

Supporting Information:

Hybridization of graphene nanosheets and carbon-coated hollow Fe₃O₄ nanoparticles as a high-performance anode material for lithium-ion batteries

Yongtao Zuo^a, Gang Wang^{a,*}, Jun Peng^a, Gang Li^a, Yanqing Ma^{a,*}, Feng Yu^a, Bin Dai^a, Xuhong Guo^{a,b}, Ching-Ping Wong^{c,d}

^a School of Chemistry and Chemical Engineering, Key Laboratory of Materials-Oriented Chemical Engineering of Xinjiang Uygur Autonomous Region, Shihezi University, Shihezi, P.R. China. Email: gwangshzu@163.com, mayanqing@shzu.edu.cn.

^b State Key Laboratory of Chemical Engineering, East China University of Science and Technology, Shanghai 200237, P. R. China

^c Department of Electronic Engineering, The Chinese University of Hong Kong, New Territories, HongKong

^d School of Materials Science and Engineering, Georgia Institute of Technology, Atlanta, GA30332, United States

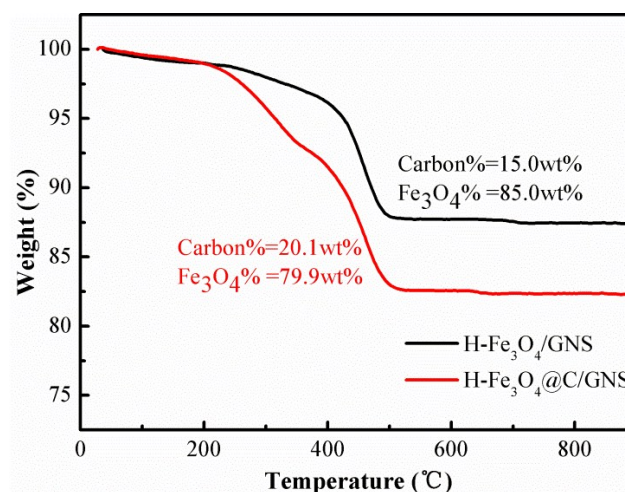


Figure S1. TGA profiles of H-Fe₃O₄/GNS nanosheets and H-Fe₃O₄@C/GNS nanosheets.

In general, a weight increase should be observed between 150°C and 400°C because of the oxidation of Fe₃O₄ to Fe₂O₃ as reported in the literatures (Yoon et al, J. Mater. Chem. 2011, 21, 17325-17330; Zhu et al, J. Phys. Chem. C 2011, 115, 9814-9820; Wang et al, RSC Adv. 2014, 4, 322-330; Dong et al, Phys. Chem. Chem. Phys. 2013, 15, 7174-7181). However, there is no sign of oxidation of Fe₃O₄ in the TGA profile in Figure S1. It may be attributed to the evaporation of the adsorbed solvents (the boil points of ethylene glycol and diethylene alcohol are 197.3°C and 245°C, respectively). This phenomenon is also observed in many Fe₃O₄/graphene composites (Li et al, J. Phys. Chem. C 2011, 115, 21567-21573; Chen et al, Adv. Mater. 2011, 23, 5679-5683; Liu et al, RSC Adv. 2014, 4, 17653-17659; Zhuo et al, J. Mater. Chem. A 2013, 1, 3954-3960; Li et al, Adv. Energy Mater. DOI: 10.1002/aenm.201500171; Fan et al, Chem. Commun. 2015, 51, 1597-1600; Guo et al, J. Mater. Chem. A DOI: 10.1039/C5TA05431A).

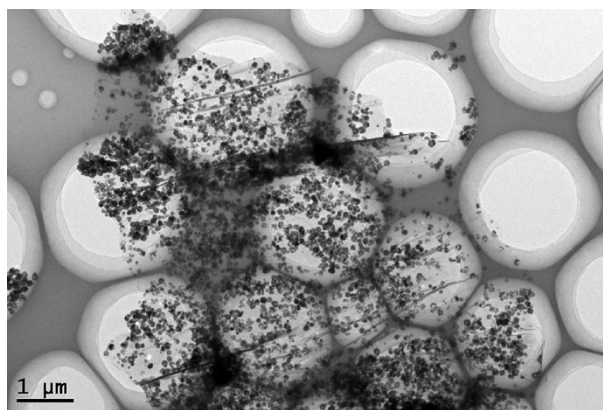


Figure S2. Low magnification TEM images of the as-prepared H-Fe₃O₄@C/GNS nanosheets.

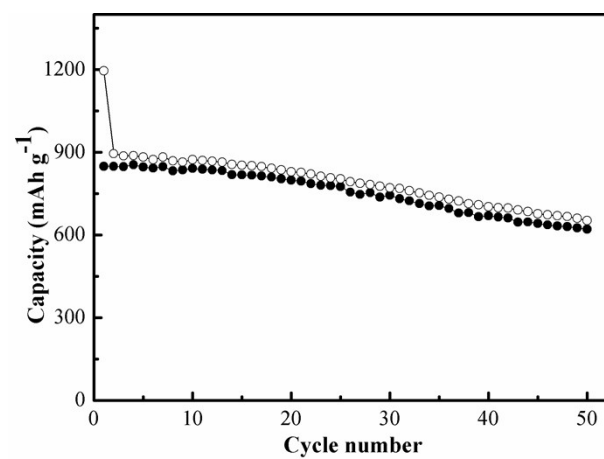


Figure S3 Cycling performance of carbon-encapsulating H-Fe₃O₄ (H-Fe₃O₄@C).

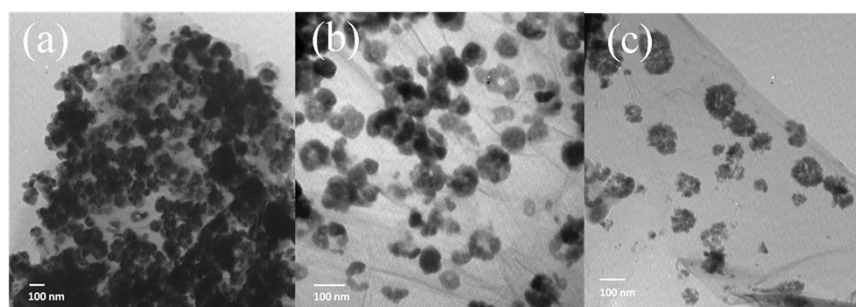


Figure S4 TEM images of H-Fe₃O₄@C/GNS hybrids with different adding amount of GO during the solvothermal process. (a) 50mg; (b) 100mg; (c) 150mg.

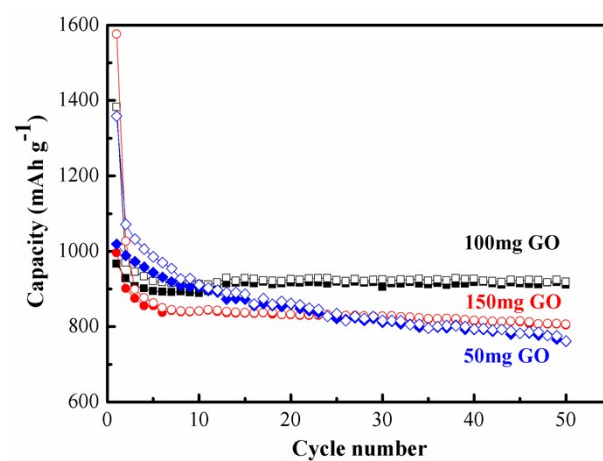


Figure S5 Cycling performance of H-Fe₃O₄@C/GNS hybrids with different adding amount of GO during the solvothermal process. (a) 50mg; (b) 100mg; (c) 150mg.

Table S1 Cycling performance of Fe₃O₄/carbon hybrids and other Fe-based materials as anodes for LIBs

Materials	Cycling Performance	Rate Performance	Ref.
C-Fe ₃ O ₄ nanospheres	712 mAh g ⁻¹ at 200 mA g ⁻¹ (60 cycles)	/	7
magnetite-C nanocomposite	776.7 mAh g ⁻¹ at 100 mA g ⁻¹ (30 cycles)	/	15
Fe ₃ O ₄ /CF composite	784 mAh g ⁻¹ at 100 mA g ⁻¹ (50 cycles)	570 mA h g ⁻¹ at 0.5C ~140 mA h g ⁻¹ at 1~2C	16
Fe ₃ O ₄ /MSU-F-C nanocomposite	~1118 mAh g ⁻¹ at 100 mA g ⁻¹ (150 cycles)	750 mA h g ⁻¹ at 1C ~500 mA h g ⁻¹ at 5C	17
Carbon-coated Fe ₃ O ₄ nanospindles	530 mAh g ⁻¹ at 0.5C (80 cycles)	190 mA h g ⁻¹ at 5C	18
carbon-Fe ₃ O ₄ nanocomposite	~1000 mAh g ⁻¹ at 100 mA g ⁻¹ (40 cycles)	418 mA h g ⁻¹ at 2500 mA g ⁻¹	19
graphene paper@Fe ₃ O ₄ nanorod array@graphene	1211 mAh g ⁻¹ at 0.2A g ⁻¹ (350 cycles)	728 mA h g ⁻¹ at 5 A g ⁻¹	20
	759 mAh g ⁻¹ at 2A g ⁻¹ (1450 cycles)		
Fe ₃ O ₄ @C	808.2 mAh g ⁻¹ at 924 mA g ⁻¹ (100 cycles)	/	21
GNS/Fe ₃ O ₄ composite	950 mAh g ⁻¹ at 35 mA g ⁻¹ (85 cycles)	520 mA h g ⁻¹ at 1750 mA g ⁻¹	22
	~600 mAh g ⁻¹ at 700 mA g ⁻¹ (100 cycles)		
RGO/Fe _x O _y nanocomposites	868.4 mAh g ⁻¹ at 500 mA g ⁻¹ (300 cycles)	585.8 mAh g ⁻¹ at 2000 mA g ⁻¹	23
Fe ₃ O ₄ -rGO composite	1039 mAh g ⁻¹ at 100 mA g ⁻¹ (170 cycles)	396 mA h g ⁻¹ at 1000 mA g ⁻¹	24
		193 mA h g ⁻¹ at 2000 mA g ⁻¹	
hollow Fe ₃ O ₄ /graphene	940 mAh g ⁻¹ at 200 mA g ⁻¹ (50 cycles)	550 mA h g ⁻¹ at 1000 mA g ⁻¹	25
	660 mAh g ⁻¹ at 500 mA g ⁻¹ (50 cycles)	420 mA h g ⁻¹ at 2000 mA g ⁻¹	
Fe ₃ O ₄ /graphene hybrids	951 mAh g ⁻¹ at 50 mA g ⁻¹ (50 cycles)	/	26
	796 mAh g ⁻¹ at 500 mA g ⁻¹ (200 cycles)		
	531 mAh g ⁻¹ at 1A g ⁻¹ (300 cycles)		
	335 mAh g ⁻¹ at 2A g ⁻¹ (300 cycles)		
	213 mAh g ⁻¹ at 5A g ⁻¹ (300 cycles)		
Fe ₃ O ₄ NCs-GAs	577 mAh g ⁻¹ at 5.2A g ⁻¹ (300 cycles)	392 mA h g ⁻¹ at 12 C	27
		118 mA h g ⁻¹ at 35C	
G-Fe ₃ O ₄ -GNRs	708 mAh g ⁻¹ at 0.4A g ⁻¹ (300 cycles)	~600 mA h g ⁻¹ at 1 A g ⁻¹	28
Fe ₃ O ₄ nanocrystals@graphene composite	538.7 mAh g ⁻¹ at 0.2C (50 cycles)	~200 mA h g ⁻¹ at 2C	29
GF@Fe ₃ O ₄	785 mAh g ⁻¹ at 1C (500 cycles)	190 mA h g ⁻¹ at 60C	30
	~400 mAh g ⁻¹ at 6C (500 cycles)		
	300 mAh g ⁻¹ at 10C (500 cycles)		
Fe ₃ O ₄ @GS/GF	1059 mAh g ⁻¹ at 93 mA g ⁻¹ (150 cycles)	363 mAh g ⁻¹ at 4800 mA g ⁻¹	31
Fe ₃ O ₄ @GN composite	826 mAh g ⁻¹ at 1A g ⁻¹ (100 cycles)	460 mAh g ⁻¹ at 5A g ⁻¹	32
Fe ₃ O ₄ -Carbon-rGO	842.7 mAh g ⁻¹ at 0.2C (100 cycles)	~600 mAh g ⁻¹ at 1C	34
		~400 mAh g ⁻¹ at 2C	
Fe ₃ O ₄ @GNSs	1010 mAh g ⁻¹ at 100 mA g ⁻¹ (50 cycles)	300 mAh g ⁻¹ at 5C	35
GN-Fe ₃ O ₄	1102 mAh g ⁻¹ at 100 mA g ⁻¹ (10 cycles)	474 mAh g ⁻¹ at 1600 mA g ⁻¹	36
Graphene-Encapsulated Hollow Porous Fe ₃ O ₄	~900 mAh g ⁻¹ at 100 mA g ⁻¹ (50 cycles)	~580 mAh g ⁻¹ at 800 mA g ⁻¹	37
G@Fe ₃ O ₄ @C	920 mAh g ⁻¹ at 200 mA g ⁻¹ (100 cycles)	550 mAh g ⁻¹ at 5000 mA g ⁻¹	39
Fe ₃ O ₄ -graphene nanocomposites	1200 mAh g ⁻¹ at 2C (100 cycles)	/	40
	460 mAh g ⁻¹ at 4C (400 cycles)		
	180 mAh g ⁻¹ at 10C (800 cycles)		
N-G-Fe ₃ O ₄ composite	800 mAh g ⁻¹ at 100 mA g ⁻¹ (100 cycles)	763 mAh g ⁻¹ at 800 mA g ⁻¹	41
		733 mAh g ⁻¹ at 1600 mA g ⁻¹	
H-Fe ₃ O ₄ -MS/GNS composite	1171.6 mAh g ⁻¹ at 200 mA g ⁻¹ (70 cycles)	935 mAh g ⁻¹ at 1000 mA g ⁻¹	42

	940.4 mAh g ⁻¹ at 500 mA g ⁻¹ (70 cycles)	745 mAh g ⁻¹ at 3000 mA g ⁻¹	
graphene/Fe ₃ O ₄ aerogel	~990 mAh g ⁻¹ at 200 mA g ⁻¹ (30 cycles)	730 mAh g ⁻¹ at 1600 mA g ⁻¹	43
Fe ₃ O ₄ /GNSs	~605 mAh g ⁻¹ at 92.5 mA g ⁻¹ (50 cycles)	/	44
nitrogen-doped graphene/Fe ₃ O ₄	1130 mAh g ⁻¹ at 100 mA g ⁻¹ (200 cycles)	648 mAh g ⁻¹ at 1600 mA g ⁻¹	45
Fe ₃ O ₄ -graphene composite	868 mAh g ⁻¹ at 200 mA g ⁻¹ (100 cycles) 539 mAh g ⁻¹ at 1000 mA g ⁻¹ (200 cycles)	282 mAh g ⁻¹ at 2000 mA g ⁻¹	46
Fe ₃ O ₄ @C@PGC nanosheets	998 mAh g ⁻¹ at 1A g ⁻¹ (100 cycles) 792 mAh g ⁻¹ at 5C (350 cycles) 556 mAh g ⁻¹ at 10C (350 cycles)	364 mAh g ⁻¹ at 15C 311 mAh g ⁻¹ at 20C	47
Mesoporous ZnFe ₂ O ₄ Microrods	~434 mAh g ⁻¹ at 1000 mA g ⁻¹ (51 cycles) 542 mAh g ⁻¹ at 1000 mA g ⁻¹ (488 cycles)	~326 mAh g ⁻¹ at 1500 mA g ⁻¹	54
ZnO/ZnFe ₂ O ₄ @C mesoporous nanospheres	~893 mAh g ⁻¹ at 500 mA g ⁻¹ (60 cycles) ~718 mAh g ⁻¹ at 1000 mA g ⁻¹ (500 cycles)	~770 mAh g ⁻¹ at 1500 mA g ⁻¹	55
ZnO/ZnFe ₂ O ₄ submicrocubes	837 mAh g ⁻¹ at 1000 mA g ⁻¹ (200 cycles)	~667 mAh g ⁻¹ at 2000 mA g ⁻¹	56
H-Fe ₃ O ₄ @C/GNS nanosheets	887.8 mAh g ⁻¹ at 0.1C (100 cycles) 744.9 mAh g ⁻¹ at 1C (200 cycles) 284.9 mAh g ⁻¹ at 10C (200 cycles)	~560 mAh g ⁻¹ at 3C ~448 mAh g ⁻¹ at 5C ~300 mAh g ⁻¹ at 10C	This work

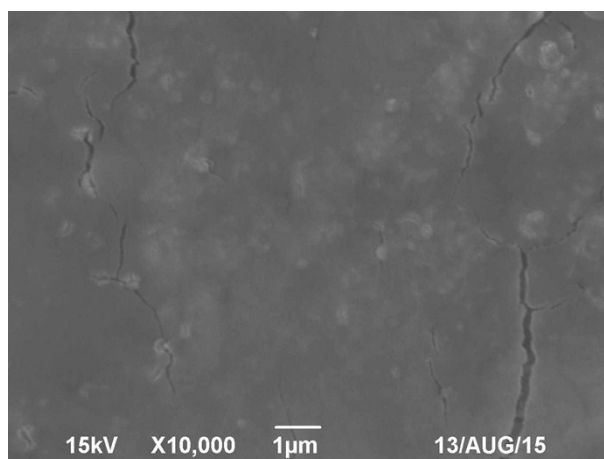


Figure S6 SEM image of the H-Fe₃O₄@C/GNS nanosheets electrodes after 200 discharge/charge cycles at a current rate of 1 C.

Table S2 Kinetic Parameters of the H-Fe₃O₄ NPs, H-Fe₃O₄/GNS nanosheets and H-Fe₃O₄@C/GNS nanosheets.

Materials	SEI film resistance (R_f)	charge-transfer resistance (R_{ct})
H-Fe ₃ O ₄ NPs	105 Ω	97.3 Ω
H-Fe ₃ O ₄ /GNS	73 Ω	28.6 Ω
H-Fe ₃ O ₄ @C/GNS	28 Ω	9.8 Ω