

Electronic Supporting Information

Fabrication of urchin-like $\text{NiCo}_2(\text{CO}_3)_{1.5}(\text{OH})_3@ \text{NiCo}_2\text{S}_4$ on the Ni foam by ion exchange route and applying to asymmetric supercapacitor

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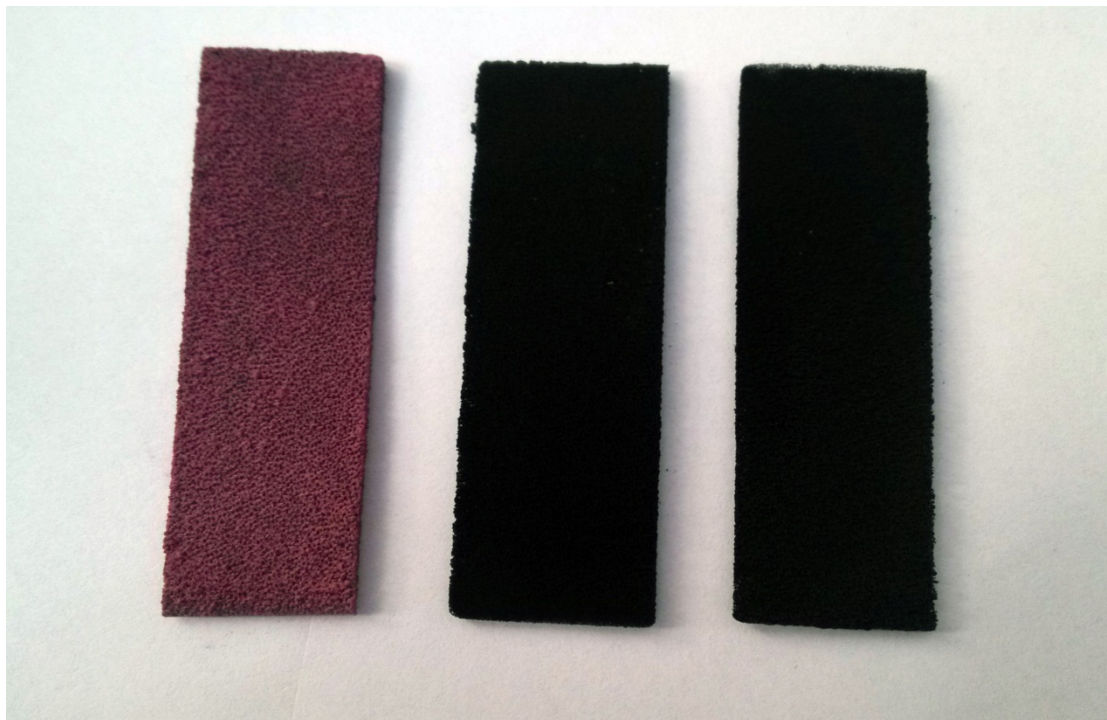


Fig. S1 the photography of urchin-like $\text{NiCo}_2(\text{CO}_3)_{1.5}(\text{OH})_3$ (left), NiCo_2S_4 (middle) and $\text{NiCo}_2(\text{CO}_3)_{1.5}(\text{OH})_3@ \text{NiCo}_2\text{S}_4$ (right).

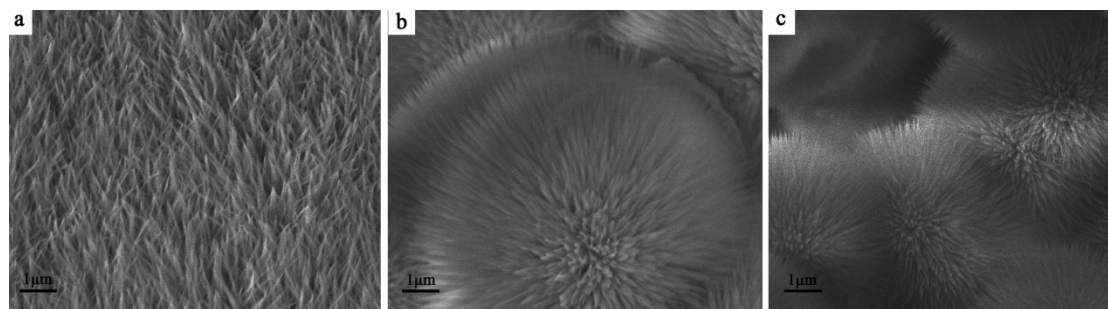


Fig. S2 the SEM images of $\text{NiCo}_2(\text{CO}_3)_{1.5}(\text{OH})_3$ nanowires (a), urchin-like $\text{NiCo}_2(\text{CO}_3)_{1.5}(\text{OH})_3$ (b) and urchin-like NiCo_2S_4 (c).

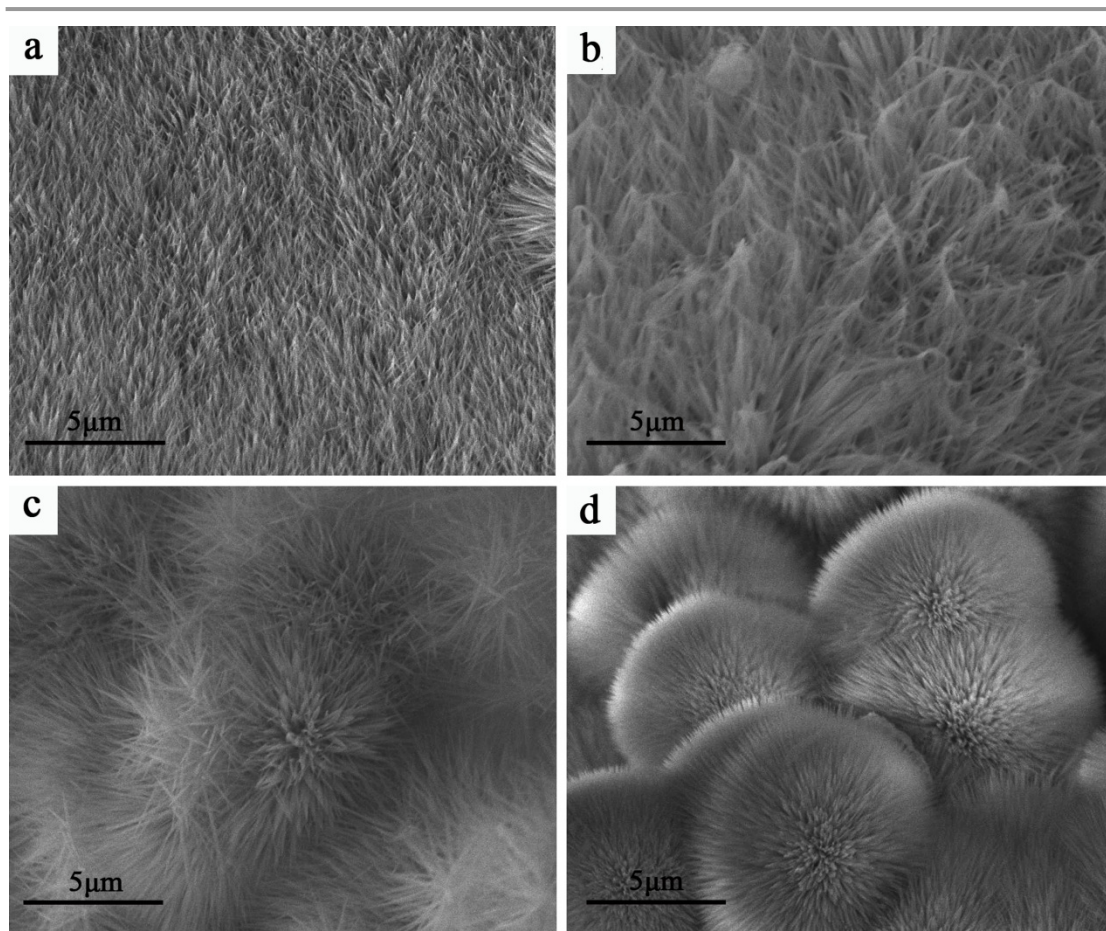


Fig. S3. the SEM images of the precursors at different reaction time, (a) 12h, (b) 16h, (c) 20h, (d) 24h,

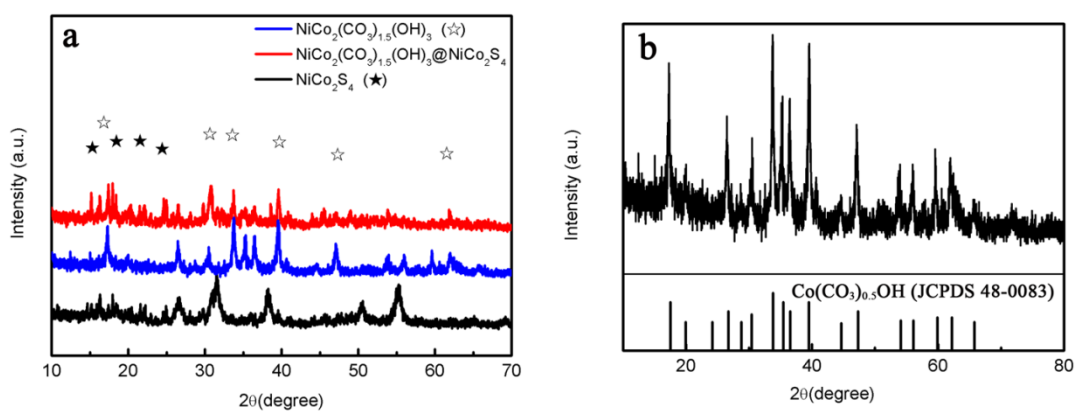


Fig. S4 (a) XRD patterns of $\text{NiCo}_2(\text{CO}_3)_{1.5}(\text{OH})_3$, NiCo_2S_4 and $\text{NiCo}_2(\text{CO}_3)_{1.5}(\text{OH})_3@$ NiCo_2S_4 , (b) XRD patterns of $\text{NiCo}_2(\text{CO}_3)_{1.5}(\text{OH})_3$.

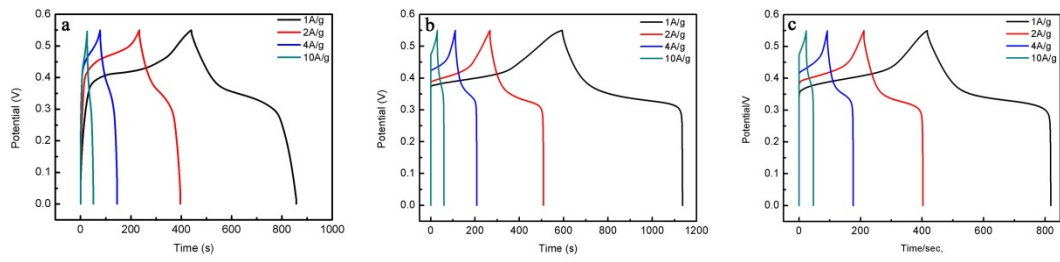


Fig. S5. The GCD curves of different electrode at current density of 1, 2, 4, 10 A g⁻¹. (a) urchin-like NiCo₂(CO₃)_{1.5}(OH)₃, (b) urchin-like NiCo₂S₄ (c) NiCo₂(CO₃)_{1.5}(OH)₃ @NiCo₂S₄ nanowires.

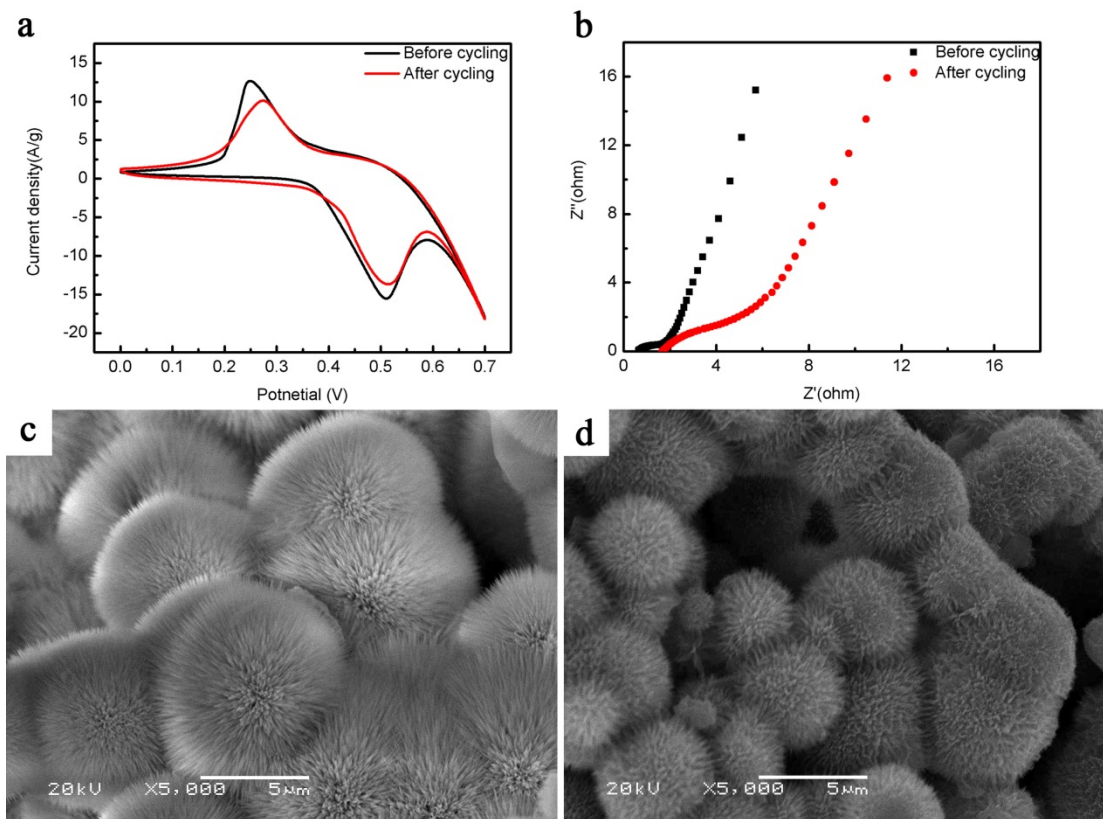


Fig. S6 (a) the CV curves of before and after cycling, (b) the EIS plots of before and after cycling (c) the SEM image of before cycling, (d) the SEM image of after cycling.

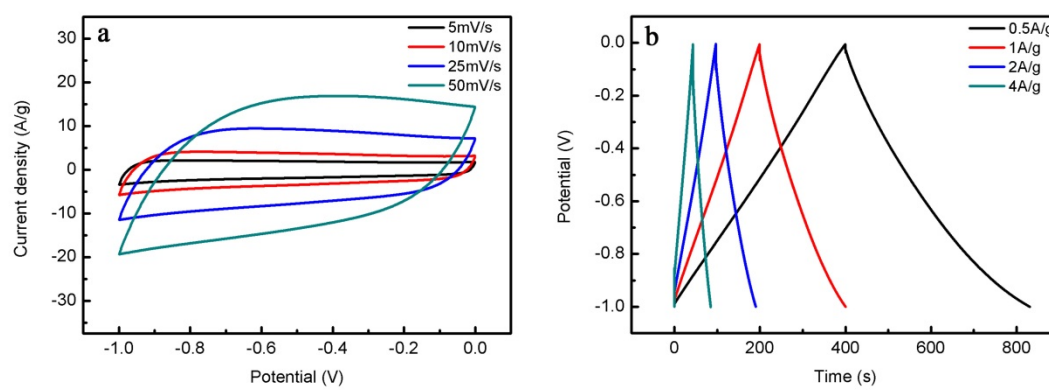


Figure S7. (a) CV curves of AC electrode at different scan rate, (b) charge-discharge curves of AC electrode at different current densities.

Table S1. Comparison of the maximum energy densities, corresponding average power densities and voltage range of the reported nickel, cobalt or bimetal oxide/hydroxide/sulfides based asymmetric supercapacitors and the present work.

Positive materials//negative materials	Energy density (W h kg ⁻¹)	Power density (W kg ⁻¹)	Voltage range	Ref.
Nickel cobalt LDHs//AC	23.7	280	0-1.2	[1]
NiCo ₂ S ₄ //RGO	31.5	156.6	0-1.6	[2]
Co _{0.45} Ni _{0.55} O-RGO//RGO	35.3	330	0-1.5	[3]
Ni ₃ S ₂ /MWCNT-NC//AC	19.8	798	0-1.6	[4]
NiCo ₂ O ₄ -CQDs//AC	27.8	128	0-1.5	[5]
CoS _x -NSA//rGO	14.68	369	0-1.55	[6]
Ni(OH) ₂ @3D Ni//AC	21.8	660	0-1.3	[7]
CoO@PPy//AC	43.5	87.5	0-1.8	[8]
Co ₃ O ₄ //AC	34	225	0-1.5	[9]
Ni(OH) ₂ //AC	35.7	490	0-1.3	[10]
NiCo ₂ O ₄ @MnO ₂ //AC	35	163	0-1.5	[11]
NiCo ₂ (CO ₃) _{1.5} (OH) ₃ @NiCo ₂ S ₄ //AC	42.55	458.8	0-1.6	This work

References for supporting information:

1. X. Wang, A. Sumboja, M. Lin, J. Yan, P. S. Lee, *Nanoscale* **2012**, *4*, 7266.
2. H. Chen, J. Jiang, L. Zhang, D. Xia, Y. Zhao, D. Guo, T. Qi and H. Wan, *J. Power Sources*, 2014, *254*, 249-257.
3. J. Xiao, S. Yang, *J. Mater. Chem.* **2012**, *22*, 12253.
4. C. S. Dai, P. Y. Chien, J. Y. Lin, S. W. Chou, W. K. Wu, P. H. Li, K. Y. Wu and T. W. Lin, *ACS Appl. Mater. Inter.*, 2013, *5*, 12168-12174.
5. Y. R. Zhu, Z. B. Wu, M. J. Jing, H. S. Hou, Y. C. Yang, Y. Zhang, X. M. Yang, W. X. Song, X. N. Jia and X. B. Ji, *J. Mater. Chem. A*, 2015, *3*, 866-877.
6. D. P. Dubal, G. S. Gungor, C. D. Lokhande and R. Holze, *Energy Technology*, 2014, *2*, 401-408.

7. Y.-Z. Su, K. Xiao, N. Li, Z.-Q. Liu and S.-Z. Qiao, *J. Mater. Chem. A*, 2014, 2, 13845-13853.
8. C. Zhou, Y. Zhang, Y. Li and J. Liu, *Nano Lett.*, 2013, 13, 2078-2085.
9. X. Zhang, Y. Q. Zhao and C. L. Xu, *Nanoscale*, 2014, 6, 3638-3646.
10. H. B. Li, M. H. Yu, F. X. Wang, P. Liu, Y. Liang, J. Xiao, C. X. Wang, Y. X. Tong and G. W. Yang, *Nat. commun.*, 2013, 4.
11. K. Xu, W. Li, Q. Liu, B. Li, X. Liu, L. An, Z. Chen, R. Zou and J. Hu, *J. Mater. Chem. A*, 2014, 2, 4795-4802.