# Supporting information for

# Freestanding CNT-WO3 Hybrid Electrodes for

## **Flexible Asymmetric Supercapacitors**

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### **Experimental section**

### Preparation of the freestanding CNTs electrode and CNT-WO<sub>3</sub> electrode

Freestanding CNT film was fabricated *via* a vacuum filtration method. 0.050 g CNTs and 0.2 g sodium dodecylbenzenesulfonate were dissolved into 150 mL deionized (DI) water and ultrasonicated for 20 min, forming a CNTs ink.<sup>1</sup> Then the as-prepared solution vacuum filtered through a filter membrane (220 nm pore size) with DI water washed can be transferred to a freestanding film on the membrane, which then be dried and peeled off to gain the freestanding CNTs electrode. The CNT films were then vacuum evaporated using WO<sub>3</sub> particle powder (Aladdin, 50 nm, 15m<sup>2</sup> g<sup>-1</sup>, 99.9%) in a physical vapor deposition system (Baoke, Shenyang) in which the working conditions were a pressure of  $3.5 \times 10^{-4}$  Pa in the chamber and a definite current of 100 amperes for 30min.<sup>2</sup>

### Fabrication of the all-solid-state CNTs// CNT-WO3 ASC device

The ASC device was assembled by CNT film as the positive electrode and CNT-WO<sub>3</sub> hybrid film as the negative electrode using  $H_2SO_4/PVA$  gel as a solid state electrolyte and separator (NKK TF40, 40 µm thickness, low ESR type, purchased from SCM industrial Chemical CO., LTD). The  $H_2SO_4/PVA$  gel electrolyte was prepared by mixing 6 g concentrated  $H_2SO_4$  and 6 g PVA powder in 60 mL deionized water and being heated at 85 °C under a constant stir for 2 h. CNTs, CNT-WO<sub>3</sub> electrodes and the separator were dipped in the electrolyte for 5 min each and then fabricated into the ASC device. Finally the device should be retained at a room temperature to evaporate excess water overnight (about 12 h).

#### **Characterization:**

The structural properties of electrode materials were characterized by field-emission scanning electron microscopy (SEM, ZEISS ULTRA 55), transmission electron microscopy (TEM, JEOL 2100F, 200 kV) equipped with an energy dispersive X-ray spectrometer (EDS), X-ray diffraction (XRD, Rigaku, MiniFlex600, Cu K $\alpha$ ) and X-ray Photoelectron Spectroscope (XPS, Ulvac-Phi, PHI X-tool). Electrochemical measurements were employed using a CHI660E electrochemical workstation and a VersaSTAT 3-400 (Princeton Applied Research). The cycle life was measured through a battery test system (Neware BTS). For a single electrode test, a piece of electrode (effective area 1 cm × 1 cm, *ca*. 1 cm<sup>2</sup>) was dipped into the 1 M H<sub>2</sub>SO<sub>4</sub> solution as a working electrode at room temperature. Ag/AgCl reference electrode and graphite rod counter electrode were used in the measurement.

## **Calculation methods**

#### **ASC device**

The volumetric capacitance  $C_v$  was calculated from the CV curves through the following equation:

$$C_v = \frac{\int I dU}{2vVU_w} = \frac{S}{2vVU_w}$$

where v is the scan rate, V the volume of the whole device, S the area of the closed CV curve and  $U_w$  is the voltage window.

And the energy density and the average power density can be gained by employing the following equation:

$$E = \frac{C_v U_w^2}{2}$$
$$P = \frac{E}{t}$$
$$t = \frac{U_w}{v}$$

in which  $C_v$  is the volumetric capacitance calculated before.<sup>3</sup>



Fig.S1 (a) Areal capacitances (C<sub>a</sub>) and Ca contribution per 10 µm of CNT-WO<sub>3</sub> electrodes with

different thickness of WO<sub>3</sub> calculated from the corresponding CV curves at a scan rate of 50 mV/s. (b) Equivalent resistances ( $R_s$ ) calculated from IR drops from the Galvanostatic charge-discharge curves of different WO<sub>3</sub> thickness at a current density of 5 mA/cm<sup>2</sup>.



**Fig.S2** CV curves of the (a) CNT-WO<sub>3</sub> electrode compared with the pure CNT electrode under a potential windows between -0.6 V to 0 V at a scan rate of 100 mV/s and (b) positive and negative electrodes at a scan rate of 100 mV/s, representing a proper capacitance ratio for asymmetric supercapacitor.



**Fig.S3** (a) Galvanostatic charge-discharge curves of the ASC device collected at different current density. (b) Nyquist plots of the solid-state ASC device.



Fig.S4 Comparison of (a) CV curves at a scan rate of 100 mV/s and (b) Galvanostatic charge-

discharge curves at current density of 1mA/cm<sup>2</sup> collected from the CNT//CNT-WO<sub>3</sub> ACS device and the CNT//CNT SC device.



Fig.S5 XPS patterns of the CNT-WO<sub>3</sub> films.

## Notes and references

1. L. Du, P. Yang, X. Yu, P. Liu, J. Song and W. Mai, J. Mater. Chem. A, 2014, 2, 17561-17567.

2. P. Yang, P. Sun, Z. Chai, L. Huang, X. Cai, S. Tan, J. Song and W. Mai, *Angew Chem Int Ed Engl*, 2014, 53, 11935-11939.

3. P. Yang and W. Mai, Nano Energy, 2014, 8, 274-290.