

Evaluation of Brønsted Acidity and Proton Topology in Zr- and Hf-based Metal–Organic Frameworks Using Potentiometric Acid–Base Titration

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Materials and Methods.

NU-1000,¹ Zr-UiO-66,² Zr-UiO-67(HCl),² Zr-PCN-57,³ and Hf-UiO-66⁴ were synthesized according to methods reported in the literature.

Instrumentation.

Activation of MOF samples was performed on a Micromeritics SmartVacPrep (Micromeritics, Norcross, GA, USA). N₂ adsorption isotherms were collected on a Tristar II 3020 (Micromeritics) at 77 K. All pore size distributions were obtained using a carbon slit pore model with a N₂ kernel (Micromeritics). Powder X-ray diffraction (PXRD) patterns were collected on a Rigaku Smartlab instrument. Thermogravimetric analyses (TGA) were carried out on a TA Instruments Q500 thermogravimetric analyzer with an evolved gas analysis furnace. Samples were heated from 20 °C to 800 °C at a rate of 10 °C/minute under flowing O₂.

MOF Ligands.

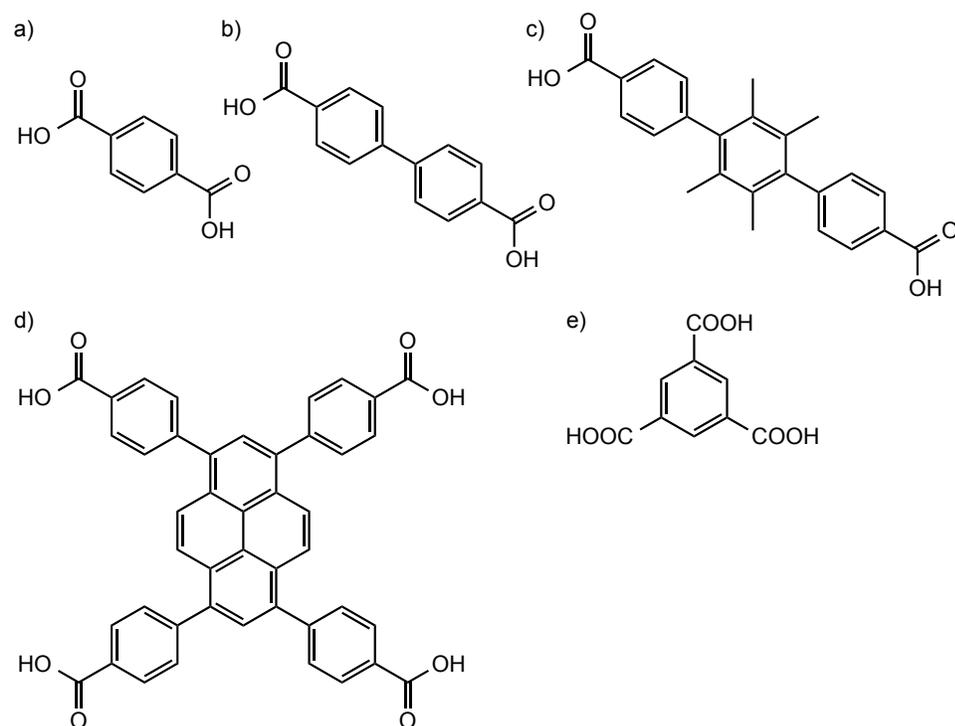


Figure S1. Ligands used for a) UiO-66, terephthalic acid, b) UiO-67, biphenyl-4,4'-dicarboxylic acid, c) PCN-57, 2',3',5',6'-tetramethyl-terphenyl-4,4''-dicarboxylic acid, d) NU-1000, 1,3,6,8-tetrakis(*p*-benzoic acid)pyrene, and e) MOF-808, benzene-1,3,5-tricarboxylic acid.

Synthesis of Hf-UiO-67. HfCl₄ (120 mg, 0.374 mmol), biphenyl-4,4'-dicarboxylic acid (BPDC) (100 mg, 0.413 mmol), 1.5 g benzoic acid (1.5 g, 12.3 mmol), and *N,N'*-dimethylformamide (DMF) (18 mL) were added to an 8-dram vial. The mixture was sonicated for 30 min, then incubated in an oven at 100 °C for 48 h. The resulting white solid was washed with fresh DMF (3 × 20 mL) and then with acetone (3 × 20 mL). After drying the sample in a vacuum oven, the sample was activated on a SmartVacPrep port by heating at 120 °C under vacuum for 12 h. Yield: 65% white microcrystalline powder.

Synthesis of Zr-UiO-67 (Benzoic Acid). The same procedure as Hf-UiO-67 was followed except ZrCl₄ (120 mg, 0.515 mmol) was used instead of HfCl₄. Yield: 27% white microcrystalline powder.

Synthesis of Hf-PCN-57. HfOCl₂·8H₂O (120 mg, 0.293 mmol), 2',3',5',6'-tetramethyl-terphenyl-4,4''-dicarboxylic acid (TPDC-4CH₃) (100 mg, 0.267 mmol), 1.0 g benzoic acid, and *N,N'*-dimethylformamide (DMF) (20 mL) were added to an 8-dram vial. The mixture was sonicated for 30 min, then incubated in an oven at 100 °C for 24 h. The sample was washed and activated using the same procedure as that of Hf-UiO-67. Yield: 72% white microcrystalline powder.

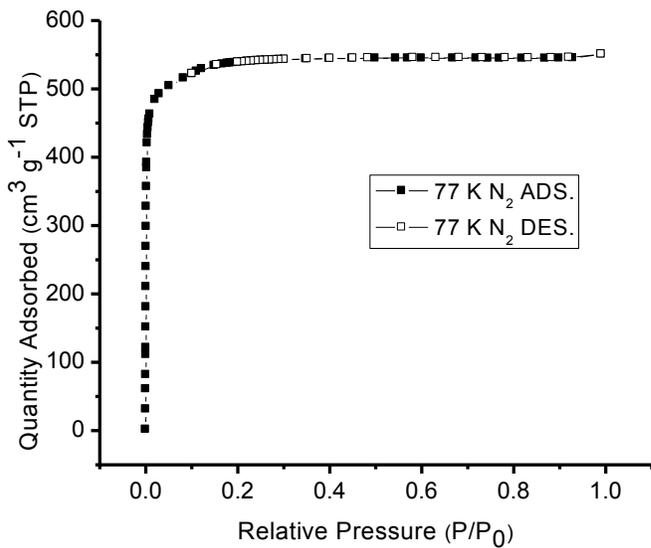


Figure S2. N_2 adsorption isotherm for Hf-UiO-67.

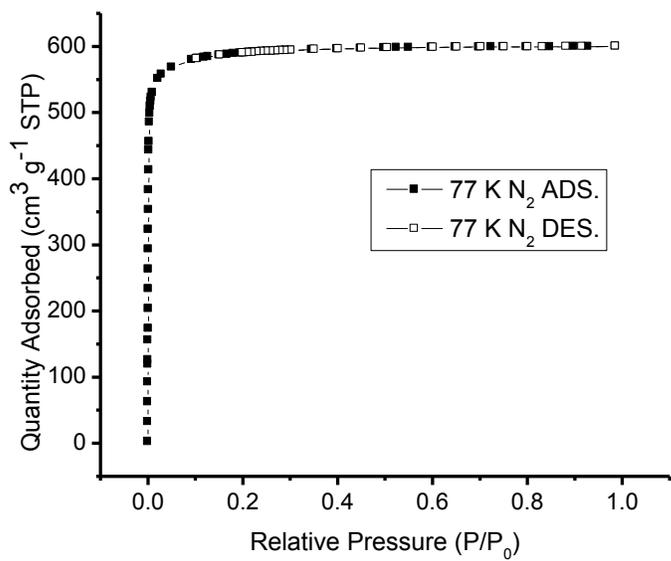


Figure S3. N_2 adsorption isotherm for Zr-UiO-67 synthesized with benzoic acid.

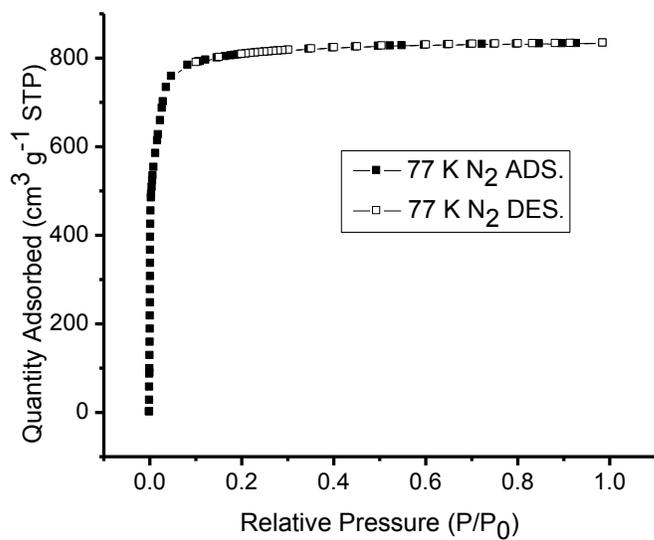


Figure S4. N₂ adsorption isotherm for Zr-PCN-57.

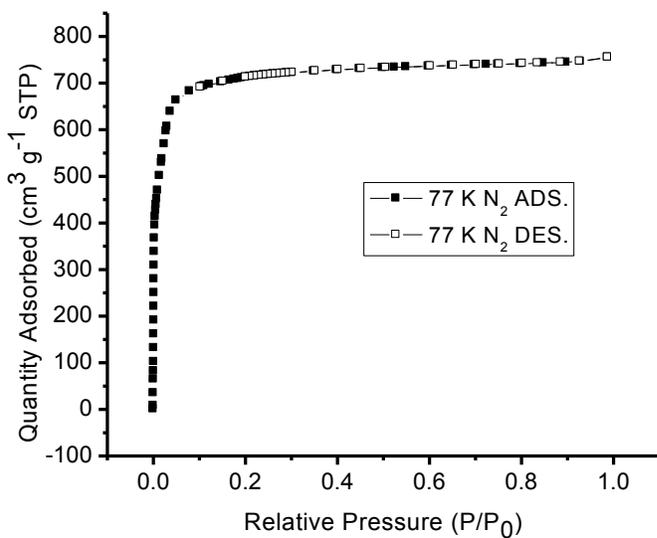


Figure S5. N₂ adsorption isotherm for Hf-PCN-57.

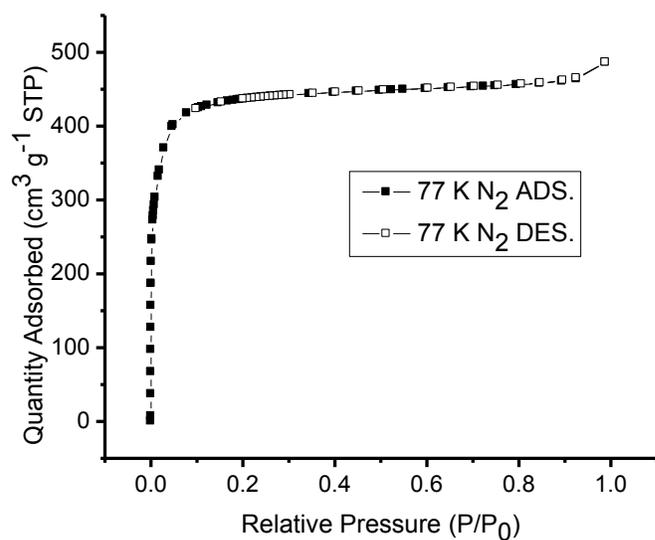


Figure S6. N₂ adsorption isotherm for Zr-UiO-66.

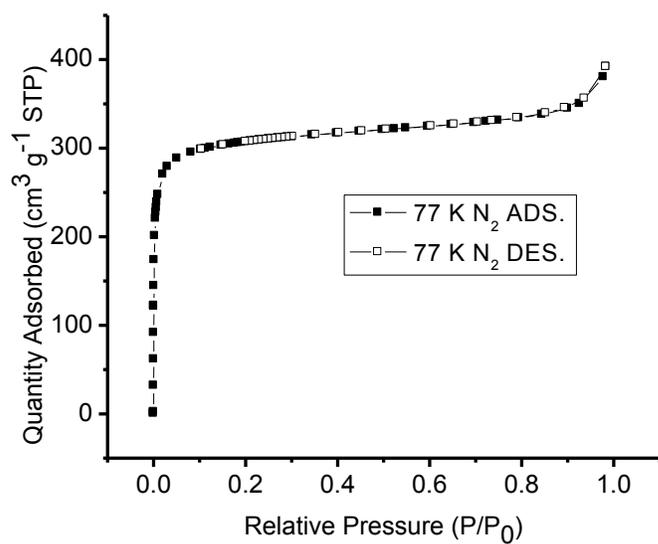


Figure S7. N₂ adsorption isotherm for Hf-UiO-66.

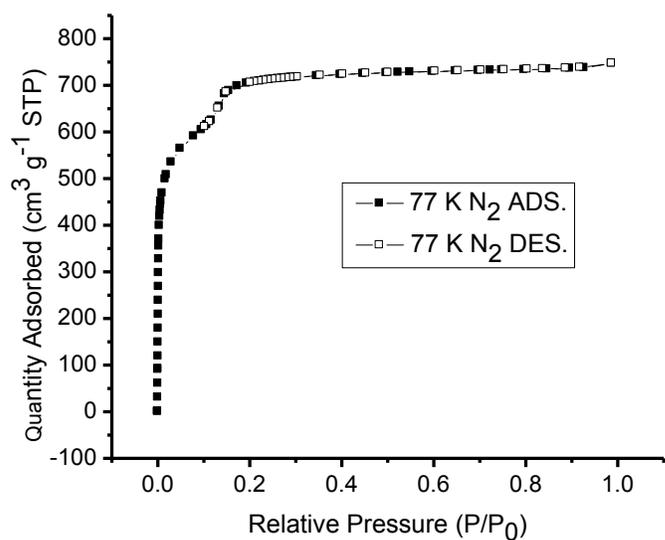


Figure S8. N₂ adsorption isotherm for Zr-UiO-67(HCl).

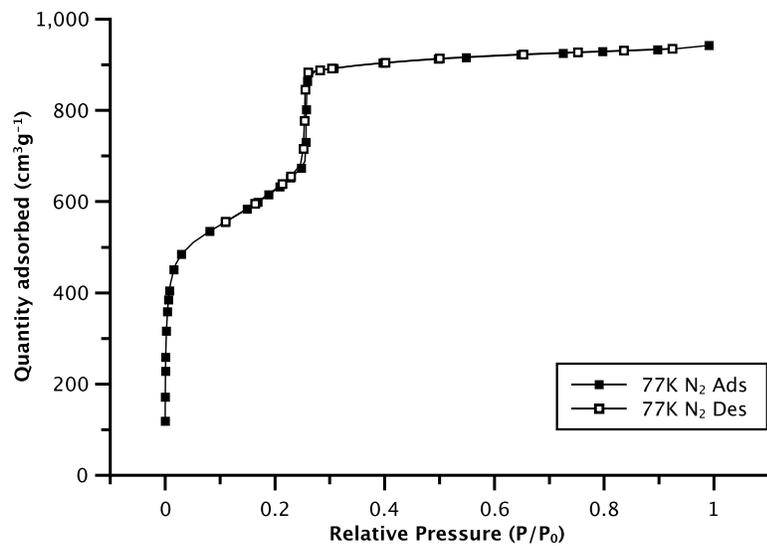


Figure S9. N₂ adsorption isotherm for NU-1000.

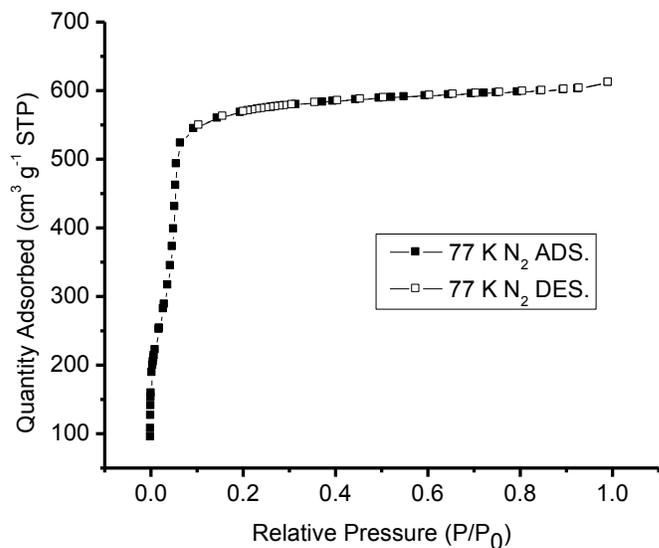


Figure S10. N₂ adsorption isotherm for MOF-808.

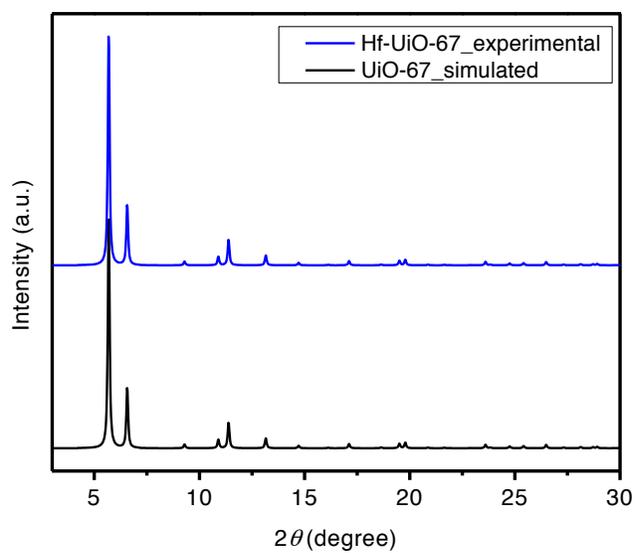


Figure S11. Simulated (black, bottom) and experimental (blue, top) PXRD patterns of Hf-UiO-67.

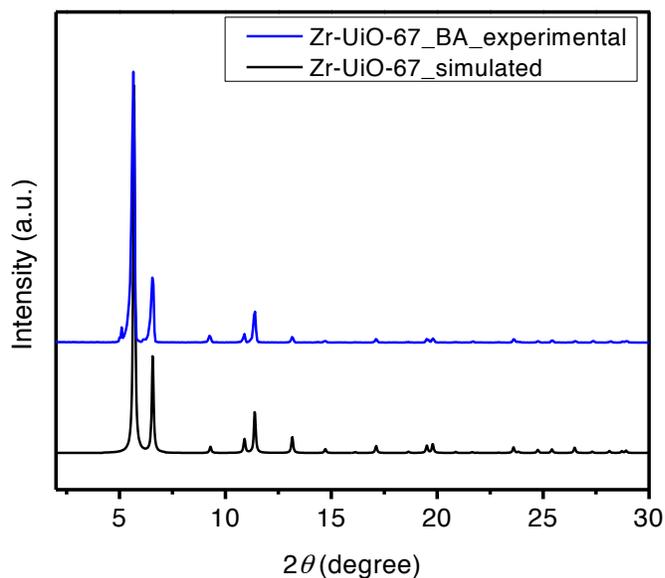


Figure S12. Simulated (black, bottom) and experimental (blue, top) PXR D patterns of Zr-UiO-67 synthesized with benzoic acid.

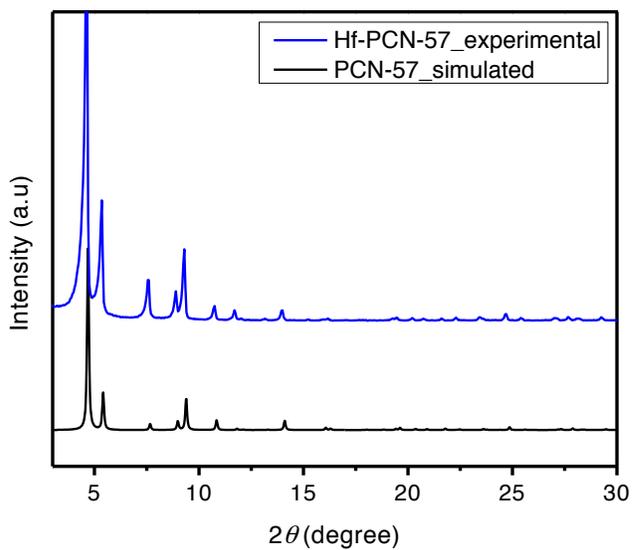


Figure S13. Simulated (black, bottom) and experimental (blue, top) PXR D patterns of Hf-PCN-57.

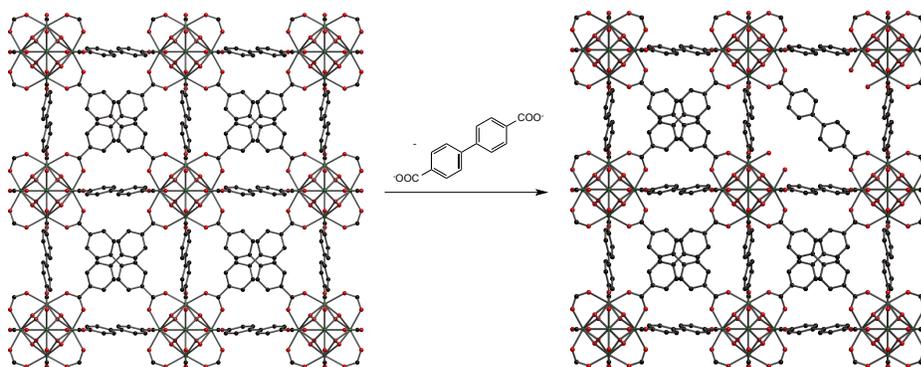


Figure S14. Representative crystal structure of UiO-67 showing a missing linker. Zr/Hf, C, and O are shown in green, black, and red, respectively. Hydrogen atoms have been omitted for clarity.

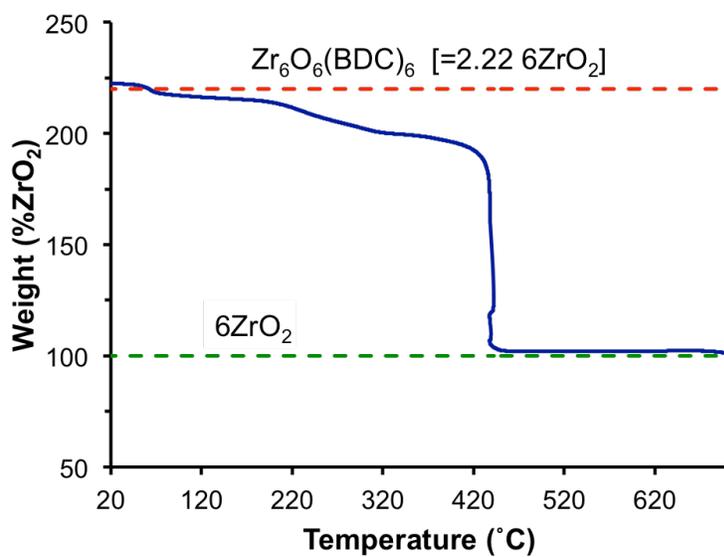


Figure S15. TGA trace of activated Zr-UiO-66.

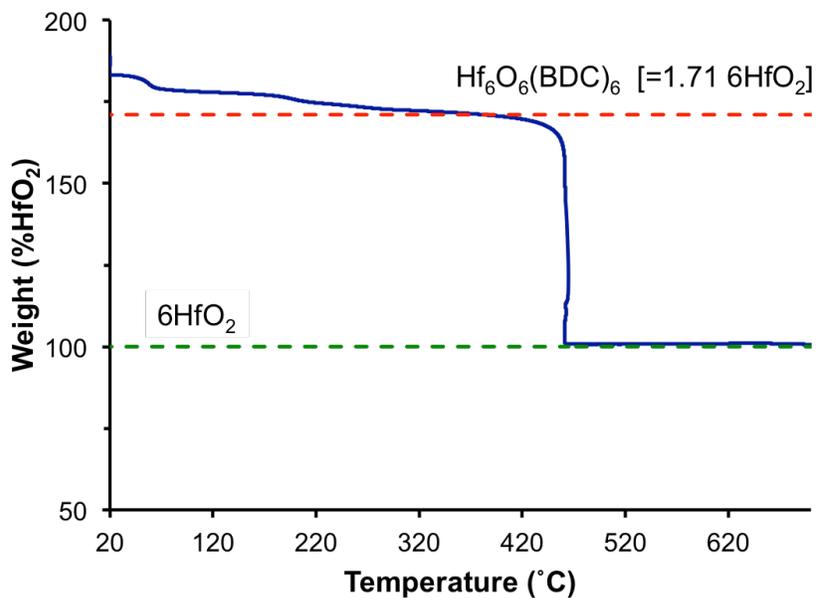


Figure S16. TGA trace of activated Hf-UiO-66.

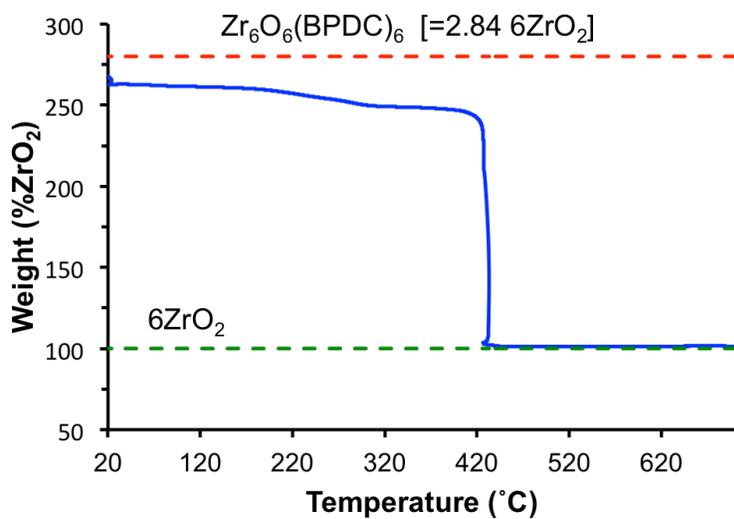


Figure S17. TGA trace of activated Zr-UiO-67(HCl) synthesized with hydrochloric acid.

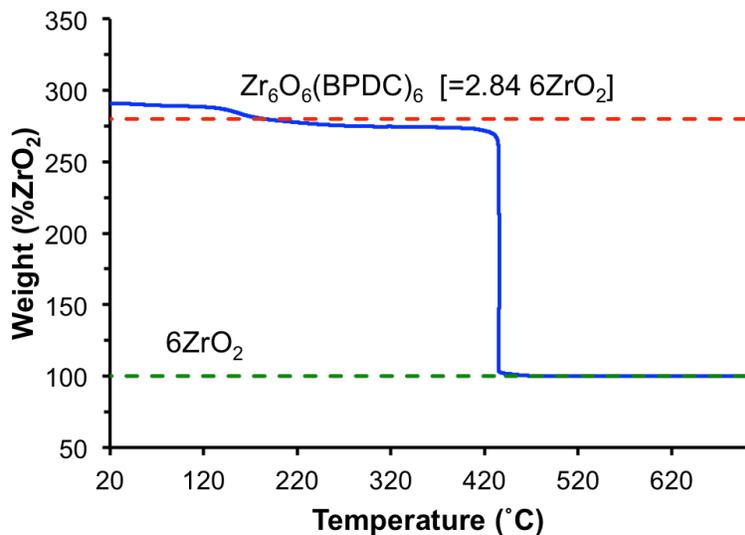


Figure S18. TGA trace of activated Zr-UiO-67 synthesized with benzoic acid.

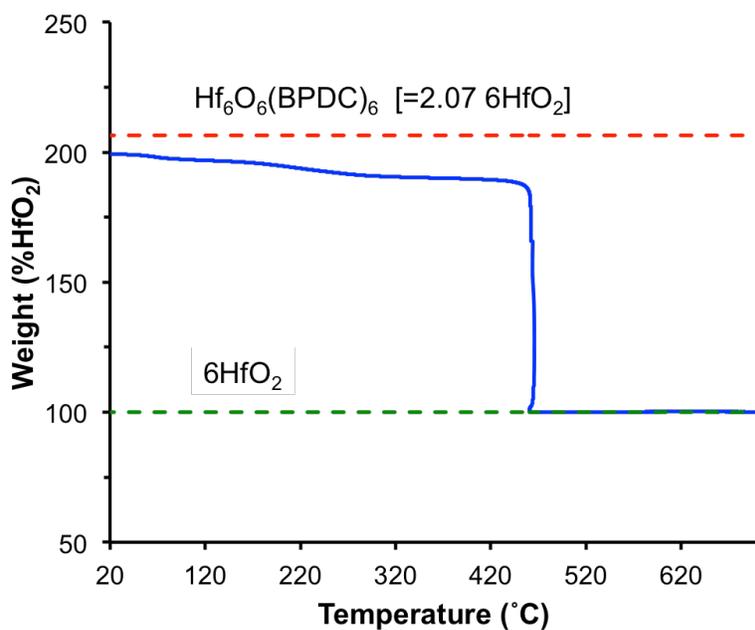


Figure S19. TGA trace of activated Hf-UiO-67.

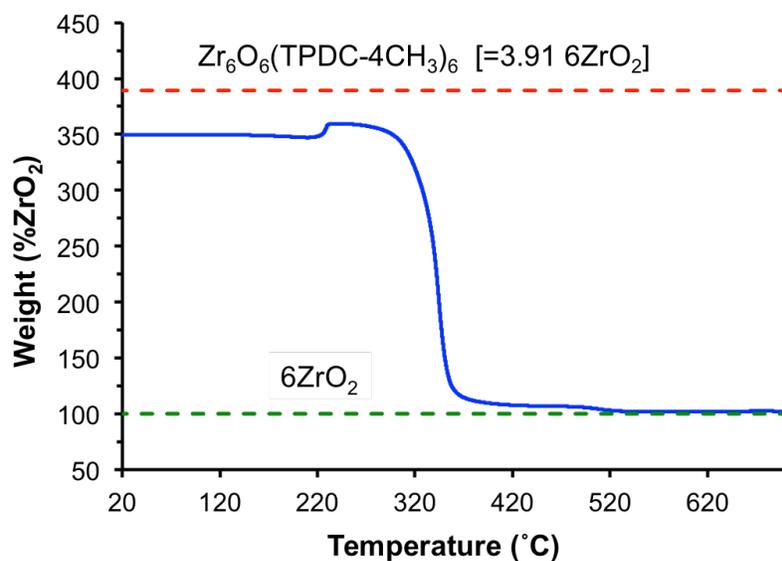


Figure S20. TGA trace of activated Zr-PCN-57. N.B.: We do not understand the weight gain observed near 220 °C for Zr-PCN-57; however, we observed similar TGA traces for multiple samples.

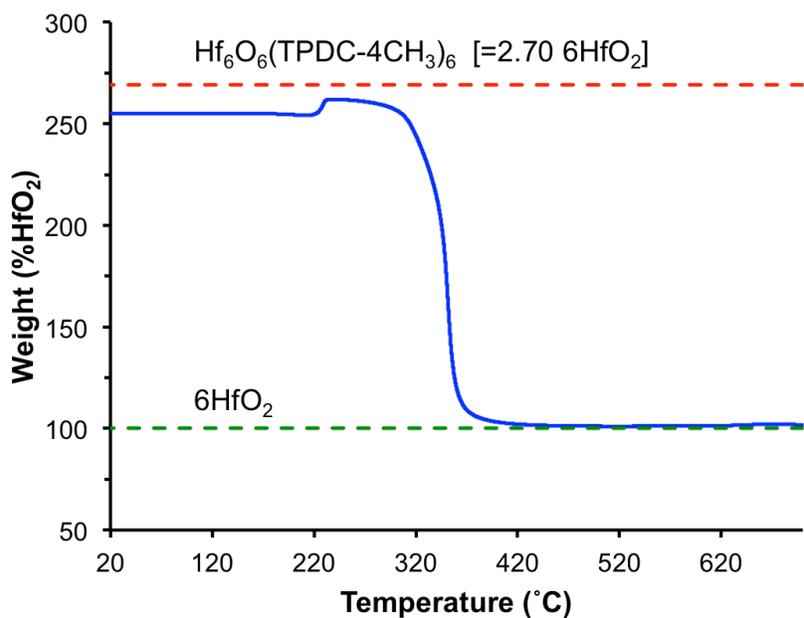


Figure S21. TGA trace of activated Hf-PCN-57. N.B.: We do not understand the weight gain observed near 220 °C for Hf-PCN-57; however, we observed similar TGA traces for multiple samples.

Table S1. Estimated Missing Linkers for UiO-type MOFs from TGA analyses.

MOF	Decomposition Starting Point (%ZrO ₂ /HfO ₂)		Estimated Missing Linkers (per M ₆ cluster)
	Experimental	Theoretical	
Zr-UiO-66	186	222	1.6
Zr-UiO-67(HCl)	242	284	1.3
Zr-UiO-67(BA)	271	284	0.4
Zr-PCN-57	351	391	0.8
Hf-UiO-66	160	171	0.8
Hf-UiO-67(BA)	186	207	1.1
Hf-PCN-57	257	270	0.4

Missing linker defects were calculated using the TGA analysis method described by Lillerud et al.⁵

PXRD Patterns of MOFs Post-Titration

PXRD assessment of the MOFs in this study titrated to pH = 10.5 show a complete loss of crystallinity. However, typically in the pH range during which node protons appear to be undergoing deprotonation, the MOFs seem to be stable. See below and the manuscript for more details.

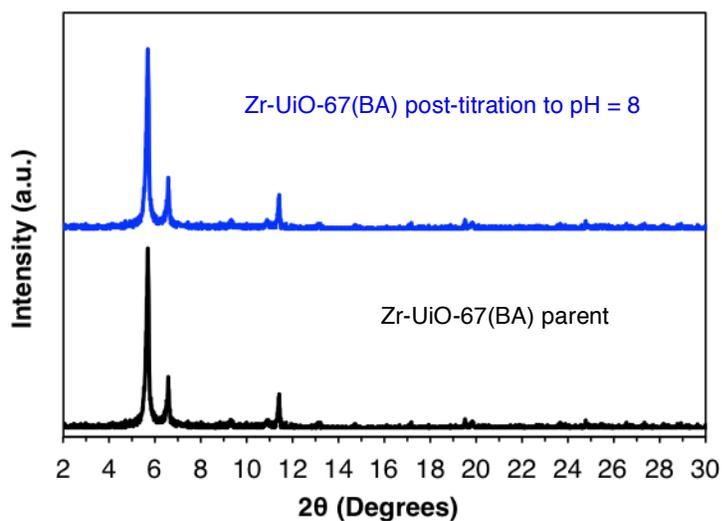


Figure S22. PXRD patterns of Zr-UiO-67(BA) parent (black, bottom) and after titration to pH = 8 (blue, top).

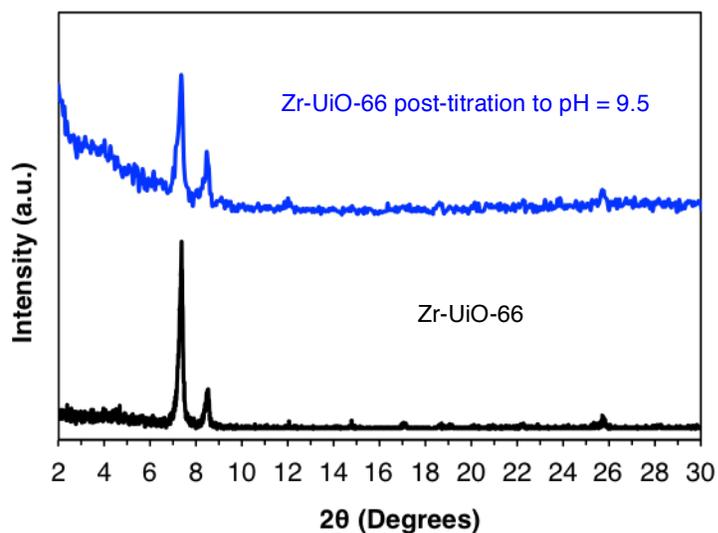


Figure S23. PXRD patterns of Zr-UiO-66 parent (black, bottom) and after titration to pH = 8 (blue, top).

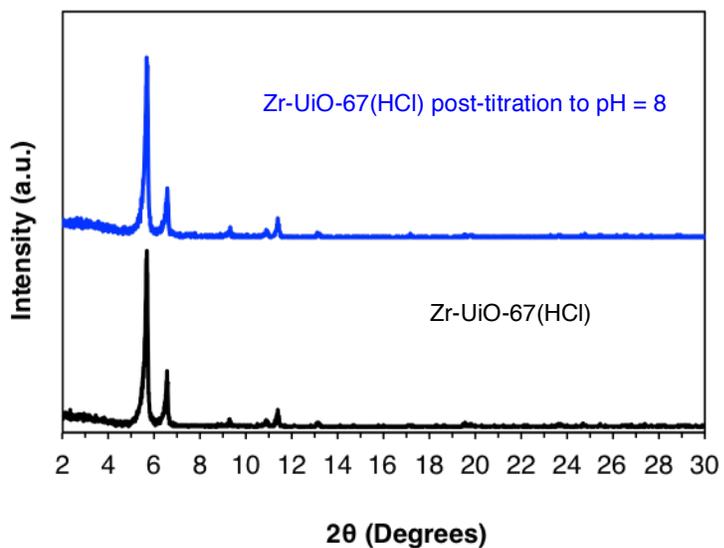


Figure S24. PXRD patterns of Zr-UiO-67(HCl) parent (black, bottom) and after titration to pH = 8 (blue, top).

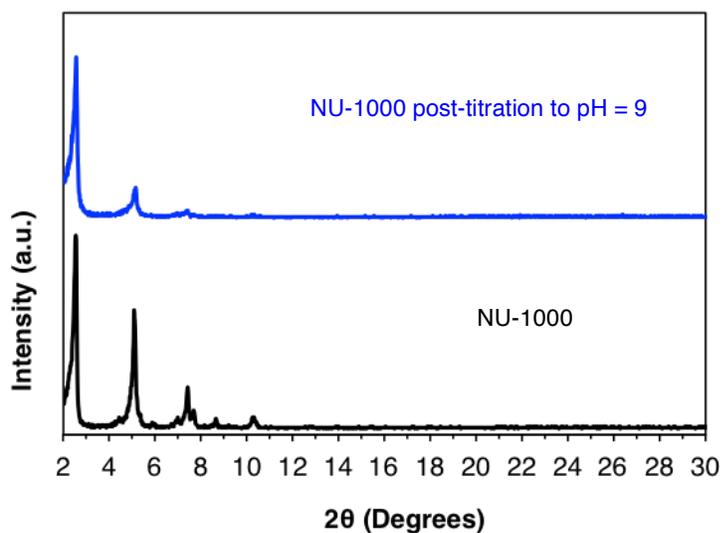


Figure S25. PXRD patterns of NU-1000 parent (black, bottom) and after titration to pH = 9 (blue, top).

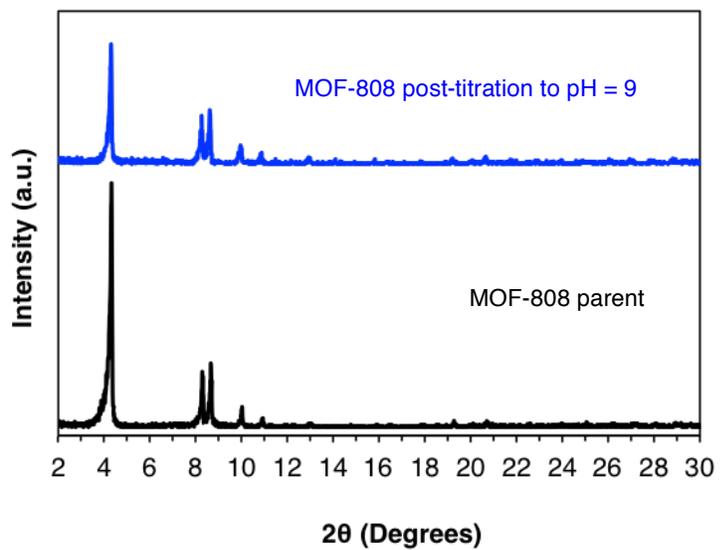


Figure S26. PXRD patterns of MOF-808 parent (black, bottom) and after titration to pH = 9 (blue, top).

Titration Data Zr-UiO-67

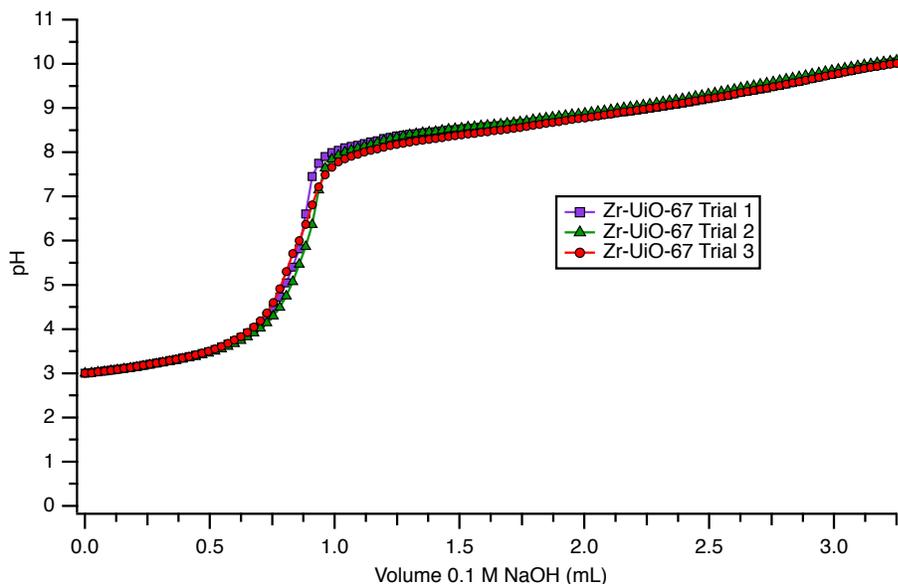


Figure S27. Overlay of three acid–base titration curves for Zr-UiO-67 showing reproducibility between samples.

Hf-UiO-67

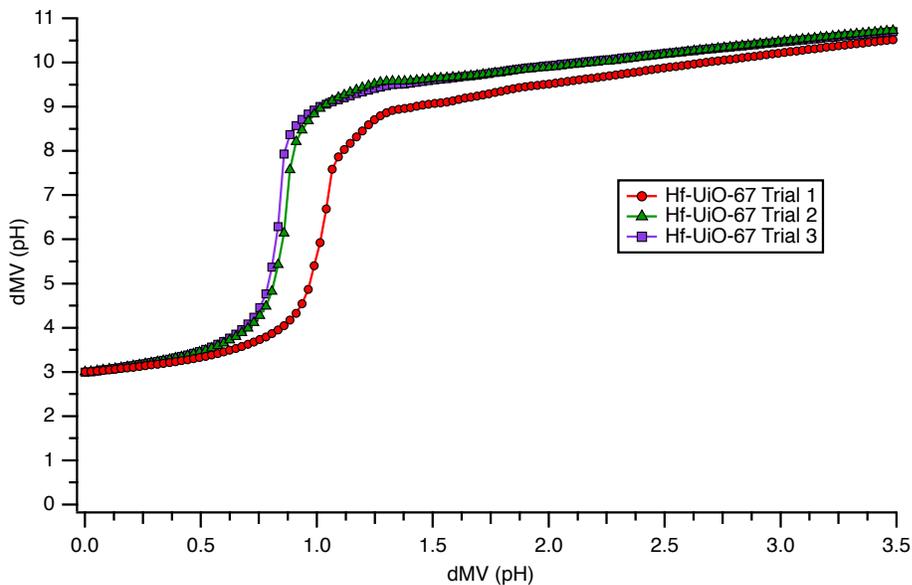


Figure S28. Overlay of three acid–base titration curves for Hf-UiO-67 showing reproducibility between samples.

Zr-PCN-57

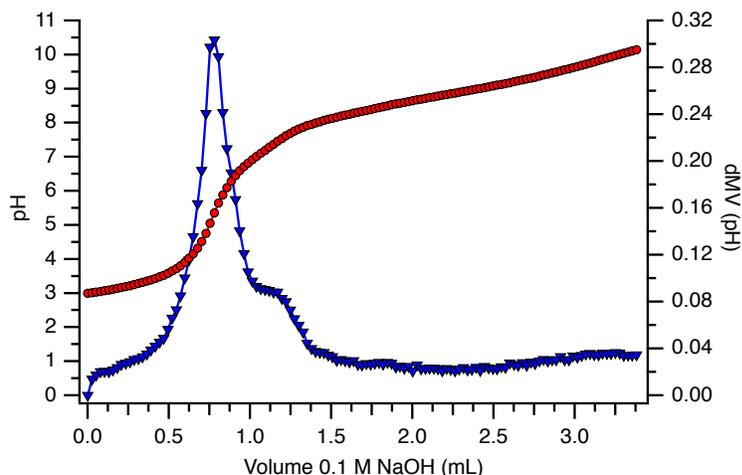


Figure S29. Acid–base titration curve of Zr-PCN-57 (red) and first derivative curve (blue).

The first derivative curve of Zr-PCN-57 may indicate a second inflection point/equivalence point around pH 7.5 suggesting the presence of some defect sites in the sample; however, due to the poor dispersion of the sample in aqueous solution we are hesitant to draw any conclusions from this data set beyond the pK_a value determined from the equivalence point at 5.4.

Hf-PCN-57

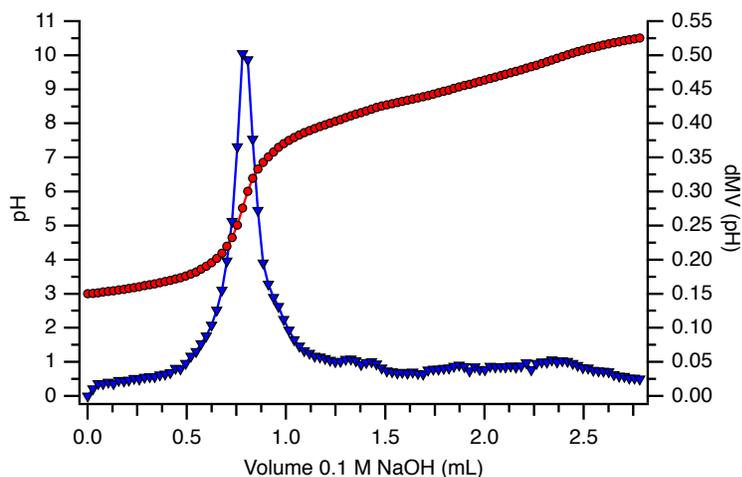


Figure S30. Acid–base titration curve of Hf-PCN-57 (red) and first derivative curve (blue).

Zr-UiO-66

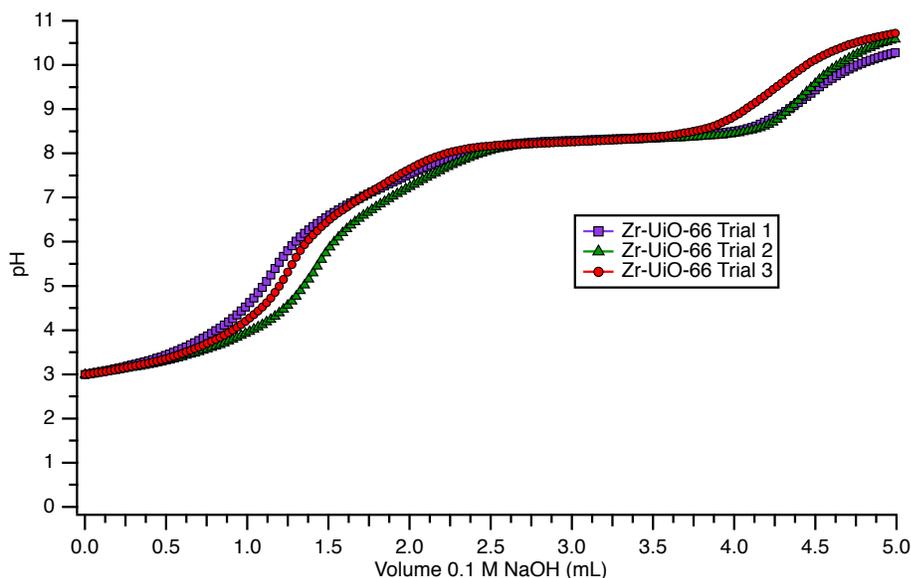


Figure S31. Overlay of three acid–base titration curves for Zr-UiO-66 showing reproducibility between samples.

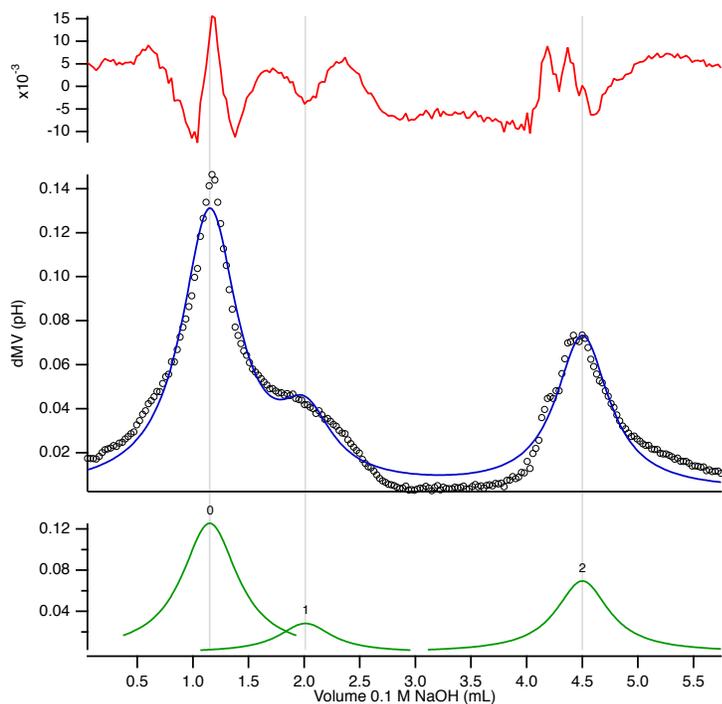


Figure S32. Representative curve fitting for the first derivative of the titration curve of pH as a function of volume of titrant added for Zr-UiO-66: Located peaks (green) (bottom); first derivative data (black circles) and fit trace (blue) (middle); residual plot (red) (top). Peaks were fit as Lorentzian functions.

Multipeak Fit report for Zr-UiO-66:

Chi square: 0.0075287

Total fitted points: 220

Multi-peak fit version 2.16

Total Peak Area = 0.20869 +/- 0.0060043

Baseline Type: Constant

y0 = 0.0023544 +/- 0.0010797

Peak 0 Type: Lorentzian

Location = 1.15 +/- 0.0055287
Height = 0.12558 +/- 0.0020745
Area = 0.12024 +/- 0.0036137
FWHM = 0.60951 +/- 0.019437

Fit function parameters

Location = 1.15 +/- 0.0055287
FWHM = 0.60951 +/- 0.019437
Area = 0.12024 +/- 0.0036137

Peak 1 Type: Lorentzian

Location = 2.0103 +/- 0.023511
Height = 0.028322 +/- 0.0021178
Area = 0.025549 +/- 0.0034409
FWHM = 0.57428 +/- 0.08251

Fit function parameters

Location = 2.0103 +/- 0.023511
FWHM = 0.57428 +/- 0.08251
Area = 0.025549 +/- 0.0034409

Peak 2 Type: Lorentzian

Location = 4.5007 +/- 0.0084137
Height = 0.069551 +/- 0.0020925
Area = 0.06291 +/- 0.0033396
FWHM = 0.57584 +/- 0.032298

Fit function parameters

Location = 4.5007 +/- 0.0084137

FWHM = 0.57584 +/- 0.032298
 Area = 0.06291 +/- 0.0033396

Table S2. Titrant consumed after the first equivalence point for Zr-UiO-66

	mmoles OH ⁻ consumed between EP1 to EP3
Trial 1	0.333
Trial 2	0.304
Trial 3	0.302
Average	0.313 (± 0.017)

Table S3. Defect calculations for Zr-UiO-66

Missing Linkers	Molecular Formula ^a	MW	mmoles H ⁺ from defects in 0.049 g sample
1	Zr ₆ O ₄ (OH) ₄ (C ₈ H ₄ O ₄) ₅ [(H ₂ O)(OH)] ₂	1570.01	0.187
2	Zr ₆ O ₄ (OH) ₄ (C ₈ H ₄ O ₄) ₄ [(H ₂ O)(OH)] ₄	1475.93	0.398
1.50	Zr ₆ O ₄ (OH) ₄ (C ₈ H ₄ O ₄) _{4.50} [(H ₂ O)(OH)] _{3.0}	1522.97	0.290
1.55	Zr ₆ O ₄ (OH) ₄ (C ₈ H ₄ O ₄) _{4.45} [(H ₂ O)(OH)] _{3.09}	1518.69	0.299
1.60	Zr ₆ O ₄ (OH) ₄ (C ₈ H ₄ O ₄) _{4.40} [(H ₂ O)(OH)] _{3.20}	1513.56	0.311
1.67	Zr ₆ O ₄ (OH) ₄ (C ₈ H ₄ O ₄) _{4.33} [(H ₂ O)(OH)] _{3.33}	1507.29	0.325
1.75	Zr ₆ O ₄ (OH) ₄ (C ₈ H ₄ O ₄) _{4.25} [(H ₂ O)(OH)] _{3.50}	1499.45	0.343

^aAll missing linkers are assumed to be replaced with 2 H₂O molecules and 2 -OH groups.

Hf-UiO-66

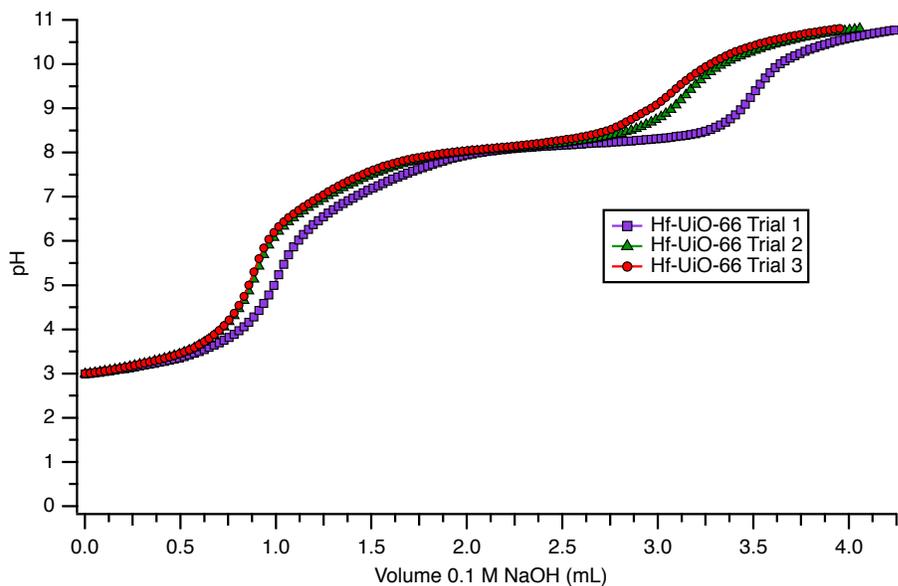


Figure S33. Overlay of three acid–base titration curves for Hf-UiO-66 showing reproducibility between samples.

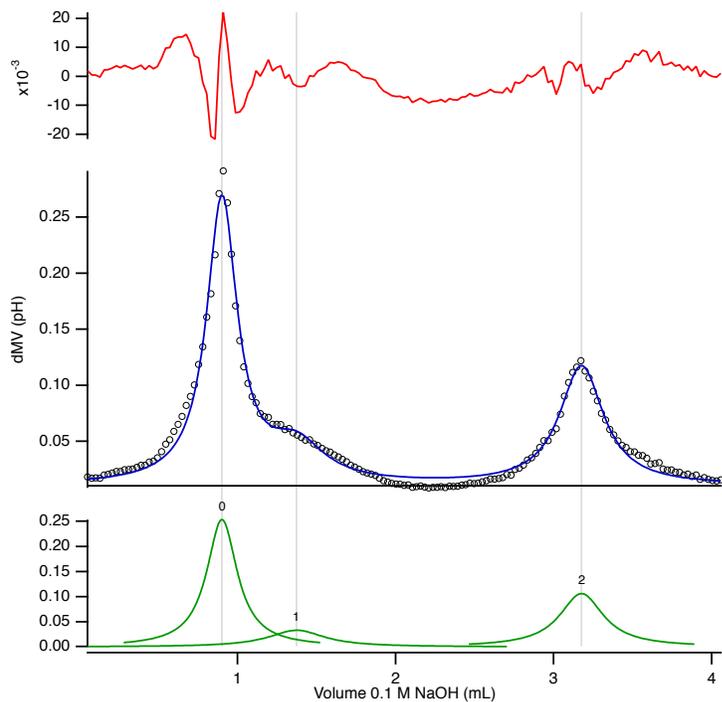


Figure S34. Representative curve fitting for the first derivative of the titration curve of pH as a function of volume of titrant added for Hf-UiO-66: Located peaks (green) (bottom); first derivative data (black circles) and fit trace (blue) (middle); residual plot (red) (top). Peaks were fit as Lorentzian functions.

Multipeak fit report for Hf-UiO-66:
 Chi square: 0.0061373
 Total fitted points: 155
 Multi-peak fit version 2.16
 Total Peak Area = 0.17198 +/- 0.0046317

Baseline Type: Constant

y0 = 0.010397 +/- 0.0011316

Peak 0 Type: Lorentzian

Location = 0.90188 +/- 0.0017373
 Height = 0.25346 +/- 0.0035568
 Area = 0.093242 +/- 0.0021403
 FWHM = 0.2342 +/- 0.0059391

Fit function parameters

Location = 0.90188 +/- 0.0017373
 FWHM = 0.2342 +/- 0.0059391
 Area = 0.093242 +/- 0.0021403

Peak 1 Type: Lorentzian

Location = 1.3735 +/- 0.019753
 Height = 0.033173 +/- 0.0026263
 Area = 0.022545 +/- 0.0033673
 FWHM = 0.43265 +/- 0.069871

Fit function parameters

Location = 1.3735 +/- 0.019753
 FWHM = 0.43265 +/- 0.069871
 Area = 0.022545 +/- 0.0033673

Peak 2 Type: Lorentzian

Location = 3.1761 +/- 0.0045788
 Height = 0.10617 +/- 0.0029212
 Area = 0.056192 +/- 0.0023521
 FWHM = 0.33693 +/- 0.01605

Fit function parameters

Location = 3.1761 +/- 0.0045788
 FWHM = 0.33693 +/- 0.01605

Area =0.056192 +/- 0.0023521

Table S4. Titrant consumed after the first equivalence point for Hf-UiO-66

	mmoles OH ⁻ consumed between EP1 to EP3
Trial 1	0.252
Trial 2	0.226
Trial 3	0.268
Average	0.249 (±0.021)

Table S5. Defect calculations for Hf-UiO-66

Missing Linkers	Molecular Formula ^a	MW	mmoles H ⁺ from defects in 0.049 g sample
1	Hf ₆ O ₄ (OH) ₄ (C ₈ H ₄ O ₄) ₅ [(H ₂ O)(OH)] ₂	2093.60	0.140
2	Hf ₆ O ₄ (OH) ₄ (C ₈ H ₄ O ₄) ₄ [(H ₂ O)(OH)] ₄	1999.52	0.294
1.55	Hf ₆ O ₄ (OH) ₄ (C ₈ H ₄ O ₄) _{4.45} [(H ₂ O)(OH)] _{3.09}	2042.28	0.222
1.60	Hf ₆ O ₄ (OH) ₄ (C ₈ H ₄ O ₄) _{4.40} [(H ₂ O)(OH)] _{3.20}	2037.15	0.231
1.65	Hf ₆ O ₄ (OH) ₄ (C ₈ H ₄ O ₄) _{4.35} [(H ₂ O)(OH)] _{3.30}	2032.45	0.239
1.70	Hf ₆ O ₄ (OH) ₄ (C ₈ H ₄ O ₄) _{4.30} [(H ₂ O)(OH)] _{3.40}	2027.74	0.246
1.75	Hf ₆ O ₄ (OH) ₄ (C ₈ H ₄ O ₄) _{4.25} [(H ₂ O)(OH)] _{3.50}	2023.04	0.254
1.80	Hf ₆ O ₄ (OH) ₄ (C ₈ H ₄ O ₄) _{4.20} [(H ₂ O)(OH)] _{3.60}	2018.34	0.262
1.85	Hf ₆ O ₄ (OH) ₄ (C ₈ H ₄ O ₄) _{4.15} [(H ₂ O)(OH)] _{3.70}	2013.6	0.270

^aAll missing linkers are assumed to be replaced with 2 H₂O molecules and 2 -OH groups.

NU-1000

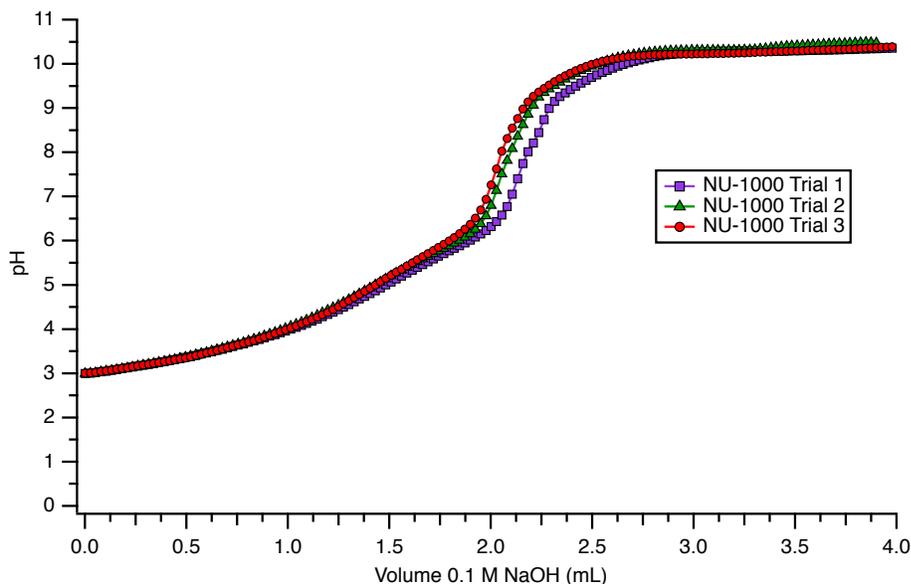


Figure S35. Overlay of three acid–base titration curves for NU-1000 showing reproducibility between samples.

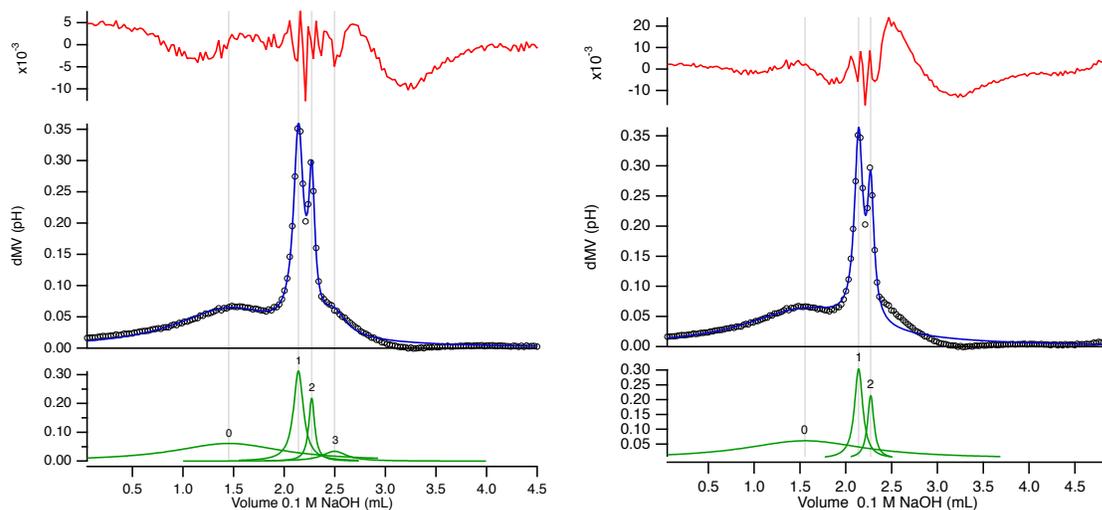


Figure S36. Representative curve fittings for the first derivative of the titration curve of pH as a function of volume of titrant added for NU-1000: Located peaks (green) (bottom); first derivative data (black circles) and fit trace (blue) (middle); residual plot (red) (top). Peaks were fit as four Lorentzian functions (left) and three Lorentzian functions (right).

Multippeak fit report for NU-1000 with 4 peaks:

Chi square: 0.0026865

Total fitted points: 172

Multi-peak fit version 2.16

Total Peak Area = 0.22374 +/- 0.004041

Baseline Type: Constant

y0 = 0 +/- 0

Peak 0 Type: Lorentzian

Location = 1.4513 +/- 0.013591
Height = 0.061526 +/- 0.00095128
Area = 0.12681 +/- 0.003338
FWHM = 1.3121 +/- 0.041091

Fit function parameters

Location = 1.4513 +/- 0.013591
FWHM = 1.3121 +/- 0.041091
Area = 0.12681 +/- 0.003338

Peak 1 Type: Lorentzian

Location = 2.1401 +/- 0.00069592
Height = 0.31358 +/- 0.0031991
Area = 0.057384 +/- 0.0011179
FWHM = 0.1165 +/- 0.0024424

Fit function parameters

Location = 2.1401 +/- 0.00069592
FWHM = 0.1165 +/- 0.0024424
Area = 0.057384 +/- 0.0011179

Peak 2 Type: Lorentzian

Location = 2.2713 +/- 0.00070087
Height = 0.21853 +/- 0.0042816
Area = 0.023995 +/- 0.0009536
FWHM = 0.069901 +/- 0.0029488

Fit function parameters

Location = 2.2713 +/- 0.00070087
FWHM = 0.069901 +/- 0.0029488
Area = 0.023995 +/- 0.0009536

Peak 3 Type: Lorentzian

Location = 2.4977 +/- 0.010343
Height = 0.034356 +/- 0.0020581
Area = 0.015556 +/- 0.0017403
FWHM = 0.28826 +/- 0.03486

Fit function parameters

Location = 2.4977 +/- 0.010343
FWHM = 0.28826 +/- 0.03486
Area = 0.015556 +/- 0.0017403

Multipeak fit report for NU-1000 with 3 peaks:

Chi square: 0.0082375

Total fitted points: 186

Multi-peak fit version 2.16

Total Peak Area = 0.23603 +/- 0.0051088

Baseline Type: Constant

y0 = 0 +/- 0

Peak 0 Type: Lorentzian

Location = 1.5559 +/- 0.023391
Height = 0.061113 +/- 0.0014903
Area = 0.15554 +/- 0.0045733
FWHM = 1.6203 +/- 0.059643

Fit function parameters

Location = 1.5559 +/- 0.023391
FWHM = 1.6203 +/- 0.059643
Area = 0.15554 +/- 0.0045733

Peak 1 Type: Lorentzian

Location = 2.1396 +/- 0.0011479
Height = 0.30434 +/- 0.0054959
Area = 0.052134 +/- 0.0017519
FWHM = 0.10905 +/- 0.0039493

Fit function parameters

Location = 2.1396 +/- 0.0011479
FWHM = 0.10905 +/- 0.0039493
Area = 0.052134 +/- 0.0017519

Peak 2 Type: Lorentzian

Location = 2.2716 +/- 0.0013596
 Height = 0.21396 +/- 0.0063174
 Area = 0.028358 +/- 0.0014543
 FWHM = 0.084377 +/- 0.0047285

Fit function parameters

Location = 2.2716 +/- 0.0013596
 FWHM = 0.084377 +/- 0.0047285
 Area = 0.028358 +/- 0.0014543

Table S6. Total moles titrant consumed for NU-1000

	Total mmoles OH ⁻ consumed to EP3	Total mmoles OH ⁻ consumed to EP4
Trial 1	0.226	0.250
Trial 2	0.216	0.247
Trial 3	0.216	0.237
Average	0.219 (± 0.006)	0.244 (± 0.007)

Table S7. Defect calculations for NU-1000

Occupancy of Mesopores (%)	Molecular Formula	MW	Total mmoles H ⁺ in 0.049 g sample
0	Zr ₆ O ₄ (μ ₃ -OH) ₄ (OH) ₄ (H ₂ O) ₄ (C ₄₄ H ₂₂ O ₈) ₂	2176.76	0.360
20	[Zr ₆ O ₄ (OH) ₄ (OH) ₄ (H ₂ O) ₄ (C ₄₄ H ₂₂ O ₈) ₂] _{0.80} [(Zr ₆ O ₄ (OH) ₄) ₂ (C ₄₄ H ₂₂ O ₈) ₆] _{0.20}	2827.54	0.252
25	[Zr ₆ O ₄ (OH) ₄ (OH) ₄ (H ₂ O) ₄ (C ₄₄ H ₂₂ O ₈) ₂] _{0.75} [(Zr ₆ O ₄ (OH) ₄) ₂ (C ₄₄ H ₂₂ O ₈) ₆] _{0.25}	2990.23	0.229
30	[Zr ₆ O ₄ (OH) ₄ (OH) ₄ (H ₂ O) ₄ (C ₄₄ H ₂₂ O ₈) ₂] _{0.70} [(Zr ₆ O ₄ (OH) ₄) ₂ (C ₄₄ H ₂₂ O ₈) ₆] _{0.30}	2949.12	0.211

The first derivative curve of NU-1000 can be fit to three or four Lorentzian peaks to give three or four equivalence points, respectively (Fig. S30). Adding a fourth peak to the fitting does not significantly shift the peak locations of peaks 0–3 (see above). We expect that NU-1000 likely has only three real pK_a values (calculated pK_{a4} = 9.43 ± 0.09), so we calculated defects based on the position of peak 2 (EP3); however, it should be noted that using EP4 in the calculation does not significantly change the occupancy of the mesopores with an estimated value of 20% instead of 25%.⁶

Zr-UiO-67(HCl)

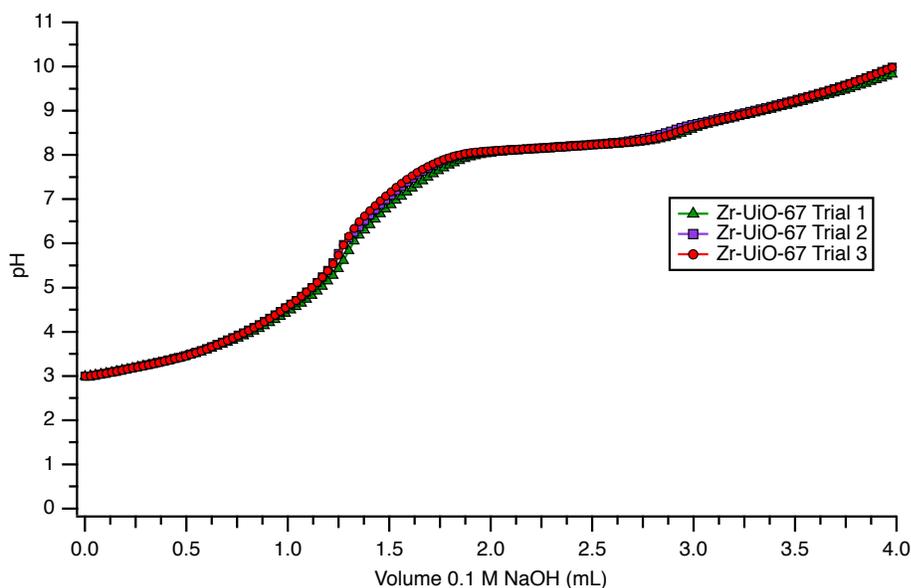


Figure S37. Overlay of three acid–base titration curves for Zr-UiO-67(HCl) showing reproducibility between samples.

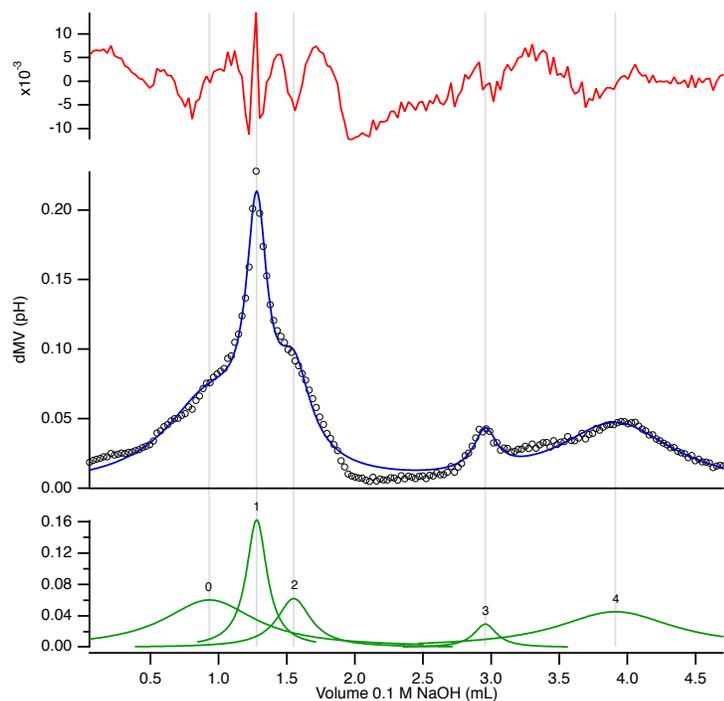


Figure S38. Representative curve fitting for the first derivative of the titration curve of pH as a function of volume of titrant added for Zr-UiO-67(HCl): Located peaks (green) (bottom); first derivative data (black circles) and fit trace (blue) (middle); residual plot (red) (top). Peaks were fit as Lorentzian functions. Peak 0 was used for fitting purposes only; calculations were done using Peak 1 as the first equivalence point.

Multipeak fit report for Zr-UiO-67(HCl).
 Chi square: 0.0041681
 Total fitted points: 180
 Multi-peak fit version 2.16
 Total Peak Area = 0.23759 +/- 0.0080377

Baseline Type: Constant

y0 = 0 +/- 0

Peak 0 Type: Lorentzian

Location = 0.93205 +/- 0.024521
 Height = 0.060447 +/- 0.0021793
 Area = 0.077706 +/- 0.0058788
 FWHM = 0.8184 +/- 0.052022

Fit function parameters

Location = 0.93205 +/- 0.024521
 FWHM = 0.8184 +/- 0.052022
 Area = 0.077706 +/- 0.0058788

Peak 1 Type: Lorentzian

Location = 1.2785 +/- 0.0022141
 Height = 0.16231 +/- 0.0042786
 Area = 0.046839 +/- 0.0030097
 FWHM = 0.18372 +/- 0.0096725

Fit function parameters

Location = 1.2785 +/- 0.0022141
 FWHM = 0.18372 +/- 0.0096725
 Area = 0.046839 +/- 0.0030097

Peak 2 Type: Lorentzian

Location = 1.5512 +/- 0.008467
 Height = 0.062172 +/- 0.0029672
 Area = 0.029369 +/- 0.002985
 FWHM = 0.30073 +/- 0.027024

Fit function parameters

Location = 1.5512 +/- 0.008467
 FWHM = 0.30073 +/- 0.027024

Area =0.029369 +/- 0.002985

Peak 3 Type: Lorentzian

Location = 2.9567 +/- 0.0099685

Height = 0.029641 +/- 0.0028731

Area = 0.0095518 +/- 0.0012787

FWHM = 0.20515 +/- 0.032995

Fit function parameters

Location = 2.9567 +/- 0.0099685

FWHM = 0.20515 +/- 0.032995

Area =0.0095518 +/- 0.0012787

Peak 4 Type: Lorentzian

Location = 3.9125 +/- 0.015982

Height = 0.045507 +/- 0.0013322

Area = 0.074124 +/- 0.003231

FWHM = 1.0369 +/- 0.059319

Fit function parameters

Location = 3.9125 +/- 0.015982

FWHM = 1.0369 +/- 0.059319

Area =0.074124 +/- 0.003231

Table S8. Titrant consumed after the first equivalence point for Zr-UiO-67(HCl)

	mmoles OH ⁻ consumed between EP1 to EP4
Trial 1	0.268
Trial 2	0.270
Trial 3	0.263
Average	0.266 (± 0.004)

Table S9. Defect calculations for Zr-UiO-67(HCl)

Missing Linkers	Molecular Formula ^a	MW	mmoles H ⁺ from defects in 0.049 g sample
1	Zr ₆ O ₄ (OH) ₄ (C ₁₄ H ₈ O ₄) ₅ [(H ₂ O)(OH)] ₂	1950.51	0.151
2	Zr ₆ O ₄ (OH) ₄ (C ₁₄ H ₈ O ₄) ₄ [(H ₂ O)(OH)] ₄	1780.33	0.330
1.60	Zr ₆ O ₄ (OH) ₄ (C ₁₄ H ₈ O ₄) _{4.40} [(H ₂ O)(OH)] _{3.20}	1848.40	0.254
1.65	Zr ₆ O ₄ (OH) ₄ (C ₁₄ H ₈ O ₄) _{4.35} [(H ₂ O)(OH)] _{3.30}	1839.89	0.264
1.70	Zr ₆ O ₄ (OH) ₄ (C ₁₄ H ₈ O ₄) _{4.30} [(H ₂ O)(OH)] _{3.40}	1831.38	0.273
1.75	Zr ₆ O ₄ (OH) ₄ (C ₁₄ H ₈ O ₄) _{4.25} [(H ₂ O)(OH)] _{3.50}	1822.88	0.282
1.80	Zr ₆ O ₄ (OH) ₄ (C ₁₄ H ₈ O ₄) _{4.20} [(H ₂ O)(OH)] _{3.60}	1814.37	0.292

^aAll missing linkers are assumed to be replaced with 2 H₂O molecules and 2 –OH groups.

Note that if the third equivalence point is used for the missing linker calculation for Zr-UiO-67(HCl) instead of the fourth apparent equivalence point, 0.168 \pm 0.002 mmoles of OH⁻ is consumed, which corresponds to approximately 1.10 missing linkers (0.167 mmoles H⁺ from defects in 0.049 g sample).

MOF-808

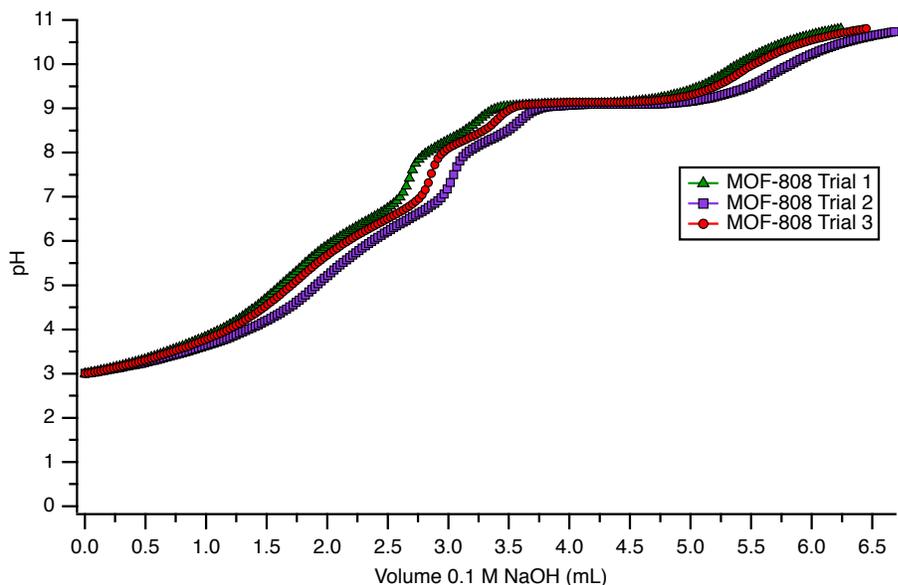


Figure S39. Overlay of three acid–base titration curves for activated MOF-808 showing reproducibility between samples.

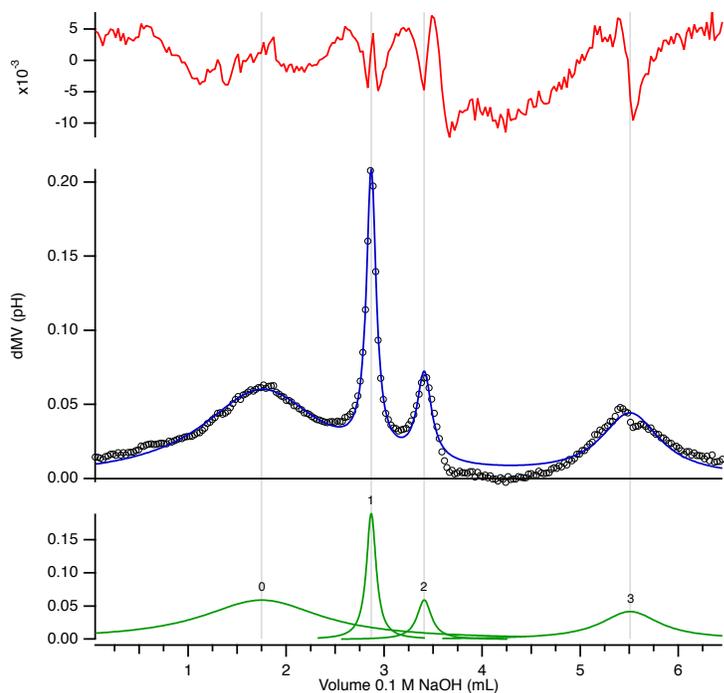


Figure S40. Representative curve fitting for the first derivative of the titration curve of pH as a function of volume of titrant added for MOF-808: Located peaks (green) (bottom); first derivative data (black circles) and fit trace (blue) (middle); residual plot (red) (top). Peaks were fit as Lorentzian functions.

Multipeak fit report for MOF-808
Chi square: 0.0054865
Total fitted points: 247
Multi-peak fit version 2.16
Total Peak Area = 0.23749 +/- 0.0038647

Baseline Type: Constant

y0 = 0 +/- 0

Peak 0 Type: Lorentzian

Location = 1.7495 +/- 0.013917
Height = 0.059156 +/- 0.0010558
Area = 0.13776 +/- 0.0031872
FWHM = 1.4825 +/- 0.046555

Fit function parameters

Location = 1.7495 +/- 0.013917
FWHM = 1.4825 +/- 0.046555
Area = 0.13776 +/- 0.0031872

Peak 1 Type: Lorentzian

Location = 2.8676 +/- 0.0011336
Height = 0.18951 +/- 0.0036161
Area = 0.035379 +/- 0.00081817
FWHM = 0.11885 +/- 0.0035296

Fit function parameters

Location = 2.8676 +/- 0.0011336
FWHM = 0.11885 +/- 0.0035296
Area = 0.035379 +/- 0.00081817

Peak 2 Type: Lorentzian

Location = 3.4083 +/- 0.0044165
Height = 0.059309 +/- 0.0029664
Area = 0.016479 +/- 0.00092191
FWHM = 0.17688 +/- 0.013238

Fit function parameters

Location = 3.4083 +/- 0.0044165
FWHM = 0.17688 +/- 0.013238

Area =0.016479 +/- 0.00092191

Peak 3 Type: Lorentzian

Location = 5.5089 +/- 0.012687
Height = 0.041943 +/- 0.001473
Area = 0.047867 +/- 0.0018049
FWHM = 0.72653 +/- 0.03812

Fit function parameters

Location = 5.5089 +/- 0.012687
FWHM = 0.72653 +/- 0.03812
Area =0.047867 +/- 0.0018049

Table S10. Total moles titrant consumed for MOF-808

	mmoles OH ⁻ consumed
Trial 1	0.541
Trial 2	0.580
Trial 3	0.551
Average	0.557 (±0.020)

Table S11. Proton quantification calculation for MOF-808

Molecular Formula	MW	Total mmoles H ⁺ expected in 0.049 g sample
Zr ₆ O ₄ (μ ₃ -OH) ₄ (OH) ₆ (H ₂ O) ₆ (C ₉ H ₃ O ₆) ₂	1303.73	0.827

$$(0.557/0.827) \times 100\% = 67\% \text{ of mmoles H}^+ \text{ expected}$$

References

1. N. Planas, J. E. Mondloch, S. Tussupbayev, J. Borycz, L. Gagliardi, J. T. Hupp, O. K. Farha and C. J. Cramer, *J. Phys. Chem. Lett.*, 2014, **5**, 3716-3723.
2. M. J. Katz, Z. J. Brown, Y. J. Colon, P. W. Siu, K. A. Scheidt, R. Q. Snurr, J. T. Hupp and O. K. Farha, *Chem. Commun.*, 2013, **49**, 9449-9451.
3. H.-L. Jiang, D. Feng, T.-F. Liu, J.-R. Li and H.-C. Zhou, *J. Am. Chem. Soc.*, 2012, **134**, 14690-14693.
4. S. Jakobsen, D. Gianolio, D. S. Wragg, M. H. Nilsen, H. Emerich, S. Bordiga, C. Lamberti, U. Olsbye, M. Tilset and K. P. Lillerud, *Phys. Rev. B*, 2012, **86**, 125429.
5. G. C. Shearer, S. Chavan, J. Ethiraj, J. G. Vitillo, S. Svelle, U. Olsbye, C. Lamberti, S. Bordiga and K. P. Lillerud, *Chem. Mater.*, 2014, **26**, 4068-4071.
6. J. E. Mondloch, W. Bury, D. Fairen-Jimenez, S. Kwon, E. J. DeMarco, M. H. Weston, A. A. Sarjeant, S. T. Nguyen, P. C. Stair, R. Q. Snurr, O. K. Farha and J. T. Hupp, *J. Am. Chem. Soc.*, 2013, **135**, 10294-10297.