

# Controlled synthesis of three-dimensional interconnected graphene-like nanosheets from graphite microspheres as high-performance anodes for lithium-ion batteries

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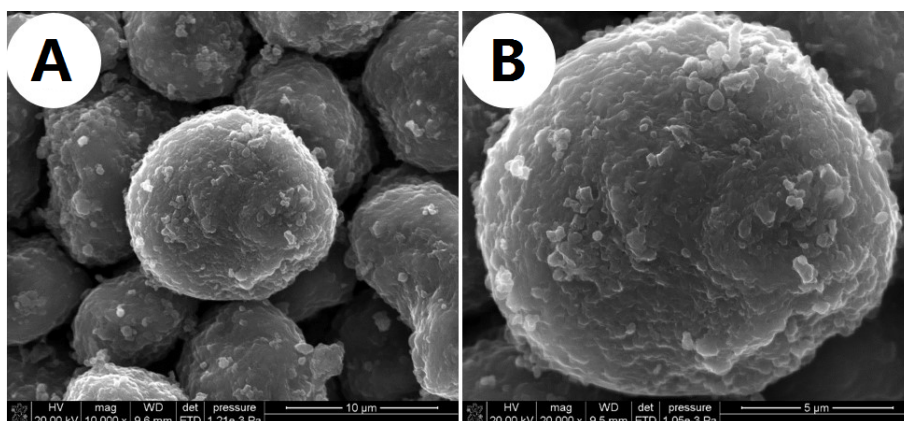


Fig. S1 Typical SEM images of GMCMBs.

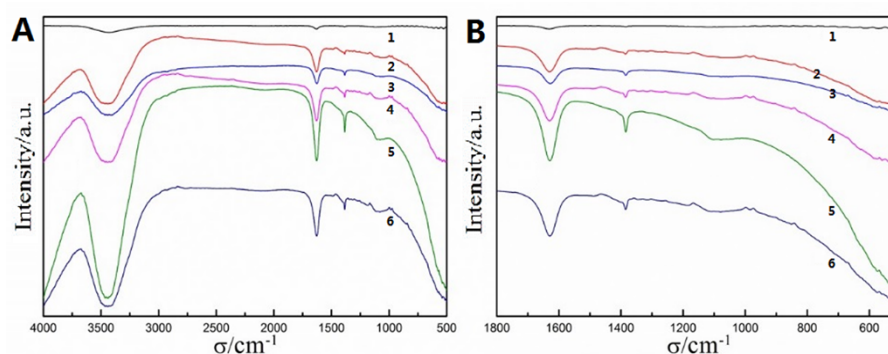


Fig. S2 FT-IR spectrums of GMCMBs (1) and 3DGNs (2-6) treated with different  $\text{KMnO}_4$  dosages (2:10g, 3:12g, 4:14g, 5:16g, 6:18g) at the same heating temperature of 950 °C.

Fig. S2 shows the FT-IR spectras of f GMCMBs and 3DGNs samles (A is the full spectrum, B is the partial spectrum of 550-1800  $\text{cm}^{-1}$ ). The absorption peak at about 3440  $\text{cm}^{-1}$  is characterized by  $-\text{OH}$  stretching vibration peaks. The absorption peak at about 1620  $\text{cm}^{-1}$  is generated by the  $\text{C}=\text{C}$  bond stretching vibration of adjacent carbon atoms in graphite. The absorption peak at about 1400  $\text{cm}^{-1}$  is the O-H bending vibration of the hydroxyl group in graphite, and the absorption peak at about 1100  $\text{cm}^{-1}$  is the C-O-C asymmetric stretching vibration peak. All the absorption peaks in GMCMBs are very weak, and these absorption peaks in 3DGNs are obviously enhanced, which indicates that, after the oxidation and high temperature treatment, substantial oxygen containing groups have been introduced in 3DGNs.