## Binary metal sulfide and polypyrrole on vertically-aligned CNTs array/carbon fiber paper as high-performance electrodes

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Figure S1: CNTs grown on CFP without coating of Al<sub>2</sub>O<sub>3</sub> buffer layer.



Figure S2: VA-CNTs array grown on CFP with 1 nm (a and b), 3nm (c and d) and 5 nm (e and f) of Al<sub>2</sub>O<sub>3</sub> buffer layer.



**Figure S3**: VA-CNTs array grown on CFP with various catalyst loading, from 0.1 (a and b), 0.5 (c and d) and 0.7 (e and f) mg mass per cm<sup>2</sup>.



**Figure S4**: The SEM images of CNTs grown on 3 nm Al<sub>2</sub>O<sub>3</sub> layer coated Ni foam with e-beam evaporation of 5 nm of Fe as catalyst.



Figure S5: CV curves of CFP (red) and VA-CNTs/CFP (black) in 1M KOH at a scan rate of 50 mV/s.



Figure S6: Energy-dispersive X-ray spectroscopy (EDS) elemental mapping of C, S, Ni and Co.



**Figure S7**: The SEM images of PPy/VA-CNTs/CFP (a and c) and VA-CNTs/CFP (b and d).



**Figure S8**. The specific areal capacitance values of PPy/VA-CNTs/CFP and Ni-Co-S/VA-CNTs/CFP with different mass loading of PPy (a) and Ni-Co-S (b) in mg per cm<sup>2</sup>.

## Table S1: Areal capacitance comparison of different electrode materials on various 3D substrates.

Electrode materials	Substrate	Areal capacitance
CoxNi <sub>1-x</sub> (OH) <sub>2</sub> /NiCo <sub>2</sub> S <sub>4</sub> [1]	Carbon fiber paper	2.86 F cm <sup>2</sup> at 4 mA/cm <sup>2</sup>
NiO nanoflower[2]	Carbon fiber paper	0.93 F/cm <sup>2</sup> at 0.1 mA/cm <sup>2</sup>
NiCo <sub>2</sub> O <sub>4</sub> nanowires[3]	Carbon fiber paper	~1.64 F/cm <sup>2</sup> at 2mA/cm <sup>2</sup> for electrode
(Co <sub>x</sub> Ni <sub>1-x</sub> ) <sub>9</sub> S <sub>8</sub> /carbon	Carbon fiber paper	1.32 F/cm <sup>2</sup> at 1mA/cm <sup>2</sup>
nanorod array[4]		
PPy [5]	Carbon fiber paper	0.42 F/cm <sup>2</sup> at 1 mA/cm <sup>2</sup>
PPy [6]	Carbon fiber paper	198.5 mF/cm <sup>2</sup> at 1 mA/cm <sup>2</sup>
Ni-Co-S/VA-CNTs (this	Carbon fiber paper	2.5 and 2.2 F/cm <sup>2</sup> at 2 and 5 mV/s
manuscript)		
PPy/VA-CNTs (this	Carbon fiber paper	3.3 and 2.7 F/cm2 at 2 and 5 mV/s
manuscript)		
NiCo <sub>2</sub> O <sub>4</sub> nanoneedles	Ni foam	3.12 and 1.44 F/cm <sup>2</sup> at 1.11 and 2.78 mA/cm <sup>2</sup>
array[7]		
CoO nano array @PPy[8]	Ni foam	6.0 and 2.51 F/cm <sup>2</sup> at 1 and 5 mA/cm <sup>2</sup> for
		electrode
Ni-Co sulfide nanowire[9]	Ni foam	$6.0 \text{ and } 2.9 \text{ F/cm}^2 \text{ at } 2.5 \text{ and } 30 \text{ mA/cm}^2 \text{ for}$
		electrode
MnO <sub>2</sub> -graphene[10]	Graphene/Ni foam	3.18 F/cm <sup>2</sup> (234.2 F/g)
Co <sub>3</sub> O <sub>4</sub> @MnO <sub>2</sub> core/shell	Stainless steel sheet	0.56F/cm <sup>2</sup> at 11.25 mA/cm <sup>2</sup>
nanowire array[11]	(2D substrate)	
TiO <sub>2</sub> @PPy[12]	Carbon cloth	64.6 mF/cm <sup>2</sup> at 10 mV/s
RGO/PPy foam[13]	RGO foam	175 mF/cm <sup>2</sup> at 10 mV/s

References:

[1] J.W. Xiao, L. Wan, S.H. Yang, F. Xiao, S. Wang, Nano Lett, 14 (2014) 831-838.

[2] S. Cheng, L. Yang, Y. Liu, W. Lin, L. Huang, D.C. Chen, C.P. Wong, M.L. Liu, J Mater Chem A, 1 (2013) 7709-7716.

[3] L. Huang, D.C. Chen, Y. Ding, S. Feng, Z.L. Wang, M.L. Liu, Nano Lett, 13 (2013) 3135-3139.

[4] L. Wan, J.W. Xiao, F. Xiao, S. Wang, Acs Appl Mater Inter, 6 (2014) 7735-7742.

[5] L.Y. Yuan, B. Yao, B. Hu, K.F. Huo, W. Chen, J. Zhou, Energ Environ Sci, 6 (2013) 470-476.

[6] C.Y. Yang, J.L. Shen, C.Y. Wang, H.J. Fei, H. Bao, G.C. Wang, J Mater Chem A, 2 (2014) 1458-1464.

[7] G.Q. Zhang, H.B. Wu, H.E. Hoster, M.B. Chan-Park, X.W. Lou, Energ Environ Sci, 5 (2012) 9453-9456.

[8] C. Zhou, Y.W. Zhang, Y.Y. Li, J.P. Liu, Nano Lett, 13 (2013) 2078-2085.

[9] Y.H. Li, L.J. Cao, L. Qiao, M. Zhou, Y. Yang, P. Xiao, Y.H. Zhang, J Mater Chem A, 2 (2014) 6540-6548.

[10] T. Zhai, F.X. Wang, M.H. Yu, S.L. Xie, C.L. Liang, C. Li, F.M. Xiao, R.H. Tang, Q.X. Wu, X.H. Lu, Y.X. Tong, Nanoscale, 5 (2013) 6790-6796.

[11] J.P. Liu, J. Jiang, C.W. Cheng, H.X. Li, J.X. Zhang, H. Gong, H.J. Fan, Adv Mater, 23 (2011) 2076-+.

[12] M.H. Yu, Y.X. Zeng, C. Zhang, X.H. Lu, C.H. Zeng, C.Z. Yao, Y.Y. Yang, Y.X. Tong, Nanoscale, 5 (2013) 10806-10810.

[13] J.T. Zhang, P. Chen, B.H.L. Oh, M.B. Chan-Park, Nanoscale, 5 (2013) 9860-9866.