Supporting Information

## Surface Cavities of Ni(OH)<sub>2</sub> Nanowires Can Host Au Nanoparticles as Supported Catalysts with High Catalytic Activity and Stability

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Figure S1 XPS spectrum of Au 4f for Au/*m*-Ni(OH)<sub>2</sub>.



**Figure S2** TEM images of (a) Pd/*m*-Ni(OH)<sub>2</sub> and (b) Ag/*m*-Ni(OH)<sub>2</sub> hybrids.



**Figure S3** TEM image of Au/*m*-Ni(OH)<sub>2</sub> hybrids after aged for six months.



**Figure S4** UV-vis spectra showing gradual reduction of 4-NP with different catalysts in the first run: (a) Au/*m*-Ni(OH)<sub>2</sub>, (b) Au/Ni(OH)<sub>2</sub>, and (c) Au-PVP/*m*-Ni(OH)<sub>2</sub>.



Figure S5 TEM image of Au-PVP/*m*-Ni(OH)<sub>2</sub> hybrids.



Figure S6 TEM image of Au/*m*-Ni(OH)<sub>2</sub> nanocatalysts after ten runs of catalytic reactions.

Support	Noble metal	Catalyst size (nm)	<i>k</i> (min <sup>-1</sup> )	TON	TOF <sup>b</sup> (s⁻	Reference
SBA-15	Ag	7.6	0.45	4.4	0.0012	1
SBA-15	Ag	9.9	0.54	5.6	0.0157	2
SBA-15	Pd	8.0	0.72	0.10	0.0002	3
SBA-15	Au	2	0.21	0.10	0.0003	4
m-Al <sub>2</sub> O <sub>3</sub>	Ag	15	0.31	0.13	0.00003	5
<i>m</i> -Ni(OH) <sub>2</sub>	Au	2	0.49	227	0.54	this work

**Table S1** Comparison of catalytic activities of different mesoporous material-supported noble metal nanoparticles involved in the reduction of 4-nitrophenol in the presence of NaBH<sub>4</sub>.

aTON = (mole of product)/(mole of catalyst); bTOF = TON/(reaction time).

## References

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