Charge storage mechanism of activated manganese oxide composites for pseudocapacitors

Supporting Information

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Synthesis method	Post treatment	Total time	Analyses	Identified	references
			during activation ^a	activated MnO _x	
Pyrolysis (ramping to 300 °C) from Mn(CH ₃ COO) ₂ precursor	N/A	~6 hr	XRD, TEM, XPS, <i>in situ</i> Raman, <i>ex</i> <i>situ</i> Raman	Amorphous MnO_2 and hausmannite- Mn_3O_4 composite	This work
Electrostatic spray deposition from Mn(CH ₃ COO) ₂ precursor at 260 °C by 10 kV	Annealing at 900 °C for 10 hr	>10 hr	XRD, XPS, SEM	Birnessite- MnO ₂	22
Electrostatic spray deposition from MnCl ₂ precursor by 12 kV	Annealing at 200-400 °C for 3 hr	>3 hr	XRD, XANES (Mn-K edge), Raman	Amorphous MnO ₂	23
Chemical bath deposition from MnSO ₄ and urea precursor for 12 hr	N/A	~12 hr	XRD, SEM, Raman, contact angle	Birnessite- MnO ₂	24
Thermal reduction of electrolytic manganese dioxide at 1050 °C for 72 hr	Ball-milled at 1000 rpm for 10 hr with acetone	~82 hr	SEM, XRD, XPS	Birnessite- MnO ₂	25

Table S1. Reported material characterisation analyses on electrochemical activation from Mn_3O_4 to MnO_2 .

^a Except for the electrochemical analyses, the methods used for identifying as-prepared Mn_3O_4 and activated MnO_2 .

	Mn ₃ O ₄	MnO ₂	MnO ₂	MnO ₂	MnO ₂	Mn ₃ O ₄	MnO ₂	Mn ₃ O ₄	I _{MnO2} / I _{Mn3O4}	Specific Capacitance (F/g)
as-	658	648	618	575	506	477	408	289	0.42	N/a
prepared	(36)	(42)	(51)	(70)	(55)	(50)	(55)	(65)		
eth 1	658	648	618	575	506	477	408	284	0.63	132
5 cycle	(36)	(45)	(45)	(61)	(55)	(55)	(55)	(65)		
10 th avala	658	648	618	575	506	477	408	285	0.89	144
$\begin{vmatrix} 10 & \text{cycle} \end{vmatrix}$ (36)	(36)	(52)	(52)	(70)	(60)	(55)	(45)	(65)		144
50 th cycle	658	648	618	575	506	477	408	285	1.62	166
	(35)	(46)	(48)	(55)	(55)	(40)	(56)	(65)		
100 th 1	658	648	618	575	506	477	408	285	2.53	160
100 cycle	(37)	(49)	(64)	(45)	(50)	(45)	(56)	(45)		109
150 th cycle	658	648	618	575	506	477	408	280	2.51	170
	(33)	(55)	(50)	(70)	(55)	(65)	(55)	(65)		170
200 th cycle	658	648	618	575	506	477	408	284	2.45	171
	(30)	(55)	(50)	(70)	(50)	(55)	(55)	(61)		1/1
500 th cycle	658	648	618	575	506	477	408	281	2.44	174
	(30)	(53)	(42)	(53)	(56)	(55)	(50)	(60)		1/4

Table S2. Peak position (cm⁻¹) and Full Width at Half Maximum (FWHM, cm⁻¹ in parenthesis) of the Mn₃O₄ and MnO₂ peaks by Lorentz fitting of *ex situ* Raman spectra.

MnO _x composite electrode	C 1s normalised area	Mn 3s normalised area	C1s/Mn3s area ratio	
As-prepared MnO _x	5792	3722.6	2.59	
Activated MnO _x	35096	34120	1.03	
(at region 1)				
Activated MnO _x	45909	43889	1.05	
(at region 2)				

Table S3. Normalised area (by atomic sensitivity factor) and area ratio calculated from XPS spectra

Specific capacitance (F g ⁻¹)	Potential window (V)	Electrolyte	Mass loading (mg cm ⁻²)	references
174 at 25 mV s ⁻¹	1.2	0.5 M Na ₂ SO ₄	0.71	This work
133 at 50 mV s^{-1}	0.9	0.2 M Na ₂ SO ₄	0.15ª	22
150 at 50 mV s ⁻¹	1.0	0.1 M Na ₂ SO ₄	0.116ª	23
${\sim}160$ at 20 mV s {-1}	1.0	1.0 M Na ₂ SO ₄	0.72 ^b	24
~160 at 10 mV s ⁻¹	1.0	0.5 M Na ₂ SO ₄	1.33°	25

Table S4. Capacitive performance of electrochemically activated birnessite-MnO₂.

 $^{\rm a}$ The electrode is prepared by ${\rm MnO}_{\rm x}$ without mixing with conductive agent and binder.

^b The specific capacitance is estimated from CV profile.

^c The electrode consists of 80% MnO_x , 10% conductive agent (carbon), and 10% poly(vinylidenefluoride) binder.



Figure S1. CV of MnO_x measured at a sweep rate of 25 mV s⁻¹ in 0.5M Na₂SO₄ against Ag/AgCl (blue) and quasi-reference electrode (red). The dash curve is measured against quasi-reference electrode after shifting by +0.1 V.



Figure S2. Lorentz fitting of *ex situ* Raman spectra recorded after (a) 5^{th} , (b) 10^{th} , (c) 50^{th} , (d) 100^{th} , (e) 150^{th} , (f) 200^{th} , and (g) 500^{th} CV cycle. Green line = envelope of total of the fitted peaks.



Figure S3. PXRD of activated MnO_x (MoK α 1 source) JCPDS CARD 18-0802 is standard birnessite-MnO₂ from database.



Figure S4. XPS Na 1s of (blue) as-prepared, (red) activated MnO_x , and (black) $Na_{0.33}MnO_2$ sample. The denoted values are the fitted peak position (eV) and Full Width at Half Maximum (FWHM, eV in parenthesis)