## **Supporting Information**

Biomass-mediated synthesis of carbon-supported nanostructured metal sulfides for ultra-high performance Lithium-ion batteries

Y. Lu and E. Fong\*



Fig. S1. SEM image of alginate after annealing (a) and alginate containing L-cysteine after annealing (b); (c - d) TEM image of sample in (a) and (b) respectively.



Fig. S2. TEM element mapping of annealed CoMoS<sub>3.13</sub>@ADC.



Fig. S3. nitrogen isothermal curve (a), pore size distribution curve (inset of a) and TGA/DTA curve (c) of  $MoS_2@ADC$ ; nitrogen isothermal curve (b), pore size distribution curve (inset of b) and TGA/DTA curve (d) of  $CoMoS_{3.13}@ADC$ .



Fig. S4 Raman spectrum of CoMoS<sub>3.13</sub>@ADC in 1000~2000 cm<sup>-1</sup>

Ref	Electrode	Initial discharge capacity	Initial charge capacity	Residual capacity	Rate capacity	Rate capacity
		(100 mA g <sup>-1</sup> )	(100 mA g <sup>-1</sup> )	(100 mA g <sup>-1</sup> )	(1A g <sup>-1</sup> )	(2A g <sup>-1</sup> )
Our material	CoMoS <sub>3.13</sub> @ADC	1439	1121	1125	800	755
1	MoS <sub>2</sub> /a-C-1.0	2100	926	961		
2	MoS <sub>2</sub> /G (1:2)	1571	1031	1100	900	
3	MoS2/N-G	1300	1021.2	1285.3	850	

Table S1. Electrochemical performance of CoMoS<sub>3.13</sub>@ADC electrode with previously reported MoS<sub>2</sub>/carbon composite anodes

## References

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