

Electronic Supplementary Information

Conformally Deposited NiO on a Hierarchical Carbon Support for High-Power and Durable Asymmetric Supercapacitor

Cao Guan¹, Yadong Wang², Yating Hu¹, Jilei Liu³, Kuan Hung Ho¹, Wei Zhao⁴, Zhanxi Fan⁴, Zexiang Shen³, Hua Zhang⁴, John Wang¹**

¹Department of Materials Science and Engineering, National University of Singapore, 117574 Singapore.

²School of Engineering, Nanyang Polytechnic, 569830 Singapore

³School of Physical and Mathematical Sciences, Nanyang Technological University, 637371 Singapore

⁴School of Materials Science and Engineering, Nanyang Technological University, 639798 Singapore

E-mail: msegc@nus.edu.sg; msewangj@nus.edu.sg

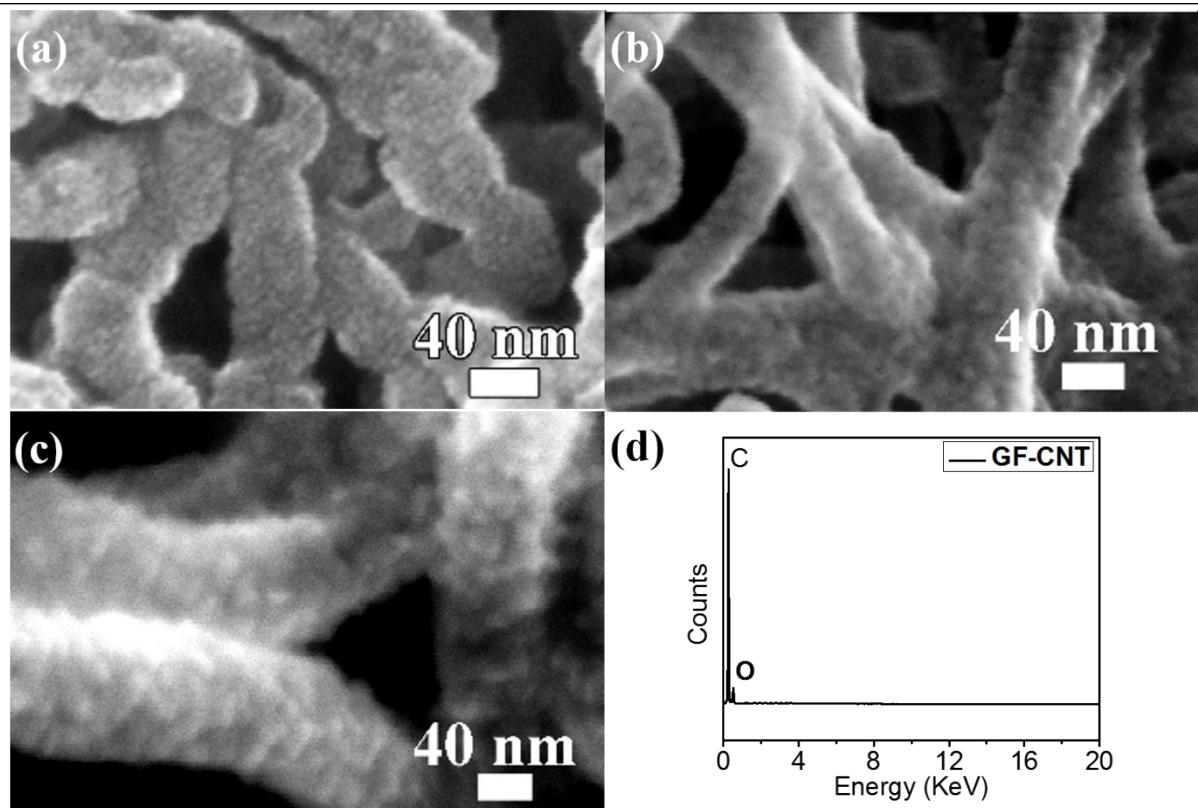


Figure S1. SEM images of ALD NiO on GF-CNT with (a) 300, (b) 600, and (c) 900 cycles, respectively. (d) EDS result of GF-CNT substrate.

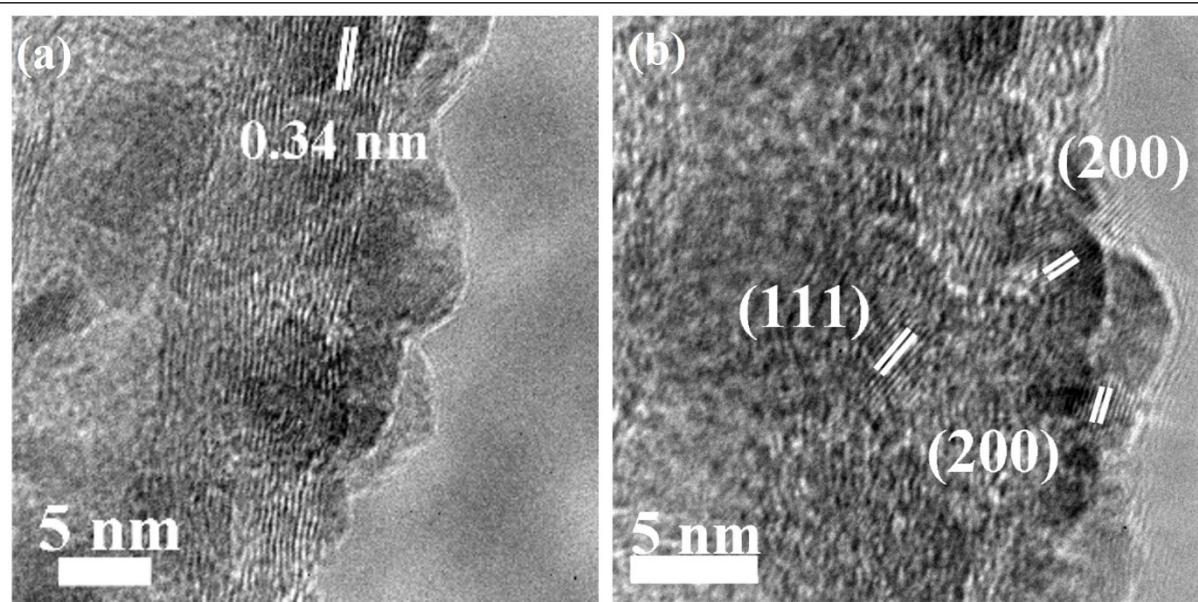
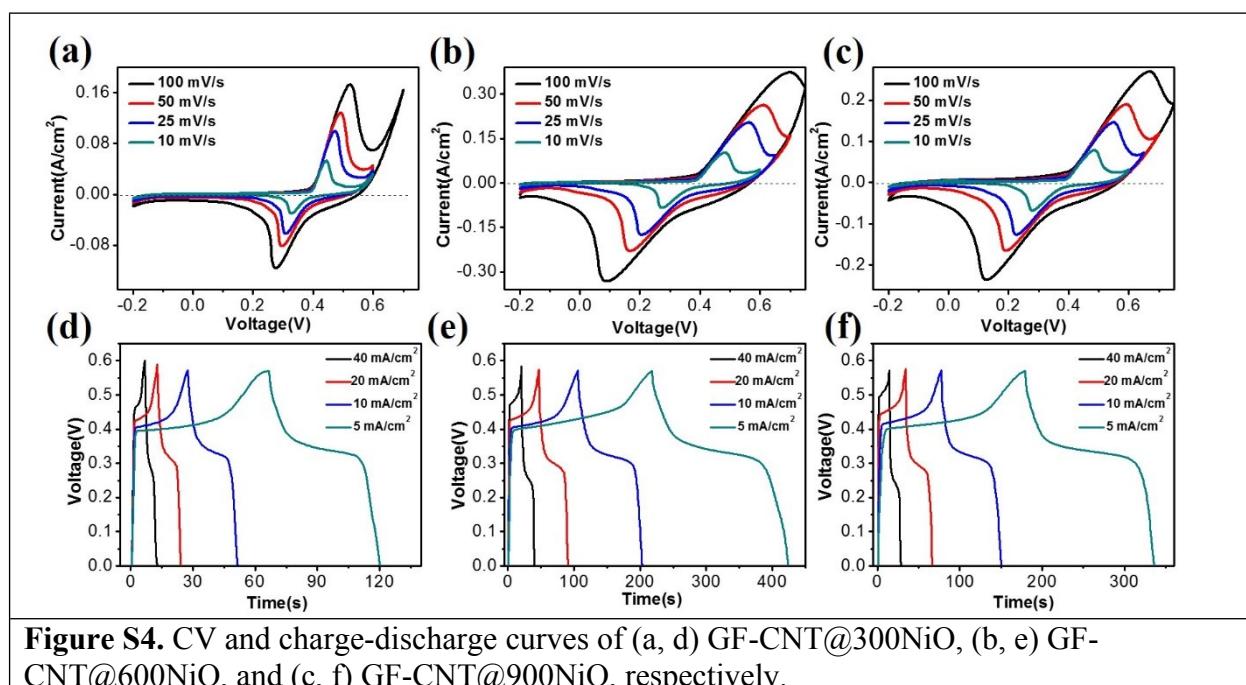
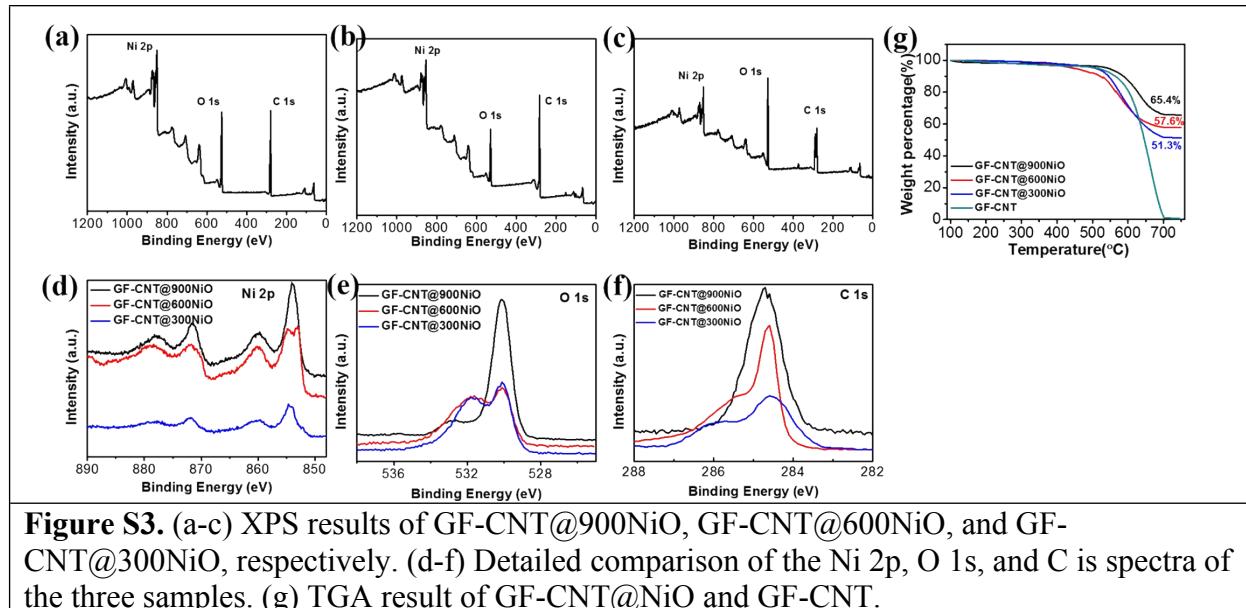


Figure S2. HRTEM images of GF-CNT@NiO.



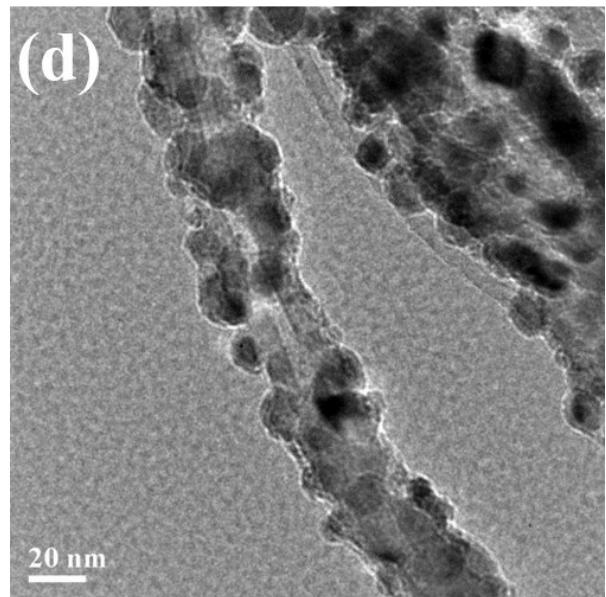
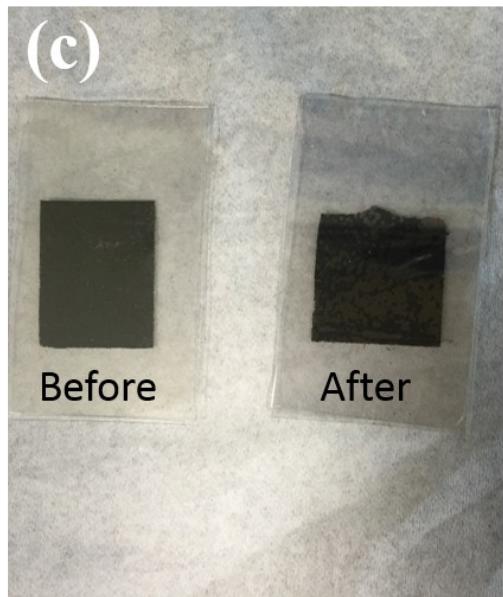
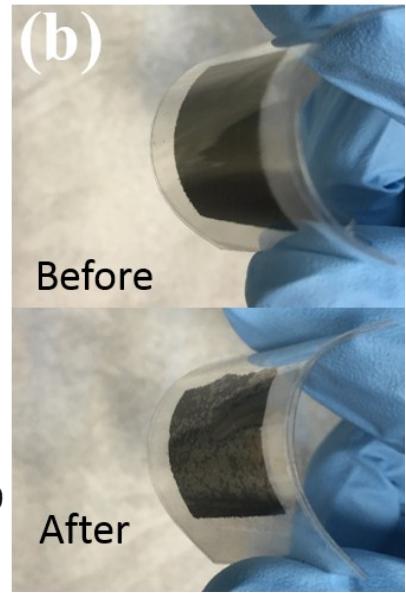
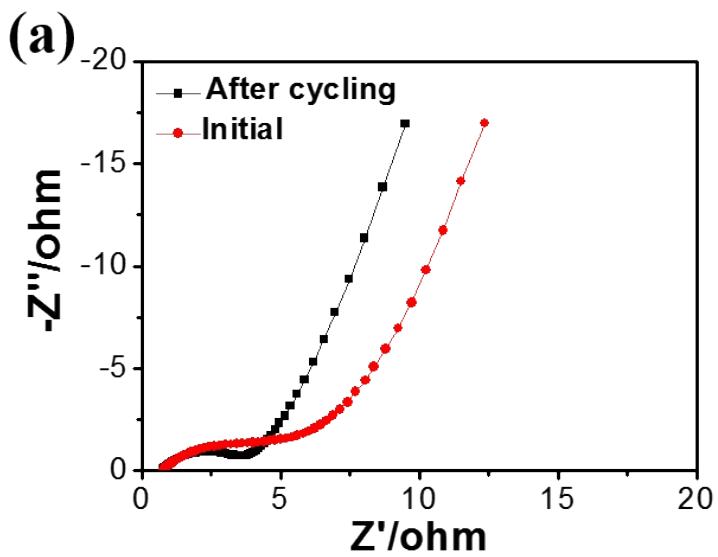


Figure S5. (a) EIS results of GF-CNT@NiO//G-CNT full-cell before and after the cycling test. (b-c) digit photos of GF-CNT@NiO electrodes before and after the cycling test. (e) TEM image of GF-CNT@NiO after the cycling test.

Table S1. NiO and Ni(OH)₂ based asymmetric supercapacitor

NiO or Ni(OH) ₂ electrode					Asymmetric supercapacitor				
Structure	Method	Loading mass per electrode	Capacitancy /Capacity	Cycling	Anode	Power density	Energy density	Cycling	Year
NiO ^a	Solvent thermal	5 mg/ cm ²			Active carbon	0.08 Wh/kg at 0.1 A/g	26.1 Wh/kg at 0.1 A/g	86% after 800 cycles	2006 ¹
NiO powder ^a	calcination	10 mg/ cm ²	310 F/g at 6 mA/cm ²		Ru _{0.36} V _{0.64} O ₂	1.42 kW/kg at 23.0 Wh/kg	41.2 Wh/kg	83.5% after 1500 cycles	2007 ²
NiO ^a	Solvent thermal	8 mg/ cm ²	2080 F/g at 5 mA/cm ²		Active carbon	1.1 kW/kg at 26.9 Wh/kg	42.3 Wh/kg at 0.11 kW/kg	82% after 1000 cycles	2010 ³
Ni(OH) ₂ /G ^a	Microwave heating	3 mg/cm ²	1735 F/g at 1 mV/s		porous graphene	15.2 kW/kg at 13.5 Wh/kg	77.8 Wh/kg at 0.175 kW/kg	94.3 % after 3000 cycles	2012 ⁴
Ni(OH) ₂ /C NT/Nickel foam ^b	chemical bath deposition		3300 F/g at 1 mV/s		Active carbon	1.8 kW/kg at 32.5 Wh/kg	50.6 Wh/kg at 0.095 kW/kg	83 % after 3000 cycles	2012 ⁵
Ni(OH) ₂ /ultrathin-graphite foam	hydrothermal	0.2 mg/cm ²	166 F/g at 0.5 A/g	65% at 1000 cycles	Activated microwave exfoliated graphite oxide	85 kW/kg at 6.5 Wh/kg	13.4 Wh/kg at 65 kW/kg	63.2% after 10 000 cycles	2013 ⁶
NiO nanoflake/carbon cloth ^b	solvothermal	2.7 mg/cm ²	392 F/g at 0.5 mA/cm ²		rGO sheets on Ni foam	~3 kW/kg at ~14 Wh/kg	39.9 Wh/kg at ~0.2 kW/kg	95 % after 3000 cycles	2013 ⁷
Ni(OH) ₂ powder ^a	chemical coprecipitation	12 mg/ cm ²	680 F/g at 1 mV/s		Porous VN	2.4 kW/kg at 26 Wh/kg	50 Wh/kg at 0.365 kW/kg	85% after 1000 cycles	2013 ⁸
amorphous Ni(OH) ₂ nanospheres	Electrochemical method	0.12 mg/ cm ²	2188 F/g at 1 mV/s	76% after 10 000 cycles	Active carbon	1.67 kW/kg at 12.6 Wh/kg	35.7 Wh/kg at 0.49 kW/kg	81% after 10 000 cycles	2013 ⁹
NiO-3D graphene foam	Pulsed laser deposition		1225 F/g at 2 A/g	89% at 1000 cycles	nitrogen-doped carbon nanotubes	42 kW/kg at 17 Wh/kg	32Wh/kg at 0.7 kW/kg	94% after 2000 cycles	2014 ¹⁰
GF-CNT@Ni O	Atomic layer deposition	1.53 mg/ cm ²	196.5 mAh/g at 5 mA/cm ²	83.5% after 30 000 cycles	Graphene-CNT paper	7.14 kW/kg at 11.9 Wh/kg	23.4 Wh/kg at 1.06 kW/kg	81.7% after 30 000 cycles	Current work

a. Powder form with metal foam/ as substrate

b. Metal/carbon substrate(such as Nickel foam, carbon cloth and carbon plate) is used

1. C. Yuan, X. Zhang, Q. Wu and B. Gao, *Solid State Ionics*, 2006, **177**, 1237-1242.
2. C.-Z. Yuan, B. Gao and X.-G. Zhang, *J. Power Sources*, 2007, **173**, 606-612.

3. J.-W. Lang, L.-B. Kong, M. Liu, Y.-C. Luo and L. Kang, *J. Solid State Electrochem.*, 2010, **14**, 1533-1539.
4. J. Yan, Z. Fan, W. Sun, G. Ning, T. Wei, Q. Zhang, R. Zhang, L. Zhi and F. Wei, *Adv. Funct. Mater.*, 2012, **22**, 2632-2641.
5. Z. Tang, C.-h. Tang and H. Gong, *Adv. Funct. Mater.*, 2012, **22**, 1272-1278.
6. J. Ji, L. L. Zhang, H. Ji, Y. Li, X. Zhao, X. Bai, X. Fan, F. Zhang and R. S. Ruoff, *ACS Nano*, 2013, **7**, 6237-6243.
7. F. Luan, G. Wang, Y. Ling, X. Lu, H. Wang, Y. Tong, X.-X. Liu and Y. Li, *Nanoscale*, 2013, **5**, 7984-7990.
8. Z.-H. Gao, H. Zhang, G.-P. Cao, M.-F. Han and Y.-S. Yang, *Electrochim. Acta*, 2013, **87**, 375-380.
9. H. B. Li, M. H. Yu, F. X. Wang, P. Liu, Y. Liang, J. Xiao, C. X. Wang, Y. X. Tong and G. W. Yang, *Nat. Commun.*, 2013, **4**, 1894.
10. H. Wang, H. Yi, X. Chen and X. Wang, *J. Mater. Chem. A*, 2014, **2**, 3223-3230.