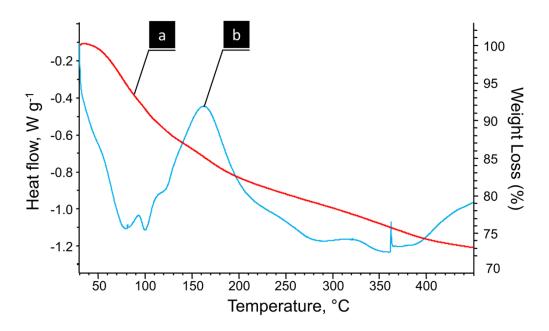
## **Electronic Supplementary Information (ESI)**

## Graphene oxide supported sodium stannate lithium ion battery anodes by the peroxide route: Low temperature and no waste processing

A. A. Mikhaylov,<sup>a,b</sup> A. G. Medvedev,<sup>a,b</sup> C. W. Mason,<sup>c</sup> A. Nagasubramanian, <sup>c,d,e</sup> S. Madhavi, <sup>c,d,e</sup> S. K. Batabyal,<sup>d</sup> Q. Zhang,<sup>e</sup> J. Gun,<sup>b</sup> P. V. Prikhodchenko<sup>\*a,d</sup> and O. Lev<sup>\*b</sup>

- <sup>a</sup>Kurnakov Institute of General and Inorganic Chemistry, Russian Academy of Sciences, Leninskii prosp.31, Moscow 119991, Russia.
- <sup>b</sup>The Casali Center of Applied Chemistry, The Institute of Chemistry, The Hebrew University of Jerusalem, Jerusalem 91904, Israel.
- <sup>c</sup>TUM CREATE, 1 CREATE Way, 10-02 Create Tower, Singapore 138602, Singapore.
- <sup>d</sup>Energy Research Institute @ NTU, Nanyang Technological University, 50 Nanyang Avenue, Singapore 639798, Singapore.
- <sup>e</sup>School of Materials Science and Engineering, Nanyang Technological University 50 Nanyang Avenue, Singapore 639798, Singapore



**Fig. S1** Thermal analysis of GO supported sodium peroxostannate a) thermogravimetry, TG; b) differential scanning calorimetry, DSC

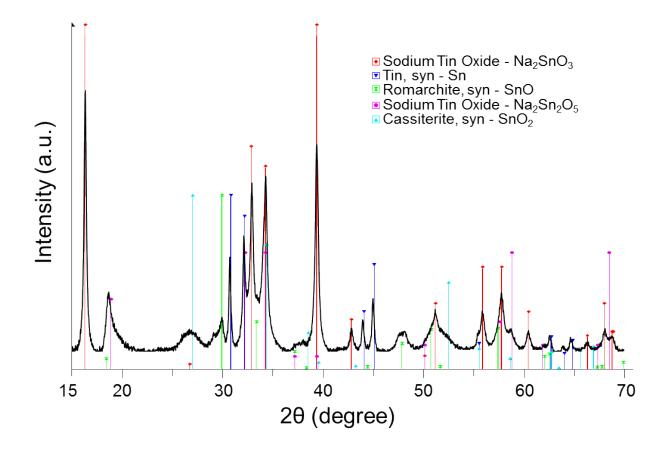


Fig. S2 X-ray diffractogram of sodium stannate heat treated in vacuum NaSnOx-GO500-2.

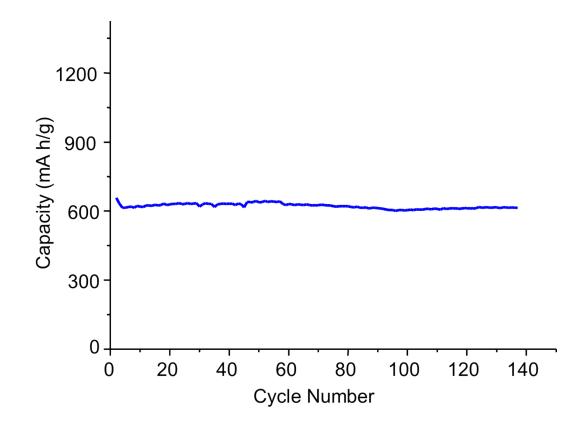
## Results of phase analysis of NaSnOx-GO500-2.

00-030-1252 Sodium Tin Oxide  $Na_2SnO_3$ Crystallite Size (Scherrer): 172.1 A System: Triclinic Space group: Cell param.: Initial Final 3.64000 3.64414 a: 5.78589 b: 5.79600 2.81300 2.81308 c: 102.326 alpha: 102.170 100.883 beta: 100.630 gamma:72.830 72.883 04-003-5664 Tin Sn Crystallite Size (Scherrer): 326.3 A System: Tetragonal Space group: I41/amd (141) Cell param.: Initial Final a: 5.81510 5.81810 c: 3.17600 3.17666

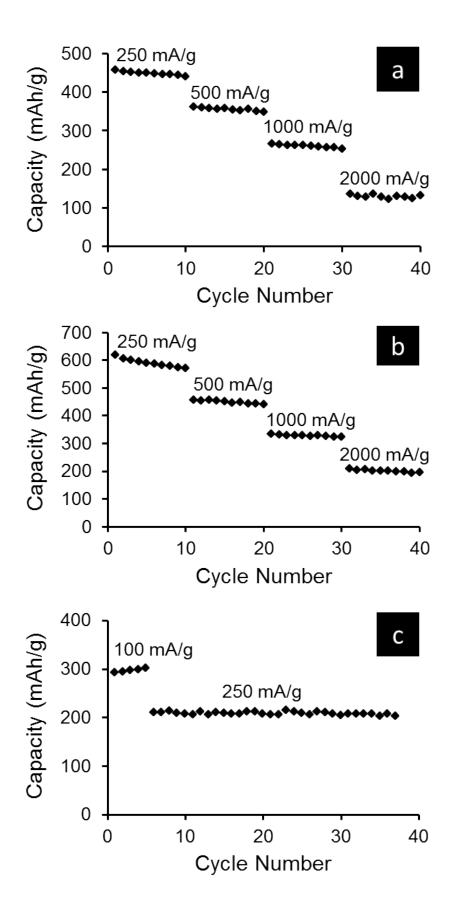
04-008-7670 Romarchite SnO Crystallite Size (Scherrer): 18.9 A System: Tetragonal Space group: P4/nmm (129) Cell param.: Initial Final a: 3.80360 3.79625 c: 4.83850 4.83850

01-071-4819 Cassiterite SnO<sub>2</sub> Crystallite Size (Scherrer): 25.0 А System: Tetragonal Space group: P42/mnm (136) Cell param .: Initial Final a: 4.68000 Fixed 3.15000 Fixed c: c: 3.15000 Fixed

00-021-1165 Sodium Tin Oxide Na<sub>2</sub>Sn<sub>2</sub>O<sub>5</sub> Crystallite Size (Scherrer): 116.7 A



**Fig. S3** Charge capacity of the NaSnOx-GO400 anodes upon repeated cycling. The cycles were conducted at a rate of 100 mA  $g^{-1}$  between 0–2.5 V vs. Li/Li<sup>+</sup>.



**Fig. S4** Charging capacities of the NaSnOx-GO400 anodes with different compositions 90% active material and 10% CMC without adding carbon (a), 80% active material, 10% CMC and 10% acetylene black (b), 80% active material and 20% CMC without adding carbon (c) conducted over the range 0–2.5 V vs. Li/Li+ electrode at different rates.

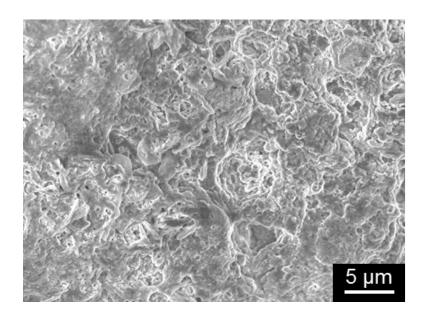


Fig. S5 STEM image of NaSnOx-GO400 anode after cycling.

## Experimental Microscopies and spectroscopies.

**High-resolution transmission electron microscope (HRTEM)**. We used a FEI Technai F20 G2 (Eindhoven, Holland) high-resolution transmission electron microscopes. HRTEM imaging was performed at 200 kV. A drop of the suspension of the sample in ethanol was deposited onto 400 mesh copper grids covered with a lacy carbon net.

**High-resolution scanning electron microscope (HRSEM).** HRSEM imaging was performed using a FEI Sirion high-resolution scanning electron microscope (Eindhoven, Holland). The accelerating voltage was set at 5–15 kV with a 5 mm working distance. Imaging was conducted using the high-resolution mode with a through-the-lens detector. The specimen was prepared by dispersion of dried samples in ethanol in an ultrasonic bath, and the suspension was then dropped on a silicon wafer to dry.

**Scanning transmission electron microscope (STEM).** STEM imaging was performed at 20 kV using a FEI extra high-resolution scanning electron microscope, MagellanTM 400L (Eindhoven, Holland). The specimen was prepared by deposition of a drop of the ethanol suspension of the sample onto 400 mesh copper grid.

**X-ray photoelectron spectroscopy (XPS).** XPS measurements were performed on a Kratos Axis Ultra x-ray photoelectron spectrometer (Manchester, UK). High-resolution spectra were acquired with a monochromated Al K $\alpha$  (1486.6 eV) x-ray source with 0° takeoff angle. The pressure in the test chamber was maintained at 1.7 × 10<sup>-9</sup> Torr during the acquisition process.

Data analysis was performed with Vision processing data reduction software (Kratos Analytical Ltd) and CasaXPS (Casa Software Ltd).

**X-ray powder diffraction (XRD).** XRD measurements were performed on a D8 Advance Diffractometer (Bruker AXS, Karlsruhe, Germany) with a goniometer radius 217.5 mm, Gobel Mirror parallel-beam optics, 2° Sollers slits and a 0.2 mm receiving slit. The powder samples were carefully filled into low-background quartz sample holders. The specimen weight was approximately 0.5 g. XRD patterns from 5° to 65° 20 were recorded at room temperature using Cu K $\alpha$  radiation (k = 1.5418 A) under the following measurement conditions: tube voltage of 40 kV, tube current of 40 mA, step scan mode with a step size 0.02° 20 and counting time of 2 s/step. XRD patterns were processed using Diffrac Plus software.