

Supplementary Information

Controllable Synthesis of $\text{Bi}_4\text{O}_5\text{Br}_2$ Ultrathin Nanosheets for Photocatalytic Removal of Ciprofloxacin and Mechanism Insight

Jun Di, Jiexiang Xia*, Mengxia Ji, Sheng Yin, Hongping Li, Hui Xu, Qi Zhang,
Huaming Li*

School of Chemistry and Chemical Engineering, Institute of Energy, Jiangsu
University, 301 Xuefu Road, Zhenjiang, 212013, P. R. China

*Corresponding author: Tel.: +86-511-88791108; Fax: +86-511-88791108;

E-mail address: xjx@ujs.edu.cn; lhm@ujs.edu.cn

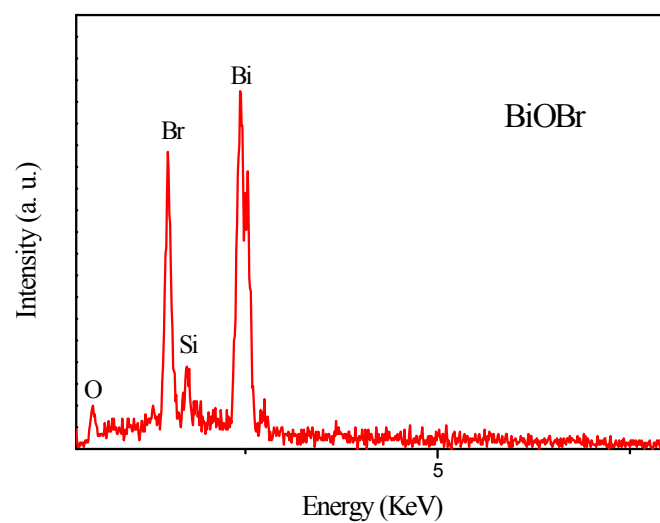


Figure S1. EDS analysis of the as-prepared BiOBr materials.

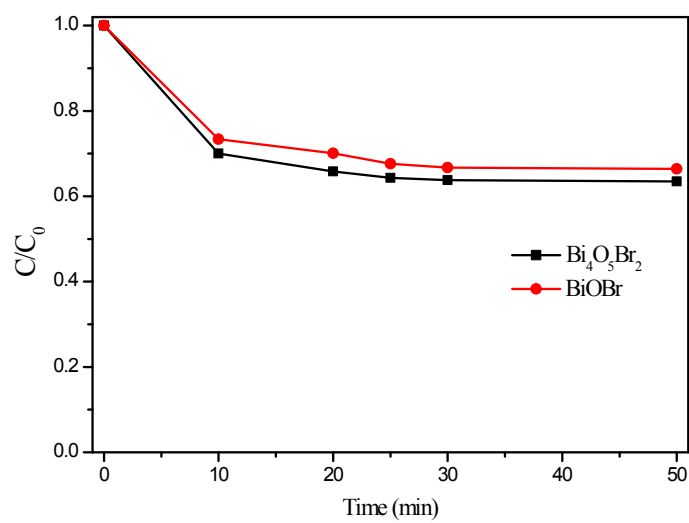


Figure S2 Time profiles of adsorption of CIP over $\text{Bi}_4\text{O}_5\text{Br}_2$ and BiOBr materials.

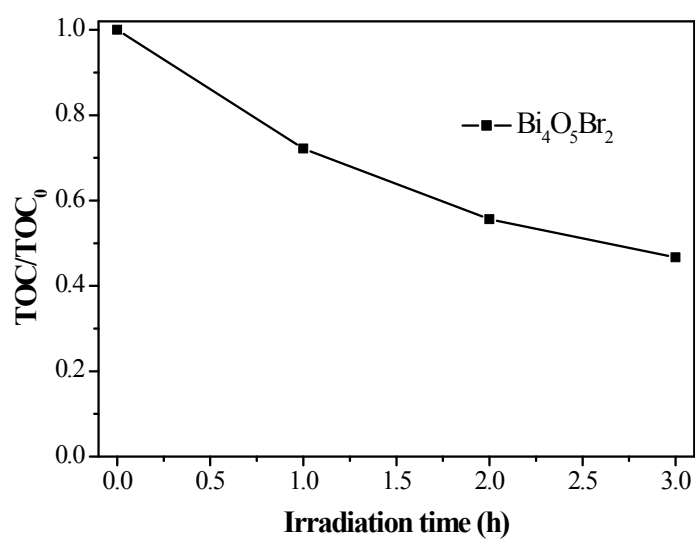


Figure S3 The decrease of TOC during photocatalytic degradation of CIP on Bi₄O₅Br₂ ultrathin nanosheets under visible light irradiation.

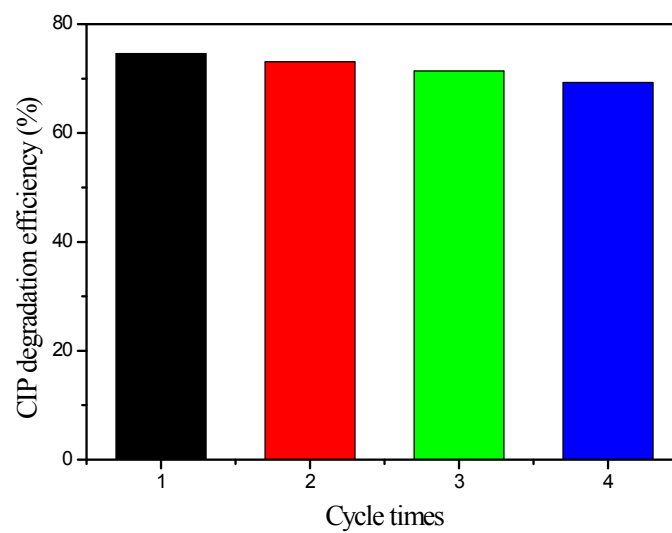


Figure S4 Cycling runs for the photodegradation of CIP in the presence of Bi₄O₅Br₂ ultrathin nanosheets under visible light irradiation.

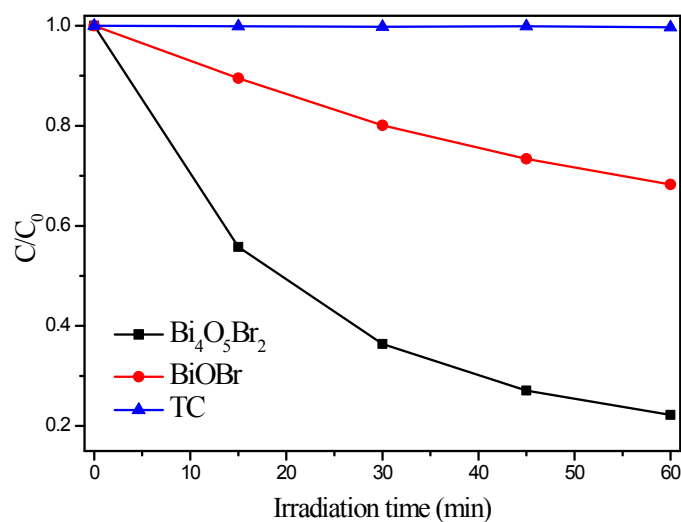


Figure S5 Photocatalytic degradation of TC in the presence of $\text{Bi}_4\text{O}_5\text{Br}_2$ ultrathin nanosheets and BiOBr materials under visible light irradiation.

Table S1 Pseudo-first-order rate constant for CIP photocatalytic oxidation under different photocatalysts

Series	Photocatalyst	The first order kinetic equation	k (min^{-1})	R^2
1	BiOBr	$-\ln(C/C_0) = 0.0059t$	0.0059	0.9998
2	$\text{Bi}_4\text{O}_5\text{Br}_2$	$-\ln(C/C_0) = 0.0113t$	0.0113	0.9993