

Electronic Supplementary Information

Binary transition metal nitrides with enhanced activity and durability for the oxygen reduction reaction

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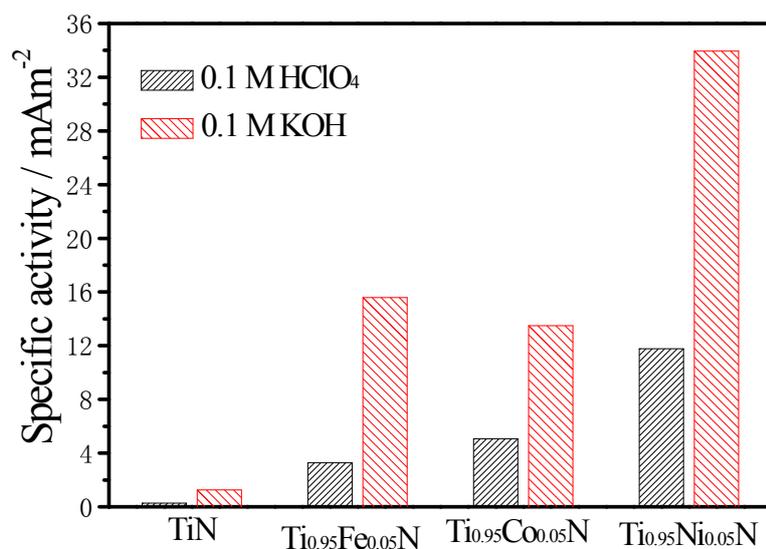


Fig. S1. The specific activity (current density was normalized to the BET surface area) of TiN and the BTMN catalysts in both 0.1 M HClO₄ and 0.1 M KOH at 0.7 V.

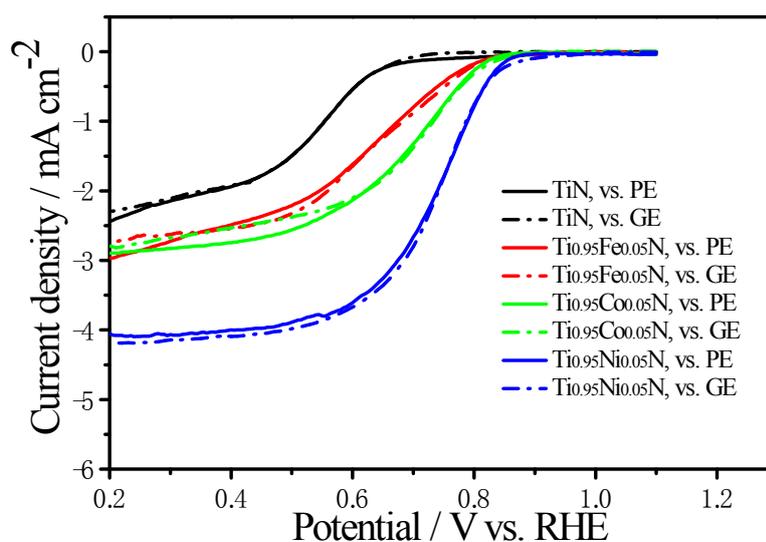


Fig. S2. Linear sweep voltammetry curves of TiN, Ti_{0.95}Fe_{0.05}N, Ti_{0.95}Co_{0.05}N, and Ti_{0.95}Ni_{0.05}N in O₂-saturated 0.1 M HClO₄ with a platinum wire as counter electrode (solid curves, PE), and a graphite rod as the counter electrode (dash curves, GE), respectively, at a scan rate of 10 mvs⁻¹.

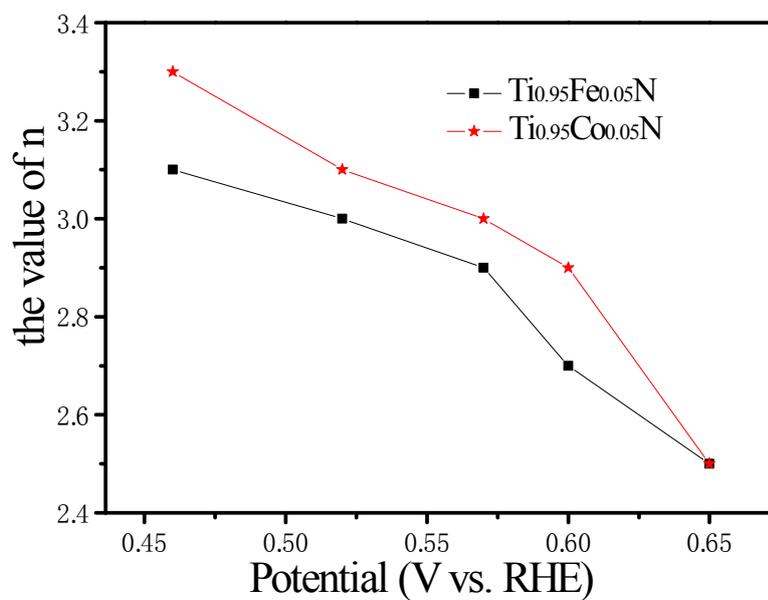


Fig. S3. The electron transfer number, n , of $\text{Ti}_{0.95}\text{Fe}_{0.05}\text{N}$ and $\text{Ti}_{0.95}\text{Co}_{0.05}\text{N}$ catalysts were derived from the slopes of the Koutecky–Levich plots at various potentials.

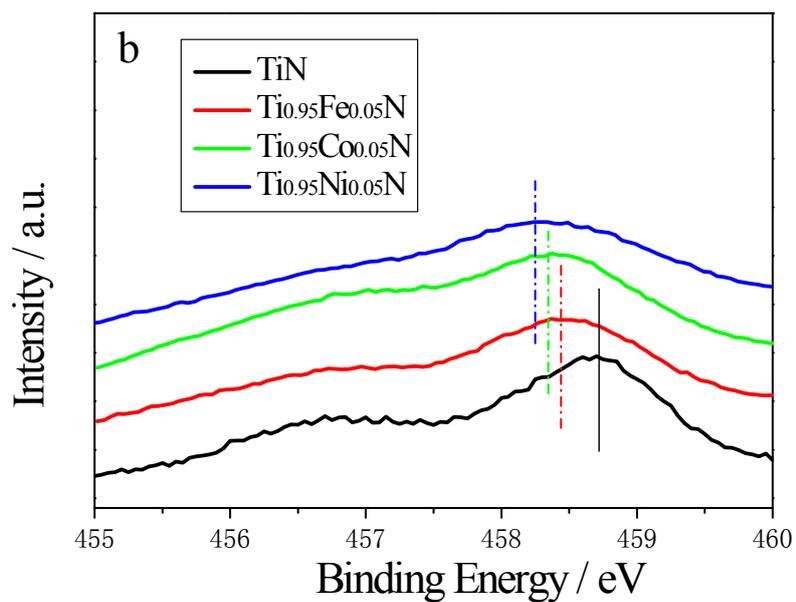
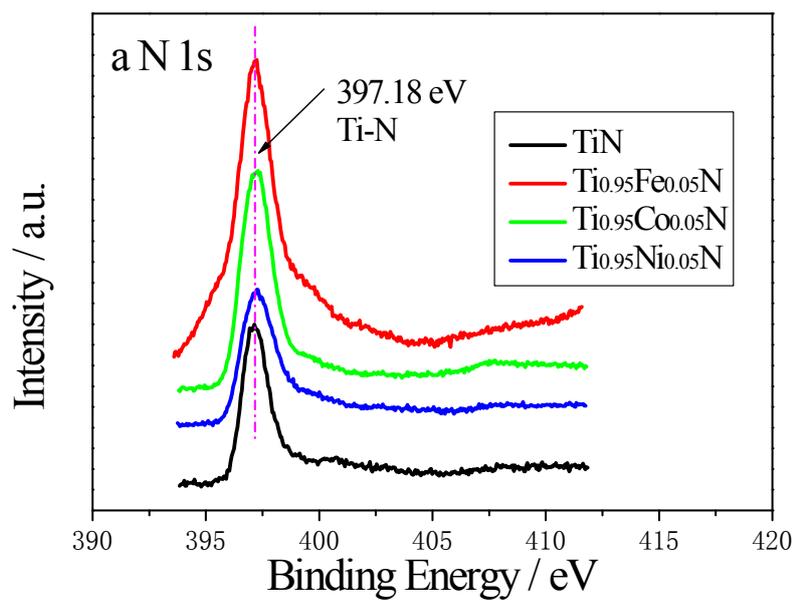


Fig. S4. (a) High-resolution N 1s spectra, and (b) the enlarged Ti 2p_{3/2} part of TiN, Ti_{0.95}Fe_{0.05}N, Ti_{0.95}Co_{0.05}N, and Ti_{0.95}Ni_{0.05}N, respectively.

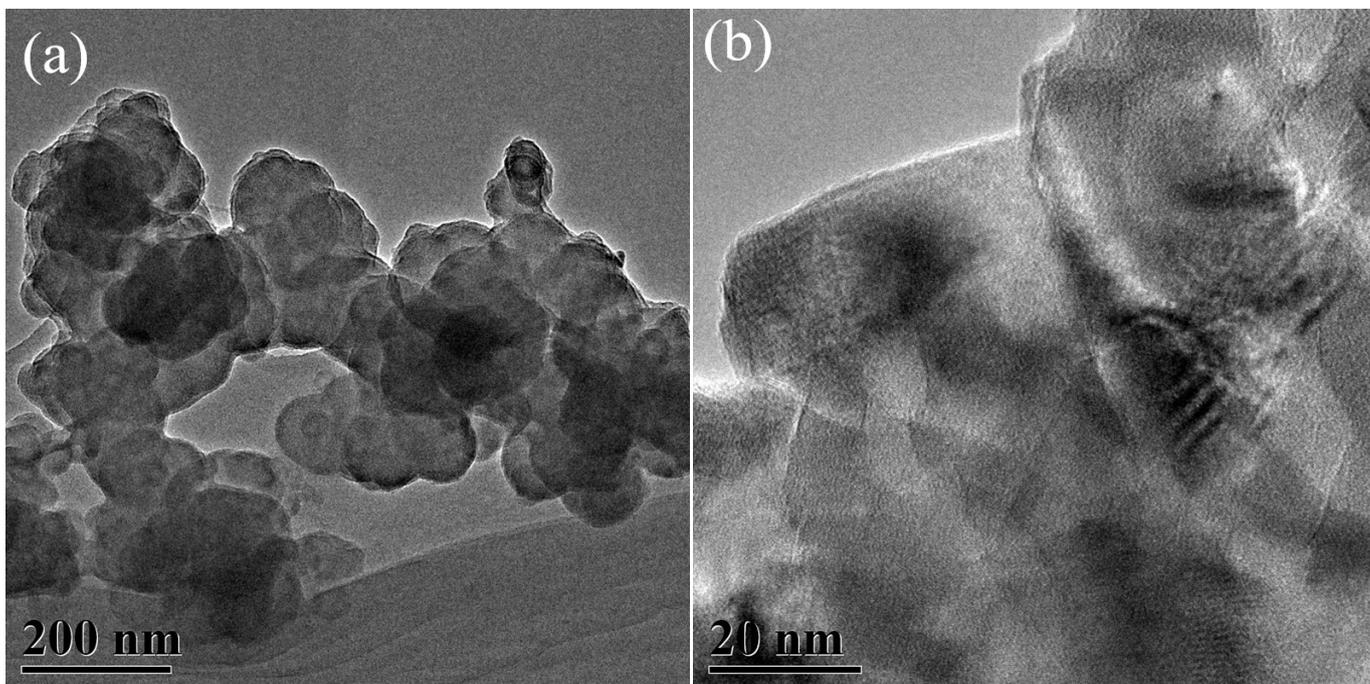


Fig. S5. (a) TEM image of and enlarged TEM image of $\text{Ti}_{0.95}\text{Ni}_{0.05}\text{N}$ annealed at 800°C

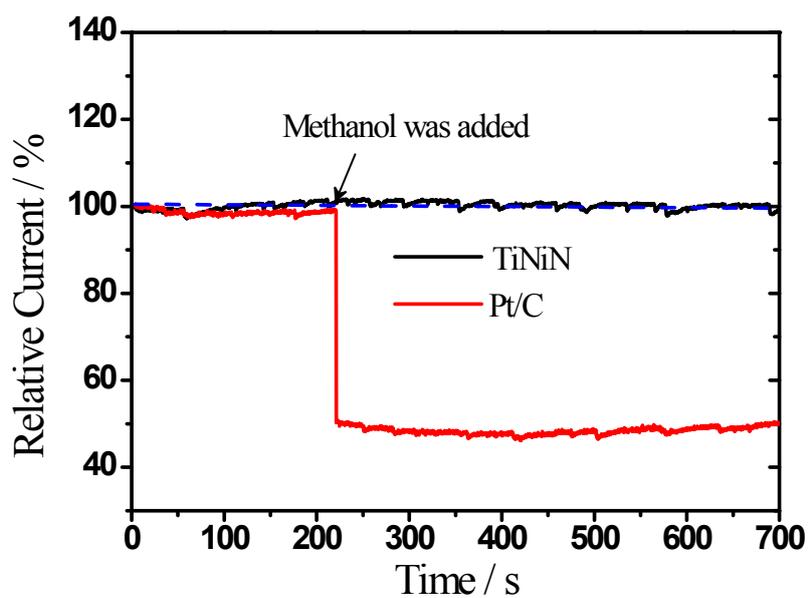


Fig. S6. Chronoamperometric response for ORR of $\text{Ti}_{0.95}\text{Ni}_{0.05}\text{N}$ and Pt/C electrodes with the addition of methanol to the electrolyte after about 230 s at 0.7 V.

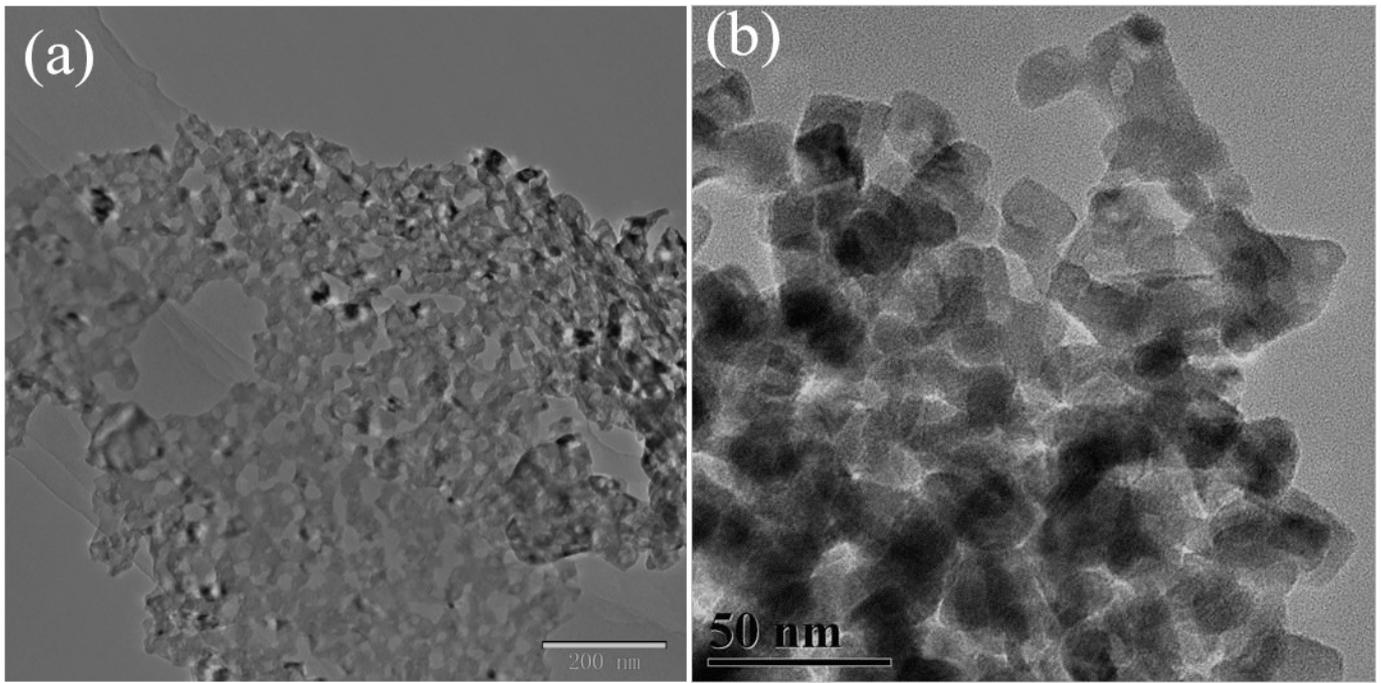


Fig. S7. (A) TEM and (b) enlarged TEM images of $\text{Ti}_{0.95}\text{Ni}_{0.05}\text{N}$ after the durability test.

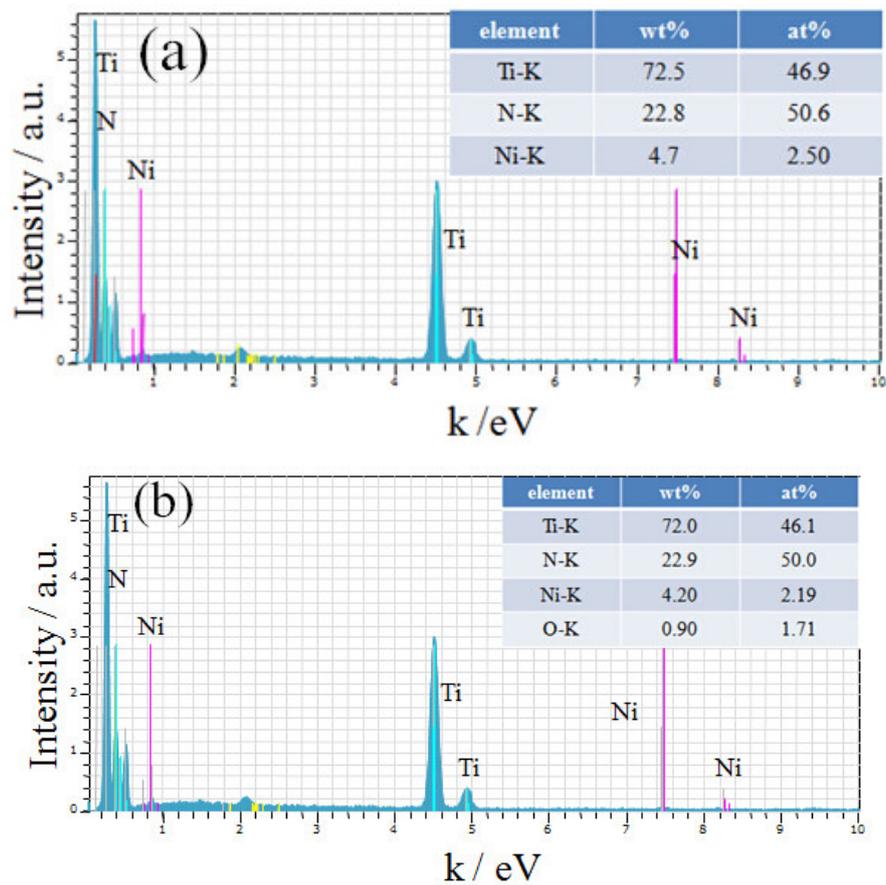


Fig. S8. The EDX profiles of $\text{Ti}_{0.95}\text{Ni}_{0.05}\text{N}$ catalyst (a) before and (b) after the durability test.

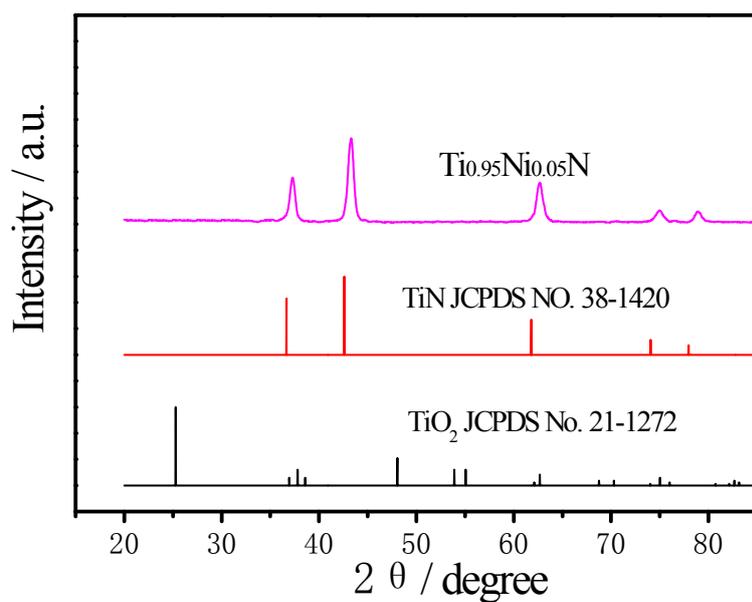


Fig. S9. XRD pattern of $\text{Ti}_{0.95}\text{Ni}_{0.05}\text{N}$ after the durability test.