

Electronic Supplementary Information (ESI)

**Highly cost-effective and sulfur/coking resistant VO<sub>x</sub>-grafted TiO<sub>2</sub> nanoparticles as an efficient anode catalyst for direct conversion of dry sour methane in solid oxide fuel cells**

Alfonso Garcia<sup>a</sup>, Ning Yan<sup>b, a\*</sup>, Adrien Vincent<sup>a</sup>, Anand Singh<sup>c</sup>, Josephine M. Hill<sup>c</sup>, Karl T. Chuang<sup>a</sup>, Jing-Li Luo<sup>a\*</sup>

<sup>a</sup> Department of Chemical and Materials Engineering, University of Alberta, Edmonton, Alberta, T6G 2V4, Canada. Fax: (+1) 780 492 2881; Tel: (+1) 780 492 2232. E-mail: [jingli.luo@ualberta.ca](mailto:jingli.luo@ualberta.ca)

<sup>b</sup> Van't Hoff Institute for Molecular Sciences (HIMS), University of Amsterdam, Amsterdam, 1098XH, The Netherlands. Tel : (+31) 020 525 6468. E-mail : [n.yan@uva.nl](mailto:n.yan@uva.nl)

<sup>c</sup> Department of Chemical and Petroleum Engineering, University of Calgary, Calgary, Alberta, T2N 1N4, Canada. Tel: (+1) 403 210 9488. E-mail: [jhill@ucalgary.ca](mailto:jhill@ucalgary.ca)

Table S1 Peak positions and atomic ratios of V and Ti from XPS

	<b>V 2p 3/2 (eV)</b>	<b>Ti 2p 3/2 (eV)</b>	<b>O 1s (eV)</b>	<b>N<sub>V</sub>/N<sub>Ti</sub> (atomic ratio)</b>
VO <sub>x</sub> /TiO <sub>2</sub> (25 °C)	517.27	458.2	529.8	0.16
VO <sub>x</sub> /TiO <sub>2</sub> (800 °C in air)	517.07	458.6	529.8	0.15
VO <sub>x</sub> /TiO <sub>2</sub> (800 °C in 0.5% H <sub>2</sub> S+ CH <sub>4</sub> )	517.13	458.2	529.8	0.21

Table S2 Comparison of SOFC non-ohmic polarization resistance using various anode materials in different methane gas fuels.

Anode Materials	Fuel Gas	Non-ohmic Polarization Resistance ( $\Omega \text{ cm}^2$ )*	Temperature ( $^{\circ}\text{C}$ )	Ref. No.
$\text{La}_{0.75}\text{Sr}_{0.25}\text{Cr}_{0.5}\text{Mn}_{0.5}\text{O}_3$	$\text{CH}_4$	0.87	900	[32]
$\text{La}_4\text{Sr}_8\text{Ti}_{11}\text{Mn}_{0.5}\text{Ga}_{0.5}\text{O}_{38}$	$\text{CH}_4$	0.57	900	[8]
$\text{Pr}_{0.5}\text{Ba}_{0.5}\text{MnO}_3$	$\text{CH}_4$	$\sim 0.4$	850	[16]
$\text{CeO}_2^{**}$	$\text{CH}_4$	$> 2$	900	[33]
$\text{Li}_{0.33}\text{La}_{0.56}\text{TiO}_3$	$0.1\%\text{H}_2\text{S} + \text{H}_2$	$> 0.35$	800	[17]
$\text{Ce}_{0.9}\text{Sr}_{0.1}\text{VO}_3$	$0.5\%\text{H}_2\text{S} + \text{CH}_4$	$> 17$	900	[34]
$\text{BaTiO}_3$	$0.5\%\text{H}_2\text{S} + \text{CH}_4$	$> 10$	900	[35]
LSBT	$0.5\%\text{H}_2\text{S} + \text{CH}_4$	11.3	850	[36]
$\text{VO}_x/\text{TiO}_2\text{-LSBT}$	$0.5\%\text{H}_2\text{S} + \text{CH}_4$	1.01	850	This work

Notes:

\*: Cathode material used in all cases was Sr-doped  $\text{LaMnO}_3$  except for work 16 and 17 which used NbSCF and BSCF, respectively.

\*\* :  $\text{CeO}_2$  was mixed with 50 wt% of  $\text{La}_{0.6}\text{Sr}_{0.4}\text{TiO}_3$ .