

## Electronic Supplementary Information

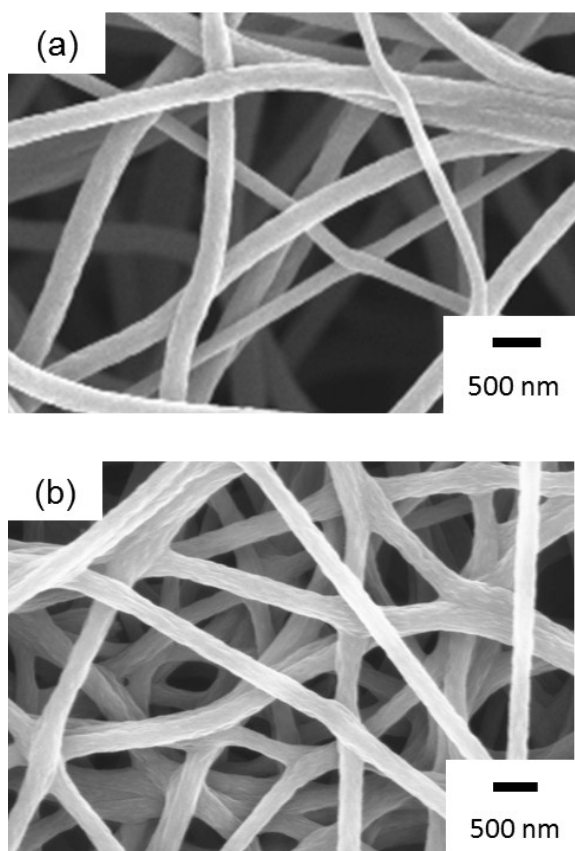
### High-energy Density Nanofiber-based Solid-state Supercapacitors

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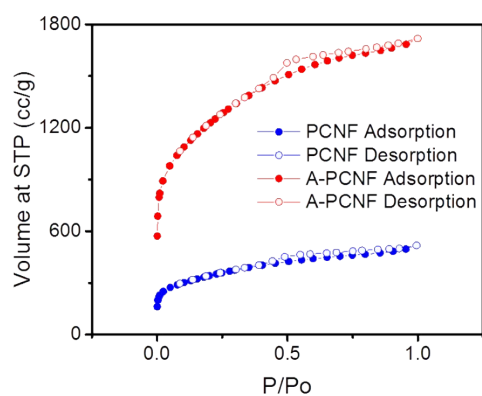
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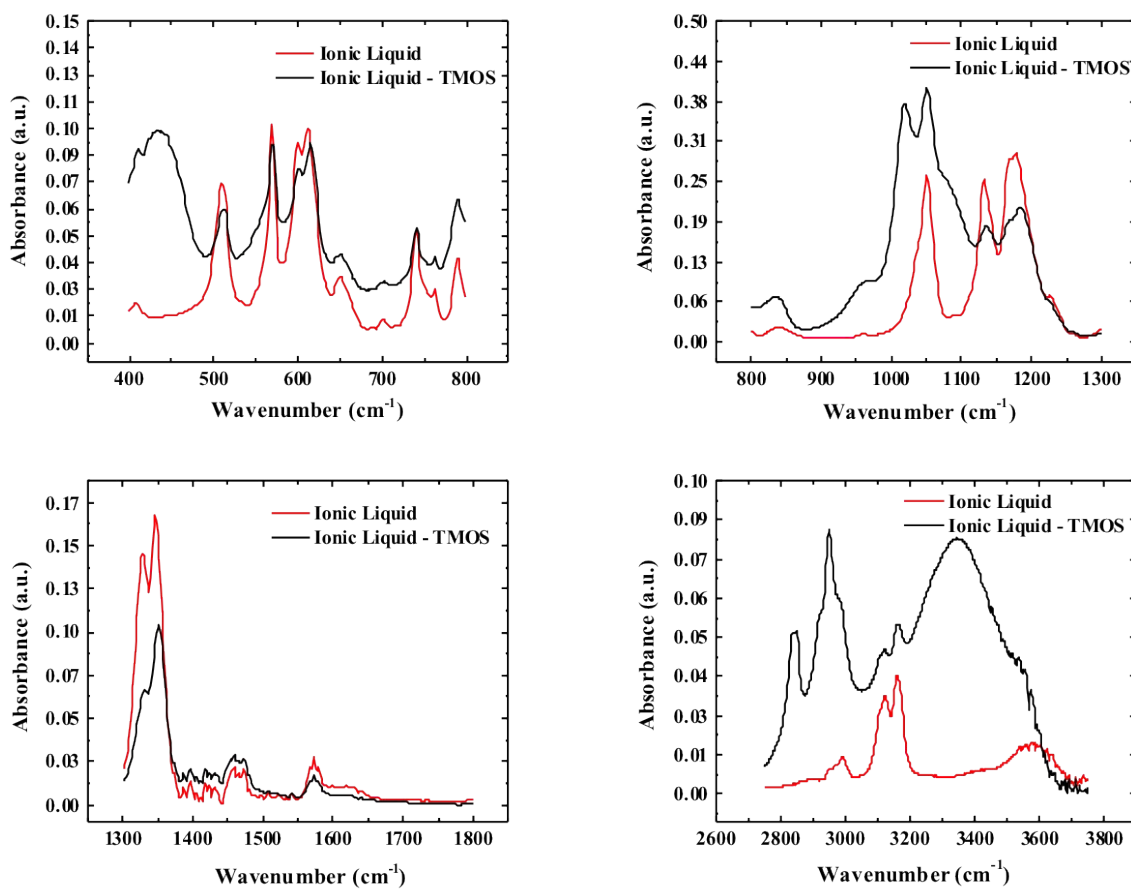
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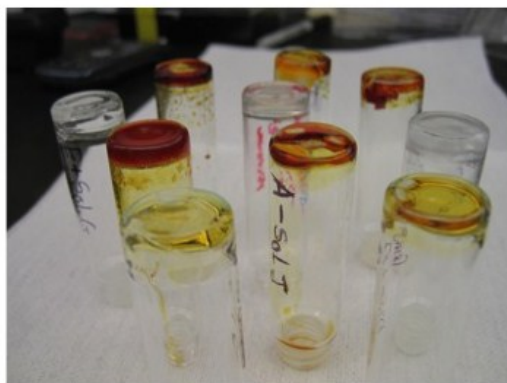
**Fig. S1.** SEM images of carbon nanofiber mats (a) PCNF and (b) A-PCNF (activated).



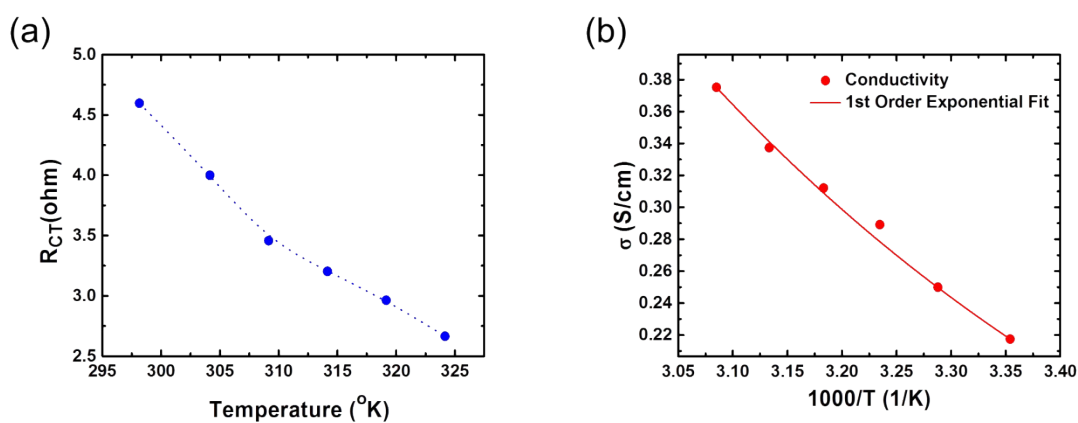
**Fig. S2.** Adsorption and desorption isotherms for PCNF and A-PCNF.



**Fig. S3.** FTIR spectra of ionic liquid electrolyte with and without TMOS gelation.



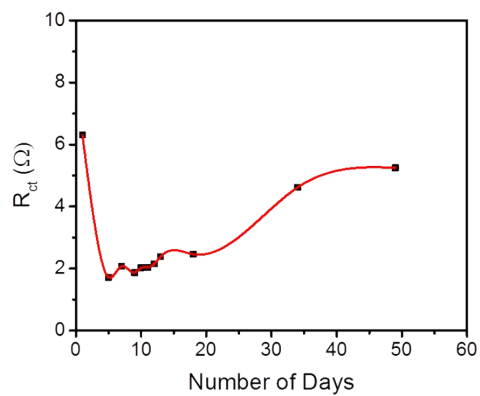
**Fig. S4.** Pictures of gelled ionic liquid in vials.



**Fig. S5.** (a) Variation of charge transfer resistance of solid-state EMIM TFSI electrolyte with temperature. The electrolyte is prepared using tetramethoxysilane (TMOS). (b) Conductivity of EMIM TFSI solid-state electrolyte as a function of inverse temperature. The data fits well to first order exponential, indicating that it tends to follow VFT behavior.

**Table S1:** Temperature dependence of the charge transfer resistance of solid-state EMIM TFSI electrolyte.

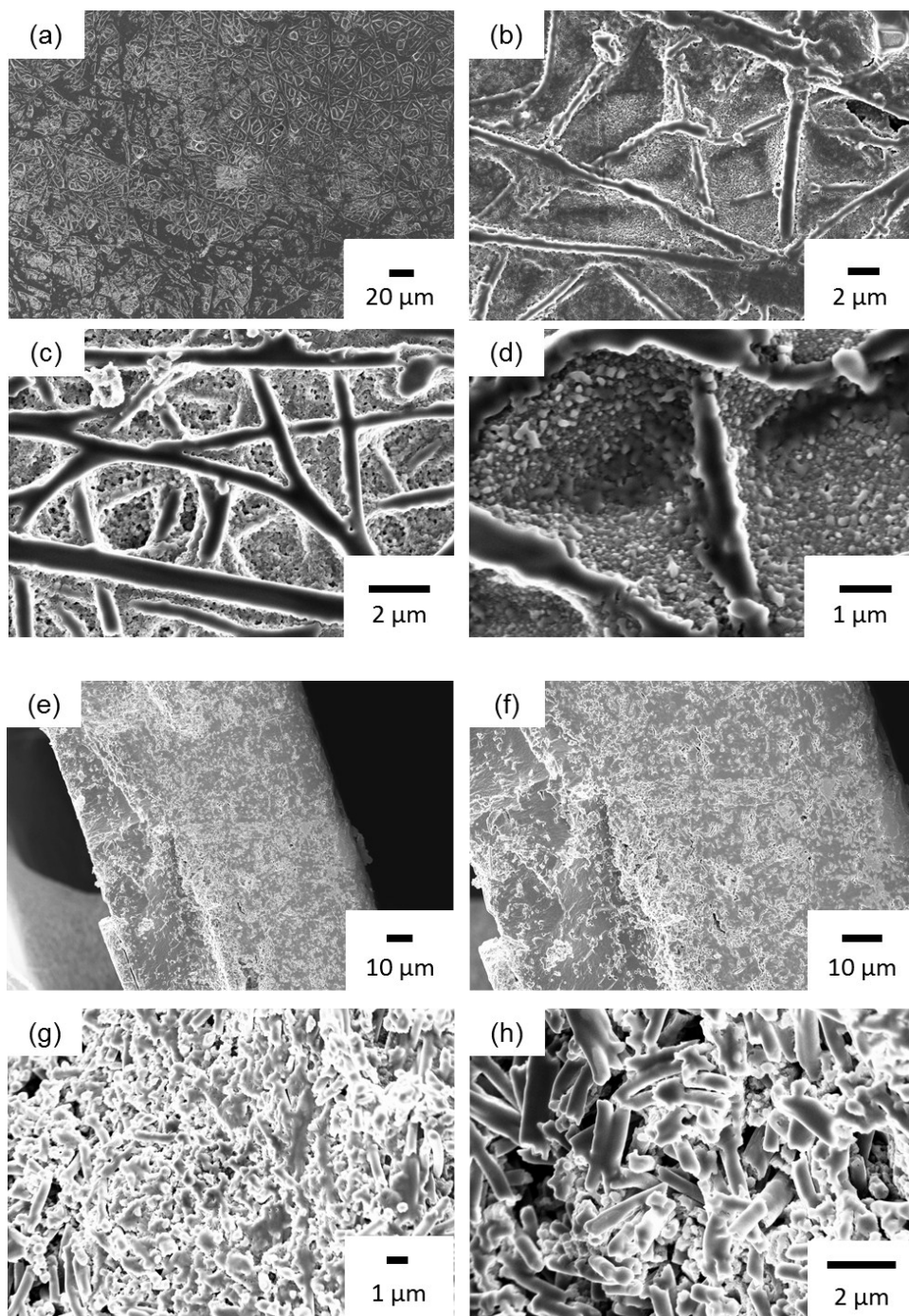
Temperature	R <sub>ct</sub> (Ω)	1000/T	σ (S/cm)
298.15	4.60	3.35	0.217
304.15	4.00	3.29	0.249
309.15	3.46	3.23	0.289
314.15	3.20	3.18	0.312
319.15	2.96	3.13	0.337
324.15	2.66	3.08	0.375



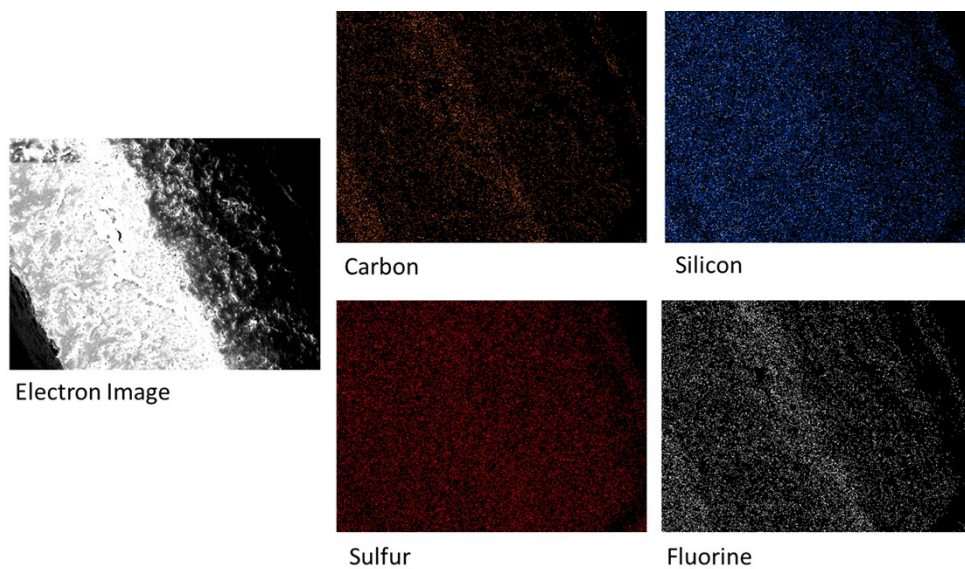
**Fig. S6.** Time dependence of charge transfer resistance of the solid-state EMIM TFSI electrolyte.

**Table S2:** Time dependence of the charge transfer resistance of solid-state EMIM TFSI electrolyte.

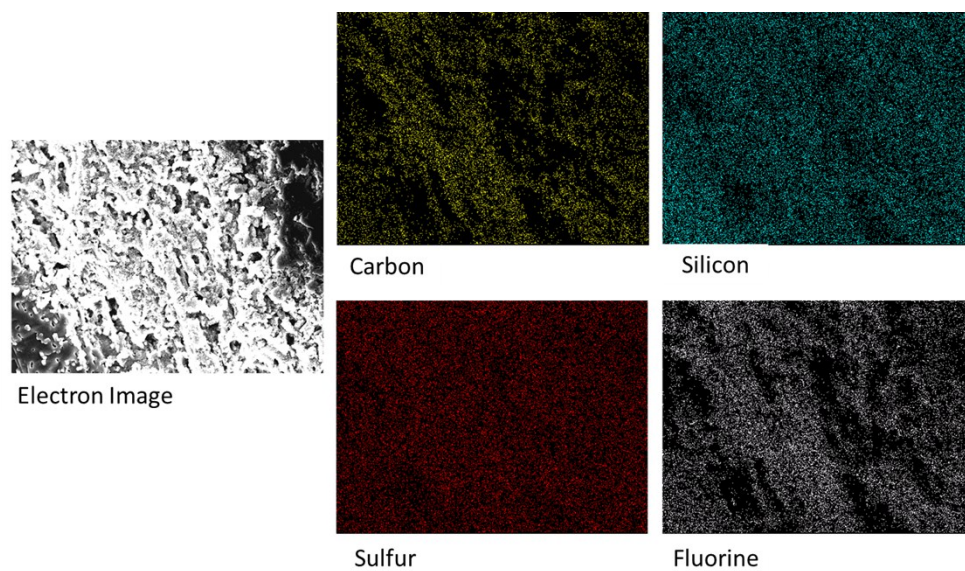
<b>Day</b>	<b>R<sub>ct</sub> (<math>\Omega</math>)</b>
1	6.305
5	1.715
7	2.079
9	1.8705
10	2.029
11	2.036
12	2.148
13	2.387
18	2.4615
34	4.616
49	5.25



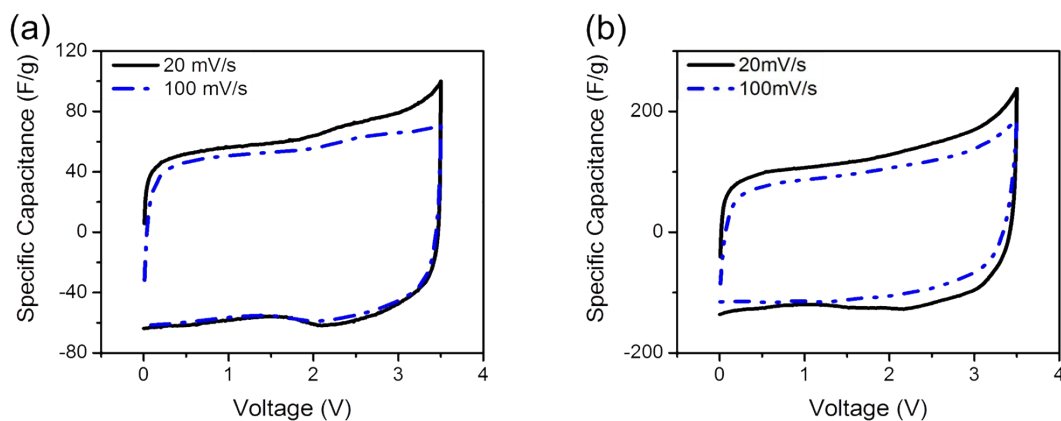
**Fig. S7.** SEM images of PCNF electrodes filled with solid electrolyte from a (a-d) top-down and (e-h) cross-sectional view.



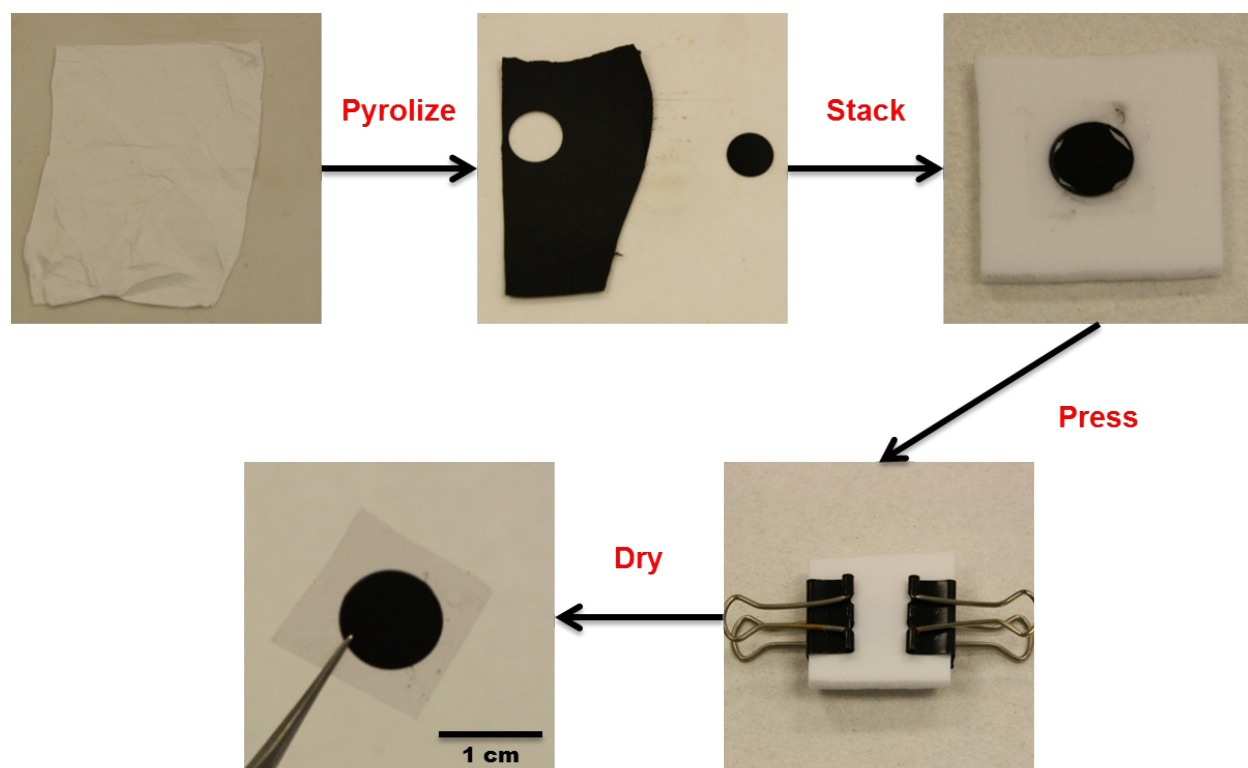
**Fig. S8.** Solid electrolyte-filled PCNF cross-section EDS showing uniform distribution of TMOS (Si) and EMIM TFSI (S, F).



**Fig. S9.** Solid electrolyte-filled PCNF EDS (top surface) showing uniform distribution of TMOS (Si) and EMIM TFSI (S, F).

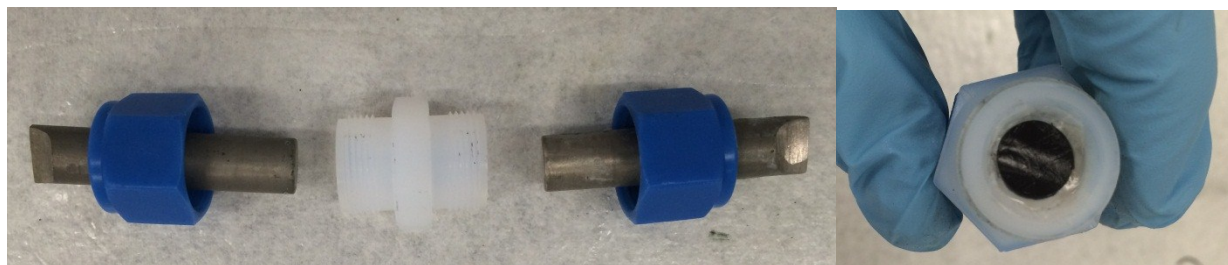


**Fig. S10.** CV curves of liquid ionic liquid electrolyte devices using the following samples as electrodes (a) PCNF and (b) A-PCNF. PCNF exhibits a capacitance of  $59.5 \text{ F g}^{-1}$  and  $54 \text{ F g}^{-1}$  at 20 and  $100 \text{ mV s}^{-1}$  respectively. A-PCNF exhibits a capacitance of  $125 \text{ F g}^{-1}$  and  $101 \text{ F g}^{-1}$  at 20 and  $100 \text{ mV s}^{-1}$  respectively.



**Fig. S11.** Schematic showing the detailed step-by-step process used for solid-state device fabrication.





**Fig. S12.** Photograph representation of the Swagelok cell and the solid-state supercapacitor insert.