

*Supporting information*

**Cocatalyst-Free CdS nanorods/ZnS nanoparticles Composite for  
High-Performance Visible-Light-Driven Hydrogen Production from  
Water**

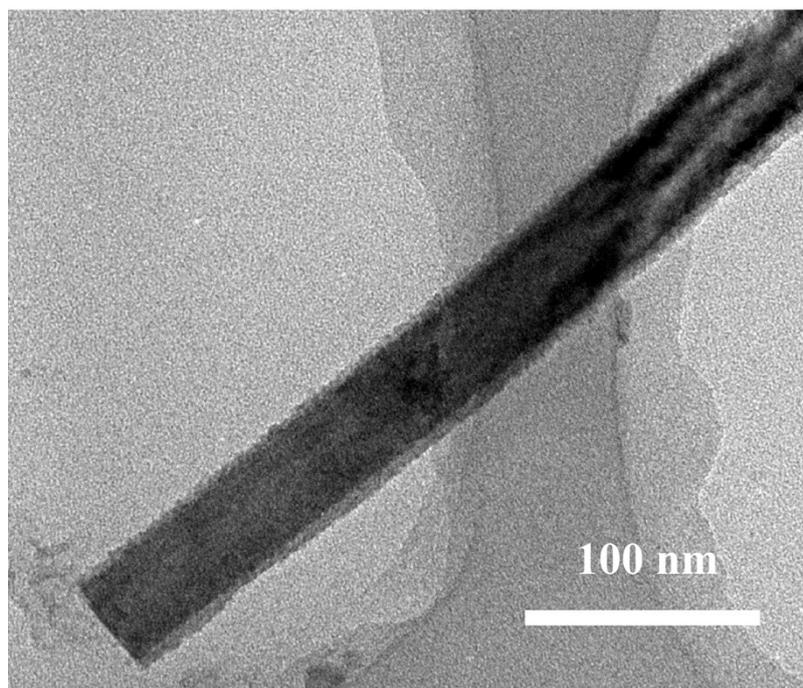
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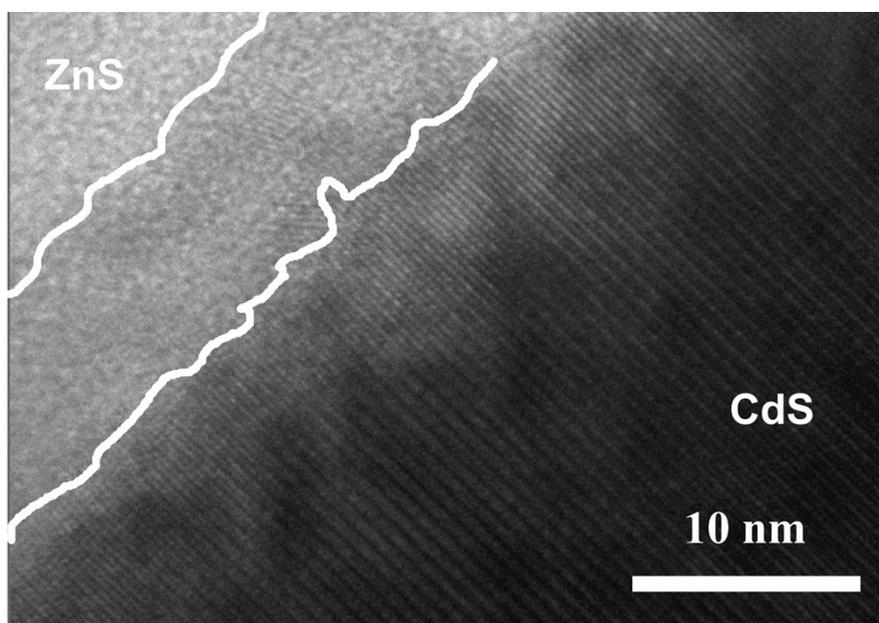
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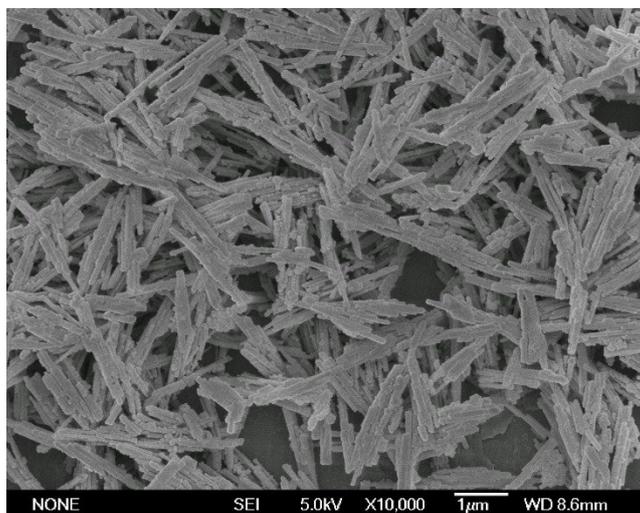
Tel/Fax: 86-551-63606207



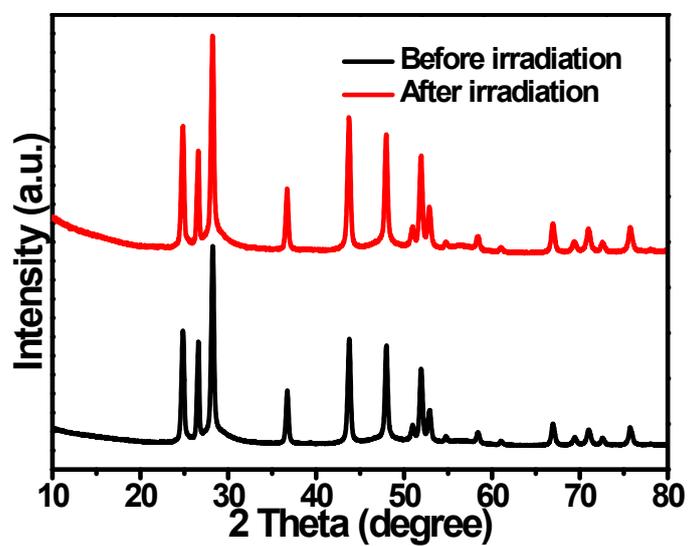
**Figure 1.** TEM image of CdS/ZnS-0.5 NRs photocatalyst.



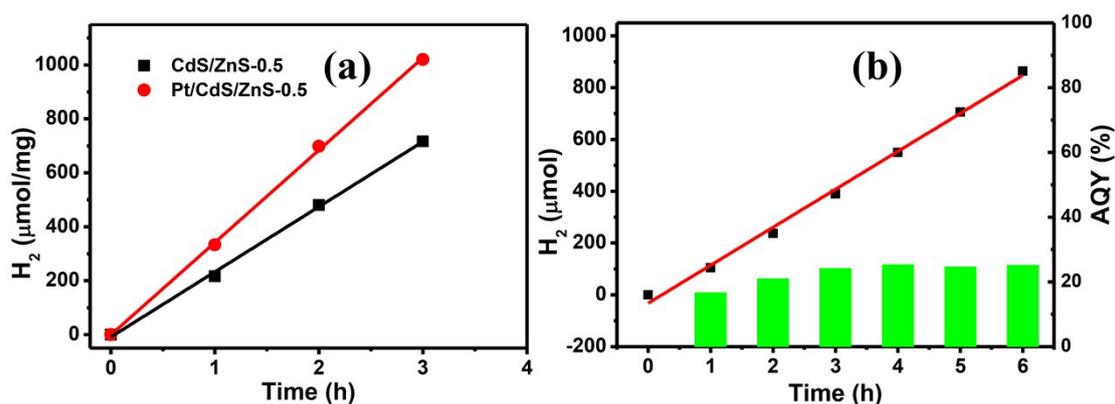
**Figure S2.** HRTEM image of CdS/ZnS-0.5 photocatalyst.



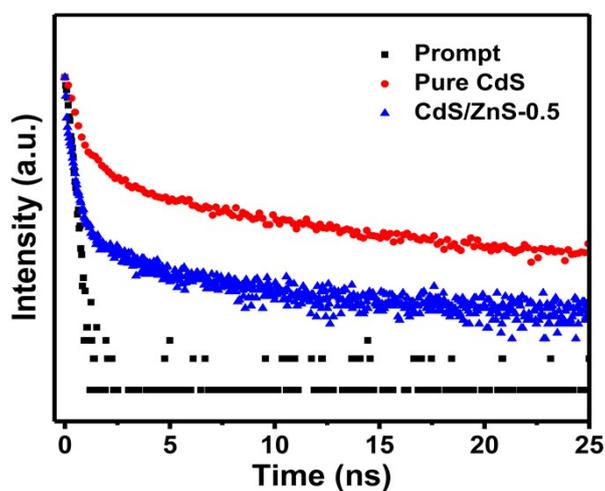
**Figure S3.** SEM image of CdS/ZnS-0.5 sample after photocatalysis.



**Figure S4.** Powder XRD spectrum of CdS/ZnS-0.5 sample after photocatalysis.



**Figure S5.** (a) Photocatalytic H<sub>2</sub> evolution over CdS/ZnS-0.5 (black line) and 1 wt% Pt/CdS/ZnS-0.5 (red line) at room temperature under visible light ( $\lambda > 420$  nm). (b) The time courses of H<sub>2</sub> production and apparent quantum yield (AQY) over 1.0 wt% Pt modified CdS/ZnS-0.5 NRs photocatalyst under monochromatic 420 nm light irradiation. The system contains 1.0 mg photocatalyst, 0.75 M Na<sub>2</sub>S and 1.05 M Na<sub>2</sub>SO<sub>3</sub> in 20 mL aqueous solution. The bars represent the apparent quantum yields.



**Figure S6.** Time-resolved photoluminescence (TRPL) decay spectra of pure CdS (red) and CdS/ZnS-0.5 (blue) excited at a wavelength of 405 nm.