

## Supplementary Information

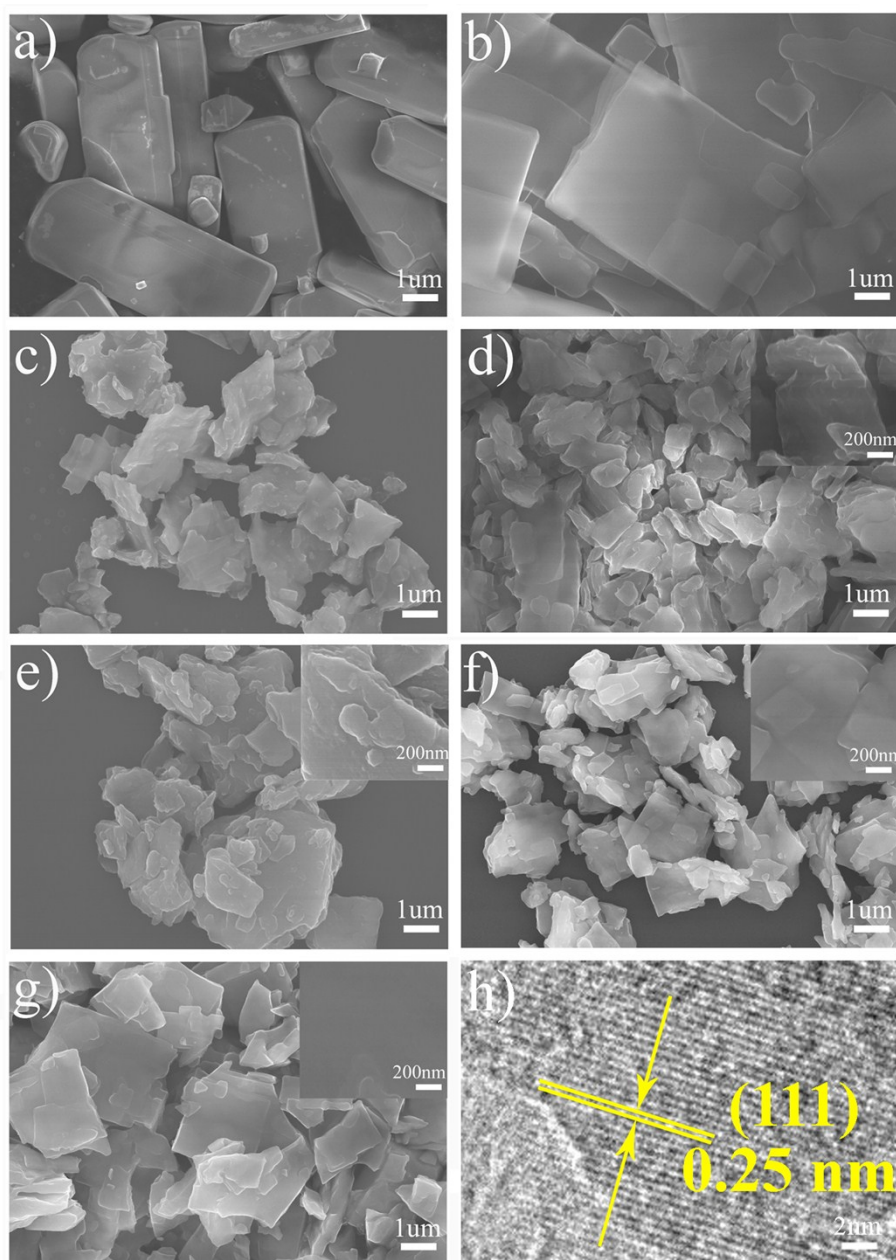
### **N-doped carbon encapsulated ultrathin MoO<sub>3</sub> nanosheets as superior anodes with high capacity and excellent rate capability for Li-ion batteries**

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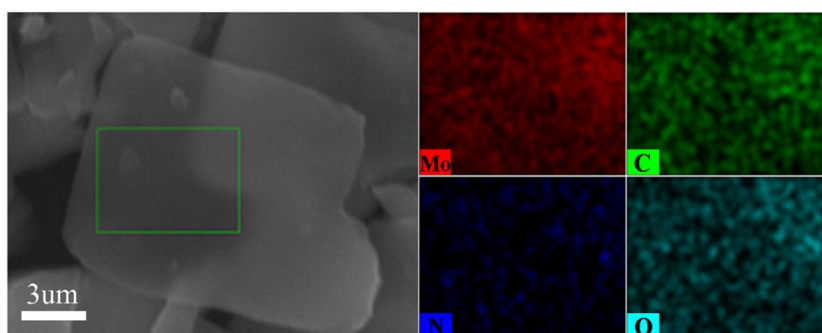
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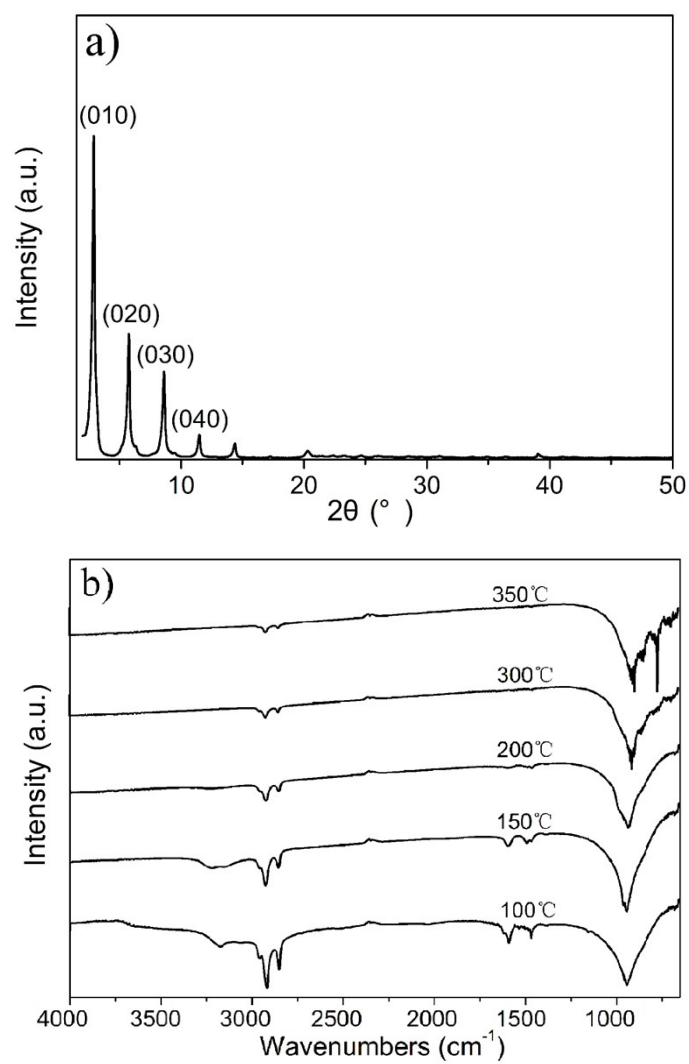
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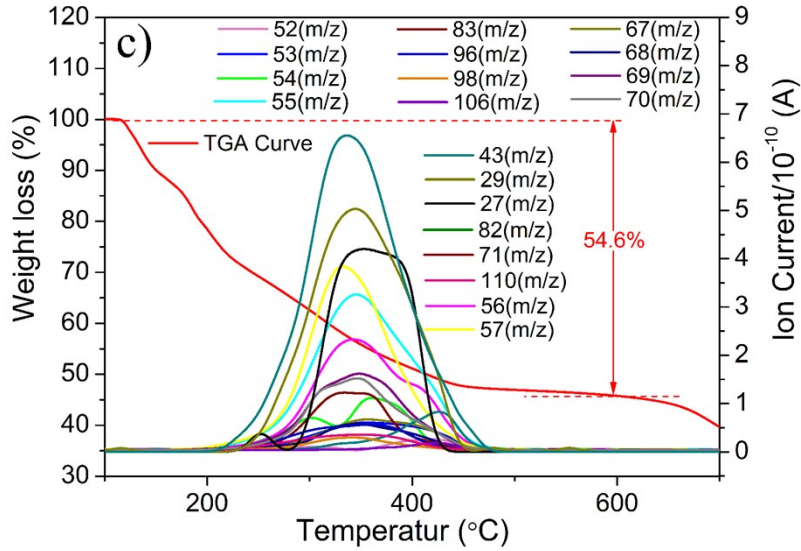


**Fig. S1** Typical SEM of (a)  $\alpha$ - $\text{MoO}_3$  and (b)  $\text{MoO}_3/\text{dodecylamine}$ . The SEM of (c) M-700, (d) M-200, (e) M-300, (f) M-400 and (g) M-500. (h) The HRTEM image of M-600.

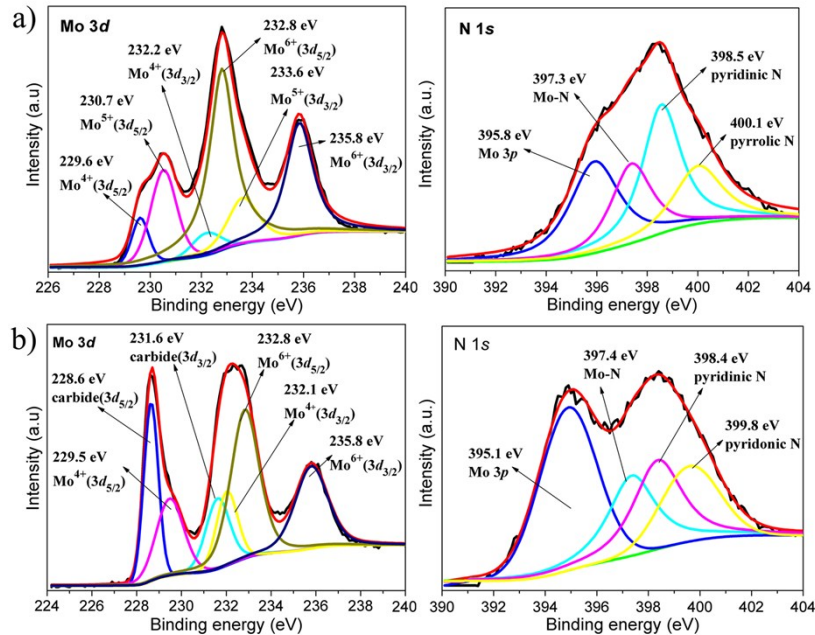


**Fig. S2** The local area magnification of elemental mapping from M-600 sample.

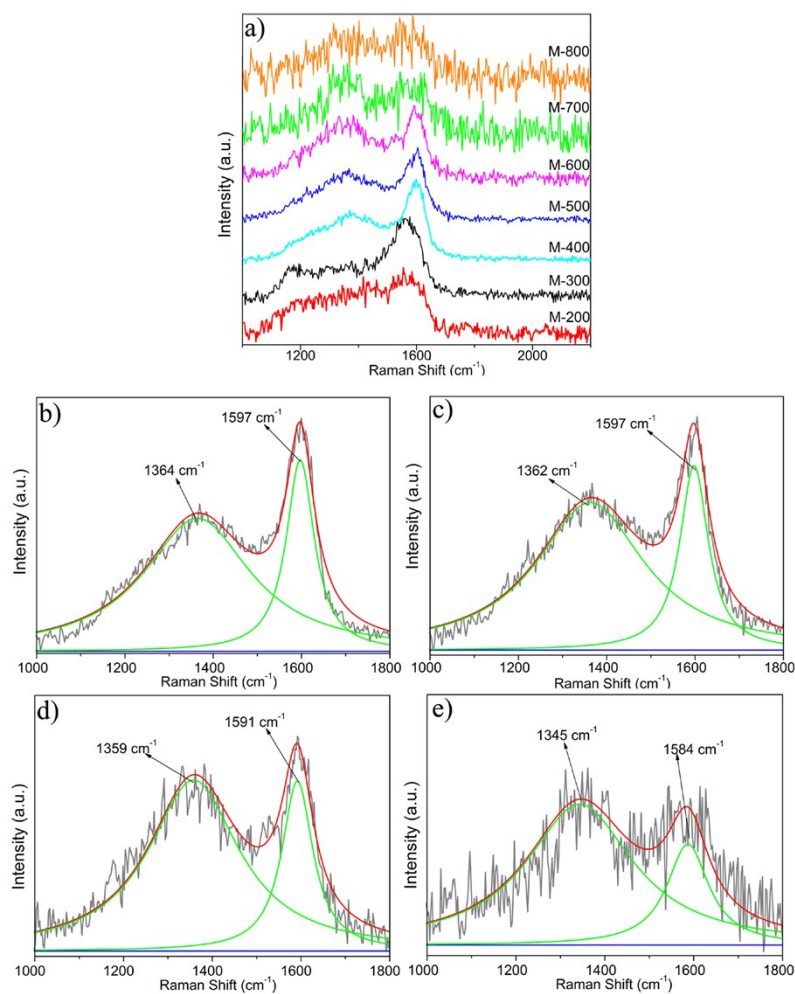




**Fig. S3** (a) XRD pattern of precursor MoO<sub>3</sub>/dodecylamine. (b) In-situ IR spectrum of composites at various annealing temperature (100~350 °C). (c) TG-MS combination analysis for the calcination of precursor MoO<sub>3</sub>/dodecylamine up to 700 °C in argon gas.



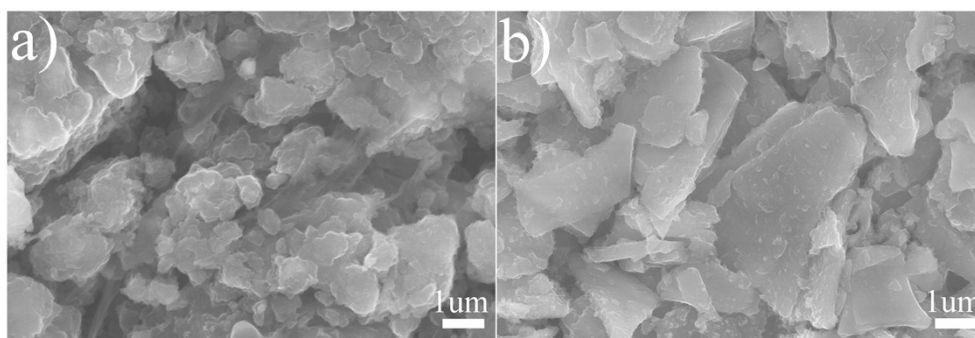
**Fig. S4** (a) XPS spectra of M-500 for Mo<sub>3d</sub> and N<sub>1s</sub>. (b) XPS spectra of M-700 for Mo<sub>3d</sub> and N<sub>1s</sub>.



**Fig. S5** (a) Raman pattern of different composites after heating treatment at various temperature with a excitation wavelength of 532 nm. The fitted D and G peaks of M-400 (b), M-500 (c), M-600 (d) and M-700 (e).

**Table S1** Characteristics of the fitted D and G band for various composites

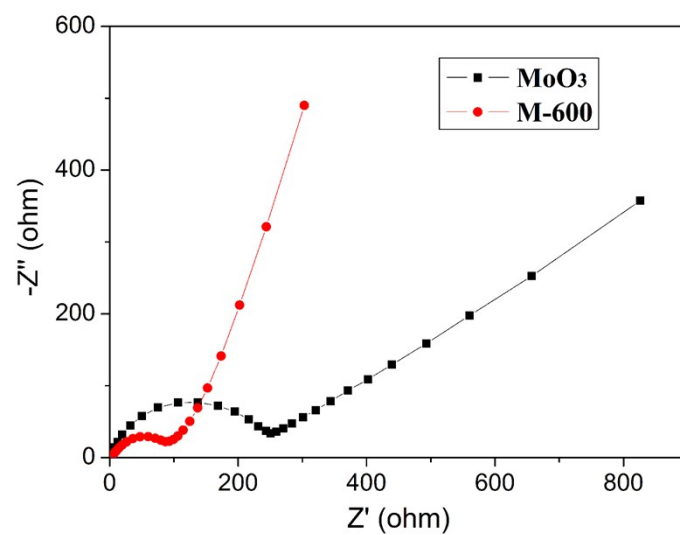
Samples	D peak position ( $\text{cm}^{-1}$ )	G peak position ( $\text{cm}^{-1}$ )	FWHM of G peak ( $\text{cm}^{-1}$ )	$I_D/I_G$
M-400	1364	1597	77.1	0.69
M-500	1362	1597	78.4	0.81
M-600	1359	1591	80.1	1.01
M-700	1345	1584	101.8	1.39



**Fig. S6** SEM images of (a)  $\text{MoO}_3$  electrode materials and (b) M-600 electrode materials after 60 cycles of charge-discharge at 1C.

**Table S2** Comparison for electrochemical properties of various MoO<sub>x</sub> materials

Materials	Current rate	Specific capacity (mAh g <sup>-1</sup> )	Cycle number	Ref
MoO <sub>2</sub> /graphene	0.2 C	1100	50	1
MoO <sub>3</sub> /MnO <sub>2</sub>	0.1 C	1127	50	2
	6 C	286	50	
MoO <sub>3</sub> nanosphere	0.1 C	1050	30	3
MoO <sub>3</sub> film	1 C	650	50	4
MoO <sub>2</sub> /C nanowires	1 A g <sup>-1</sup>	327	20	5
MoO <sub>3</sub> microsphere	1 C	780	100	6
MoO <sub>3</sub> /C microballs	2 A g <sup>-1</sup>	733	300	7
MoO <sub>3</sub> /C nanofiber	0.2 A g <sup>-1</sup>	500	100	8
MoO <sub>3-x</sub> nanowire arrays	0.05 A g <sup>-1</sup>	630	20	9
Core-shell MoO <sub>2</sub>	1 C	624	50	10
Mesoporous MoO <sub>2</sub>	0.05 C	750	30	11
MoO <sub>2</sub> /C nanosphere	3 C	410	60	11
MoO <sub>2</sub> /C nanobelts	0.1 A g <sup>-1</sup>	617	30	12
MoO <sub>2</sub> /MWCNT	0.1 A g <sup>-1</sup>	1143	200	13
MoO <sub>3</sub> @C	1 A g <sup>-1</sup>	502	100	14
MoO <sub>3</sub> @C	0.1 C	1064	50	15
MoO <sub>3</sub> @C	0.1 A g <sup>-1</sup>	500	100	16
MoO <sub>3</sub> @C	0.2 C	700	120	17
MoO <sub>3</sub> /NC	6 C (6.2 A g <sup>-1</sup> )	605	150	Our
nanosheets	0.3 C (0.46 A g <sup>-1</sup> )	1250	60	work



**Fig. S7** Nyquist plots of M-600 and MoO<sub>3</sub> over the frequency range from 100 kHz to 0.01 Hz at the discharged potential of 2.5 V after the 15th cycle.



## References

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