

Supporting information

Topotactical Conversion of Carbon Coated Fe-based Electrodes on Graphene

Aerogels for Lithium Ion Storage

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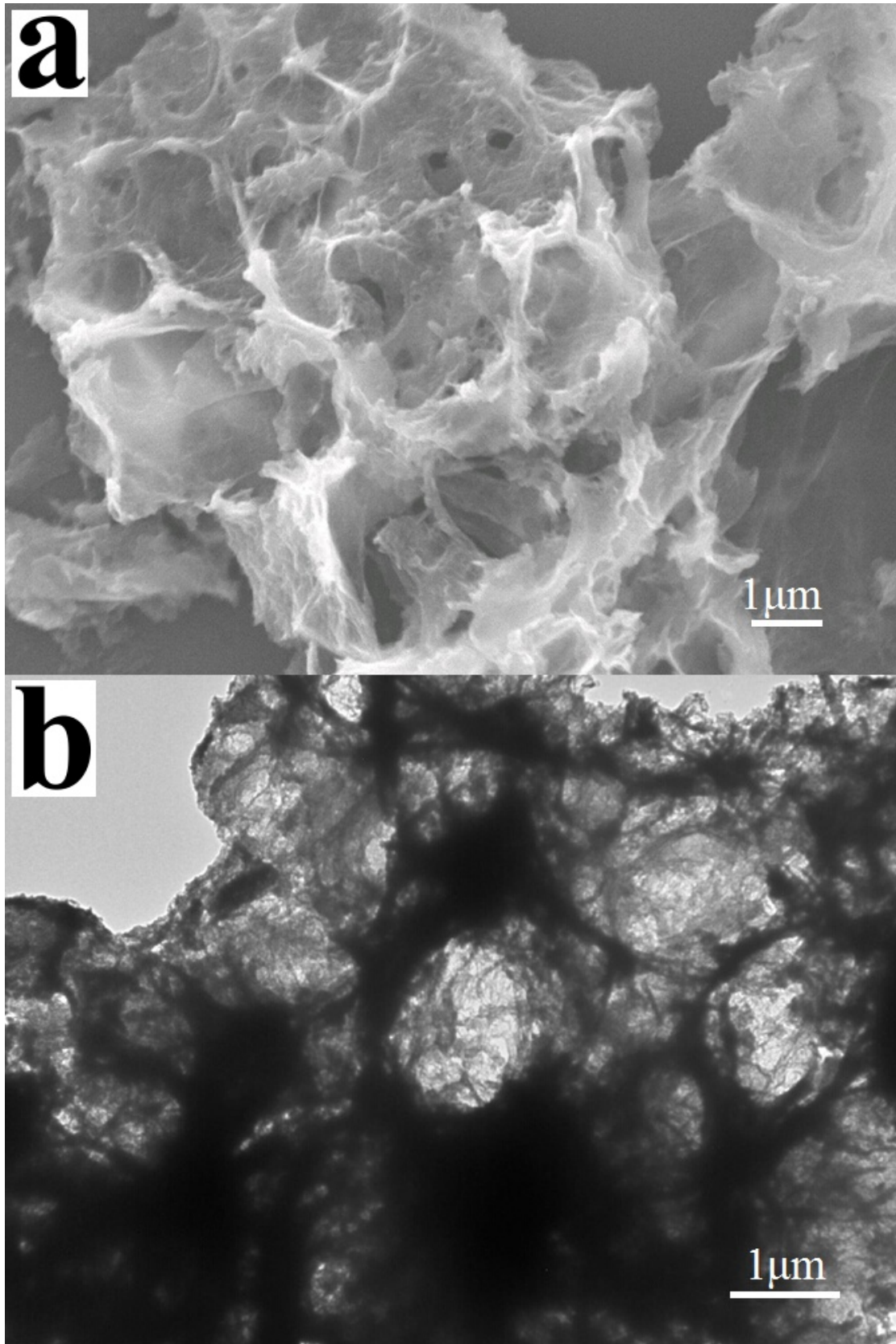


Fig. S1 Graphene aerogels (GAs) : (a) SEM image and (b) TEM image.

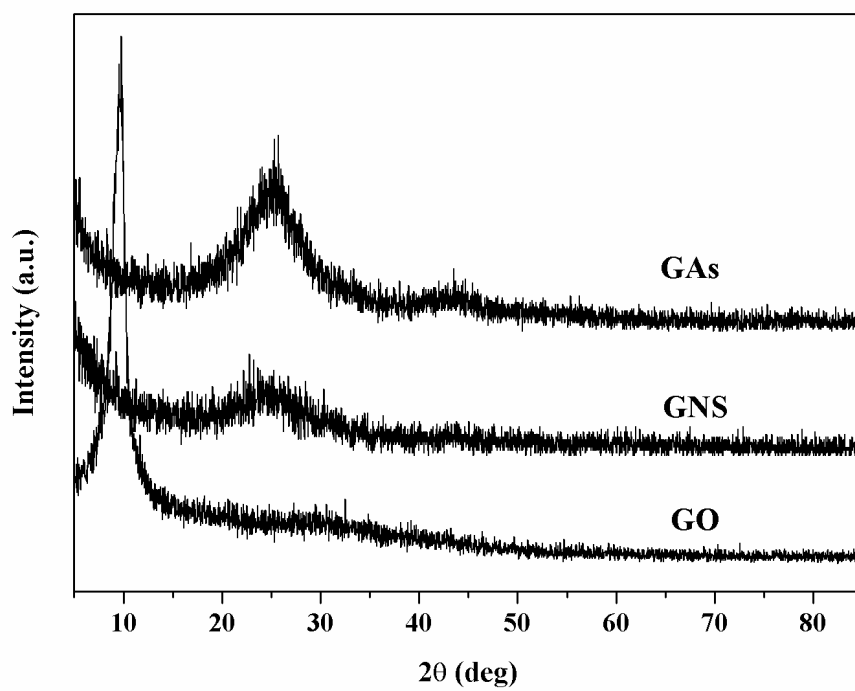


Fig. S2 XRD patterns of graphene aerogels (GAs), graphene nanosheets (GNS) and graphene oxide (GO).

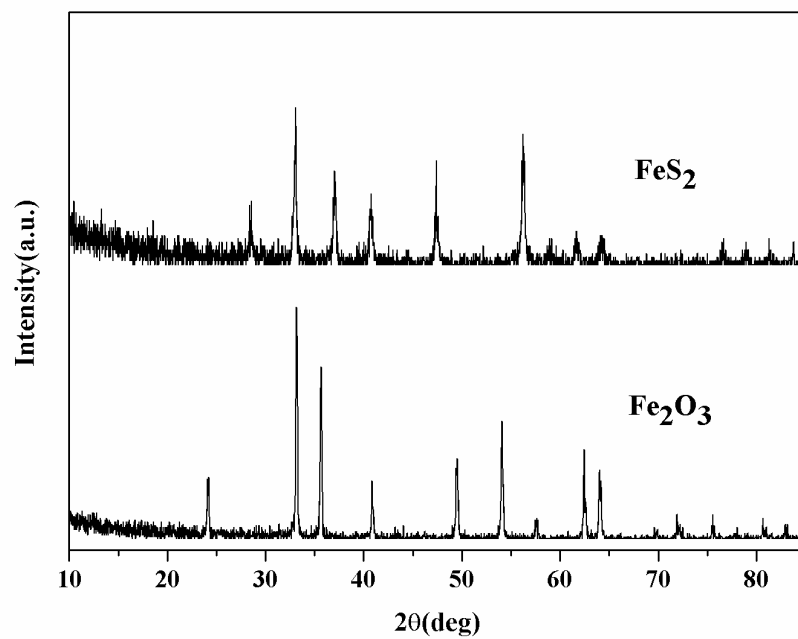


Fig. S3 XRD patterns of pristine Fe₂O₃ and FeS₂.

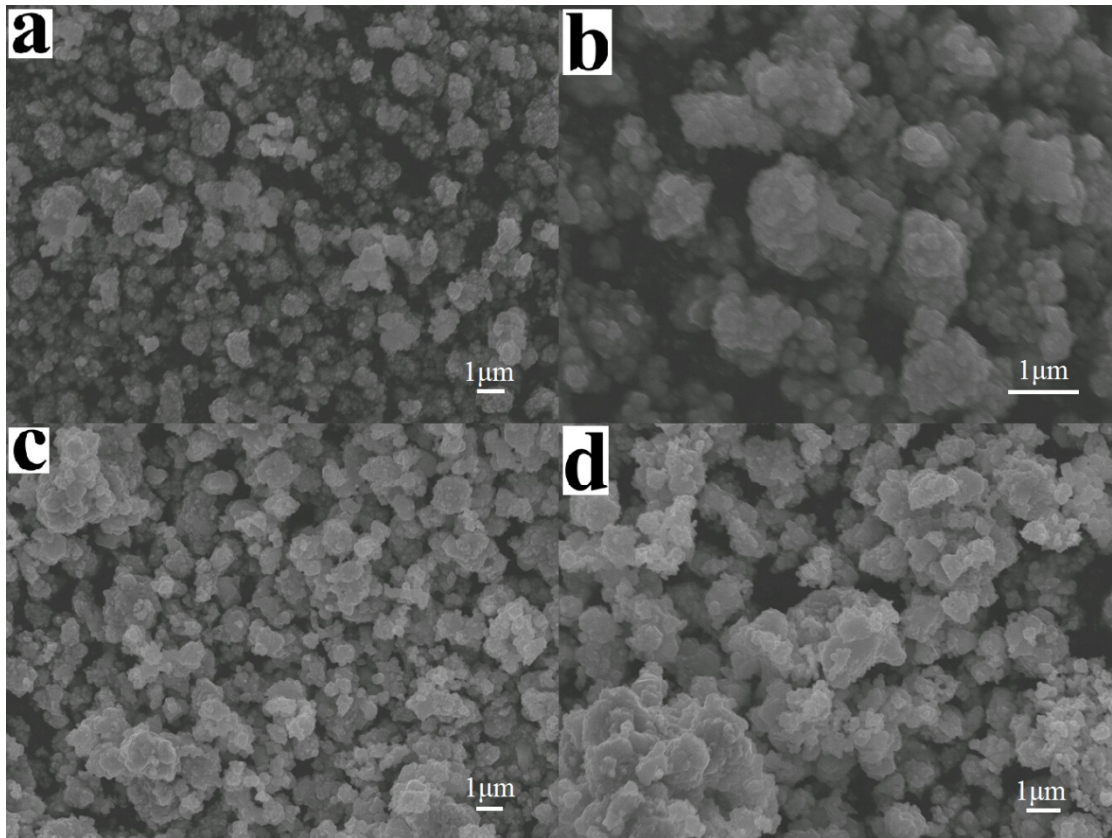


Fig. S4 (a, b) SEM images of pristine Fe_2O_3 and (c, d) SEM images of FeS_2 .

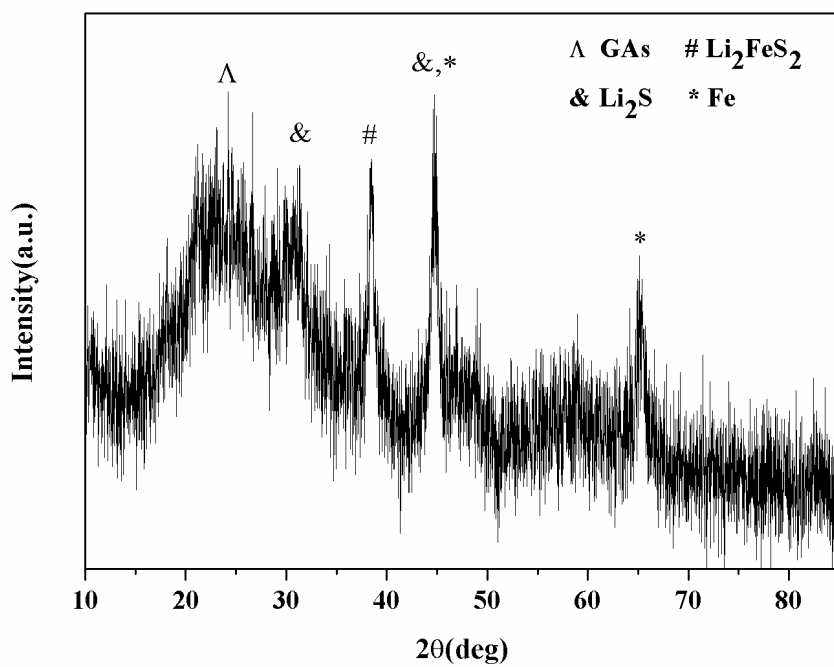


Fig. S5 XRD patterns of the FeS₂@C/S-GAs electrode, discharged at ~1.3 V after 80 cycles.

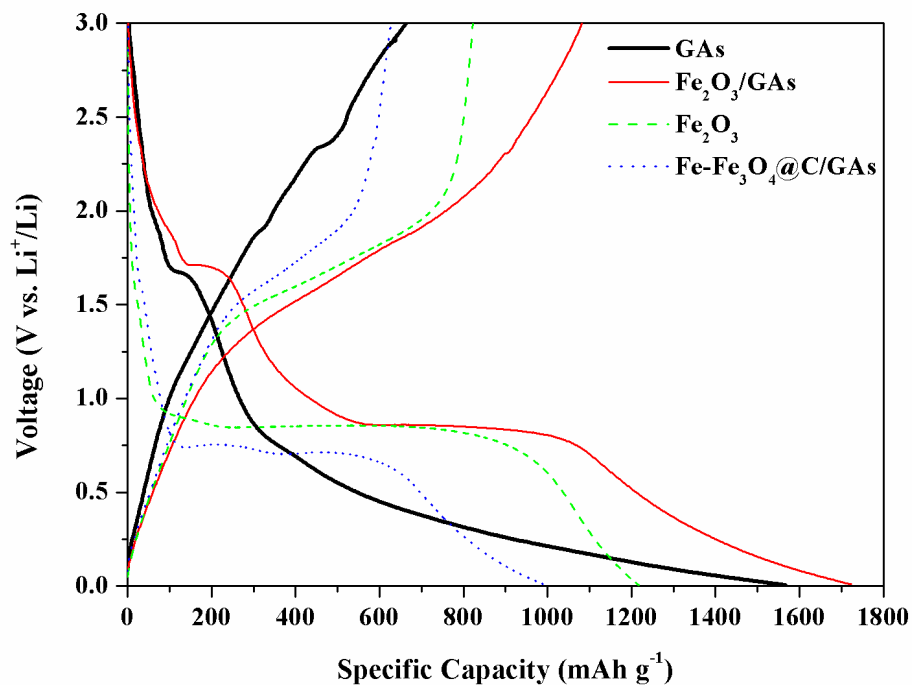


Fig. S6 First-cycle discharge (lithium insertion) and charge (lithium extraction) curves of the GAs, Fe₂O₃/GAs, Fe₂O₃, Fe-Fe₃O₄@C/GAs electrodes. Initial discharge capacities of 1567.1, 1723.9, 1216.9, 995.9 mAh g⁻¹ and charge capacities of 664.1, 1082.0, 822.1, 627.4 mAh g⁻¹ are observed for the GAs, Fe₂O₃/GAs, Fe₂O₃, Fe-Fe₃O₄@C/GAs electrodes respectively.

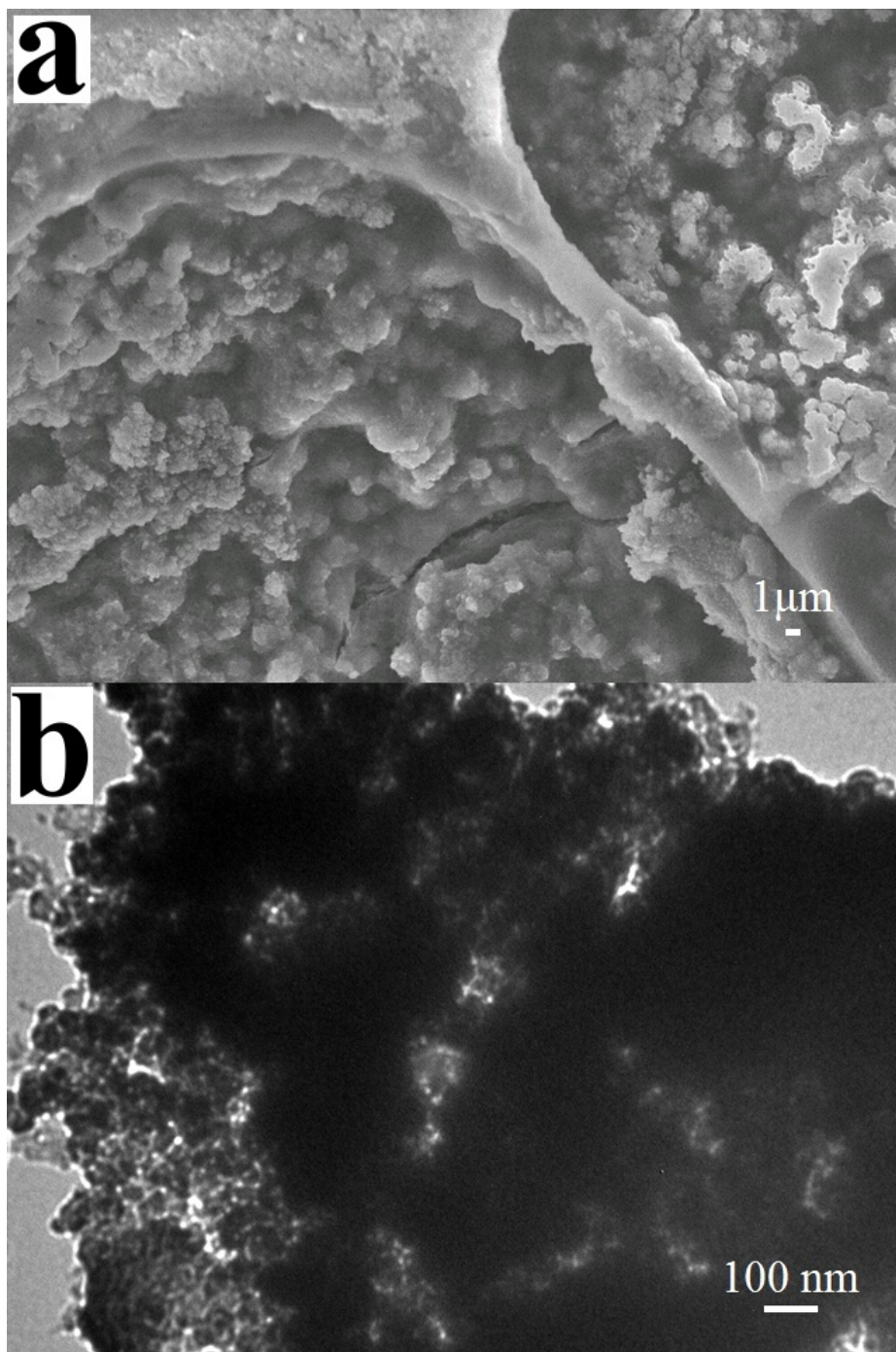


Fig. S7 The $\text{FeS}_2@\text{C}/\text{S-GAs}$ electrode after 80 cycles (a) SEM image and (b) TEM image. The conductive agent of carbon black and the PVDF binder are also present in the electrode.