Electronic Supplementary Material (ESI) for Journal of Materials Chemistry A. This journal is © The Royal Society of Chemistry 2016

## **Electronic Supplementary Information**

N-type cathode interlayer based on dicyanomethylenated quinacridone derivative for high-performance polymer solar

cells

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Fig. S1. Normalized UV-vis absorption spectra of QA-PyBr and DCNQA-PyBr in dilute solutions ( $1 \times 10^{-5}$  M in DMF) and in thin films.



Fig. S2. Cyclic voltammograms of QA-PyBr and DCNQA-PyBr measured in DMF.



Fig. S3. TGA curves of QA-PyBr and DCNQA-PyBr.



**Fig. S4.** *J-V* curves of the QA-PyBr, DCNQA-PyBr incorporated, methanol-treated, and Al-only devices under dark.



Fig. S5.  $J^{1/2}$ -V characteristics of the electron-only devices.



**Fig. S6.** The absorption spectra of (a) QA-PyBr films and (b) DCNQA-PyBr films with different thickness on the quartz substrates. The thickness of 20 nm for QA-PyBr film and 40 nm for DCNQA-PyBr film were determined by the profilometer.



**Fig. S7.** *J-V* characteristics of the devices with various thicknesses of QA-PyBr (a) and DCNQA-PyBr (b) interlayer under dark.



**Fig. S8.** Atomic force microscopy (AFM) images  $(5 \times 5 \ \mu\text{m})$  of with various thicknesses QA-PyBr interlayers atop of the PCDTBT:PC<sub>71</sub>BM layer. (a) 4 nm, RMS: 0.57 nm; (b) 10 nm, RMS: 0.34 nm; (c) 14 nm, RMS: 0.46 nm; (d) 20 nm, RMS: 0.30 nm.

 Table S1. Photophysical data, cyclic voltammetric data and HOMO/LUMO energy levels.

	$\lambda_{ m onset}$	$E_g^{ m opt}$	$E_{re}^{onset}$	LUMO	НОМО
	$(nm)^a$	$(eV)^b$	$(\mathbf{V})^{c}$	$(eV)^d$	(eV) <sup>e</sup>
QA-PyBr	553	2.24	-1.58	-3.22	-5.46
DCNQA-PyBr	685	1.81	-1.35	-3.45	-5.26

<sup>*a*</sup> The lowest-energy absorption edge of the absorption spectra. <sup>*b*</sup> Optical band gap estimated from the lowest-energy absorption edge. <sup>*c*</sup> Onset reduction potential (*vs*.ferrocene/ferrocenium) measured in DMF. <sup>*d*</sup> Deduced from the onset reduction potentials, assuming that the energy level of ferrocene lies 4.8 eV below the vacuum level.  $E_{LUMO} = -(E_{re}^{onset} + 4.80) \text{ eV}$ . <sup>*e*</sup> Calculated from LUMO and  $E_g^{opt}$ .  $E_{HOMO} = (E_{HOMO} - E_g^{opt}) \text{ eV}$ .

**Table S2.** The measured  $J_{sc}$  values (from the *J-V* curves) and calculated  $J_{sc}$  values (from the EQE spectra) of the PCDTBT:PC<sub>71</sub>BM devices under 100 mW cm<sup>-2</sup> AM 1.5G irradiation.

	none	LiF	methanol	QA-PyBr	DCNQA-PyBr
Measured $J_{\rm sc}$ (mA cm <sup>-2</sup> )	10.24	10.64	10.30	11.19	11.33
Calculated $J_{\rm sc}$ (mA cm <sup>-2</sup> )	10.07	10.51	10.12	10.80	11.00

**Table S3.** The measured  $J_{sc}$  values (from the *J-V* curves) and calculated  $J_{sc}$  values (from the EQE spectra) of the P3HT:PC<sub>61</sub>BM devices under 100 mW cm<sup>-2</sup> AM 1.5G irradiation.

	none	LiF	DCNQA-PyBr
Measured $J_{\rm sc}$ (mA cm <sup>-2</sup> )	9.12	10.13	10.25
Calculated $J_{\rm sc}$ (mA cm <sup>-2</sup> )	9.06	10.00	10.15

**Table S4.** The measured  $J_{sc}$  values (from the *J-V* curves) and calculated  $J_{sc}$  values (from the EQE spectra) of the PTBT:PC<sub>71</sub>BM devices under 100 mW cm<sup>-2</sup> AM 1.5G irradiation.

	none	LiF	DCNQA-PyBr
Measured $J_{\rm sc}$ (mA cm <sup>-2</sup> )	14.62	15.19	16.51
Calculated $J_{\rm sc}$ (mA cm <sup>-2</sup> )	14.48	15.07	16.20