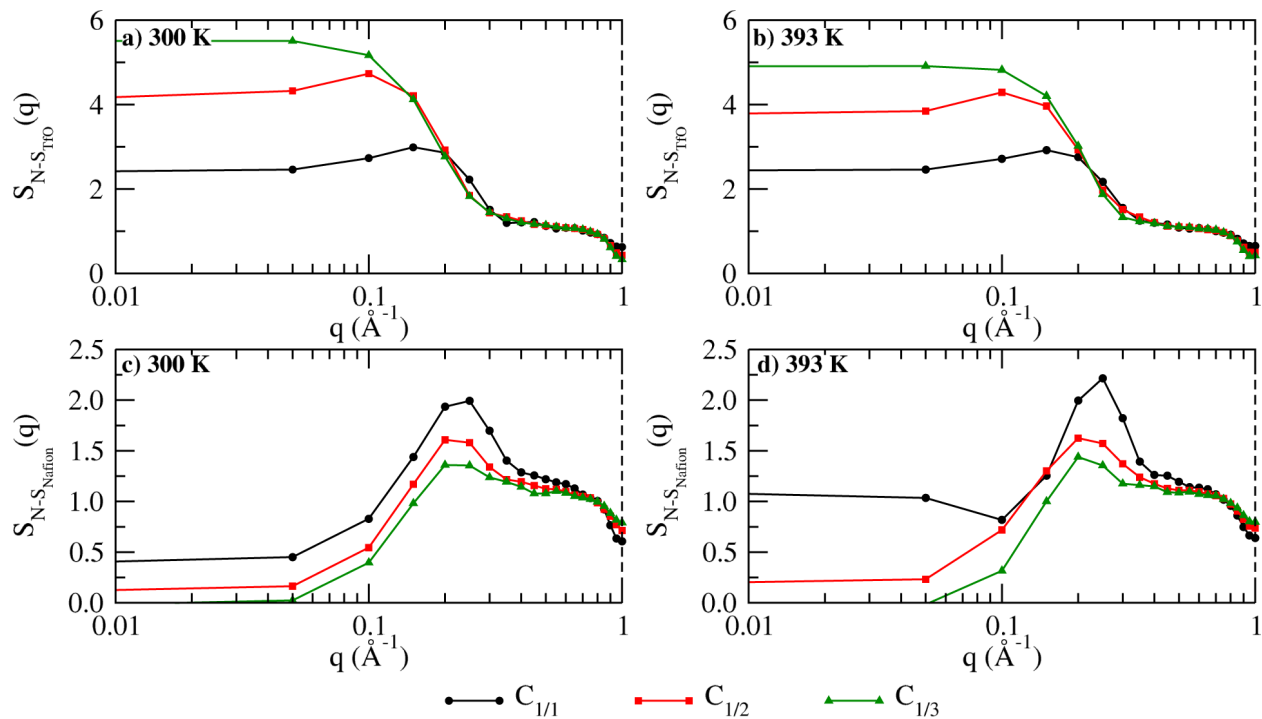


Figure S4: Structure factor calculated at 300 K and 393 K for (a, b) N-S_{TfO}, and (c, d) N-S_{Nafion} interactions respectively.

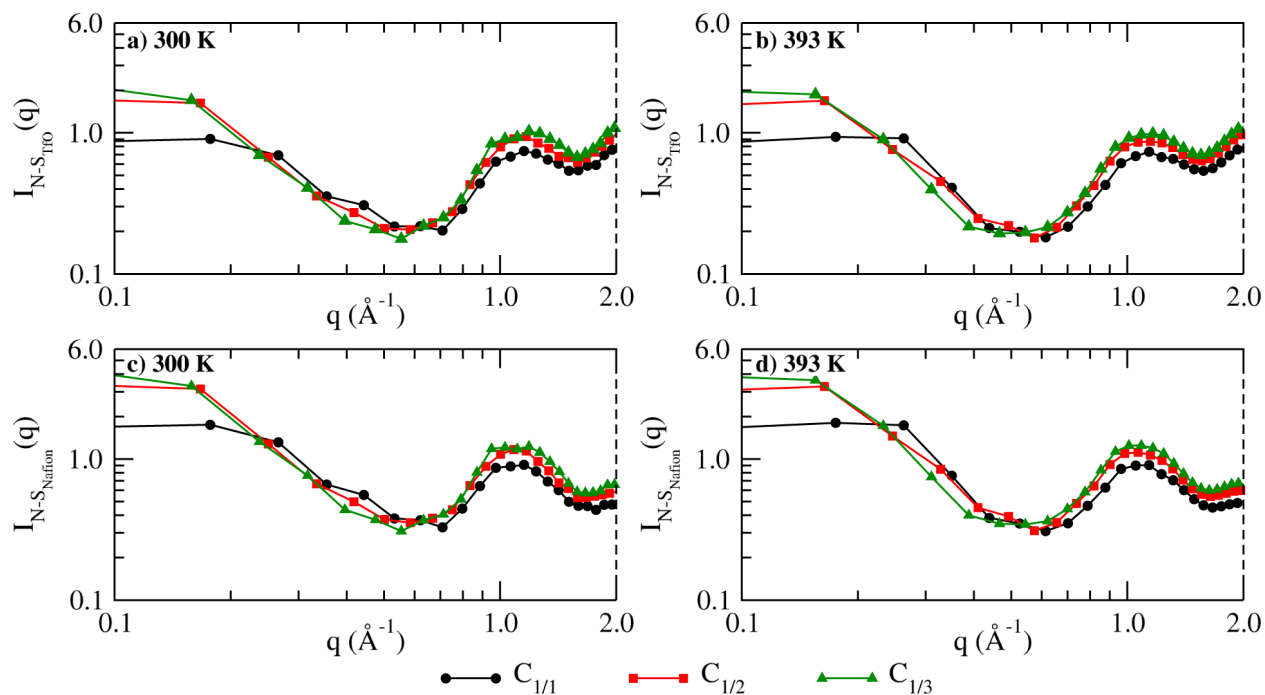


Figure S5: Scattering intensities calculated at 300 K and 393 K for (a, b) N-S_{TfO}, and (c, d) N-S_{Nafion} interactions respectively.

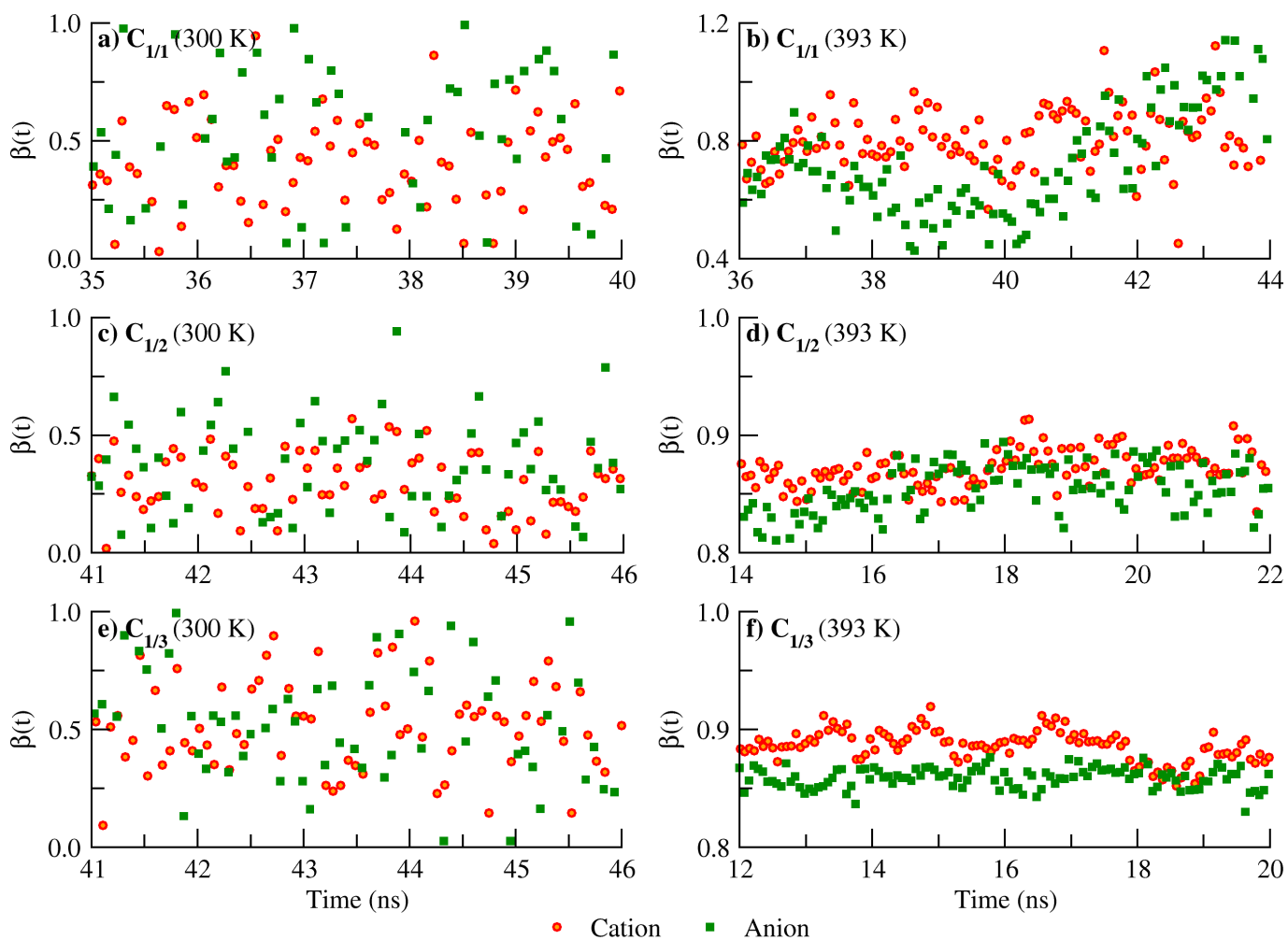


Figure S6: The β plot for cations and anions of PILs at varying Nafion:PIL ratios (a, b) $C_{1/1}$, (c, d) $C_{1/2}$, and (e, f) $C_{1/3}$ at 300 K and 393 K respectively. (Time window shown here was used for determining diffusion coefficients)