Supportive information

A highly stable CuS and CuS--Pt modified Cu₂O/CuO heterostructure as an efficient photocathode for the hydrogen evolution reaction

Amare Aregahegn Dubale, ¹ Andebet Gedamu Tamirat, ¹ Hung-Ming Chen, ¹ Taame Abraha Berhe, ² Chun-Jern Pan, ¹ Wei-Nien Su^{2,*} and Bing-Joe Hwang^{1,3*}

^{*} Corresponding authors: bjh@mail.ntust.edu.tw

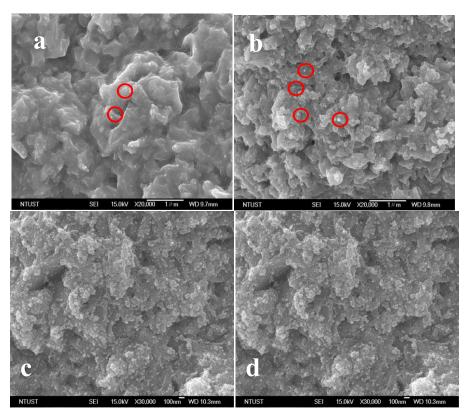


Figure S1. Scanning electron microscopy images of 3 (a) and 6 (b). 12 (c) and 15 (d) times SILAR of CuS on Cu₂O/CuO.

¹ NanoElectrochemistry Laboratory, Department of Chemical Engineering, National Taiwan University of Science and Technology, Taipei 106, Taiwan.

² NanoElectrochemistry Laboratory, Graduate Institute of Applied Science and Technology, National Taiwan University of Science and Technology, Taipei 106, Taiwan.

³ National Synchrotron Radiation Research Center, Hsinchu 30076, Taiwan.

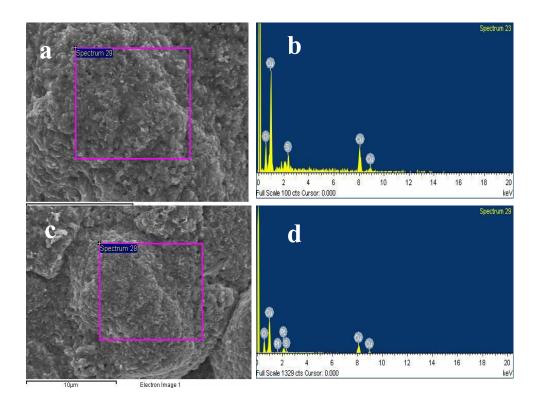
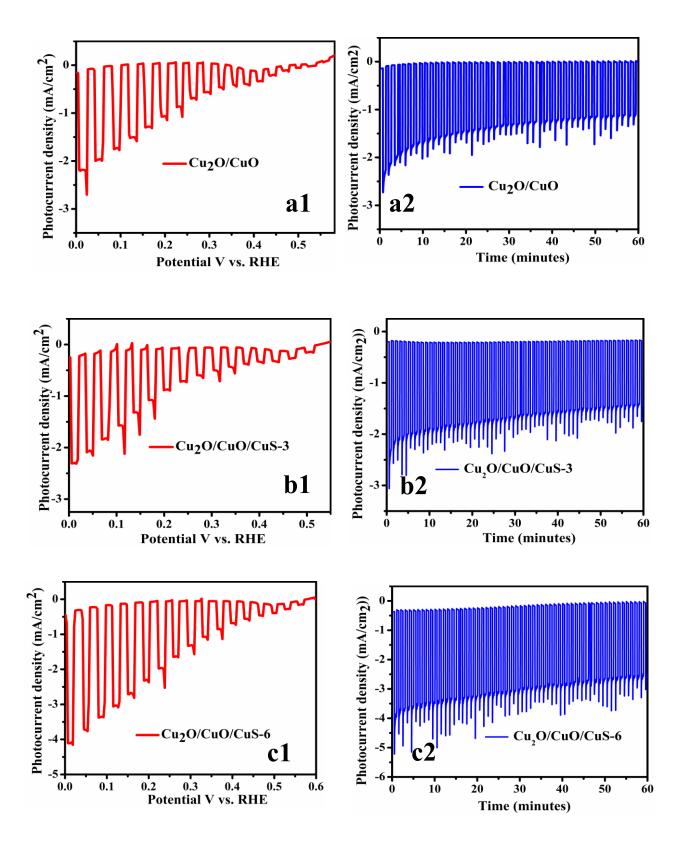


Figure S2. SEM images (a and c) and Typical EDX analysis (b and d) acquired from the $Cu_2O/CuO/CuS-9$ and $Cu_2O/CuO/CuS-9/Pt$ photocathode film. It can be seen that a weak S peak was detected together with Cu and O from the $Cu_2O/CuO/CuS-9$ while both S and Pt was detected from the $Cu_2O/CuO/CuS-9/Pt$.



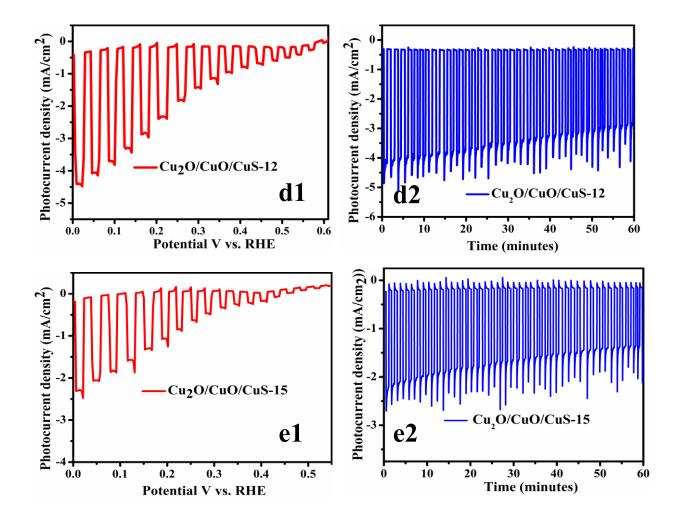


Figure S3. PEC performances and stability measurement of bare Cu_2O/CuO (a1 and a2) $Cu_2O/CuO/CuS-3$ (b1 and b2), $Cu_2O/CuO/CuS-6$ (c1 and c2), $Cu_2O/CuO/CuS-12$ (d1 and d2) and $Cu_2O/CuO/CuS-15$ (e1 and e2).

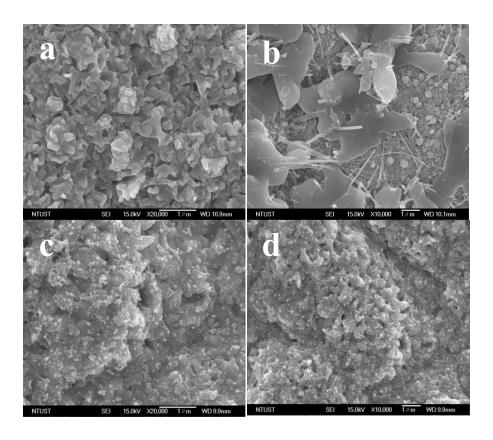


Figure S4 FE SEM images of Cu_2O/CuO and $Cu_2O/CuO/CuS-9$ before and after 20 minute photo-stability measurement under illumination of AM 1.5 G (a) Cu_2O/CuO before PEC (b) Cu_2O/CuO after PEC (c) $Cu_2O/CuO/CuS-9$ before PEC and (d) $Cu_2O/CuO/CuS-9$ after PEC.