

Electronic Supplementary Information

Energetic compounds consisting of 1,2,5- and 1,3,4-oxadiazole rings

Yongxing Tang,^a Chunlin He,^a Lauren A. Mitchell,^b Damon A. Parrish^b and Jean'ne M. Shreeve^{*a}

[a] Department of Chemistry, University of Idaho, Moscow, Idaho 83844-2343, United States

[b] Naval Research Laboratory, 4555 Overlook Avenue, Washington, D.C. 20375, United States

Table of Contents	S2
Table S1 Crystal data and structure refinement for 1 , 2 ·H ₂ O and 5 ·H ₂ O	S3
Table S2 Atomic coordinates and equivalent isotropic displacement parameters for 1	S4
Table S3 Bond lengths and angles for 1 .	S5
Table S4 Anisotropic displacement parameters for 1 .	S6
Table S5 Hydrogen coordinates and isotropic displacement parameters for 1	S7
Table S6 Torsion angles for 1	S8
Table S7 Hydrogen bonds for 1	S9
Table S8 Atomic coordinates and equivalent isotropic displacement parameters for 2 ·H ₂ O	S10
Table S9 Bond lengths and angles for 2 ·H ₂ O.	S11
Table S10 Anisotropic displacement parameters for 2 ·H ₂ O	S12
Table S11 Hydrogen coordinates and isotropic displacement parameters for 2 ·H ₂ O	S13
Table S12 Torsion angles for 2 ·H ₂ O	S14
Table S13 Hydrogen bonds for 2 ·H ₂ O	S15
Table S14 Atomic coordinates and equivalent isotropic displacement parameters for 5 ·H ₂ O	S16
Table S15 Bond lengths and angles for 5 ·H ₂ O	S17
Table S16 Anisotropic displacement parameters for 5 ·H ₂ O	S18
Table S17 Hydrogen coordinates and isotropic displacement parameters for 5 ·H ₂ O	S18
Table S18 Torsion angles for 5 ·H ₂ O	S19
Table S19 Hydrogen bonds for 5 ·H ₂ O	S20
Computational Details (Isodesmic Reactions)	S21

Table S1. Crystal data and structure refinement for **1**, **2**·H₂O and **5**·H₂O.

Identification code	1	2 ·H ₂ O	5 ·H ₂ O
CCDC number	1419914	1419915	1419916
Empirical formula	C ₄ H ₄ N ₆ O ₂	C ₄ H ₅ N ₇ O ₅	C ₄ H ₈ N ₈ O ₆
Formula weight	168.13	231.15	264.18
Temperature/K	293(2) K	296(2) K	293(2) K
Crystal system	Monoclinic	Monoclinic	Monoclinic
Space group	P2 ₁ /c	P2 ₁ /n	C2/c
a/Å	7.0958(2)	6.5964(3)	26.968(2)
b/Å	8.6167(2)	7.4775(3)	4.0185(3)
c/Å	11.0839(3)	17.4904(7)	18.7676(14)
α/°	90	90	90
β/°	102.3650(10)	97.043(2)	99.579(5)
γ/°	90	90	90
Volume/Å ³	661.98(3)	856.20(6)	2005.5(3)
Z	4	4	8
ρ _{calc} /g/cm ³	1.687	1.793	1.750
μ/mm ⁻¹	0.140	0.163	0.161
F(000)	344	472	1088
Crystal size/mm ³	0.222 x 0.200 x 0.120	0.112 x 0.103 x 0.056	0.200 x 0.180 x 0.150
2θ range for data collection/°	3.764 to 26.440	3.188 to 26.426	1.532 to 26.480
Index ranges	-8<=h<=8, -10<=k<=10, -13<=l<=13	-8<=h<=8, -7<=k<=9, -21<=l<=21	-33<=h<=33, - 5<=k<=5, - 18<=l<=23
Reflections collected	6119	7971	9159
Independent reflections	1355 [R _(int) = 0.0144]	1763 [R _(int) = 0.0387]	2031 [R _{int} = 0.0604]
Data/restraints/parameters	1355 / 0 / 121	1763 / 0 / 160	2031 / 5 / 181
Goodness-of-fit on F ²	1.065	1.008	1.081
Final R indexes [I>=2σ (I)]	R ₁ = 0.0305, wR ₂ = 0.0858	R ₁ = 0.0363, wR ₂ = 0.0810	R ₁ = 0.0455, wR ₂ = 0.1107
Final R indexes [all data]	R ₁ = 0.0335, wR ₂ = 0.0887	R ₁ = 0.0691, wR ₂ = 0.0943	R ₁ = 0.0969, wR ₂ = 0.1598
Largest diff. peak/hole / e Å ⁻³	0.194 and -0.225	0.221 and -0.159	0.240 and -0.259

Table S2. Atomic coordinates ($\times 10^4$) and equivalent isotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for **1**. $U(\text{eq})$ is defined as one third of the trace of the orthogonalized U^{ij} tensor.

	x	y	z	U(eq)
N(1)	4865(2)	1091(1)	-1622(1)	45(1)
C(2)	5819(2)	2010(1)	-745(1)	32(1)
N(3)	6509(1)	1690(1)	420(1)	36(1)
N(4)	7392(1)	3055(1)	942(1)	35(1)
C(5)	7174(1)	4065(1)	71(1)	29(1)
O(6)	6177(1)	3506(1)	-1038(1)	31(1)
C(7)	7876(1)	5642(1)	179(1)	29(1)
N(8)	7520(2)	6652(1)	-717(1)	38(1)
O(9)	8469(1)	7980(1)	-254(1)	44(1)
N(10)	9451(2)	7769(1)	972(1)	40(1)
C(11)	9082(2)	6336(1)	1249(1)	32(1)
N(12)	9788(2)	5661(2)	2337(1)	53(1)

Table S3. Bond lengths [Å] and angles [°] for **1**.

N(1)-C(2)	1.3224(15)
N(1)-H(1A)	0.866(18)
N(1)-H(1B)	0.854(18)
C(2)-N(3)	1.3096(15)
C(2)-O(6)	1.3659(13)
N(3)-N(4)	1.3986(13)
N(4)-C(5)	1.2842(14)
C(5)-O(6)	1.3680(12)
C(5)-C(7)	1.4435(15)
C(7)-N(8)	1.3029(15)
C(7)-C(11)	1.4365(14)
N(8)-O(9)	1.3704(13)
O(9)-N(10)	1.3993(14)
N(10)-C(11)	1.3122(15)
C(11)-N(12)	1.3362(16)
N(12)-H(12A)	0.855(19)
N(12)-H(12B)	0.826(19)
C(2)-N(1)-H(1A)	117.0(11)
C(2)-N(1)-H(1B)	119.4(11)
H(1A)-N(1)-H(1B)	120.8(15)
N(3)-C(2)-N(1)	128.55(11)
N(3)-C(2)-O(6)	112.57(10)
N(1)-C(2)-O(6)	118.88(10)
C(2)-N(3)-N(4)	105.69(9)
C(5)-N(4)-N(3)	106.73(9)
N(4)-C(5)-O(6)	113.13(9)
N(4)-C(5)-C(7)	126.30(10)
O(6)-C(5)-C(7)	120.58(9)
C(2)-O(6)-C(5)	101.87(8)
N(8)-C(7)-C(11)	109.58(10)
N(8)-C(7)-C(5)	123.95(10)
C(11)-C(7)-C(5)	126.45(10)
C(7)-N(8)-O(9)	105.83(9)
N(8)-O(9)-N(10)	110.96(8)
C(11)-N(10)-O(9)	105.47(9)
N(10)-C(11)-N(12)	124.49(11)
N(10)-C(11)-C(7)	108.15(10)
N(12)-C(11)-C(7)	127.32(11)
C(11)-N(12)-H(12A)	118.2(12)
C(11)-N(12)-H(12B)	118.4(13)
H(12A)-N(12)-H(12B)	121.6(18)

Table S4. Anisotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for **1**. The anisotropic displacement factor exponent takes the form: $-2\pi^2 [h^2 a^{*2} U^{11} + \dots + 2 h k a^* b^* U^{12}]$

	U ¹¹	U ²²	U ³³	U ²³	U ¹³	U ¹²
N(1)	61(1)	38(1)	31(1)	0(1)	-6(1)	-16(1)
C(2)	33(1)	29(1)	31(1)	0(1)	2(1)	-2(1)
N(3)	42(1)	31(1)	31(1)	0(1)	-2(1)	-5(1)
N(4)	39(1)	32(1)	29(1)	-1(1)	-2(1)	-4(1)
C(5)	28(1)	31(1)	26(1)	-4(1)	1(1)	0(1)
O(6)	35(1)	30(1)	26(1)	-1(1)	-2(1)	-3(1)
C(7)	28(1)	29(1)	28(1)	-3(1)	4(1)	2(1)
N(8)	43(1)	32(1)	36(1)	0(1)	0(1)	-1(1)
O(9)	55(1)	31(1)	42(1)	2(1)	2(1)	-4(1)
N(10)	47(1)	33(1)	37(1)	-4(1)	3(1)	-4(1)
C(11)	36(1)	29(1)	31(1)	-6(1)	6(1)	0(1)
N(12)	78(1)	42(1)	29(1)	-2(1)	-8(1)	-14(1)

Table S5. Hydrogen coordinates ($\times 10^4$) and isotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for **1**.

	x	y	z	U(eq)
H(1A)	4470(20)	210(20)	-1390(16)	55
H(1B)	4370(20)	1471(19)	-2330(16)	55
H(12A)	9330(20)	4780(20)	2481(16)	63
H(12B)	10420(30)	6190(20)	2898(17)	63

Table S6. Torsion angles [°] for **1**.

N(1)-C(2)-N(3)-N(4)	179.75(12)
O(6)-C(2)-N(3)-N(4)	-0.20(12)
C(2)-N(3)-N(4)-C(5)	-0.13(12)
N(3)-N(4)-C(5)-O(6)	0.42(12)
N(3)-N(4)-C(5)-C(7)	-179.13(10)
N(3)-C(2)-O(6)-C(5)	0.42(12)
N(1)-C(2)-O(6)-C(5)	-179.53(11)
N(4)-C(5)-O(6)-C(2)	-0.52(12)
C(7)-C(5)-O(6)-C(2)	179.07(9)
N(4)-C(5)-C(7)-N(8)	-175.80(10)
O(6)-C(5)-C(7)-N(8)	4.68(16)
N(4)-C(5)-C(7)-C(11)	5.48(18)
O(6)-C(5)-C(7)-C(11)	-174.04(9)
C(11)-C(7)-N(8)-O(9)	0.17(12)
C(5)-C(7)-N(8)-O(9)	-178.74(10)
C(7)-N(8)-O(9)-N(10)	0.28(12)
N(8)-O(9)-N(10)-C(11)	-0.64(12)
O(9)-N(10)-C(11)-N(12)	178.61(12)
O(9)-N(10)-C(11)-C(7)	0.71(12)
N(8)-C(7)-C(11)-N(10)	-0.59(13)
C(5)-C(7)-C(11)-N(10)	178.29(10)
N(8)-C(7)-C(11)-N(12)	-178.41(12)
C(5)-C(7)-C(11)-N(12)	0.47(19)

Table S7. Hydrogen bonds for **1** [\AA and $^\circ$].

D-H...A	d(D-H)	d(H...A)	d(D...A)	$\angle(\text{DHA})$
N(1)-H(1A)...N(3)#1	0.866(18)	2.154(18)	3.0032(15)	166.8(15)
N(1)-H(1B)...N(8)#2	0.854(18)	2.295(18)	3.1081(15)	159.4(15)
N(12)-H(12A)...N(4)	0.855(19)	2.450(18)	3.0327(16)	126.0(15)
N(12)-H(12A)...N(10)#3	0.855(19)	2.459(18)	3.0949(16)	131.8(15)
N(12)-H(12B)...N(3)#4	0.826(19)	2.579(19)	3.3270(15)	151.3(16)
N(12)-H(12B)...N(4)#4	0.826(19)	2.402(19)	3.2035(15)	164.0(16)

Symmetry transformations used to generate equivalent atoms:

#1 $-x+1, -y, -z$ #2 $-x+1, y-1/2, -z-1/2$ #3 $-x+2, y-1/2, -z+1/2$

#4 $-x+2, y+1/2, -z+1/2$

Table S8. Atomic coordinates ($\times 10^4$) and equivalent isotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for $2 \cdot \text{H}_2\text{O}$. $U(\text{eq})$ is defined as one third of the trace of the orthogonalized U^{ij} tensor.

	x	y	z	U(eq)
N(1)	13839(3)	5064(3)	7108(1)	43(1)
C(2)	12237(3)	4610(2)	6647(1)	32(1)
N(3)	11693(2)	4962(2)	5913(1)	36(1)
N(4)	9826(2)	4176(2)	5666(1)	37(1)
C(5)	9354(3)	3393(3)	6262(1)	32(1)
O(6)	10788(2)	3582(2)	6897(1)	35(1)
C(7)	7504(3)	2401(3)	6338(1)	33(1)
N(8)	7241(2)	1647(3)	6987(1)	45(1)
O(9)	5354(2)	877(2)	6861(1)	52(1)
N(10)	4431(2)	1177(2)	6104(1)	43(1)
C(11)	5761(3)	2125(3)	5774(1)	33(1)
N(12)	5683(2)	2825(2)	5045(1)	34(1)
N(13)	3970(2)	2498(2)	4576(1)	39(1)
O(14)	2548(2)	1587(2)	4772(1)	51(1)
O(15)	3855(2)	3169(2)	3919(1)	56(1)
O(1S)	9082(3)	976(3)	3640(1)	58(1)

Table S9. Bond lengths [Å] and angles [°] for 2·H₂O.

N(1)-C(2)	1.294(2)	N(1)-H(1A)	0.89(2)
N(1)-H(1B)	0.86(2)	C(2)-N(3)	1.317(2)
C(2)-O(6)	1.341(2)	N(3)-N(4)	1.385(2)
N(3)-H(3)	0.84(2)	N(4)-C(5)	1.267(2)
C(5)-O(6)	1.375(2)	C(5)-C(7)	1.448(3)
C(7)-N(8)	1.298(2)	C(7)-C(11)	1.434(2)
N(8)-O(9)	1.364(2)	O(9)-N(10)	1.406(2)
N(10)-C(11)	1.315(2)	C(11)-N(12)	1.374(2)
N(12)-N(13)	1.334(2)	N(13)-O(14)	1.241(2)
N(13)-O(15)	1.248(2)	O(1S)-H(1S)	0.84(3)
O(1S)-H(2S)	0.77(3)		
<hr/>			
C(2)-N(1)-H(1A)	116.5(14)	C(2)-N(1)-H(1B)	114.1(14)
H(1A)-N(1)-H(1B)	129(2)	N(1)-C(2)-N(3)	131.30(19)
N(1)-C(2)-O(6)	120.48(16)	N(3)-C(2)-O(6)	108.22(15)
C(2)-N(3)-N(4)	110.15(16)	C(2)-N(3)-H(3)	129.2(14)
N(4)-N(3)-H(3)	120.6(14)	C(5)-N(4)-N(3)	103.83(14)
N(4)-C(5)-O(6)	113.73(16)	N(4)-C(5)-C(7)	127.34(16)
O(6)-C(5)-C(7)	118.91(15)	C(2)-O(6)-C(5)	104.06(13)
N(8)-C(7)-C(11)	110.87(16)	N(8)-C(7)-C(5)	120.44(16)
C(11)-C(7)-C(5)	128.67(16)	C(7)-N(8)-O(9)	105.03(15)
N(8)-O(9)-N(10)	111.47(13)	C(11)-N(10)-O(9)	105.26(15)
N(10)-C(11)-N(12)	131.40(17)	N(10)-C(11)-C(7)	107.37(16)
N(12)-C(11)-C(7)	121.22(16)	N(13)-N(12)-C(11)	115.63(15)
O(14)-N(13)-O(15)	120.82(16)	O(14)-N(13)-N(12)	122.88(16)
O(15)-N(13)-N(12)	116.30(16)	H(1S)-O(1S)-H(2S)	108(3)

Table S10. Anisotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for $2 \cdot \text{H}_2\text{O}$. The anisotropic displacement factor exponent takes the form: $-2\pi^2 [h^2 a^2 U^{11} + \dots + 2 h k a^* b^* U^{12}]$

	U11	U22	U33	U23	U13	U12
N(1)	42(1)	52(1)	33(1)	0(1)	-3(1)	-12(1)
C(2)	32(1)	32(1)	30(1)	-4(1)	0(1)	1(1)
N(3)	33(1)	43(1)	30(1)	4(1)	-2(1)	-7(1)
N(4)	32(1)	46(1)	33(1)	5(1)	-2(1)	-5(1)
C(5)	30(1)	39(1)	26(1)	0(1)	-2(1)	5(1)
O(6)	32(1)	44(1)	27(1)	1(1)	-1(1)	-3(1)
C(7)	31(1)	37(1)	32(1)	3(1)	3(1)	3(1)
N(8)	36(1)	60(1)	41(1)	12(1)	4(1)	-4(1)
O(9)	42(1)	70(1)	43(1)	19(1)	4(1)	-10(1)
N(10)	37(1)	50(1)	42(1)	8(1)	2(1)	-6(1)
C(11)	30(1)	32(1)	36(1)	3(1)	6(1)	1(1)
N(12)	27(1)	41(1)	33(1)	0(1)	-4(1)	-4(1)
N(13)	33(1)	47(1)	37(1)	-6(1)	2(1)	-2(1)
O(14)	34(1)	68(1)	49(1)	-3(1)	0(1)	-17(1)
O(15)	48(1)	83(1)	33(1)	7(1)	-6(1)	-7(1)
O(1S)	43(1)	86(1)	45(1)	-13(1)	-2(1)	-15(1)

Table S11. Hydrogen coordinates ($\times 10^4$) and isotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for $2 \cdot \text{H}_2\text{O}$.

	x	y	z	U(eq)
H(1A)	13900(30)	4690(30)	7592(13)	51
H(1B)	14680(30)	5740(30)	6900(13)	51
H(3)	12300(30)	5580(30)	5607(12)	43
H(1S)	8300(50)	400(40)	3889(17)	88
H(2S)	10050(50)	1200(40)	3914(17)	88

Table S12. Torsion angles [°] for **2**·H₂O.

N(1)-C(2)-N(3)-N(4)	180.0(2)
O(6)-C(2)-N(3)-N(4)	0.9(2)
C(2)-N(3)-N(4)-C(5)	-0.2(2)
N(3)-N(4)-C(5)-O(6)	-0.5(2)
N(3)-N(4)-C(5)-C(7)	177.87(18)
N(1)-C(2)-O(6)-C(5)	179.68(18)
N(3)-C(2)-O(6)-C(5)	-1.14(19)
N(4)-C(5)-O(6)-C(2)	1.1(2)
C(7)-C(5)-O(6)-C(2)	-177.47(17)
N(4)-C(5)-C(7)-N(8)	179.2(2)
O(6)-C(5)-C(7)-N(8)	-2.5(3)
N(4)-C(5)-C(7)-C(11)	-2.4(3)
O(6)-C(5)-C(7)-C(11)	175.87(18)
C(11)-C(7)-N(8)-O(9)	0.1(2)
C(5)-C(7)-N(8)-O(9)	178.69(17)
C(7)-N(8)-O(9)-N(10)	-0.1(2)
N(8)-O(9)-N(10)-C(11)	0.0(2)
O(9)-N(10)-C(11)-N(12)	-178.97(19)
O(9)-N(10)-C(11)-C(7)	0.0(2)
N(8)-C(7)-C(11)-N(10)	-0.1(2)
C(5)-C(7)-C(11)-N(10)	-178.5(2)
N(8)-C(7)-C(11)-N(12)	179.05(17)
C(5)-C(7)-C(11)-N(12)	0.6(3)
N(10)-C(11)-N(12)-N(13)	-0.3(3)
C(7)-C(11)-N(12)-N(13)	-179.13(17)
C(11)-N(12)-N(13)-O(14)	-1.6(3)
C(11)-N(12)-N(13)-O(15)	178.02(16)

Table S13. Hydrogen bonds for **2**·H₂O [\AA and $^\circ$].

D-H...A	d(D-H)	d(H...A)	d(D...A)	$\angle(\text{DHA})$
N(1)-H(1A)...O(1S)#1	0.89(2)	1.89(2)	2.774(2)	177(2)
N(1)-H(1B)...O(15)#2	0.86(2)	2.00(2)	2.821(3)	159(2)
N(3)-H(3)...N(12)#2	0.84(2)	2.21(2)	3.041(2)	170.7(19)
O(1S)-H(1S)...N(10)#3	0.84(3)	2.15(3)	2.901(2)	149(3)
O(1S)-H(2S)...O(14)#4	0.77(3)	2.11(3)	2.872(2)	172(3)

Symmetry transformations used to generate equivalent atoms:

#1 $x+1/2, -y+1/2, z+1/2$ #2 $-x+2, -y+1, -z+1$ #3 $-x+1, -y, -z+1$

#4 $x+1, y, z$

Table S14. Atomic coordinates ($\times 10^4$) and equivalent isotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for $5 \cdot \text{H}_2\text{O}$. $U(\text{eq})$ is defined as one third of the trace of the orthogonalized U^{ij} tensor.

	x	y	z	U(eq)
N(1)	3775(1)	723(7)	4498(1)	32(1)
N(2)	3853(1)	725(7)	5252(1)	34(1)
C(3)	4243(1)	-1148(8)	5453(2)	29(1)
O(4)	4439(1)	-2391(6)	4886(1)	30(1)
C(5)	4121(1)	-1113(8)	4307(2)	26(1)
C(6)	4200(1)	-1949(8)	3583(2)	27(1)
N(7)	4562(1)	-3888(7)	3465(1)	34(1)
O(8)	4518(1)	-4080(6)	2726(1)	42(1)
N(9)	4117(1)	-2144(7)	2376(2)	35(1)
C(10)	3915(1)	-838(8)	2903(2)	26(1)
N(11)	4463(1)	-1884(8)	6111(2)	47(1)
N(12)	3505(1)	1208(7)	2897(1)	29(1)
N(13)	3264(1)	2106(7)	2259(1)	28(1)
O(14)	2883(1)	3924(6)	2261(1)	39(1)
O(15)	3397(1)	1271(6)	1681(1)	40(1)
O(1S)	2825(1)	6398(6)	4210(2)	45(1)
N(2S)	2770(1)	3514(7)	3786(2)	31(1)
O(3S)	3033(1)	5711(6)	557(1)	33(1)

Table S15. Bond lengths [Å] and angles [°] for 5·H₂O.

N(1)-C(5)	1.285(4)	N(1)-N(2)	1.396(4)
N(2)-C(3)	1.297(4)	C(3)-N(11)	1.311(4)
C(3)-O(4)	1.359(4)	O(4)-C(5)	1.369(4)
C(5)-C(6)	1.448(4)	C(6)-N(7)	1.297(4)
C(6)-C(10)	1.446(4)	N(7)-O(8)	1.374(3)
O(8)-N(9)	1.403(3)	N(9)-C(10)	1.317(4)
C(10)-N(12)	1.375(4)	N(11)-H(11A)	0.8600
N(11)-H(11B)	0.8600	N(12)-N(13)	1.315(3)
N(13)-O(15)	1.244(3)	N(13)-O(14)	1.260(3)
O(1S)-N(2S)	1.400(4)	O(1S)-H(1S)	0.98(4)
N(2S)-H(2SA)	0.8800(11)	N(2S)-H(2SB)	0.8801(11)
N(2S)-H(2SC)	0.8801(11)	O(3S)-H(3SA)	0.8402(11)
O(3S)-H(3SB)	0.8402(11)		
<hr/>			
C(5)-N(1)-N(2)	106.8(2)	C(3)-N(2)-N(1)	105.8(2)
N(2)-C(3)-N(11)	128.5(3)	N(2)-C(3)-O(4)	112.8(3)
N(11)-C(3)-O(4)	118.7(3)	C(3)-O(4)-C(5)	102.1(2)
N(1)-C(5)-O(4)	112.4(3)	N(1)-C(5)-C(6)	128.4(3)
O(4)-C(5)-C(6)	119.2(3)	N(7)-C(6)-C(10)	109.9(3)
N(7)-C(6)-C(5)	122.1(3)	C(10)-C(6)-C(5)	128.0(3)
C(6)-N(7)-O(8)	105.3(2)	N(7)-O(8)-N(9)	112.0(2)
C(10)-N(9)-O(8)	104.5(2)	N(9)-C(10)-N(12)	131.6(3)
N(9)-C(10)-C(6)	108.4(3)	N(12)-C(10)-C(6)	120.0(3)
C(3)-N(11)-H(11A)	120.0	C(3)-N(11)-H(11B)	120.0
H(11A)-N(11)-H(11B)	120.0	N(13)-N(12)-C(10)	116.6(3)
O(15)-N(13)-O(14)	120.9(2)	O(15)-N(13)-N(12)	123.3(3)
O(14)-N(13)-N(12)	115.9(3)	N(2S)-O(1S)-H(1S)	106(2)
O(1S)-N(2S)-H(2SA)	112(2)	O(1S)-N(2S)-H(2SB)	107(2)
H(2SA)-N(2S)-H(2SB)	120(3)	O(1S)-N(2S)-H(2SC)	105(2)
H(2SA)-N(2S)-H(2SC)	113(3)	H(2SB)-N(2S)-H(2SC)	97(3)
H(3SA)-O(3S)-H(3SB)	81(3)		

Table S16. Anisotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for $5 \cdot \text{H}_2\text{O}$. The anisotropic displacement factor exponent takes the form: $-2\pi^2 [h^2 a^{*2} U^{11} + \dots + 2 h k a^* b^* U^{12}]$

	U11	U22	U33	U23	U13	U12
N(1)	33(2)	41(2)	22(2)	2(1)	6(1)	9(1)
N(2)	34(2)	46(2)	21(1)	0(1)	4(1)	10(1)
C(3)	29(2)	36(2)	22(2)	-1(2)	4(1)	4(1)
O(4)	27(1)	41(1)	21(1)	2(1)	2(1)	11(1)
C(5)	23(2)	30(2)	23(2)	0(1)	1(1)	2(1)
C(6)	26(2)	30(2)	27(2)	4(1)	5(1)	0(1)
N(7)	33(2)	45(2)	26(2)	0(1)	9(1)	10(1)
O(8)	40(1)	56(2)	32(1)	2(1)	15(1)	19(1)
N(9)	39(2)	44(2)	25(2)	0(1)	11(1)	12(1)
C(10)	28(2)	30(2)	21(2)	1(1)	8(1)	-3(1)
N(11)	43(2)	70(2)	24(2)	-2(2)	-2(1)	24(2)
N(12)	28(1)	37(2)	20(1)	4(1)	2(1)	9(1)
N(13)	31(2)	34(2)	20(2)	1(1)	2(1)	2(1)
O(14)	35(1)	52(2)	29(1)	6(1)	0(1)	14(1)
O(15)	52(2)	51(2)	18(1)	1(1)	6(1)	12(1)
O(1S)	69(2)	27(1)	36(2)	-2(1)	-1(1)	3(1)
N(2S)	29(2)	31(2)	34(2)	-4(1)	4(1)	5(1)
O(3S)	31(1)	42(1)	26(1)	5(1)	2(1)	-5(1)

Table S17. Hydrogen coordinates ($\times 10^4$) and isotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for $5 \cdot \text{H}_2\text{O}$.

	x	y	z	U(eq)
H(11A)	4348	-1083	6476	56
H(11B)	4722	-3166	6177	56
H(1S)	2910(14)	5630(100)	4710(20)	54
H(2SA)	3049(6)	2340(60)	3836(18)	38
H(2SB)	2486(6)	2560(70)	3846(18)	38
H(2SC)	2670(12)	4220(80)	3341(6)	38
H(3SA)	3316(5)	6630(70)	650(20)	40
H(3SB)	3166(12)	4500(60)	901(11)	40

Table S18. Torsion angles [°] for **5**·H₂O.

C(5)-N(1)-N(2)-C(3)	-0.2(4)
N(1)-N(2)-C(3)-N(11)	179.4(4)
N(1)-N(2)-C(3)-O(4)	0.7(4)
N(2)-C(3)-O(4)-C(5)	-0.9(3)
N(11)-C(3)-O(4)-C(5)	-179.7(3)
N(2)-N(1)-C(5)-O(4)	-0.3(4)
N(2)-N(1)-C(5)-C(6)	179.0(3)
C(3)-O(4)-C(5)-N(1)	0.7(4)
C(3)-O(4)-C(5)-C(6)	-178.7(3)
N(1)-C(5)-C(6)-N(7)	-178.9(3)
O(4)-C(5)-C(6)-N(7)	0.3(5)
N(1)-C(5)-C(6)-C(10)	1.8(6)
O(4)-C(5)-C(6)-C(10)	-179.0(3)
C(10)-C(6)-N(7)-O(8)	-0.4(4)
C(5)-C(6)-N(7)-O(8)	-179.8(3)
C(6)-N(7)-O(8)-N(9)	1.0(3)
N(7)-O(8)-N(9)-C(10)	-1.2(3)
O(8)-N(9)-C(10)-N(12)	-178.2(3)
O(8)-N(9)-C(10)-C(6)	0.9(3)
N(7)-C(6)-C(10)-N(9)	-0.3(4)
C(5)-C(6)-C(10)-N(9)	179.0(3)
N(7)-C(6)-C(10)-N(12)	178.9(3)
C(5)-C(6)-C(10)-N(12)	-1.8(5)
N(9)-C(10)-N(12)-N(13)	-0.7(5)
C(6)-C(10)-N(12)-N(13)	-179.8(3)
C(10)-N(12)-N(13)-O(15)	-2.0(4)
C(10)-N(12)-N(13)-O(14)	178.6(3)

Table S19. Hydrogen bonds for **5**·H₂O [Å and °].

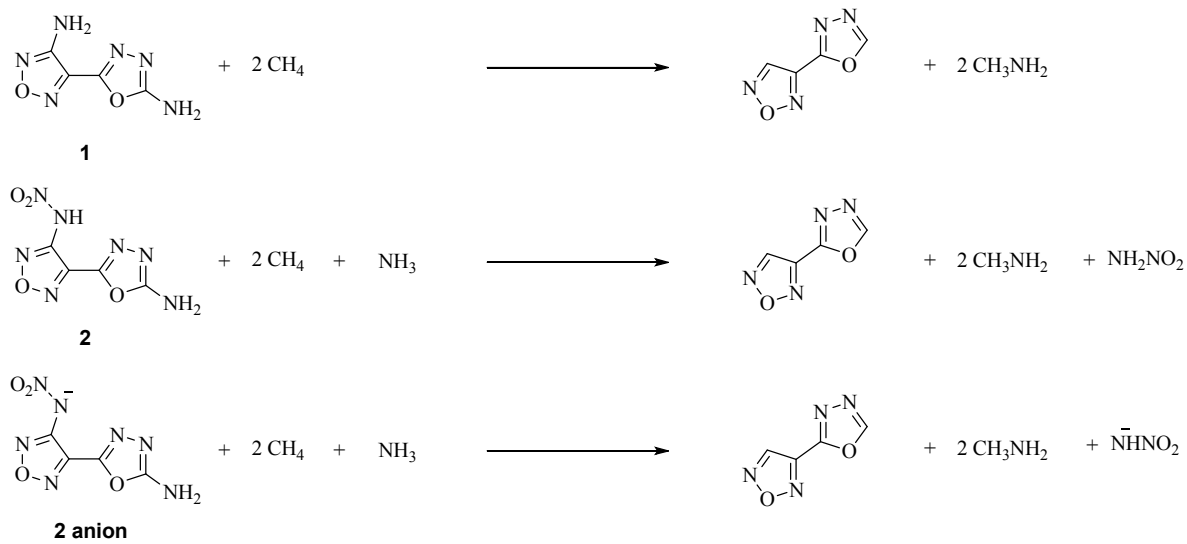
D-H...A	d(D-H)	d(H...A)	d(D...A)	<(DHA)
N(11)-H(11A)...N(9)#1	0.86	2.30	3.140(4)	167.2
N(11)-H(11B)...N(7)#2	0.86	2.27	3.121(4)	170.3
O(1S)-H(1S)...O(3S)#3	0.98(4)	1.65(5)	2.637(3)	178(4)
N(2S)-H(2SA)...N(1)	0.8800(11)	2.234(17)	3.031(4)	150(3)
N(2S)-H(2SB)...O(3S)#40	0.8801(11)	2.071(15)	2.892(4)	155(3)
N(2S)-H(2SC)...O(14)	0.8801(11)	2.20(2)	2.933(4)	140(3)
O(3S)-H(3SA)...N(2)#50	0.8402(11)	2.04(2)	2.773(3)	146(3)
O(3S)-H(3SB)...O(15)	0.8402(11)	1.977(5)	2.812(3)	172(3)

Symmetry transformations used to generate equivalent atoms:

#1 x,-y,z+1/2 #2 -x+1,-y-1,-z+1 #3 x,-y+1,z+1/2

#4 -x+1/2,y-1/2,-z+1/2 #5 x,-y+1,z-1/2

Computational Details



Scheme S1. Isodesmic reactions for **1**, **2** and **2 anion**