

Supporting Information

Open-top TiO₂ Nanotube Arrays with Enhanced
Photovoltaic and Photochemical Performances *via* a
Micromechanical Cleavage Approach

Yulong Liao^{a,}, Dainan Zhang^b, Qi Wang^a, Tianlong Wen^a, Lijun Jia^a, Zhiyong Zhong^a,
Feiming Bai^a, Longhuang Tang^a, Wenxiu Que^c and Huaiwu Zhang^{a,*}*

^a State Key Laboratory of Electronic Thin Film and Integrated Devices, University of Electronic Science and Technology,
Chengdu 610054, China

^b Department of Electrical and Computer Engineering, University of Delaware, Newark, Delaware 19716, USA

^c School of Electronic and Information Engineering, Xi'an Jiaotong University, Xi'an 710049, Shaanxi, China

* Corresponding author:

Email address: yulong.liao@uestc.edu.cn (Y.L.), hwzhang@uestc.edu.cn (H.Z.)

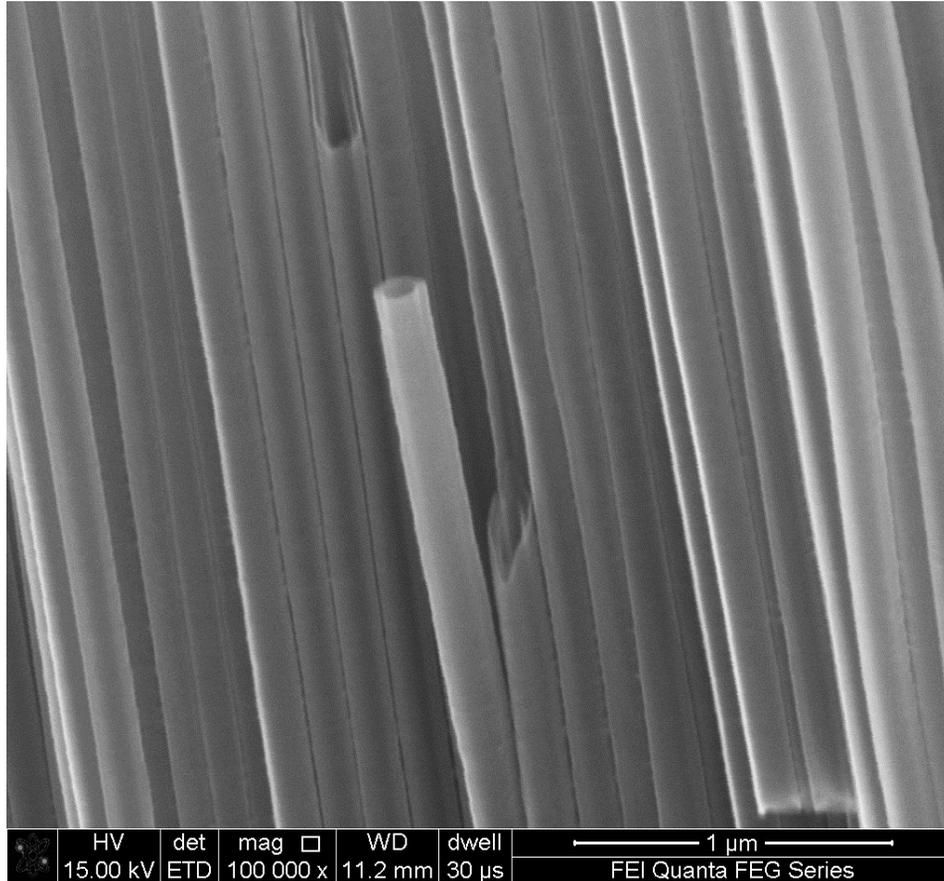


Figure S1. SEM image of the side view of anodic TiO₂ NTs, showing the TiO₂ NTs are uniform and well aligned with smooth wall.

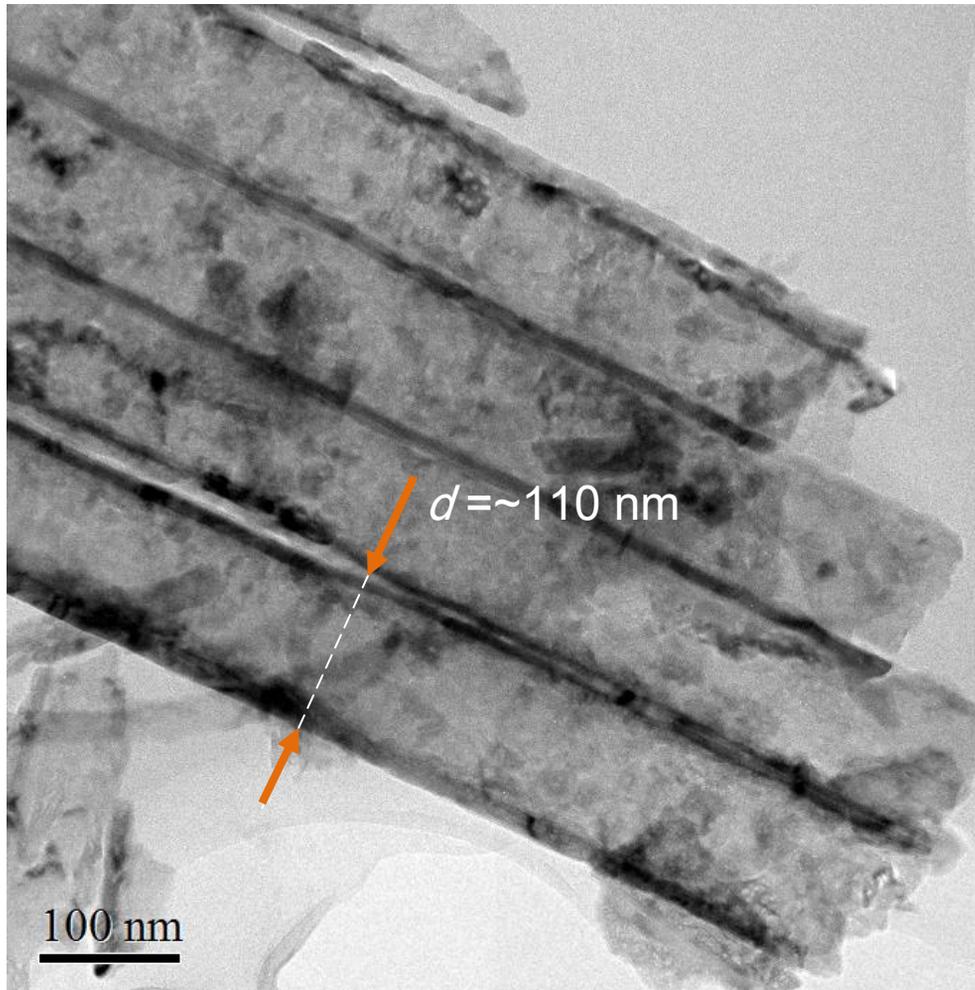


Figure S2. TEM image of the anodic TiO₂ NTs, showing highly ordered and uniform the TiO₂ NTs with diameter of ~110 nm.

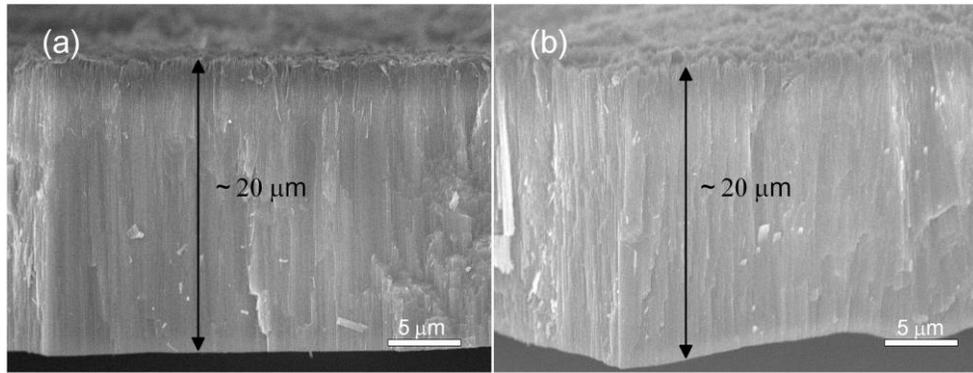


Figure S3. SEM images of the side view of anodic TiO₂ NT arrays (a) before and (b) after peeling off the nanoglass layer, showing negligible change in thickness.

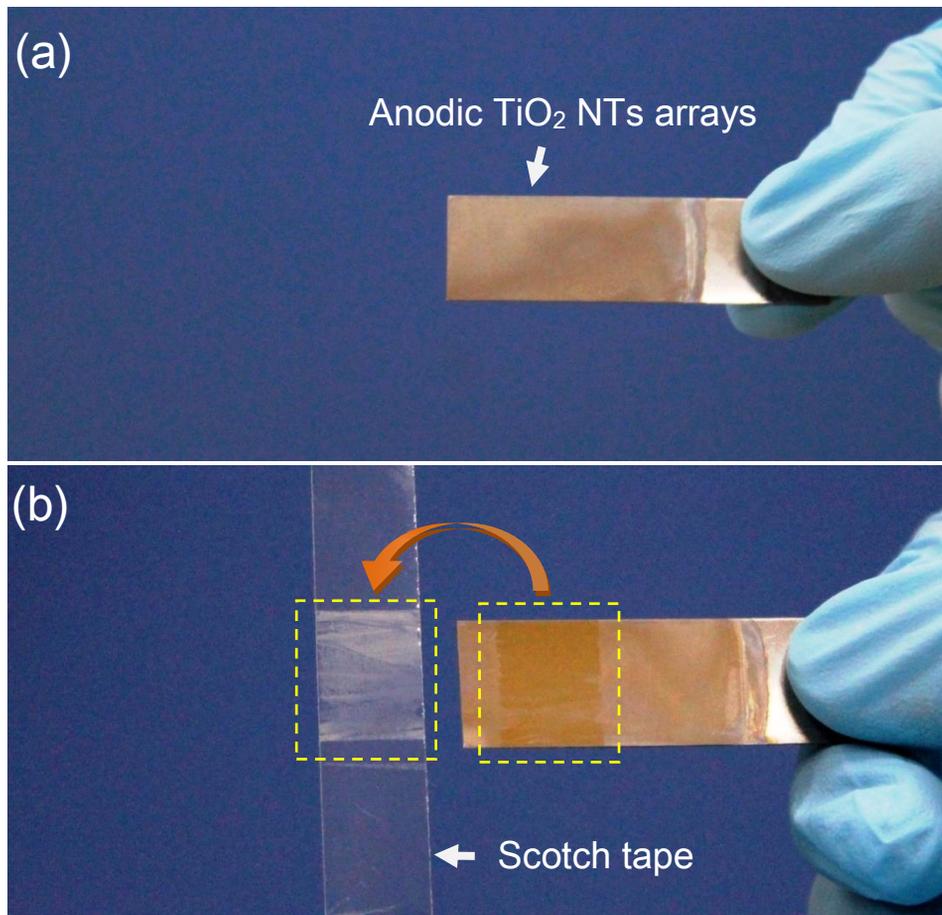


Figure S4. Photograph of (a) the as-prepared anodic TiO₂ NT arrays on metal Ti substrate and (b) the peeling off process to remove the nanograss top layer by using scotch tape technique.

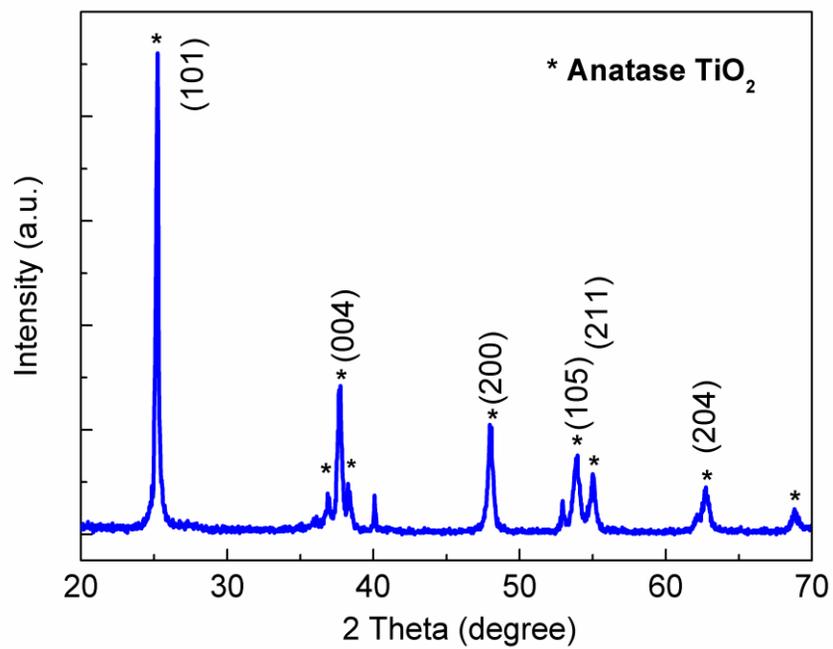


Figure S5. XRD patterns of the anodic TiO₂ NT arrays after sintered at 450 °C for 2 h, indicating the anodic TiO₂ NT arrays was crystallized to pure anatase phase.

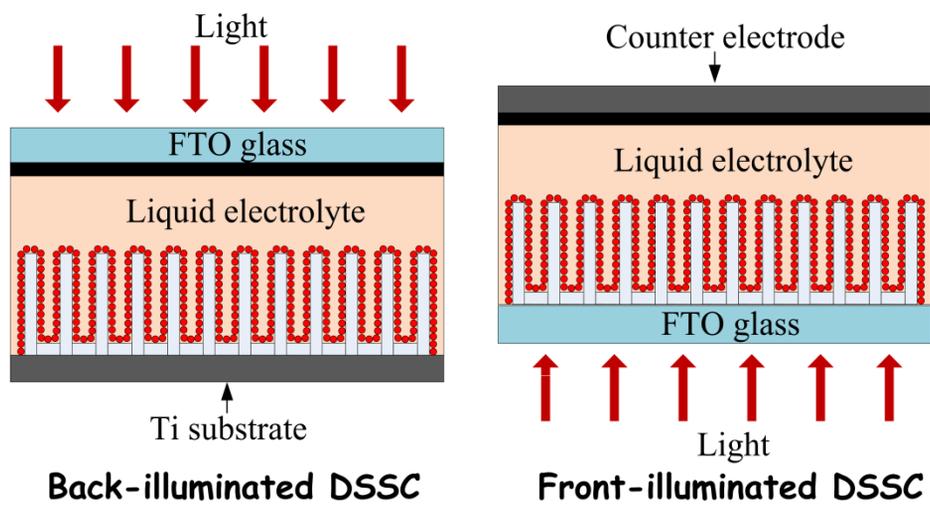


Figure S6. Illustration of the structure of a back-illuminated DSSC and a front-illuminated DSSC.

More Experimental Details:

Photocatalytic testing: The photocatalytic activities of the TiO₂ NT arrays with and without nanoglass top-layer were evaluated on the basis of the degradation of methyl orange (MO, 20 mM L⁻¹) as model organic pollutants in aqueous solutions (40 mL total volume). The TiO₂ NT films (3.0 × 1.5 cm² surface area, total mass ~ 0.04 g) were immersed in the solution and irradiated with six 4-W UV bulbs (Eiko, Black Light Blue, Shawnee, KS; 18 W/m² measured illumination intensity). Before testing, the samples were soaked for 30 min in dark with magnetic stirring to reach adsorption/desorption equilibrium. The photo-degradation testing lasted for 120 min and were carried out in an open quartz photoreaction vessel with rapid stirring under room temperature. The concentration of the residual MO was measured by a Uv-vis spectrometer at 463 nm.

In order to simulate the molecular transportation (e.g. MO molecule) inside the TiO₂ NTs, we also solve the molecular diffusion equation by Finite-difference Methods for nanoglass-capped TiO₂ NTs and top opened TiO₂ NTs.