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## **Supplementary information:**

		Temperature, °C							
Composition		900	950	1000	1050	1100	1150	1200	1250
Li <sub>2</sub> O×	0.5B <sub>2</sub> O <sub>3</sub>	$\checkmark$	$\checkmark$	$\checkmark$	$\checkmark$	$\checkmark$	$\checkmark$	$\checkmark$	$\checkmark$
	1.0B <sub>2</sub> O <sub>3</sub>	$\checkmark$	$\checkmark$	$\checkmark$	$\checkmark$	$\checkmark$	$\checkmark$	$\checkmark$	$\checkmark$
	2.0B <sub>2</sub> O <sub>3</sub>	$\checkmark$	$\checkmark$	$\checkmark$	$\checkmark$	$\checkmark$	$\checkmark$	$\checkmark$	$\checkmark$
	3.0B <sub>2</sub> O <sub>3</sub>	$\checkmark$	$\checkmark$	$\checkmark$	$\checkmark$	$\checkmark$	$\checkmark$	$\checkmark$	$\checkmark$
Na <sub>2</sub> O×	1.0B <sub>2</sub> O <sub>3</sub>	$\checkmark$	$\checkmark$	$\checkmark$	$\checkmark$	$\checkmark$	$\checkmark$	$\checkmark$	$\checkmark$
	2.0B <sub>2</sub> O <sub>3</sub>	$\checkmark$	$\checkmark$	$\checkmark$	$\checkmark$	$\checkmark$	$\checkmark$	$\checkmark$	$\checkmark$
	3.0B <sub>2</sub> O <sub>3</sub>	$\checkmark$	$\checkmark$	$\checkmark$	$\checkmark$	$\checkmark$	$\checkmark$	$\checkmark$	$\checkmark$
K <sub>2</sub> O×	1.0B <sub>2</sub> O <sub>3</sub>			$\checkmark$		$\checkmark$		$\checkmark$	$\checkmark$
	2.0B <sub>2</sub> O <sub>3</sub>			$\checkmark$		$\checkmark$		$\checkmark$	$\checkmark$
	3.0B <sub>2</sub> O <sub>3</sub>			$\checkmark$		$\checkmark$		$\checkmark$	$\checkmark$

Table S1. Studied alkali borate compositions and synthesis temperatures employed.

Table S2. Studied alkaline earth borate compositions and synthesis temperatures employed.

		Temperature, °C					
Composition		1000	1100	1200	1250		
	0,5B <sub>2</sub> O <sub>3</sub>	$\checkmark$	$\checkmark$	$\checkmark$	$\checkmark$		
	1.0B <sub>2</sub> O <sub>3</sub>	$\checkmark$	$\checkmark$	$\checkmark$	$\checkmark$		
	1,5B <sub>2</sub> O <sub>3</sub>	$\checkmark$	$\checkmark$	$\checkmark$	$\checkmark$		
MgO×	2.0B <sub>2</sub> O <sub>3</sub>	$\checkmark$	$\checkmark$	$\checkmark$	$\checkmark$		
	2,5B <sub>2</sub> O <sub>3</sub>	$\checkmark$	$\checkmark$	$\checkmark$	$\checkmark$		
	3,5B <sub>2</sub> O <sub>3</sub>	$\checkmark$	$\checkmark$	$\checkmark$	$\checkmark$		
	5.0B <sub>2</sub> O <sub>3</sub>	$\checkmark$	$\checkmark$	$\checkmark$	$\checkmark$		
	1,5B <sub>2</sub> O <sub>3</sub>	$\checkmark$	$\checkmark$	$\checkmark$	$\checkmark$		
MgO×*)	2,5B <sub>2</sub> O <sub>3</sub>	$\checkmark$	$\checkmark$	$\checkmark$	$\checkmark$		
	3,5B <sub>2</sub> O <sub>3</sub>	$\checkmark$	$\checkmark$	$\checkmark$	$\checkmark$		
CaO×	0,5B <sub>2</sub> O <sub>3</sub>	$\checkmark$	$\checkmark$	$\checkmark$	$\checkmark$		
	1.0B <sub>2</sub> O <sub>3</sub>	$\checkmark$	$\checkmark$	$\checkmark$	$\checkmark$		

	1,5B <sub>2</sub> O <sub>3</sub>	$\checkmark$	$\checkmark$	$\checkmark$	$\checkmark$
	0,5B <sub>2</sub> O <sub>3</sub>	$\checkmark$	$\checkmark$	$\checkmark$	$\checkmark$
SrO×	1.0B <sub>2</sub> O <sub>3</sub>	$\checkmark$	$\checkmark$	$\checkmark$	$\checkmark$
	1,5B <sub>2</sub> O <sub>3</sub>	$\checkmark$	$\checkmark$	$\checkmark$	$\checkmark$
	0,5B <sub>2</sub> O <sub>3</sub>	$\checkmark$	$\checkmark$	$\checkmark$	$\checkmark$
BaO×	1.0B <sub>2</sub> O <sub>3</sub>	$\checkmark$	$\checkmark$	$\checkmark$	$\checkmark$
	1,5B <sub>2</sub> O <sub>3</sub>	$\checkmark$	$\checkmark$	$\checkmark$	$\checkmark$

\*) Heating in argon.



Fig. S1 X-ray diffraction pattern of the reaction products of lithium tetraborate with ammonia at 1200 °C. All reflections except the one marked with an asterisk were assigned to a hexagonal phase with the lattice constants a= 2.50 Å and c= 6.69 Å.



Fig. S2. SEM image of BNGPs grown on BNNTs surface from lithium tetraborate at 1200 °C.



Fig. S3 SEM image of MgO·1.5B<sub>2</sub>O<sub>3</sub> reacted with ammonia at 1200 °C.



Fig. S4 SEM images of samples heated in argon and synthesized in ammonia at 1200 °C from  $MgO\cdot 1.5B_2O_3$  (a) and  $MgO\cdot 3.5B_2O_3$  (b).