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Supporting Information

New hybrid nanocatalyst based on Cu-doped Pd -Fe₃O₄

for tandem synthesis of 2-phenylbenzofurans

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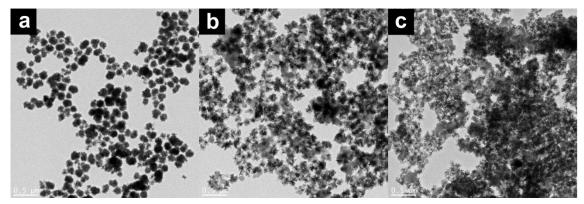


Fig. S1 Low magnified TEM images of the (a) Cu-doped Pd-Fe $_3$ O $_4$ -0, (b) Cu-doped Pd-Fe $_3$ O $_4$ -0.3 and (c) Cu-doped Pd-Fe $_3$ O $_4$ -0.5 nanocomposites.

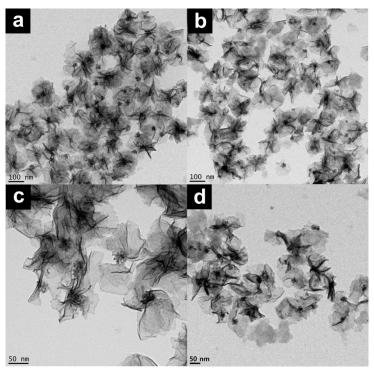


Fig. S2 TEM images of the Cu-doped Pd-Fe $_3$ O $_4$ -0.3 nanocomposites synthesized at different temperature (a,c) 160 °C and (b,d) 200 °C.

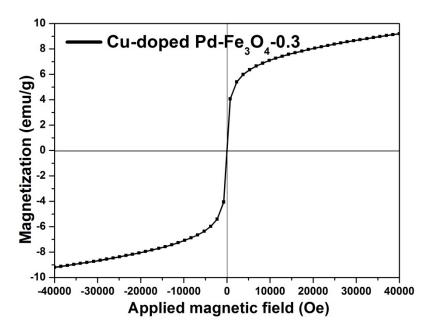


Fig. S3 SQUID data of Cu-doped Pd-Fe₃O₄ nanocomposites.

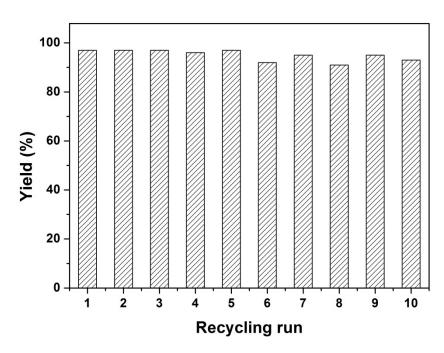


Fig. S4 GC-MS yields during ten recycling runs. Reaction condition: Cu-doped Pd- Fe_3O_4 -0.3 catalyst (Cu base: 3.0 mol%), 2-iodophenol (0.5 mmol), phenylpropiolic acid (0.6 mmol), NaOAc (1.0 mmol), DMSO (5.0 ml) and 130 °C for 3 h.

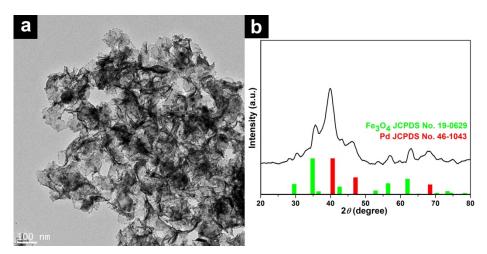


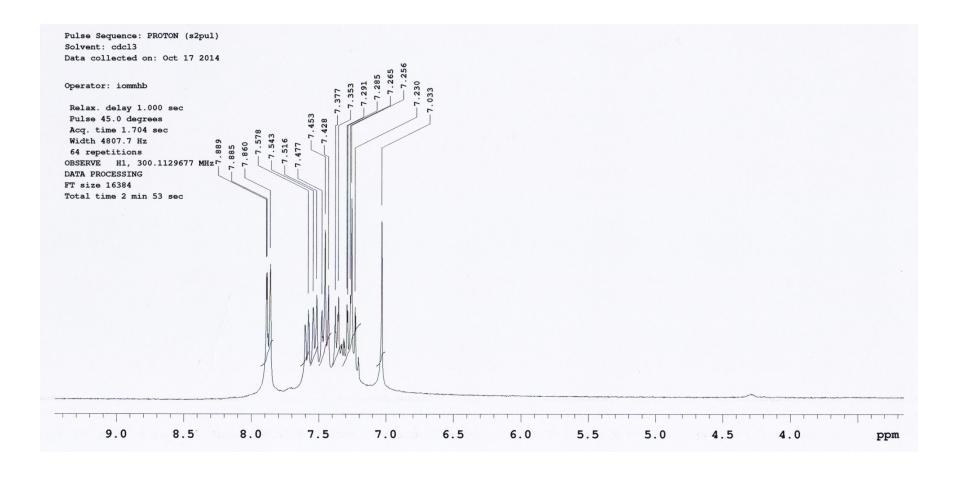
Fig. S5 (a,b) TEM image and XRD pattern of Cu-doped Pd-Fe $_3$ O $_4$ -0.3 catalyst after first reaction.

Data for ¹H and GC-MS spectra are reported as follows:

Chemical shift (δ ppm), multiplicity (s = singlet, d = doublet, t = triplet, m = multiplet), coupling constant (Hz), and integration.

2-phenybenzofuran¹ (**Table 1**): ¹H-NMR (CDCl₃, 300 MHz): $\delta = 7.87$ (d, J = 8.1 Hz, 2H), 7.59 (d, J = 7.2 Hz, 1H), 7.53 (d, J = 8.1 Hz, 1H), 7.45 (t, J = 7.5 Hz, 2H), 7.35 (t, J = 7.2 Hz, 1H), 7.29 (t, J = 6.9 Hz, 1H), 7.23 (t, J = 6.9 Hz, 1H), 7.03 (s, 1H). MS (EI) m/z: 194(100), 165(60), 139(9), 97(11), 82(13), 28(16).

1. D. Zhao, C. Gao, X. Su, Y. He, J. You and Y. Xue, *Chem. Commun.* **2010**, *46*, 9049-9051.



Spectrum



