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## **Electronic Supplementary Information (ESI)**

## Flexible Metal-organic Frameworks as Superior Cathodes for

## **Rechargeable Sodium-ion Batteries**

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**Figure S1.** Photographs of (a) bare  $FeFe(CN)_6$  nanoparticles and (b, c)  $FeFe(CN)_6$ /carbonfiber composites.



Figure S2. SEM images of acid-treated carbon cloth.



Figure S3. TEM image and SAED pattern of FeFe(CN)<sub>6</sub>.



Figure S4. SEM image of bare FeFe(CN)<sub>6</sub> nanoparticles.



Figure S5. XRD pattern of the carbon textiles.



Figure S6. Thermogravimetric curve of FeFe(CN)\_6 samples at a rate of 10  $\,^\circ\!\!\mathbb{C}$  min<sup>-1</sup> under air.



**Figure S7.** Cycling performance of the carbon textiles at a current density of 50 mA  $g^{-1}$  in the voltage window of 2.0-4.0 V.



**Figure S8.** (a) Nyquist impedance plots of the  $FeFe(CN)_6$ /carbon textiles and bare  $FeFe(CN)_6$  electrodes measured before cycling, (b) Electrochemical impedance spectra of the the  $FeFe(CN)_6$ / carbon cloth electrode.



**Figure S9.** XRD pattern of the  $FeFe(CN)_6$ /carbon cloth after 50 cycles at a current density of 0.5C.

To examine phase structure of the electrodes after cycling, X-ray diffraction (XRD) was investigated using a Rigaku Miniflex II diffractometer using Cu K $\alpha$  radiation operated at 30 kV and 15 mA. Figure S9 shows XRD pattern of the FeFe(CN)<sub>6</sub>/carbon textiles after 50 cycles at a current density of 0.5C in the voltage range of 2.0-4.0 V.



**Figure S10.** Aqueous sodium-ion battery performance of  $FeFe(CN)_6$ /carbon cloth composites assessed from three-electrode measurements in 1 M NaNO<sub>3</sub> (pH=2): a) CV curves, b) galvanostatic charge-discharge voltage profiles at different current densities.



**Figure S11.** Galvanostatic charge-discharge curves and Mössbauer spectra of the  $FeFe(CN)_6$  electrode discharged at different depths (Voltage = 2.0, 2.8, and 3.1 V).