

Electronic Supplementary Information for:

Design and Synthesis of Ni-MOF/CNTs composites and
rGO/Carbon Nitride composites for An Asymmetric
Supercapacitor with High Energy and Power Density

Ping Wen^a, Peiwei Gong^b, Jinfeng Sun^b, Jinqing Wang^{a*} and Shengrong Yang^a

^a State Key Laboratory of Solid Lubrication, Lanzhou Institute of Chemical Physics,
Chinese Academy of Sciences, Lanzhou, 730000, P. R. China.

^b University of Chinese Academy of Sciences, Beijing 100080, P. R. China.

* Corresponding author, jqwang@licp.ac.cn;

Fax: +86 931 8277088

Tel.: +86 931 4968076

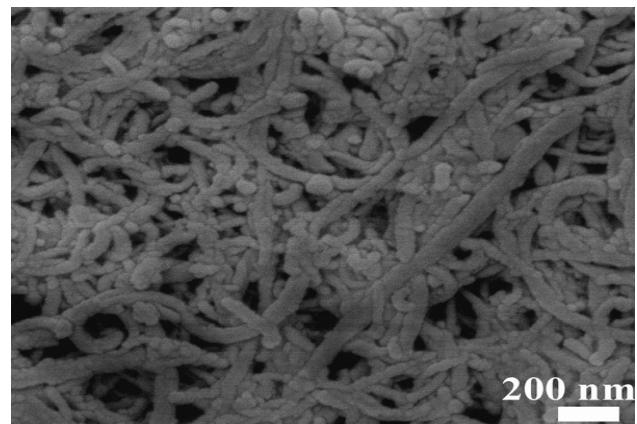


Figure S1. SEM image of Ni-MOF/CNTs-5

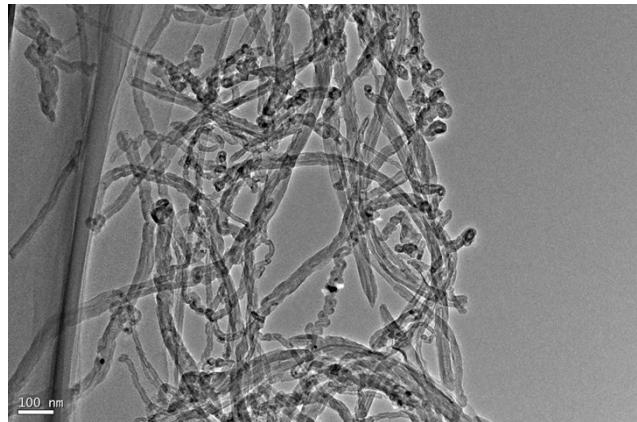


Figure S2. TEM image of mildly oxidized CNTs.

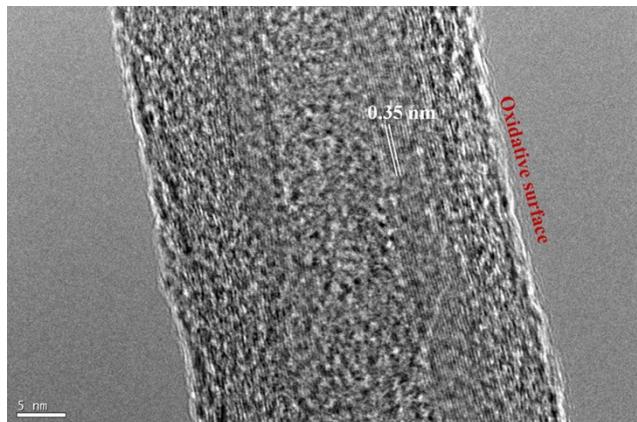


Figure S3. HRTEM image of mildly oxidized CNTs.

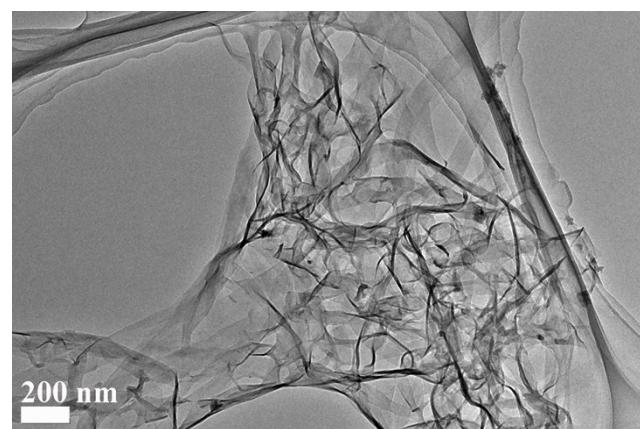


Figure S4. TEM image of Ni-MOF.



Figure S5. HRTEM image of Ni-MOF.

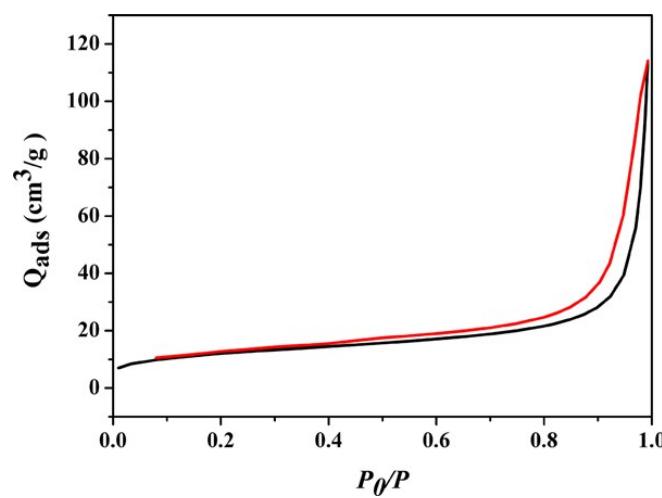


Figure S6. Nitrogen adsorption-desorption isotherm of Ni-MOF/CNTs-5 at 77.3 K.

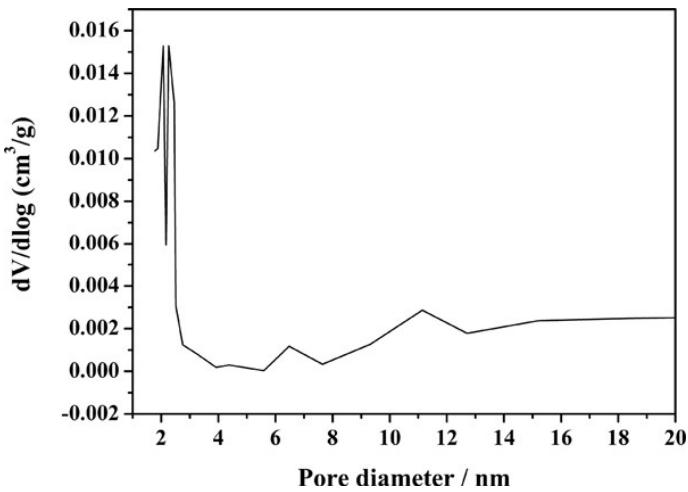


Figure S7. The pore-size distribution of Ni-MOF/CNTs-5.

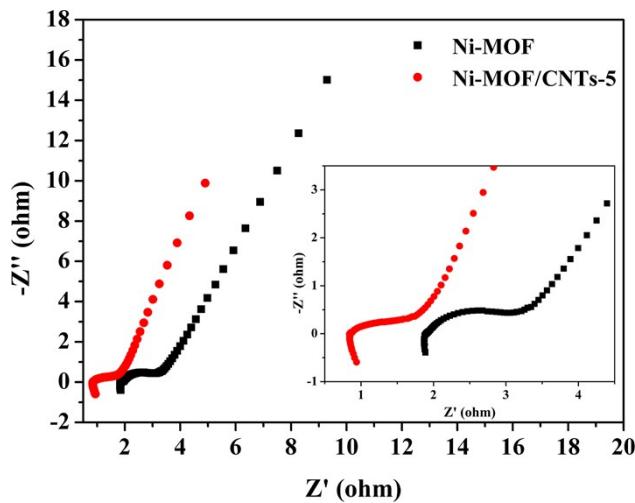


Figure S8. Nyquist plots of Ni-MOF and Ni-MOF/CNTs-5 composite.

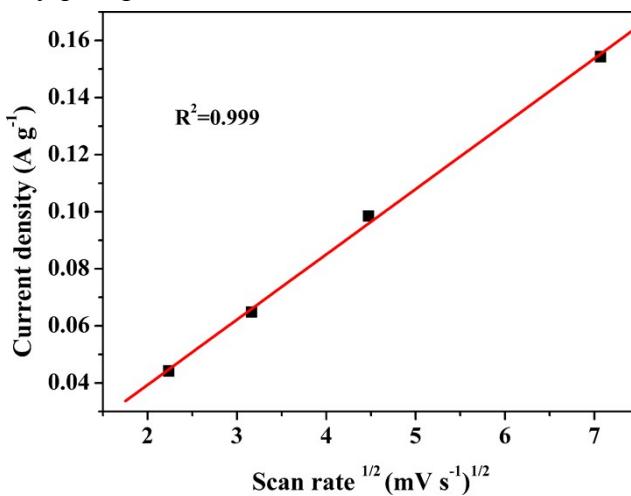


Figure S9. The relationship of current density vs the square root of scan rates in pure Ni-MOF.

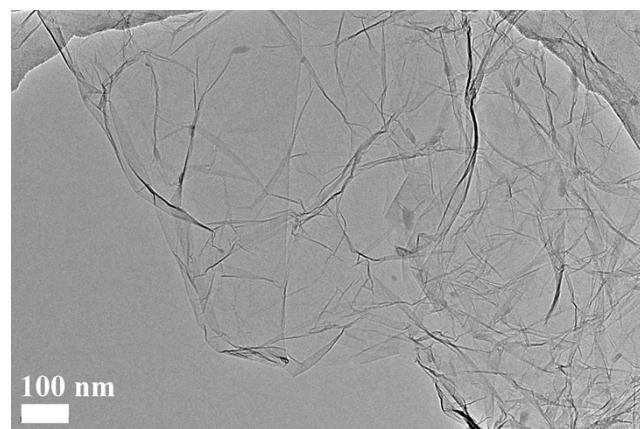


Figure S10. TEM image of GO sheets.

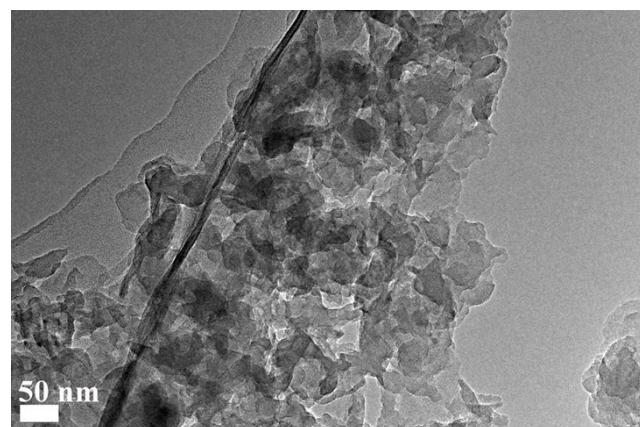


Figure S11. TEM image of $g\text{-C}_3\text{N}_4$ sheets.

Table S1. Comparison of capacity retention of asymmetric supercapacitors fabricated in our work with others reported.

Asymmetric Supercapacitors	Cycling Number	Current Density or Scan Rate	Capacity Retention	Reference
Ni ₃ S ₂ /MWCNT-NC//AC	5000	4 A g ⁻¹	90%	[1]
Co ₃ O ₄ @Ni(OH) ₂ //RGO	1000	25 mA cm ⁻²	86%	[2]
Co ₉ S ₈ //Co ₃ O ₄ @RuO ₂	2000	2.5 mA cm ⁻²	90.2%	[3]
RuO ₂ /graphene//graphene	2000	1.0 A g ⁻¹	95%	[4]
Ni ₃ S ₂ /MWCNT-NC//AC	5000	4.0 A g ⁻¹	90%	[5]
Graphite/Ni/Co ₂ NiO ₄ //graphite/Ni/AC	5000	10 mA cm ⁻²	96.4%	[6]
Ni-Co LDH//rGO	5000	5 A g ⁻¹	82%	[7]
CoO@C//AC	10000	50 mA cm ⁻²	96.9%	[8]
Ni(OH) ₂ /graphene//porous graphene	3000	100 mV s ⁻¹	94%	[9]
Ni(OH) ₂ //activated carbon	1000	5 mV s ⁻¹	82%	[10]
LiNi _{1/3} Co _{1/3} Mn _{1/3} O ₂ //AC	1000	100 mA g ⁻¹	80%	[11]
Graphene/MnO ₂ //activated carbon nanofiber	1000	200 mV s ⁻¹	97%	[12]
Ni-MOF/CNTs//rGO/C3N4	5000	2 A g ⁻¹	95%	Our Work

References:

- [1] C. S. Dai, P. Y. Chien, J. Y. Lin, S. W. Chou, W. K. Wu, P. H. Li, K. Y. Wu and T. W. Lin, *ACS Appl. Mater. Interfaces*, 2013, **5**, 12168.
- [2] C. H. Tang, X. S. Yin and H. Gong, *ACS Appl. Mater. Interfaces*, 2013, **5**, 10574.
- [3] J. Xu, Q. Wang, X. Wang, Q. Xiang, B. Liang, D. Chen and G. Shen, *ACS Nano*, 2013, **7**, 5453.
- [4] B. G. Choi, S. J. Chang, H. W. Kang, C. P. Park, H. J. Kim, W. H. Hong, S. Lee and Y. S. Huh, *Nanoscale*, 2012, **4**, 4983.
- [5] C. Dai, P. Chien, J. Lin, S. Chou, W. Wu, P. Li, K. Wu and T. Lin, *ACS Appl.*

Mater. Interfaces, 2013, **5**, 12168.

[6] J. X. Feng, S. H. Ye, A. L. Wang, X. F. Lu, Y. X. Tong and G. R. Li, *Adv. Funct. Mater.*, 2014, **24**, 7093.

[7] H. Chen, L. F. Hu, M. Chen, Y. Yan and L. M. Wu, *Adv. Funct. Mater.*, 2014, **24**, 934.

[8] H. Wang, C. Qing, J. L. Guo, A. A. Aref, D. M. Sun, B. X. Wang and Y. W. Tang, *J. Mater. Chem. A*, 2014, **2**, 11776.

[9] J. Yan, Z. J. Fan, W. Sun, G. Q. Ning, T. Wei, Q. Zhang, R. F. Zhang, L. J. Zhi and F. Wei, *Adv. Funct. Mater.*, 2012, **22**, 2632.

[10] J. W. Lang, L. B. Kong, M. Liu, Y. C. Luo and L. Kang, *J. Solid State Electron.*, 2010, **14**, 1533.

[11] Y. Zhao, Y. Y. Wang, Q. Y. Lai, L. M. Chen, Y. J. Hao and X. Y. Ji, *Synth. Met.*, 2009, **159**, 331.

[12] Z. Fan, J. Yan, T. Wei, L. Zhi, G. Ning, T. Li and F. Wei, *Adv. Funct. Mater.*, 2011, **21**, 2366.