

Supplementary information

**TiNb₂O₇/Graphene Composites as a High-Rate Anode Material for
Lithium/Sodium Ion Batteries**

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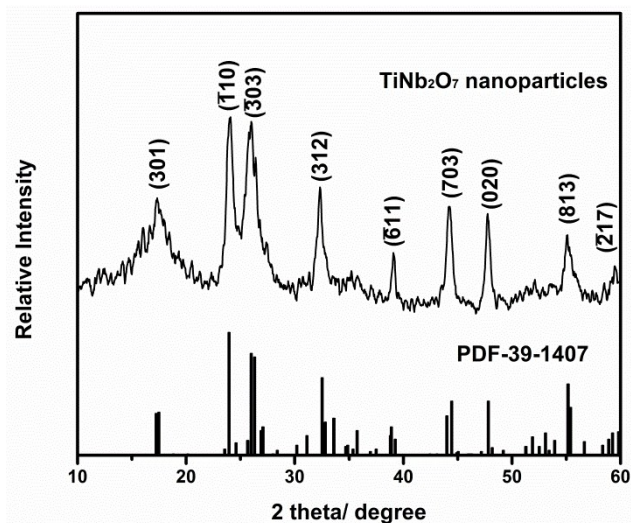


Figure S1. XRD patterns of pure TNO before freeze drying and thermal reduction. The vertical lines at the bottom correspond to the standard XRD reflections of monoclinic TiNb₂O₇.

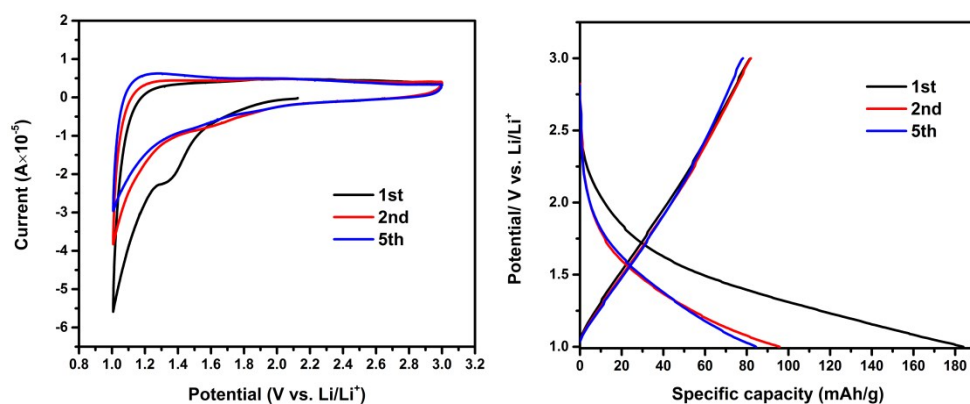


Figure S2. (a) CV curves of pure graphene electrodes at 0.2 mV s^{-1} in lithium coin cells; (b) discharge-charge profiles of pure graphene electrodes in lithium coin cells at current rate of 0.1 C .

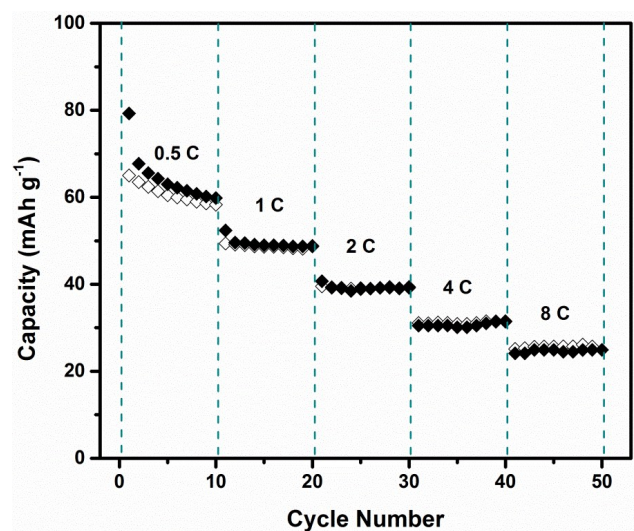


Figure S3. Rate capacities of pure graphene sheets synthesized in this work

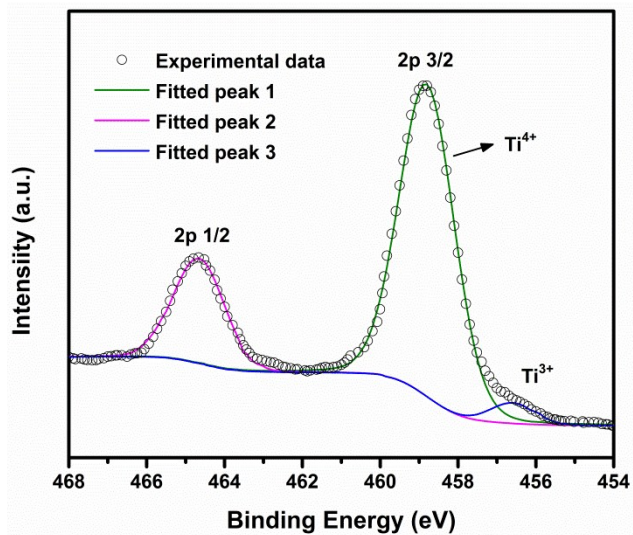


Figure S4. Ti2p XPS spectra of TNO/G composite after annealed in Ar with Ti^{3+} available.

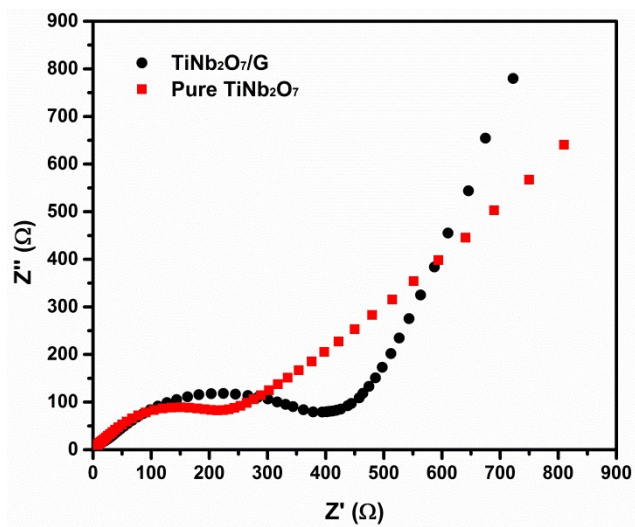


Figure S5. Nyquist plots of EIS results of TNO/G and pure TNO electrodes after 5 discharge and charge cycles.

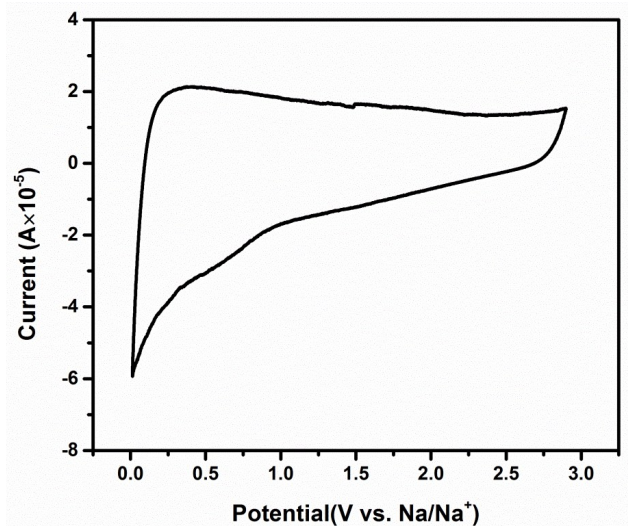


Figure S6. The 100th cycle CV curve of TNO/G composite at 0.2 mV s⁻¹.

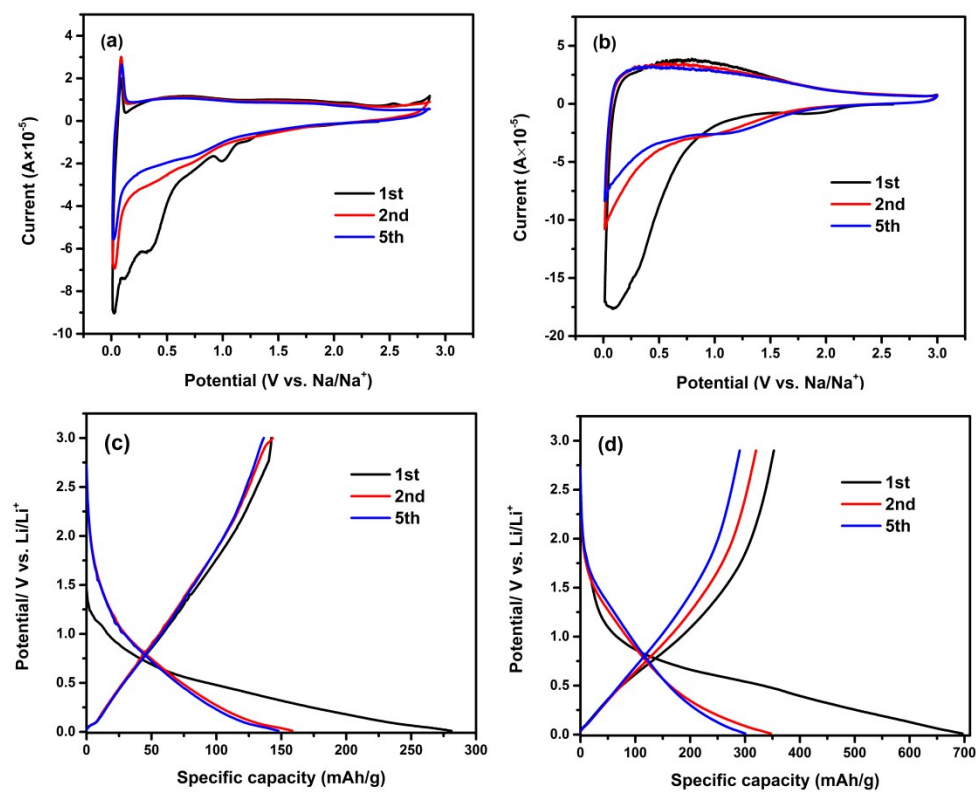


Figure S7. CV curves of (a) pure TNO and (b) pure graphene electrodes at 0.2 mV s⁻¹; discharge-charge profiles of (c) pure TNO and (d) pure graphene electrodes in sodium coin cells at current density of 25 mA g⁻¹.

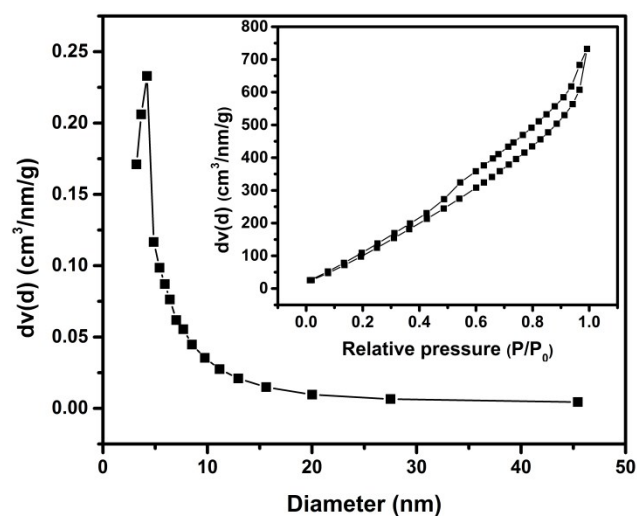


Figure S8. BJH desorption pore size distribution for the as synthesized graphene sheets (inset: N₂ adsorption-desorption isotherm).

Table S1. Specific capacity values of pure TNO (C^{TNO}), graphene (C^G) and enhanced capacities ($C^{TNO/G}_{exp} - C^{TNO/G}_{cal}$) from Figure 6.

Capacity Values (mA h g ⁻¹)	Discharge-Charge Rate (1C=387 mA g ⁻¹)				
	0.5 C	1 C	2 C	4 C	8 C
C^{TNO}	136.5	108.7	81.2	54.9	33.4
C^G	24.3	19.5	15.7	12.5	10.2
$C^{TNO/G}_{cal}$	160.8	128.2	96.9	67.4	43.6
$C^{TNO/G}_{exp}$	201.2	186.1	170.6	154.3	133.9
$C^{TNO/G}_{exp} - C^{TNO/G}_{cal}$	40.4	57.9	73.7	86.9	90.3