

Supplementary Information

High-capacity organic cathode active materials of 2,2'-bis-*p*-benzoquinone derivatives for rechargeable batteries

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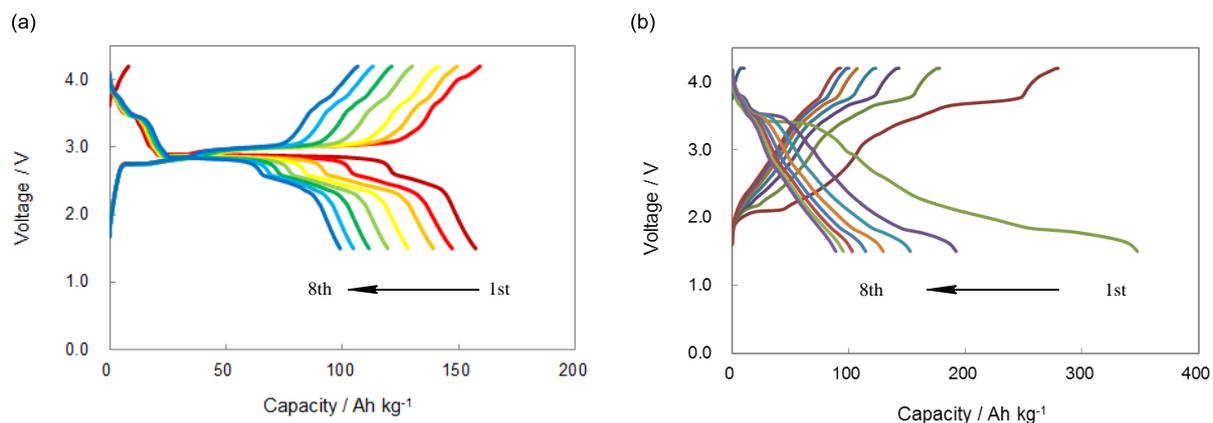


Fig. S1 Charge-discharge curves and cycle-life performance over eight cycles for (a) BQ and (b) NDQ cells in 1.0 M LiPF₆-EC/DEC.

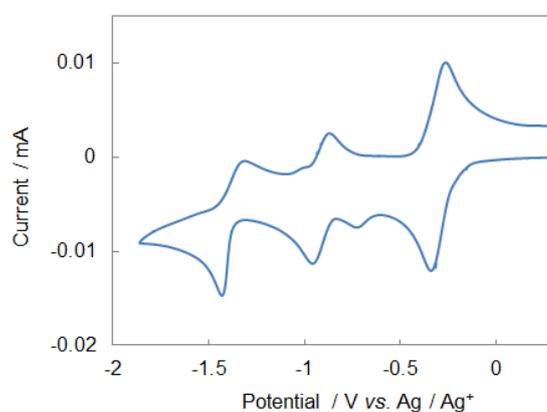


Fig. S2 CV of NDQ in acetonitrile.

Table S1 Electrochemical parameters and the energy levels of LUMO and HOMO for BQ and BBQ.

Sample	$E_{1p,c}$ in solution CV (V)	Discharge voltage (V)	LUMO level (eV)	HOMO level (eV)
BQ	-0.91	2.8	-3.947	-7.797
BBQ	-0.76	2.9	-4.251	-7.736
NDQ	-0.34	1.8-3.4	-4.645	-7.404

DFT calculations were carried out on the B3LYP/6-31+G* level.

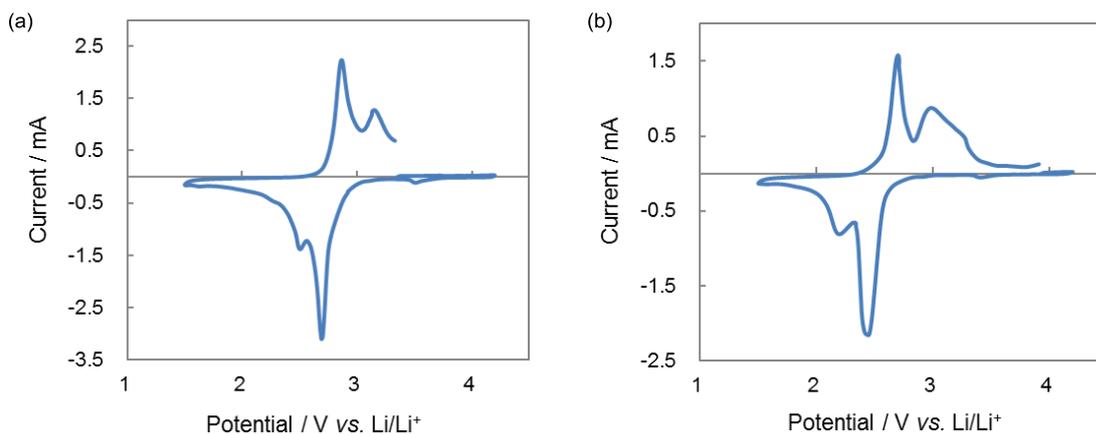


Fig. S3 Electrode CVs of (a) Br₂- and (b) Me₂-BBQ cells (1.0 M LiPF₆-EC/DEC) at initial cycle.

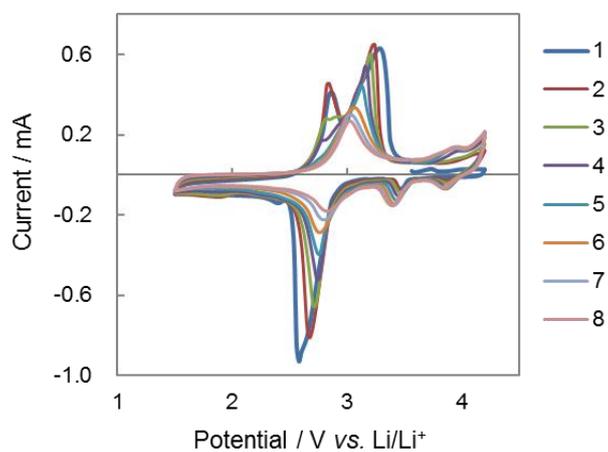


Fig. S4 Electrode CVs of BBQ cells (1.0 M LiPF₆-EC/DEC) from first to eighth cycles.

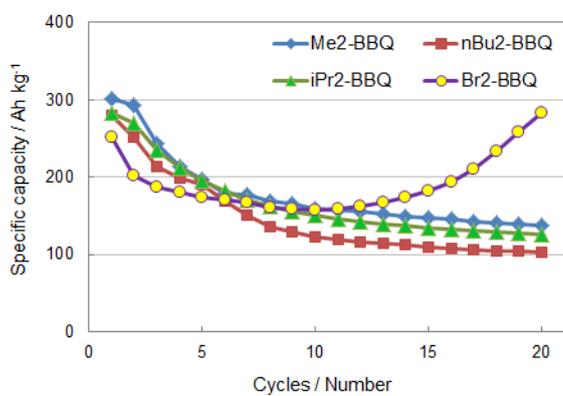


Fig. S5 Cycle-life performance over twenty cycles for cells based on R₂-BBQ (R = Br, Me, *n*Bu, and *i*Pr) in 2.75 M LiTFSI-tetraglyme systems.

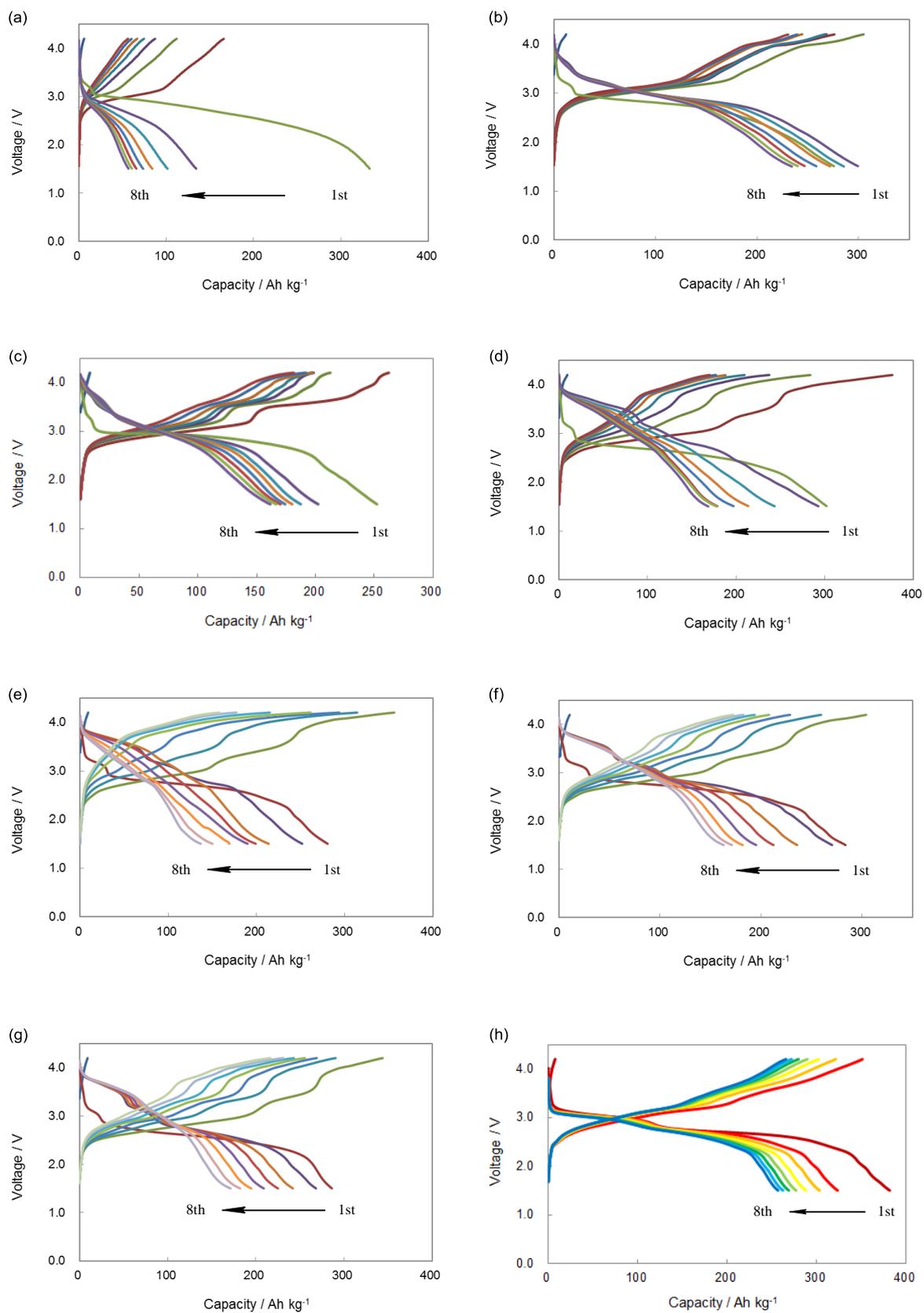


Fig. S6 Charge-discharge curves and cycle-life performance over eight cycles for (a) F_2^- , (b) Cl_2^- , (c) Br_2^- , (d) Me_2^- , (e) nBu_2^- , (f) iPr_2^- , (g) tBu_2^- , and (h) $(OMe)_2$ -BBQ cells in 2.75 M LiTFSI-tetraglyme systems.

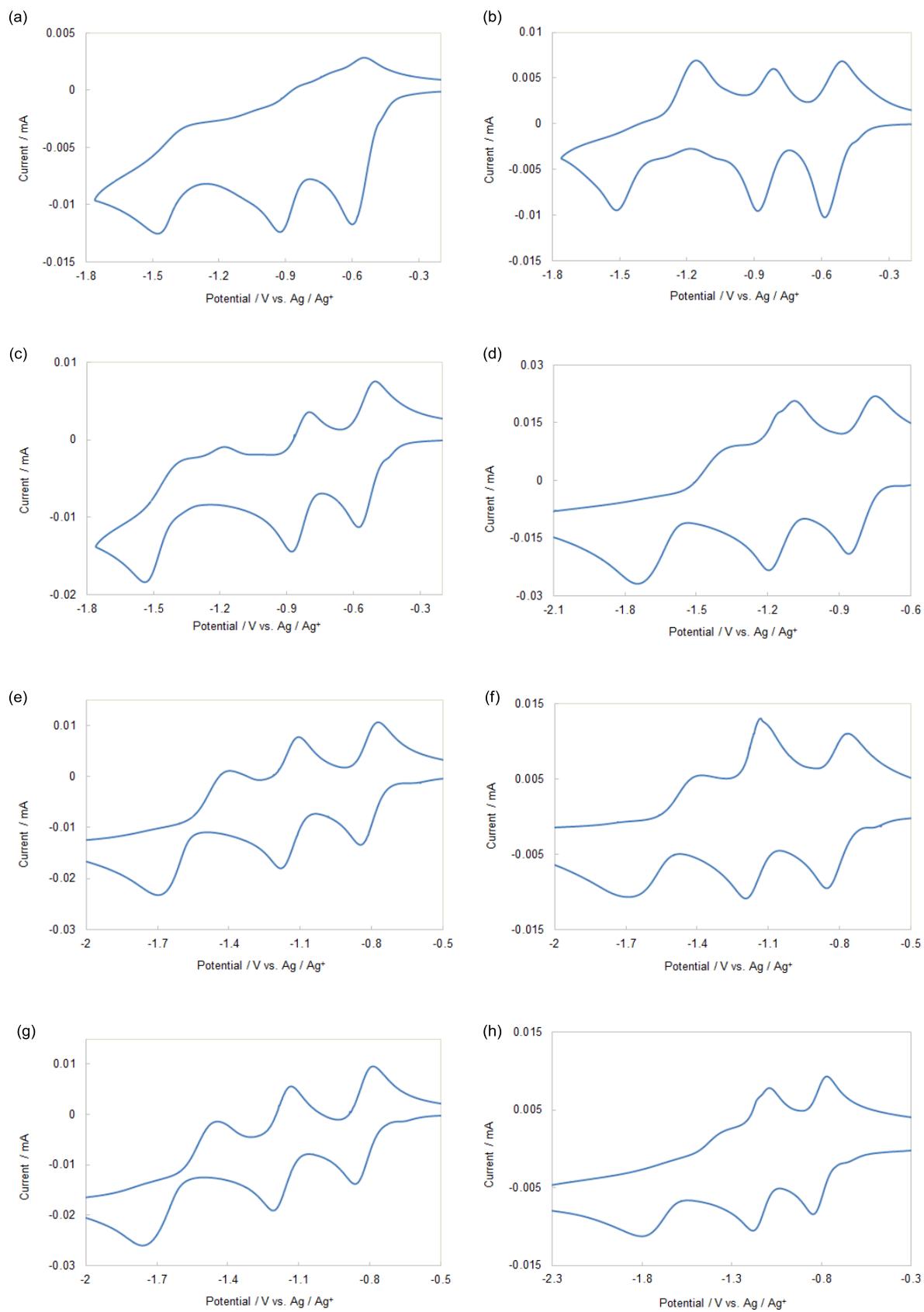


Fig. S7 CV of (a) F_2^- , (b) Cl_2^- , (c) Br_2^- , (d) Me_2^- , (e) nBu_2^- , (f) iPr_2^- , (g) tBu_2^- , and (h) $(OMe)_2-BBQ$ in acetonitrile.

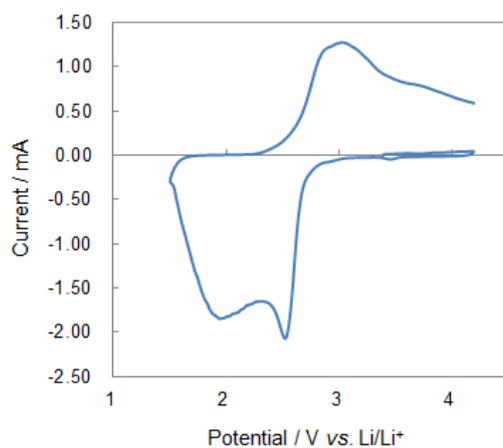


Fig. S8 Electrode CV of $(\text{OMe})_2\text{-BBQ}$ cells at initial cycle.

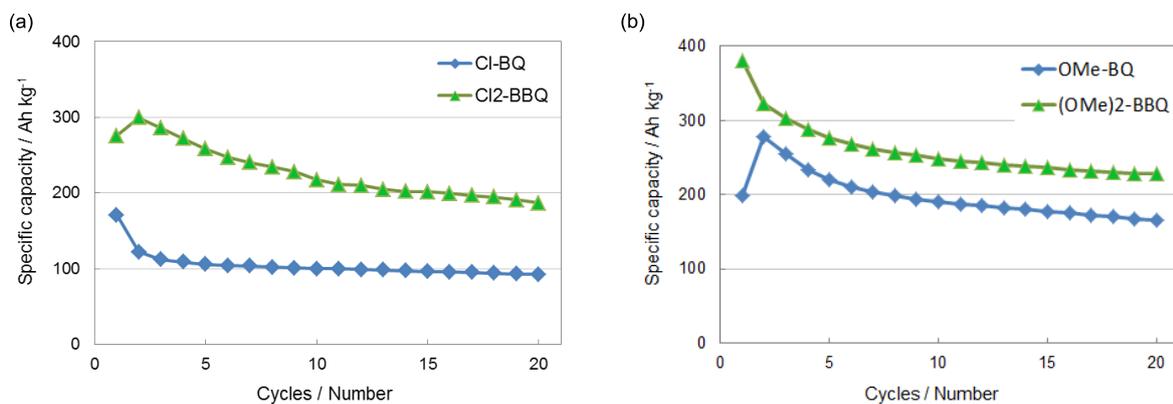
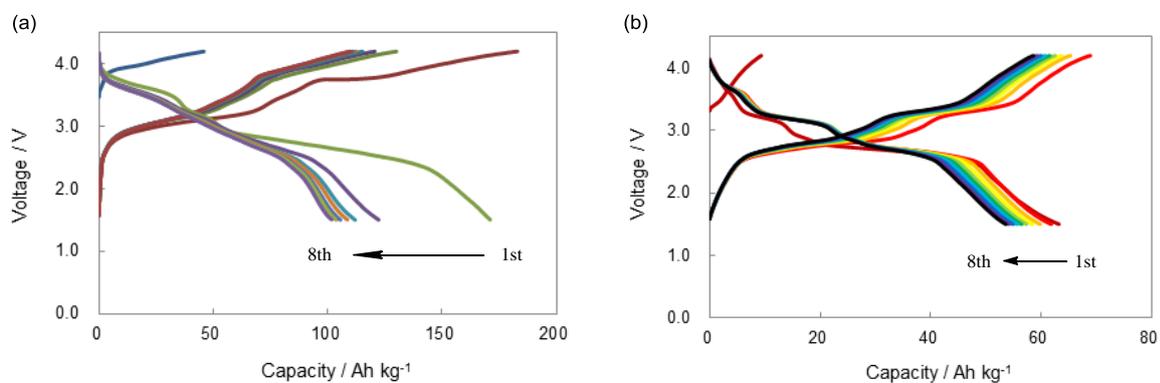


Fig. S9 Cycle-life performance over twenty cycles for cells based on (a) Cl-BQ and Cl₂-BBQ, and (b) OMe-BQ and $(\text{OMe})_2\text{-BBQ}$ in 2.75 M LiTFSI-tetraglyme systems.



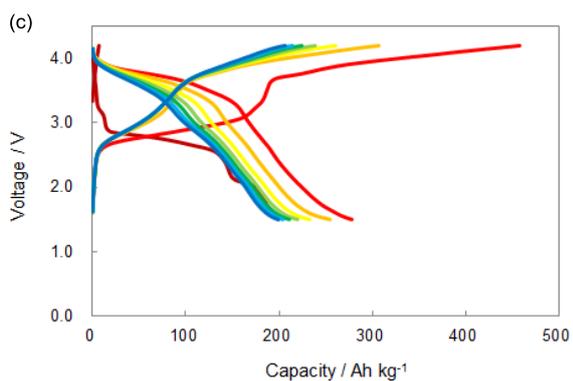


Fig. S10 Charge-discharge curves and cycle-life performance over eight cycles for (a) Cl-BQ, (b) Me-BQ, and (c) OMe-BQ in 2.75 M LiTFSI-tetraglyme systems.

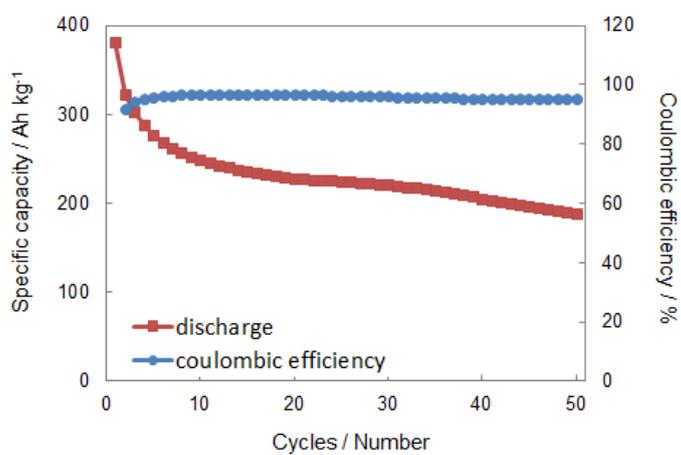


Fig. S11 Cycle-life performance over 50th cycles for cells based on $(\text{OMe})_2\text{-BBQ}$ in 2.75 M LiTFSI-tetraglyme systems.

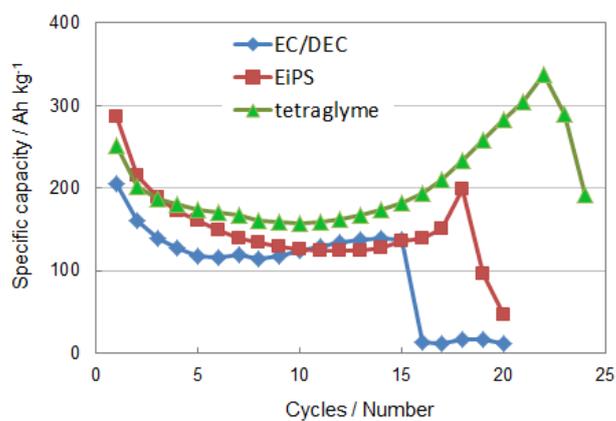


Fig. S12 Cyclability for $\text{Br}_2\text{-BBQ}$ cells in 1.0 M $\text{LiPF}_6\text{-EC/DEC}$, 1.0 M LiTFSI-EiPS, and 2.75 M LiTFSI-tetraglyme systems.

Electrolyte effects in charge-discharge curves and cycle performance

Effects of the electrolyte in the BBQ cell was examined by preparing cells with three different electrolytes: EiPS with 1.0 M $\text{LiN}(\text{SO}_2\text{CF}_3)_2$ (LiTFSI), tetraglyme with 2.75 M LiTFSI, and a mixture of EC (30 vol%) and DEC (70 vol%) with 1.0 M LiPF_6 . The cyclability for the BBQ cells in different electrolytes is shown in Fig. S12, and the charge-discharge curves over eight cycles are depicted in Fig. S13. The LiTFSI-EiPS system provided the highest initial capacity of 429 Ah kg^{-1} , which exceeded the theoretical value involving three electrons (376 Ah kg^{-1}). However, its discharge capacity significantly faded to 370 Ah kg^{-1} in the second cycle. In the LiTFSI-tetraglyme system, the cell afforded an initial capacity of 358 Ah kg^{-1} , and surprisingly increased to 368 Ah kg^{-1} in the second cycle. Although the initial capacity in the tetraglyme system was smaller than in the EiPS system, tetraglyme afforded the highest cycle retention rate (the ratios of the 20th capacity to the 1st in the tetraglyme, EiPS, and EC/DEC systems were 67, 47, and 52%, respectively), and its capacity remained at 239 Ah kg^{-1} at the 20th cycle.

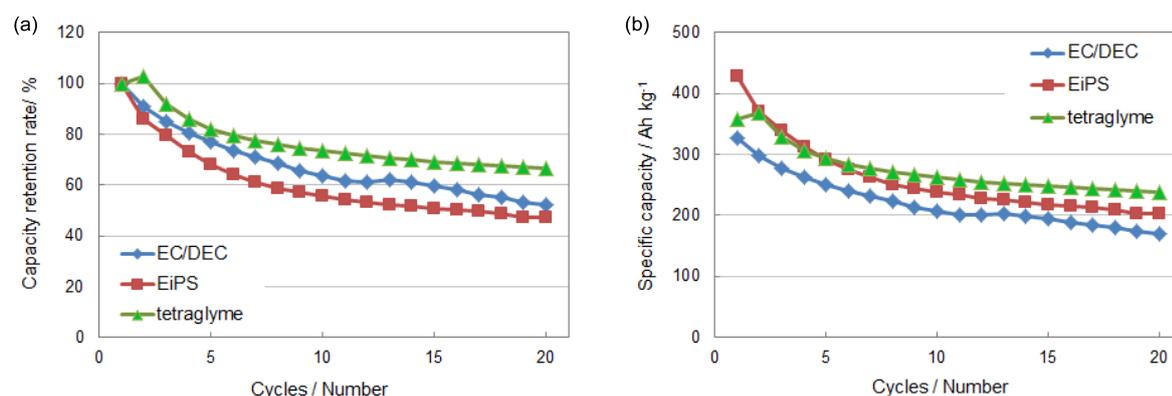


Fig. S13 Cyclability over twenty cycles for BBQ cells in 1.0 M LiPF_6 -EC/DEC, 1.0 M LiTFSI-EiPS, and 2.75 M LiTFSI-tetraglyme systems: plots of (a) capacity retention rate versus cycle number and (b) capacity retention rate versus cycle number.

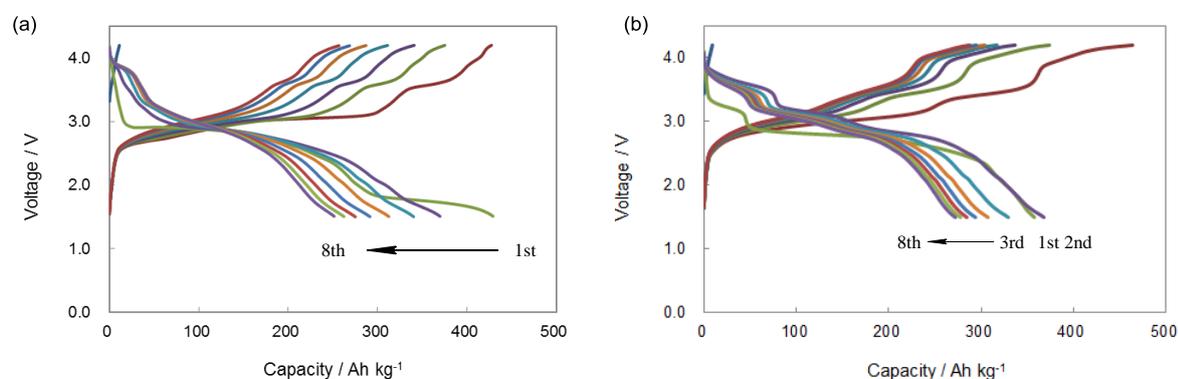


Fig. S14 Charge-discharge curves and cycle-life performance over eight cycles for BBQ cells in (a) 1.0 M LiTFSI-EiPS, and (b) 2.75 M LiTFSI-tetraglyme systems.

Experimental details of preparations for cathode active materials

Melting points were obtained with a Yanako micro melting point apparatus and are not corrected. Products were purified by flash chromatography on silica gel (Kanto Chemical Co., Inc., Silica Gel 60N (spherical, neutral), 63-210 μm). ^1H NMR spectra were recorded with JEOL-ECP-500 (500 MHz) and JEOL-ECP-400 (400 MHz) spectrometers in CDCl_3 or DMSO-d_6 . Chemical shifts were reported in parts per million (δ) referenced to the solvent peak at 7.26 or 2.49 ppm. ^{13}C NMR spectra were recorded with JEOL-ECP-500 (126 MHz) and JEOL-ECP-400 (100 MHz) spectrometers in CDCl_3 or DMSO-d_6 , and referenced to the solvent peak at 77.0 or 39.7 ppm. Coupling constants, J , were reported in Hertz (Hz), and splitting patterns were designated as s (singlet), d (doublet), t (triplet), sext (sextet), sept (septet), dd (double doublet), dt (double triplet) and m (multiplet). IR spectra were obtained on a JASCO FT/IR-4100 or a JASCO FT/IR-5300 spectrometer; absorptions were reported in reciprocal centimeters. Conventional mass spectra were recorded with SHIMADZU GCMS-QP2010Plus and JEOL MS-700 spectrometers, and high-resolution mass spectra were recorded with a JEOL MS-700 spectrometer. Unless otherwise noted, materials obtained from commercial suppliers were used without further purification.

1,4,5,8-Naphthodiquinone (NDQ)

A mixture of **1** (570 mg, 3.00 mmol) and [bis(trifluoroacetoxy)iodo]benzene (3.02 g, 7.02 mmol) in acetone (20 mL) was stirred for 1.5 h at room temperature. Then, [bis(trifluoroacetoxy)iodo]benzene (3.44 g, 8.00 mmol) was added to the reaction mixture, and the mixture was stirred at room temperature. After stirring for 2.5 h, the mixture was filtered through a glass filter, and the residue was washed with acetone and benzene, and then dried over under reduced pressure to yield **NDQ** (86 mg, 15%) as a brown solid. M.p. 204.5–208.0 $^\circ\text{C}$; IR (KBr): $\nu = 3065, 1685, 1642, 1606, 1299, 1132, 865 \text{ cm}^{-1}$; ^1H NMR (500 MHz, DMSO-d_6): $\delta = 6.89$ (s, 4H; vinyl-H); ^{13}C NMR (126 MHz, DMSO-d_6): $\delta = 186.0, 136.1, 126.3$ ppm.

2,2'-Bis-*p*-benzoquinone (BBQ)

A mixture of **2** (2.75 g, 19.9 mmol), iodine (2.03 g, 8.01 mmol), iodic acid (1.04 g, 5.92 mmol), acetic acid (33 mL), water (7 mL), and sulfuric acid (1 mL) was heated to 60 $^\circ\text{C}$ for 5 h. After cooling to room temperature, the reaction mixture was poured into 10% aqueous NaHSO_3 . The insoluble precipitate was filtered off using a glass filter, and then washed with diethyl ether. The aqueous layer was then separated and extracted with diethyl ether. The combined organic layer was washed with brine, dried over sodium sulfate, and evaporated under reduced pressure. The residue was purified by column chromatography on silica gel (eluent: hexane–chloroform = 2/1) to yield **3** (3.90 g, 74%) as a colorless liquid. ^1H NMR (500 MHz, CDCl_3): $\delta = 7.33$ (d, $J = 2.8$ Hz, 1H; ClCH), 6.85 (dd, $J = 9.2, 3.2$ Hz, 1H; $\text{ClC}(\text{OCH}_3)\text{CHCH}$), 6.74 (d, $J = 8.7$ Hz, 1H; $\text{ClC}(\text{OCH}_3)\text{CH}$), 3.81 (s, 3H; OCH_3), 3.74 (s,

3H; OCH₃) ppm; ¹³C NMR (126 MHz, CDCl₃): δ = 154.1, 152.5, 124.7, 114.6, 111.5, 85.9, 56.9, 55.8 ppm; LRMS (EI): *m/z* (%): 264 (100) [M⁺], 249 (78), 122 (18), 107 (23), 79 (22).

A mixture of **3** (5.84 g, 22.1 mmol), copper (7.76 g, 122 mmol), 2,2'-bipyridine (347 mg, 2.22 mmol), and DMSO (40 mL) was heated at 150 °C under N₂ for 15 h. After cooling to room temperature, water (80 mL) was added with cooling by an ice bath. The mixture was then filtered through a pad of Celite and the pad was washed with diethyl ether. The aqueous layer was separated and extracted with diethyl ether. The combined organic layer was washed with brine, dried over sodium sulfate, and evaporated under reduced pressure. The residue was purified by column chromatography on silica gel (eluent: hexane–chloroform = 1/1 and 100% chloroform) to yield **4** (1.99 g, 66%) as a white crystalline solid. M.p. 96.0–98.0 °C; ¹H NMR (500 MHz, CDCl₃): δ = 6.91 (d, *J* = 8.7 Hz, 2H; CC(OCH₃)CH), 6.86 (dd, *J* = 8.7, 3.2 Hz, 2H; CC(OCH₃)CHCH), 6.84 (d, *J* = 3.3 Hz, 2H; CCHC(OCH₃)), 3.78 (s, 6H; OCH₃), 3.73 (s, 6H; OCH₃) ppm; ¹³C NMR (126 MHz, CDCl₃): δ = 153.2, 151.2, 128.5, 117.1, 113.3, 112.4, 56.4, 55.6 ppm; LRMS (EI): *m/z* (%): 274 (100) [M⁺], 259 (42), 244 (26), 228 (37), 213 (17).

To a mixture of **4** (551 mg, 2.01 mmol) in acetonitrile (15 mL), a solution of CAN (11.0 g, 6.00 mmol) in water (15 mL) was added at 0°C, the mixture was then warmed slowly to room temperature. After stirring for 1 h, water (60 mL) and dichloromethane (30 mL) was added to the mixture. The aqueous layer was then separated and extracted with dichloromethane. The combined organic layer was washed with brine, dried over sodium sulfate, and evaporated under reduced pressure. The residue was purified by column chromatography on silica gel (eluent: 100% chloroform and chloroform–ethyl acetate = 20/1) to yield **BBQ** (294 mg, 68%) as a yellow crystalline solid. M.p. 184.5–191.0 °C; IR (KBr): ν = 1657, 1594, 1351, 1293, 1280, 1101, 928, 918, 847 cm⁻¹; ¹H NMR (500 MHz, CDCl₃): δ = 6.89 (d, *J* = 10.1 Hz, 2H; CCOCH), 6.86 (dd, *J* = 10.1, 2.3 Hz, 2H; CCOCHCH), 6.83 (d, *J* = 2.3 Hz, 2H; CCHCO) ppm; ¹³C NMR (126 MHz, CDCl₃): δ = 186.2, 184.4, 139.9, 136.8, 136.6, 135.8 ppm; LRMS (EI): *m/z* (%): 214 (63) [M⁺], 186 (23), 130 (28), 102 (36), 82 (58), 54 (69), 50 (27), 44 (100).

5,5'-Difluoro-2,2'-bis-*p*-benzoquinone (F₂-BBQ)

To a mixture of CAN (7.24 g, 13.2 mmol) in water (12 mL), a solution of **5** (625 mg, 4.00 mmol) in acetonitrile (12 mL) was added dropwise at room temperature. After stirring for 1.5 h, water (60 mL) was added to the mixture, and the mixture was filtered through a glass filter. The residue was washed with water and chloroform / hexane = 1 : 1, then dried over under reduced pressure to yield **F₂-BBQ** (195 mg, 39%) as a yellow solid. M.p. 183.0–185.0 °C; IR (KBr): ν = 1685, 1656, 1604, 1357, 1163, 921 cm⁻¹; ¹H NMR (400 MHz, CDCl₃): δ = 6.87 (d, *J* = 7.6 Hz, 2H; CCHCO), 6.57 (d, *J* = 10.0 Hz, 2H; CFCH) ppm; ¹³C NMR (100 MHz, CDCl₃): δ = 184.1 (d, *J* = 14.3 Hz), 178.3 (d, *J* = 24.8 Hz), 159.6 (d, *J* = 292.2 Hz), 139.4, 134.4 (d, *J* = 2.8 Hz), 115.7 (d, *J* = 9.5 Hz) ppm; LRMS (EI): *m/z* (%): 250 (100) [M⁺], 129 (58), 73 (77), 69 (64), 60 (53), 57 (61), 55 (62); HRMS (EI): found [M⁺] 250.0073, C₁₂H₄F₂O₄⁺ requires 250.0078.

5,5'-Dichloro-2,2'-bis-*p*-benzoquinone (Cl₂-BBQ)

To a mixture of CAN (7.24 g, 13.2 mmol) in water (12 mL), a solution of **6** (690 mg, 4.00 mmol) in acetonitrile (12 mL) was added dropwise at room temperature. After stirring for 1.5 h, water (60 mL) was added to the mixture, and the mixture was filtered through a glass filter. The residue was washed with water, and dried over under reduced pressure. Then, the residue was purified by column chromatography on silica gel (eluent: 100% chloroform), and by washing with water and chloroform / hexane = 1 : 1, then dried over under reduced pressure to yield **Cl₂-BBQ** (176 mg, 31%) as a yellow solid. M.p. 196.0–199.0 °C; IR (KBr): ν = 1670, 1653, 1608, 1570, 1320, 1194, 1010, 910 cm⁻¹; ¹H NMR (400 MHz, CDCl₃): δ = 7.12 (s, 2H; CCICH), 7.00 (s, 2H; CCHCO) ppm; ¹³C NMR (100 MHz, DMSO-d₆): δ = 182.8, 179.0, 143.1, 139.6, 136.0, 134.2 ppm; LRMS (EI): *m/z* (%): 286 (25), 284 (44), 282 (56) [M⁺], 149 (75), 84 (44), 81 (50), 73 (37), 69 (100), 57 (48), 55 (40).

5,5'-Dibromo-2,2'-bis-*p*-benzoquinone (Br₂-BBQ)

To a mixture of CAN (7.24 g, 13.2 mmol) in water (12 mL), a solution of **7** (868 mg, 4.00 mmol) in acetonitrile (12 mL) was added dropwise at room temperature. After stirring for 1.5 h, water (60 mL) was added to the mixture, and the mixture was filtered through a glass filter. The residue was washed with water and chloroform / hexane = 1 : 1, then dried over under reduced pressure to yield **Br₂-BBQ** (187 mg, 25%) as a yellow solid. M.p. 188.5–191.0 °C; IR (KBr): ν = 1664, 1604, 1571, 1318, 1189, 984, 910, 438 cm⁻¹; ¹H NMR (400 MHz, CDCl₃): δ = 7.40 (s, 2H; CBrCH), 7.04 (s, 2H; CCHCO) ppm; ¹³C NMR (100 MHz, CDCl₃): δ = 181.7, 178.4, 139.0, 138.2, 137.7, 135.6 ppm; LRMS (EI): *m/z* (%): 374 (57), 372 (85), 370 (41) [M⁺], 149 (40), 86 (44), 84 (63), 81 (49), 73 (41), 69 (100), 57 (45), 55 (43).

5,5'-Dimethyl-2,2'-bis-*p*-benzoquinone (Me₂-BBQ)

To a mixture of CAN (3.61 g, 6.60 mmol) in water (6 mL), a solution of **8** (304 mg, 2.00 mmol) in acetonitrile (6 mL) was added dropwise at room temperature. After stirring for 1 h, water (40 mL) was added to the mixture, and the mixture was filtered through a glass filter. The crude product obtained was washed with water, and then dried over under reduced pressure. The residue was recrystallized from chloroform / ethanol to yield **Me₂-BBQ** (164 mg, 68%) as a yellow solid. Decomposition 167.0 °C; IR (KBr): ν = 1649, 1629, 1577, 1234, 1205 cm⁻¹; ¹H NMR (400 MHz, CDCl₃): δ = 6.81 (s, 2H; CCHCO), 6.70 (d, *J* = 1.6 Hz, 2H; C(CH₃)CH), 2.10 (d, *J* = 1.2 Hz, 6H; CH₃) ppm; ¹³C NMR (100 MHz, CDCl₃): δ = 186.9, 184.7, 146.2, 139.5, 135.9, 133.6, 15.6 ppm; LRMS (EI): *m/z* (%): 242 (100) [M⁺], 146 (82), 118 (24), 68 (41), 40 (37).

5,5'-Dimethoxy-2,2'-bis-*p*-benzoquinone ((OMe)₂-BBQ)

To a mixture of CAN (18.1 g, 33.0 mmol) in water (30 mL), a solution of **9** (1.68 g, 10.0 mmol) in acetonitrile (30 mL) was added dropwise at room temperature. After stirring for 1.5 h, water (150

mL) was added to the mixture, and the mixture was filtered through a glass filter. The residue was washed with water and methanol, then dried over under reduced pressure to yield **(OMe)₂-BBQ** (93 mg, 7%) as a yellow solid. Decomposition 214.0 °C; IR (KBr): $\nu = 1671, 1643, 1620, 1574, 1225, 1197, 1173, 998, 910 \text{ cm}^{-1}$; ¹H NMR (400 MHz, CDCl₃): $\delta = 6.77 \text{ (s, 2H; C(OCH}_3\text{)CH)}$, $6.04 \text{ (s, 2H; CCHCO)}$, $3.87 \text{ (s, 6H; OCH}_3\text{)}$ ppm; Measurement of ¹³C NMR was unsuccessful because of low solubility of the product in general organic solvents; LRMS (EI): m/z (%): 274 (100) [M⁺], 245 (47), 231 (37), 201 (21), 69 (24).

5,5'-Dibutyl-2,2'-bis-*p*-benzoquinone (*n*Bu₂-BBQ)

To a mixture of **2** (277 mg, 2.00 mmol) and TMEDA (621 mg, 5.40 mmol) in dry hexane (4 mL), *n*-butyllithium (1.6 M hexane solution, 2.91 g, 6.00 mmol) was added dropwise at room temperature under N₂. After stirring for 20 h, the reaction mixture was cooled by an ice bath, 1-bromobutane (1.154 g, 8.40 mmol) was then added dropwise to the mixture, and the mixture was warmed slowly to room temperature. After stirring for 12 h, the mixture was quenched by adding excess amount of water dropwise. The aqueous layer was separated and extracted with diethyl ether. The combined organic layer was then washed with brine, dried over sodium sulfate, and evaporated under reduced pressure. The residue was purified by column chromatography on silica gel (eluent: hexane–chloroform = 4/1) to yield **10** (84 mg, 22%) as a colorless liquid. IR (neat): $\nu = 2954, 2931, 1501, 1465, 1224, 1050 \text{ cm}^{-1}$; ¹H NMR (400 MHz, CDCl₃): $\delta = 6.76 \text{ (d, } J = 8.4 \text{ Hz, 1H; CC(OCH}_3\text{)CH)}$, $6.73 \text{ (d, } J = 3.2 \text{ Hz, 1H; CCHC(OCH}_3\text{))}$, $6.68 \text{ (dd, } J = 8.8, 3.2 \text{ Hz, 1H; CC(OCH}_3\text{)CHCH)}$, $3.78 \text{ (s, 3H; OCH}_3\text{)}$, $3.76 \text{ (s, 3H; OCH}_3\text{)}$, $2.58 \text{ (t, } J = 7.2 \text{ Hz, 2H; Ar-CH}_2\text{)}$, $1.59\text{--}1.52 \text{ (m, 2H; ArCH}_2\text{CH}_2\text{)}$, $1.37 \text{ (sext, } J = 7.2 \text{ Hz, 2H; CH}_2\text{CH}_3\text{)}$, $0.93 \text{ (t, } J = 7.6 \text{ Hz, 3H; CH}_3\text{)}$ ppm; ¹³C NMR (100 MHz, CDCl₃): $\delta = 153.3, 151.7, 132.4, 116.1, 110.9, 110.3, 55.6, 55.3, 32.0, 29.9, 22.5, 13.9$ ppm; LRMS (EI): m/z (%): 194 (90) [M⁺], 152 (30), 151 (100), 137 (68), 121 (51), 91 (27), 77 (19).

To a mixture of CAN (3.77 g, 6.87 mmol) in water (15 mL), a solution of **10** (403 mg, 2.10 mmol) in acetonitrile (6 mL) was added dropwise at room temperature. After stirring for 1 h, water (60 mL) was added to the mixture. The mixture was filtered through a glass filter, and the residue was washed with water and then dried over under reduced pressure. The crude product obtained was recrystallized from chloroform / ethanol to yield ***n*Bu₂-BBQ** (220 mg, 65%) as a yellow crystalline solid. M.p. 167.5–169.0 °C; IR (KBr): $\nu = 2953, 2926, 1661, 1646, 923 \text{ cm}^{-1}$; ¹H NMR (400 MHz, CDCl₃): $\delta = 6.80 \text{ (s, 2H; CCHCO)}$, $6.65 \text{ (t, } J = 1.6 \text{ Hz, 2H; C(nBu)CH)}$, $2.46 \text{ (dt, } J = 7.8, 1.2 \text{ Hz, 4H; CCH}_2\text{)}$, $1.55\text{--}1.48 \text{ (m, 4H; CCH}_2\text{CH}_2\text{)}$, $1.41 \text{ (sext, } J = 7.2 \text{ Hz, 4H; CH}_2\text{CH}_3\text{)}$, $0.95 \text{ (t, } J = 7.2 \text{ Hz, 6H; CH}_3\text{)}$ ppm; ¹³C NMR (100 MHz, CDCl₃): $\delta = 186.8, 185.0, 150.0, 139.2, 136.2, 132.6, 29.8, 28.5, 22.4, 13.8$ ppm; LRMS (EI): m/z (%): 328 (11), 327 (11) [M+H]⁺, 155 (22), 138 (28), 137 (69), 136 (100), 107 (33), 91 (24), 90 (26), 89 (34), 77 (32).

5,5'-Diisopropyl-2,2'-bis-*p*-benzoquinone (*i*Pr₂-BBQ)

To a mixture of *n*-butyllithium (1.6 M hexane solution, 2.1 mL, 3.3 mmol) in dry THF (1 mL), a

solution of **2** (279 mg, 2.00 mmol) in dry THF (1.6 mL) was added at room temperature under argon. After stirring for 1 h, acetone (480 mg, 8.20 mmol) dissolved in dry THF (2.2 mL) was slowly added. After stirring for 1 h, the mixture was quenched by adding excess amount of water dropwise. The aqueous layer was then separated and extracted with diethyl ether. The combined organic layer was washed with brine, dried over sodium sulfate, and evaporated under reduced pressure. The residue was purified by column chromatography on silica gel (eluent: hexane–ether = 1/2) to yield **11** (186 mg, 47%) as a yellow liquid. IR (neat): $\nu = 3444, 2965, 2939, 2834, 1491, 1464, 1420, 1281, 1218, 1178, 1070, 1050, 1026 \text{ cm}^{-1}$; $^1\text{H NMR}$ (400 MHz, CDCl_3): $\delta = 6.91$ (d, $J = 3.2$ Hz, 1H; $\text{CCHC}(\text{OCH}_3)$), 6.85 (d, $J = 8.6$ Hz, 1H; $\text{CC}(\text{OCH}_3)\text{CH}$), 6.74 (dd, $J = 8.8, 3.2$ Hz, 1H; $\text{CC}(\text{OCH}_3)\text{CHCH}$), 4.25 (s, 1H; OH), 3.88 (s, 3H; OCH_3), 3.78 (s, 3H; OCH_3), 1.59 (s, 6H; CH_3) ppm; $^{13}\text{C NMR}$ (100 MHz, CDCl_3): $\delta = 153.5, 151.0, 137.1, 113.0, 111.9, 111.1, 72.4, 55.7, 55.6, 29.5$ ppm; LRMS (EI): m/z (%): 196 (47) [M^+], 181 (98), 43 (100).

To a mixture of 10% palladium on carbon (43 mg) and **11** (372 mg, 1.90 mmol) in ethyl acetate (5 mL), a drop of sulfuric acid was added, then the mixture was stirred at room temperature under the atmosphere of hydrogen using a balloon. After stirring for 24 h, water (50 mL) was added dropwise to the mixture. The palladium on carbon was filtered off, and the aqueous layer was separated and extracted with diethyl ether. The combined organic layer was washed with brine, dried over sodium sulfate, and evaporated under reduced pressure. The residue was purified by column chromatography on silica gel (eluent: hexane–chloroform = 4/1) to yield **12** (284 mg, 79%) as a yellow liquid. IR (neat): $\nu = 2959, 1501, 1464, 1281, 1220, 1048 \text{ cm}^{-1}$; $^1\text{H NMR}$ (400 MHz, CDCl_3): $\delta = 6.84$ (d, $J = 2.4$ Hz, 1H; $\text{CCHC}(\text{OCH}_3)$), 6.80 (d, $J = 8.4$ Hz, 1H; $\text{CC}(\text{OCH}_3)\text{CH}$), 6.70 (dd, $J = 8.4, 2.4$ Hz, 1H; $\text{CC}(\text{OCH}_3)\text{CHCH}$), 3.81 (s, 3H; OCH_3), 3.80 (s, 3H; OCH_3), 3.33 (sept, $J = 6.8$ Hz, 1H; CH), 1.23 (d, $J = 7.6$ Hz, 6H; CH_3) ppm; $^{13}\text{C NMR}$ (100 MHz, CDCl_3): $\delta = 153.7, 151.0, 138.4, 113.0, 111.2, 109.9, 56.0, 55.5, 26.8, 22.7$ ppm; LRMS (EI): m/z (%): 180 (74) [M^+], 165 (100), 150 (50), 105 (17).

To a mixture of CAN (1.15 g, 2.10 mmol) in water (4 mL), a solution of **12** (115 mg, 0.63 mmol) in acetonitrile (4 mL) was added dropwise at room temperature. After stirring for 1 h, water (40 mL) was added to the mixture. The mixture was filtered through a glass filter, and the residue was washed with water and then dried over under reduced pressure. The crude product obtained was recrystallized from chloroform / ethanol to yield ***i*Pr₂-BBQ** (76 mg, 81%) as a yellow crystalline solid. M.p. 173.5–175.0 °C; IR (KBr): $\nu = 2966, 1653, 1600, 1233, 1048, 925 \text{ cm}^{-1}$; $^1\text{H NMR}$ (400 MHz, CDCl_3): $\delta = 6.81$ (s, 2H; CCHCO), 6.63 (s, 2H; $\text{C}(\textit{iPr})\text{CH}$), 3.07 (sept, $J = 6.4$ Hz, 2H; CH), 1.16 (d, $J = 6.8$ Hz, 12H; CH_3) ppm; $^{13}\text{C NMR}$ (100 MHz, CDCl_3): $\delta = 186.4, 185.4, 155.3, 139.0, 136.4, 130.7, 26.8, 21.3$ ppm; LRMS (EI): m/z (%): 300 (19), 299 (15), 298 (8) [M^+], 154 (27), 149 (100), 136 (26), 83 (27), 81 (28), 71 (44), 69 (43), 57 (35), 55 (27); HRMS (EI): found [M^+] 298.1193, $\text{C}_{18}\text{H}_{18}\text{O}_4^+$ requires 298.1205.

5,5'-Di-*tert*-butyl-2,2'-bis-*p*-benzoquinone (*t*Bu₂-BBQ)

To a mixture of sodium hydride (178 mg, 4.40 mmol) in DMF (3 mL), a solution of **13** (334 mg, 2.00

mmol) in DMF (3 mL) was added at room temperature under N₂. Then, methyl iodide (1.36 g, 9.60 mmol) was slowly added to the mixture, and stirred for 4 h at 40 °C. After cooling to room temperature, the mixture was quenched by adding excess amount of water dropwise. The aqueous layer was then separated and extracted with diethyl ether. The combined organic layer was washed with brine, dried over sodium sulfate, and evaporated under reduced pressure. The residue was purified by column chromatography on silica gel (eluent: hexane–ether = 95/5) to yield **14** (323 mg, 83%) as a yellow liquid. IR (neat): $\nu = 2952, 2927, 1499, 1488, 1464, 1283, 1221, 1059, 1032 \text{ cm}^{-1}$; ¹H NMR (400 MHz, CDCl₃): $\delta = 6.89 \text{ (d, } J = 3.2 \text{ Hz, 1H; C}(t\text{Bu})\text{CH)}$, $6.88 \text{ (d, } J = 8.8 \text{ Hz, 1H; C}(t\text{Bu})\text{C(OCH}_3\text{)CH)}$, $6.79 \text{ (dd, } J = 8.4, 3.2 \text{ Hz, 1H; CC(OCH}_3\text{)CHCH)}$, $3.80 \text{ (s, 3H; OCH}_3\text{)}$, $3.77 \text{ (s, 3H; OCH}_3\text{)}$, $1.36 \text{ (s, 9H; CH}_3\text{)}$ ppm; ¹³C NMR (100 MHz, CDCl₃): $\delta = 153.2, 152.8, 139.8, 114.2, 112.2, 109.7, 55.6, 55.5, 34.9, 29.6 \text{ ppm}$; LRMS (EI): m/z (%): 194 (56) [M⁺], 179 (100), 164 (36), 151 (36).

To a mixture of CAN (2.71 g, 4.90 mmol) in water (10 mL), a solution of **14** (320 mg, 1.70 mmol) in acetonitrile (10 mL) was added dropwise at room temperature. After stirring for 1 h, water (60 mL) was added to the mixture. The mixture was filtered through a glass filter, and the residue was washed with water and then dried over under reduced pressure. The crude product obtained was recrystallized from chloroform / ethanol to yield **tBu₂-BBQ** (150 mg, 56%) as a yellow crystalline solid. M.p. 194.0–196.0 °C; IR (KBr): $\nu = 2956, 2924, 1658, 1630, 1594, 1362, 1237, 914 \text{ cm}^{-1}$; ¹H NMR (400 MHz, CDCl₃): $\delta = 6.75 \text{ (s, 2H; CCHCO)}$, $6.68 \text{ (s, 2H; C}(t\text{Bu})\text{CH)}$, $1.31 \text{ (s, 18H; CH}_3\text{)}$ ppm; ¹³C NMR (100 MHz, CDCl₃): $\delta = 186.6, 185.6, 156.4, 138.0, 137.6, 131.9, 35.3, 29.0 \text{ ppm}$; LRMS (EI): m/z (%): 326 (48) [M⁺], 311 (56), 285 (36), 284 (87), 283 (34), 269 (100), 201 (48), 41 (40).

Details of DFT calculations and Gaussian archives of optimized structures at the B3LYP/6-31+G* level

DFT calculations were carried out using GAUSSIAN 09 program.¹ Geometry optimization were performed using standard gradient techniques at the B3LYP level of theory using restricted (RB3LYP) and unrestricted (UB3LYP) methods for closed- and open-shell systems, respectively.² In every case, 6-31+G* was used as the basis set, and all ground states were verified by vibration frequency analysis. The calculated molecular structures were visualized by Gauss View 5.0.³

The singlet and triplet states of the BQ dianion, the doublet and quartet states of the trianion, and singlet, triplet, and quintet states of the tetraanion were calculated and summarised in the following Table. BQ dianion and trianion are more stable in singlet and doublet states than in triplet and quartet states, respectively, while BBQ dianion and trianion are more stable in triplet and doublet states than in singlet and quartet states, respectively. Although the BBQ tetraanion had similar energies in singlet and triplet states, the singlet was determined as the most stable state considering that the triplet state was highly spin contaminated.

Species	Spin state	Energy (kJ mol ⁻¹)	<S ² > value ^a
BQ dianion	singlet	-1001446.2	—
	triplet	-1001326.5	2.082
BQ trianion	doublet	-1000698.6	0.751
	quartet	-1000623.0	3.822
BBQ dianion	singlet	-2000099.7	—
	triplet	-2000134.6	2.278
BBQ trianion	doublet	-1999622.3	1.360
	quartet	-1999553.6	4.089
BBQ tetraanion	singlet	-1998790.5	—
	triplet	-1998789.7	2.598
	quintet	-1998752.0	6.406

^a Determined by the HF/6-31+G* level.

BQ

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I\1\GINC-HEX\FOpt\RB3LYP\6-31+G(d)\C6H4O2\YOKOJI\10-Aug-2015\0\#B3LYP/6-31+G* opt=(ReadFC,Maxcycle=100) NoSymm freq=NoRaman geom=check guess=read\bqb3+\0,I\C,1.1473960993,0.6385839984,0.0000000008\C,-1.2455936404,1.2457168671,0.0000000035\C,-0.512664242,-1.1884288198,-0.0000000034\C,-1.6176902387,-0.1937182215,-0.0000000006\C,0.7752942945,-0.8008521598,0\C,0.0423659446,1.6332890664,-0.0000000026\H,-2.0720010581,1.9520647157,0.0000000074\H,-0.8119264774,-2.2335912849,-0.0000000048\H,0.341651651,2.6784450456,-0.0000000088\O,2.3221669073,0.9923242349,0.0000000025\O,-2.7924601291,-0.5474613541,0.\H,1.6017108891,-1.5071845781,0.0000000059\\Version=EM64L-G09RevC.01\HF=-381.4704834\RMSD=3.710e-09\RMSF=1.597e-04\Dipole=0.000008,0.000037,0.\Quadrupole=-9.3959157,6.969247,2.4266688,-5.4190299,0.,0.\PG=C01 [X(C6H4O2)]\@
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BQ anion radical

1\1\GINC-HEX\FOpt\UB3LYP\6-31+G(d)\C6H4O2(1,-2)\YOKOJI\07-Sep-2015\0\#B3LYP/6-31+G* opt=(ReadFC,Maxcycle=100) NoSymm freq=NoRaman geom=check guess=read\bqradanib3+\-1,2\C,1.1701074551,0.6453203105,0.000000002\C,-1.2465743574,1.1972606258,-0.0000000005\C,-0.5404529244,-1.1485372011,-0.0000000005\C,-1.6403988766,-0.2004556137,0.000000001\C,0.7762825226,-0.7523958326,0.0000000001\C,0.0701615082,1.5934015139,0.0000000002\H,-2.0533614742,1.9300581383,-0.0000000004\H,-0.8087995645,-2.2049257995,-0.0000000005\H,0.3385081895,2.6497901472,-0.0000000005\O,2.3885404091,1.0121158865,0.0000000007\O,-2.8588325568,-0.567251678,0.0000000007\H,1.5830696694,-1.4851929873,-0.0000000004\Version=EM64L-G09RevC.01\HF=-381.5500259\S2=0.761125\S2-1=0.\S2A=0.75004\RMSD=5.840e-09\RMSF=4.801e-05\Dipole=0.0000022,0.0000005,0.\Quadrupole=-17.0104226,8.1357382,8.8746844,-8.129647,0.,0.\PG=CS [SG(C6H4O2)]\@\

BQ dianion (singlet)

1\1\GINC-HEX\FOpt\RB3LYP\6-31+G(d)\C6H4O2(2-)\YOKOJI\07-Sep-2015\0\#B3LYP/6-31+G* opt=(ReadFC,Maxcycle=100) NoSymm freq=NoRaman geom=check guess=read\bqdianib3+\-2,1\C,1.2060409451,0.6560230462,0.0000000015\C,-1.2528691485,1.1589064936,0.0000000013\C,-0.5671534805,-1.1200526782,-0.0000000006\C,-1.6763312212,-0.2111611783,0.0000000037\C,0.7825791502,-0.71404129,0.0000000021\C,0.0968597503,1.5649167725,0.0000000004\H,-2.0417433894,1.919837133,0.0000000022\H,-0.8051207041,-2.190000515,-0.0000000042\H,0.3348278472,2.6348643575,-0.0000000013\O,2.466028291,1.0351184497,-0.0000000061\O,-2.9363209113,-0.5902504951,-0.0000000047\H,1.5714528711,-1.4749725858,0.0000000058\Version=EM64L-G09RevC.01\HF=-381.4306621\RMSD=2.621e-09\RMSF=3.335e-05\Dipole=0.0000002,0.0000007,0.\Quadrupole=-26.5357374,10.0961101,16.4396273,-11.7316984,0.,0.\PG=C01 [X(C6H4O2)]\@\

BQ dianion (triplet)

1\1\GINC-HEX\FOpt\UB3LYP\6-31+G(d)\C6H4O2(2,-3)\YOKOJI\07-Sep-2015\0\#B3LYP/6-31+G* opt=(ReadFC,Maxcycle=100) NoSymm freq=NoRaman geom=check guess=read\bqdiani_tb3+\-2,3\C,1.1517051528,0.6398139779,-0.0000000037\C,-1.2504952668,1.2090974392,-0.0000000021\C,-0.5371206073,-1.1605380364,-0.0000000008\C,-1.6219724052,-0.1949419362,-0.0000000034\C,0.7801736459,-0.7642009423,-0.0000000077\C,0.0668213364,1.6053601848,-0.0000000016\H,-2.0531257789,1.9492676641,0.0000000039\H,-0.7978887924,-2.2207948058,0.0000000053\H,0.3275890478,2.6656204715,-0.0000000041\O,2.3794228588,1.0094182017,0.0000000131\O,-2.8496660064,-0.5645403031,0.0000000041\H,1.5828068154,-1.5043744054,-0.0000000029\Version=EM64L-G09RevC.01\HF=-381.3850578\S2=2.010858\S2-1=0.\S2A=2.000029\RMSD=4.382e-09\RMSF=6.300e-05\Dipole=0.0006022,0.000176,0.\Quadrupole=-9.4420042,-14.5613075,24.0033117,2.0941194,0.,0.\PG=C01 [X(C6H4O2)]\@\

BQ radical trianion (doublet)

1\1\GINC-I7\FOpt\UB3LYP\6-31+G(d)\C6H4O2(3,-2)\YOKOJI\24-Sep-2015\0\#B3LYP/6-31+G* opt=(ReadFC,Maxcycle=100) NoSymm freq=NoRaman geom=check guess=read\bqradtrianib3+\-3,2\C,1.1759334645,0.6469238006,-0.0000000037\C,-1.2589570636,1.1787564209,-0.0000000025\C,-0.5613350778,-1.139927229,-0.0000000013\C,-1.6462010964,-0.2020868079,-0.0000000027\C,0.7886323578,-0.7338498394,-0.0000000007\C,0.0910377929,1.5847931815,-0.0000000011\H,-2.0486933569,1.9375794324,-0.0000000015\H,-0.801197311,-2.2085933933,0.0000000037\H,0.3308731859,2.6534668673,0.0000000096\O,2.4515378956,1.0306295218,-0.0000000046\O,-2.921783745,-0.5858559938,0.0000000007\H,1.5784029539,-1.4926484511,0.0000000042\Version=EM64L-G09RevC.01\HF=-381.1459255\S2=0.750385\S2-1=0.\S2A=0.75\RMSD=6.496e-09\RMSF=1.409e-05\Dipole=0.0008225,-0.0000944,0.\Quadrupole=-19.332406,-14.5575759,33.8899819,-1.0103823,0.,0.\PG=C01 [X(C6H4O2)]\@\

BQ trianion (quartet)

1\1\GINC-I7\FOpt\UB3LYP\6-31+G(d)\C6H4O2(3,-4)\YOKOJI\24-Sep-2015\0\#B3LYP/6-31+G* opt=(ReadFC,Maxcycle=100) NoSymm freq=NoRaman geom=check guess=read\bqtriani_qb3+\-3,4\C,1.1465935269,0.6375382051,0.0005394

374\C,-1.2518446907,1.2151280769,0.0000604907\C,-0.5361929961,-1.1660233197,-0.0004491558\C,-1.6168881742,-0.1927050116,-0.0005580095\C,0.7815707376,-0.7702772742,0.0001821872\C,0.0659073767,1.6109034044,0.0007657824\H,-2.0462786929,1.9631124019,-0.0000934307\H,-0.7868316804,-2.228014512,-0.0007447715\H,0.3165371537,2.6728997251,0.0004527883\O,2.3805113446,1.0084339042,0.0007273884\O,-2.8508343641,-0.5635454061,-0.0012267105\H,1.5760004589,-1.518258194,0.0003440037\Version=EM64L-G09RevC.01\HF=-381.1171206\S2=3.760769\S2-1=0.\S2A=3.750024\RMSD=9.435e-09\RMSF=2.076e-05\Dipole=-0.0003198,-0.0000959,-0.00115\Quadrupole=-5.8660167,-34.7434279,40.6094447,10.1303205,-0.0116159,-0.0205312\PG=C01 [X(C6H4O2)]\@

BBQ

1\1\GINC-HEX\FOpt\RB3LYP\6-31+G(d)\C12H6O4\YOKOJI\12-Jun-2014\0\#B3LYP/6-31+G* opt=(ReadFC,Maxcycle=100) NoSymm freq=NoRaman geom=check guess=read\bbqb3+\0,1\C,-1.2778582671,-1.4267324274,0.4361316144\C,-3.5932714274,-0.6508160007,0.0264002517\C,-1.6022139124,0.8575889265,-0.4311019121\C,-3.0712339914,0.6550742193,-0.4414232863\C,-0.7317500355,-0.1032558785,-0.0381814097\C,-2.7508698711,-1.6163630807,0.4284331341\H,-1.2700518354,1.8306300507,-0.7749048935\H,-3.0920156125,-2.5870845395,0.7778938327\O,-0.5487332502,-2.3290135418,0.831570026\O,-3.8252386423,1.5407150279,-0.8337297144\H,-4.6728577919,-0.7740987873,0.0189141313\C,3.0712338981,-0.6550740049,-0.4414242718\C,0.7317500293,0.1032558913,-0.0381815187\C,2.7508699601,1.6163628783,0.428433068\C,1.2778583585,1.4267322155,0.436132027\C,3.5932714312,0.6508159938,0.0263997755\C,1.6022138223,-0.8575887245,-0.4311026702\H,3.0920157774,2.5870841691,0.7778943972\H,1.2700516728,-1.8306296847,-0.7749060479\O,3.8252384678,-1.5407146211,-0.8337312928\O,0.5487334258,2.329013131,0.8315710445\H,4.6728577939,0.7740987877,0.018913476\Version=EM64L-G09RevC.01\HF=-761.742748\RMSD=6.903e-09\RMSF=2.699e-06\Dipole=0.,0.,0.0415335\Quadrupole=3.0018799,-4.9835492,1.9816693,16.7248217,0.0000039,-0.0000035\PG=C01 [X(C12H6O4)]\@

BBQ anion radical

1\1\GINC-HEX\FOpt\UB3LYP\6-31+G(d)\C12H6O4(1,-2)\YOKOJI\12-Jun-2014\0\#B3LYP/6-31+G* opt=(ReadFC,Maxcycle=100) NoSymm freq=NoRaman geom=check guess=read\bbqradanib3+\-1,2\C,-1.2585144815,-1.4673858664,0.3295123518\C,-3.5747921519,-0.6318463078,0.0269395058\C,-1.6330455883,0.882427057,-0.331957872\C,-3.0705603905,0.6980890418,-0.3485093217\C,-0.7244018479,-0.1165374765,-0.0304275964\C,-2.7242339941,-1.6331054684,0.341691391\H,-1.2876360973,1.8755217011,-0.5887735852\H,-3.0709169042,-2.6275520533,0.6146420435\O,-0.5468158108,-2.43822992,0.6386828011\O,-3.8653397769,1.610626389,-0.6787163803\H,-4.6551040822,-0.7647790775,0.0269144395\C,3.0705605998,-0.6980890962,-0.3485086705\C,0.7244020481,0.1165373891,-0.0304275161\C,2.7242337216,1.6331054991,0.3416917827\C,1.2585142377,1.4673858553,0.3295123638\C,3.5747919957,0.6318463987,0.0269401689\C,1.6330459295,-0.8824271656,-0.3319575022\H,3.0709165321,2.6275521259,0.6146424294\H,1.287636588,-1.8755218653,-0.5887731975\O,3.8653402742,-1.6106263761,-0.6787154101\O,0.5468152842,2.4382298955,0.6386824182\H,4.6551039147,0.7647793208,0.0269153563\Version=EM64L-G09RevC.01\HF=-761.8470123\S2=0.770726\S2-1=0.\S2A=0.750203\RMSD=5.461e-09\RMSF=1.914e-06\Dipole=-0.0000018,-0.0000004,0.1935648\Quadrupole=-14.0494731,-2.0548402,16.1043133,19.101606,-0.000003,0.0000029\PG=C01 [X(C12H6O4)]\@

BBQ dianion (singlet)

1\1\GINC-HEX\FOpt\RB3LYP\6-31+G(d)\C12H6O4(2-)\YOKOJI\12-Jun-2014\0\#B3LYP/6-31+G* opt=(CalcFC,Maxcycle=100) NoSymm freq=NoRaman geom=check guess=read\bbqdiani_sb3+\-2,1\C,-1.2495756318,-1.4986003841,0.2213382639\C,-3.5676619366,-0.6127674767,0.0473699671\C,-1.665904756,0.8981794996,-0.2603430762\C,-3.0872454416,0.7315898072,-0.2643001902\C,-0.7233088325,-0.1271184099,-0.0422303632\C,-2.7056199132,-1.6414525824,0.2803577067\H,-1.3071095909,1.9020696644,-0.4418755316\H,-3.0584530652,-2.6502695869,0.4971730927\O,-0.548198019,-2.5285388873,0.4154036359\O,-3.9225344585,1.6660807464,-0.525783855\H,-4.6496882494,-0.7576217608,0.0694310953\C,3.087325577,-0.731656962,-0.2631720628\C,0.7233219275,0.1271083688,-0.0420413817\C,2.7055348726,1.6415237519,0.2807681309\C,1.2495084574,1.498656885,0.2213407273\C,3.5676473773,0.6127792531,0.0483036658\C,1.6659837043,-0.8982451374,-0.2596069785\H,3.0583023939,2.6503953434,0.4974361296\H,1.3072428039,-1.902180638,-0.4409956269\O,3.9226939096,-1.6662145503,-0.5241635154\O,0.5480718568,2.5286443166,0.4149325839\H,4.6496670144,0.757638739

4,0.0706575824\\Version=EM64L-G09RevC.01\\HF=-761.7976553\\RMSD=2.897e-09\\RMSF=3.408e-06\\Dipole=-0.000050
4,0.0000422,0.3337398\\Quadrupole=-33.0328518,0.7610427,32.2718091,22.5885661,-0.0128397,0.0074347\\PG=C01 [X(C
12H6O4)]\\@

BBQ dianion (triplet)

1\\1\\GINC-HEX\\FOpt\\UB3LYP\\6-31+G(d)\\C12H6O4(2-,3)\\YOKOJI\\12-Jun-2014\\0\\#B3LYP/6-31+G* opt=(ReadFC,Maxc
ycle=100) NoSymm freq=NoRaman geom=check guess=read\\bbqdiani_tb3+\\-2,3\\C,-1.3258715512,-1.0347011127,1.030
4103049\\C,-3.5446631596,-0.6070129483,-0.0136795176\\C,-1.5394365194,0.4558584498,-0.9240795707\\C,-2.975065382,
0.2627335761,-1.0193718036\\C,-0.7372796096,-0.12742625,0.0352309775\\C,-2.7648495522,-1.210422278,0.9459969114
\\H,-1.0942372801,1.1204710879,-1.663504999\\H,-3.2045837273,-1.8743365846,1.6916141165\\O,-0.639969088,-1.662844
9436,1.8934913871\\O,-3.6862381957,0.8204962627,-1.924442379\\H,-4.6232213382,-0.7691712374,-0.0516736792\\C,2.97
50660583,-0.262733432,-1.0193699972\\C,0.7372796172,0.1274260559,0.0352314807\\C,2.7648489209,1.2104221918,0.94
59987501\\C,1.3258708885,1.0347008597,1.0304112614\\C,3.5446631432,0.6070131063,-0.013677319\\C,1.53943715,-0.45
58584297,-0.9240786687\\H,3.204582575,1.874336504,1.6916162582\\H,1.0942384003,-1.1204709252,-1.663504522\\O,3.6
862394819,-0.8204959817,-1.9244401648\\O,0.6399678509,1.6628444686,1.8934920499\\H,4.6232213171,0.7691715604,-0.
0516708777\\Version=EM64L-G09RevC.01\\HF=-761.8109459\\S2=2.021559\\S2-1=0.\\S2A=2.000233\\RMSD=9.051e-09\\RM
SF=3.493e-06\\Dipole=-0.0000002,0.,0.5073304\\Quadrupole=-25.9708775,20.3556857,5.6151918,7.7762014,-0.0000092,0.0
000041\\PG=C01 [X(C12H6O4)]\\@

BBQ radical trianion (doublet)

1\\1\\GINC-HEX\\FOpt\\UB3LYP\\6-31+G(d)\\C12H6O4(3-,2)\\YOKOJI\\12-Jun-2014\\0\\#B3LYP/6-31+G* opt=(ReadFC,Maxc
ycle=100) NoSymm freq=NoRaman geom=check guess=read\\bbqradtrianib3+\\-3,2\\C,-1.256072923,-1.4807978561,0.42
48186724\\C,-3.5537288613,-0.6072436752,-0.0552327445\\C,-1.6537937136,0.8577107886,-0.3135345976\\C,-3.074889599
6,0.7000816849,-0.4218240357\\C,-0.7305694082,-0.1456219035,0.0590220951\\C,-2.6920538729,-1.6193755417,0.346855
7947\\H,-1.2612500439,1.8445720016,-0.5394508919\\H,-3.0806170153,-2.6056100772,0.62014438\\O,-0.5452388681,-2.48
10256276,0.8178273305\\O,-3.8880000752,1.6407010305,-0.8268030191\\H,-4.6344904314,-0.781851115,-0.1118256021\\C,
3.074889702,-0.7000817351,-0.4218234483\\C,0.7305694783,0.1456219712,0.0590222453\\C,2.6920537407,1.6193755048,
0.3468564383\\C,1.2560728089,1.4807979142,0.4248189048\\C,3.5537288352,0.6072436098,-0.055231944\\C,1.653793857
4,-0.8577107738,-0.3135343365\\H,3.0806168876,2.6056100186,0.6201451212\\H,1.2612502161,-1.8445719604,-0.5394508
111\\O,3.8880003193,-1.640701099,-0.8268022957\\O,0.5452385453,2.4810257649,0.8178272531\\H,4.6344904217,0.78185
10755,-0.111824509\\Version=EM64L-G09RevC.01\\HF=-761.6157972\\S2=0.768778\\S2-1=0.\\S2A=0.750205\\RMSD=8.474
e-09\\RMSF=2.784e-06\\Dipole=-0.0000019,-0.0000004,0.3268645\\Quadrupole=-49.2702799,5.9054319,43.364848,21.0406
65,-0.0000072,0.0000001\\PG=C01 [X(C12H6O4)]\\@

BBQ trianion (quartet)

1\\1\\GINC-HEX\\FOpt\\UB3LYP\\6-31+G(d)\\C12H6O4(3-,4)\\YOKOJI\\17-Jun-2014\\0\\#B3LYP/6-31+G* opt=(ReadFC,Maxc
ycle=100) NoSymm freq=NoRaman scf=xqc geom=check guess=read\\bbqtriani_qb3+\\-3,4\\C,-1.2842671189,-1.4051815
606,-0.6060173101\\C,-3.5458451484,-0.6334066644,0.0858514723\\C,-1.5995559077,0.8087176347,0.435200155\\C,-3.029
1632316,0.6269466423,0.5764245278\\C,-0.7356827313,-0.1324022159,-0.1002233141\\C,-2.7205192325,-1.5818598722,-0.
4698280489\\H,-1.1985191344,1.7562671575,0.7869898712\\H,-3.1347058875,-2.5215159617,-0.8395474335\\O,-0.5727530
962,-2.3124873439,-1.1477562906\\O,-3.7836940096,1.5174783662,1.1121725234\\H,-4.6191213913,-0.817558773,0.16655
9936\\C,3.0291615906,-0.6269462465,0.5764587169\\C,0.735685319,0.132395031,-0.1002095913\\C,2.7205187002,1.58186
57442,-0.4697860907\\C,1.284269886,1.4051791028,-0.6059933114\\C,3.5458418615,0.6334130493,0.0858988385\\C,1.599
5569216,-0.808724409,0.4352177817\\H,3.1347050507,2.5215252624,-0.8394964783\\H,1.1985211688,-1.7562788881,0.78
69949815\\O,3.7836927431,-1.5174772269,1.1122082172\\O,0.5727572501,2.3124813694,-1.1477396286\\H,4.6191163978,0.
8175698022,0.1666204761\\Version=EM64L-G09RevC.01\\HF=-761.5896338\\S2=3.770541\\S2-1=0.\\S2A=3.750158\\RMSD
=5.488e-07\\RMSF=3.439e-05\\Dipole=-0.0000024,-0.0000033,0.6381637\\Quadrupole=-81.1018574,19.618495,61.4833625,-
23.7080612,-0.0009921,-0.000295\\PG=C01 [X(C12H6O4)]\\@

BBQ tetraanion (singlet)

1\1\GINC-HEX\FOpt\RB3LYP\6-31+G(d)\C12H6O4(4-)\YOKOJI\16-Jun-2014\0\#B3LYP/6-31+G* opt=(ReadFC,Maxcyc cle=100) NoSymm freq=NoRaman scf=xqc geom=check guess=read\bbqtetraanib3+\-4,1\C,-1.3158815861,-1.30350600 61,0.8745945234\C,-3.532418039,-0.6331056249,-0.1363302079\C,-1.5893838612,0.6660062056,-0.6155060674\C,-3.0031 653406,0.4924983054,-0.8295389648\C,-0.7451874462,-0.1574182512,0.1576396787\C,-2.7287418197,-1.4681381578,0.6 728848325\H,-1.1287456698,1.5178645905,-1.1220294858\H,-3.1837422558,-2.336773485,1.1687691311\O,-0.642246323 6,-2.1220309819,1.6416649036\O,-3.7357929631,1.2944304007,-1.6014723876\H,-4.6140459882,-0.8206688787,-0.21062 27685\C,3.0031623435,-0.4925004681,-0.8295483905\C,0.7451879769,0.1574187701,0.1576364636\C,2.7287441939,1.46 81400442,0.6728710926\C,1.3158848589,1.3035079014,0.8745868904\C,3.5324175028,0.6331052917,-0.1363444837\C,1. 5893815907,-0.6660077082,-0.6155101862\H,3.183746406,2.3367769366,1.1687510771\H,1.1287415366,-1.5178672212,- 1.1220299982\O,3.7357872108,-1.2944344648,-1.6014824544\O,0.6422525711,2.1220340307,1.641658677\H,4.614045102 3,0.8206687708,-0.210641875\Version=EM64L-G09RevC.01\HF=-761.2989724\RMSD=4.016e-09\RMSF=4.539e-06\Dip ole=-0.0000001,0.0000002,0.180132\Quadrupole=-66.6988359,28.2781056,38.4207304,3.2724201,0.0001762,0.0000013\PG =C01 [X(C12H6O4)]\@

BBQ tetraanion (triplet)

1\1\GINC-HEX\FOpt\UB3LYP\6-31+G(d)\C12H6O4(4-,3)\YOKOJI\19-Jun-2014\0\#B3LYP/6-31+G* opt=(CalcFC,Maxc ycle=100) NoSymm freq=NoRaman scf=xqc geom=check guess=read\bbqtetraani_tb3+\-4,3\C,-1.2734713324,-1.53934 93684,0.0033267464\C,-3.590627229,-0.6119734496,0.0048424361\C,-1.6999291012,0.9014650636,0.0010233927\C,-3.12 33706761,0.7518266013,0.0026712716\C,-0.7351071286,-0.1502141051,0.0010623659\C,-2.7137372459,-1.6830613331,0. 0054381154\H,-1.3185397769,1.9152733109,-0.0003225933\H,-3.093640016,-2.7070756421,0.0072607166\O,-0.56993745 42,-2.6251415142,0.003606555\O,-3.9653492816,1.7622527222,0.0022507977\H,-4.6710914604,-0.793271028,0.00628530 92\C,3.1234396256,-0.7518288444,-0.0045526817\C,0.7351282461,0.1501901297,-0.001218827\C,2.7136503186,1.683055 7987,-0.0035609879\C,1.2734013615,1.5393352363,-0.0018669063\C,3.5906041639,0.6119910936,-0.0049341399\C,1.700 0153728,-0.9014706728,-0.0025961448\H,3.0935318762,2.7070888662,-0.0038129616\H,1.3186927956,-1.9153012486,-0. 0020326426\O,3.9654694232,-1.762247446,-0.0058600534\O,0.5698023984,2.6251112063,-0.0005612992\H,4.671065120 4,0.7933446235,-0.0064484689\Version=EM64L-G09RevC.01\HF=-761.2986676\S2=2.022721\S2-1=0.\S2A=2.000187\R MS D=9.681e-09\RMSF=1.545e-05\Dipole=0.0008642,0.0008478,-0.0023441\Quadrupole=-102.6508154,10.4564442,92.19 43711,-21.8329575,0.2587467,0.0717221\PG=C01 [X(C12H6O4)]\@

BBQ tetraanion (quintet)

1\1\GINC-HEX\FOpt\UB3LYP\6-31+G(d)\C12H6O4(4-,5)\YOKOJI\26-Dec-2014\0\#B3LYP/6-31+G* opt=(ReadFC,Max cycle=100) NoSymm freq=NoRaman scf=xqc geom=check guess=read\bbqtetraani_qub3+\-4,5\C,-1.2933177273,-1.394 5278716,-0.6371311391\C,-3.5599124684,-0.6404868713,0.0841121338\C,-1.6019532903,0.7936590894,0.4541259933\C,- 3.03313509,0.6131201025,0.5877373773\C,-0.7374064256,-0.136612997,-0.1018949959\C,-2.7292916921,-1.5817515962,- 0.477118843\H,-1.1993911149,1.7342876322,0.8255850801\H,-3.1517758884,-2.5183974536,-0.8455260193\O,-0.5862809 126,-2.2826843206,-1.2202463807\O,-3.7808157637,1.509185332,1.1341765195\H,-4.6304686621,-0.8375760675,0.16502 93996\C,3.0316271257,-0.6125883409,0.5959692105\C,0.7376591111,0.1365338264,-0.100151927\C,2.7305022728,1.581 3236531,-0.4716345174\C,1.2949361644,1.3939582249,-0.6351207027\C,3.5596886419,0.6405617503,0.0925523182\C,1. 6007860499,-0.7932409899,0.4588999543\H,3.1539266631,2.5176326202,-0.8398188142\H,1.1972797877,-1.733530596,0. 8301918053\O,3.7779198473,-1.5081708285,1.1450906158\O,0.5893845299,2.281589938,-1.2208284544\H,4.6300388416, 0.8377157641,0.1760013858\Version=EM64L-G09RevC.01\HF=-761.2843424\S2=6.020381\S2-1=0.\S2A=6.000126\RMS D=3.539e-09\RMSF=5.177e-06\Dipole=-0.0012267,0.0004061,0.9345762\Quadrupole=-131.7106807,36.654948,95.055732 7,-52.9685106,-0.2627032,-0.0405227\PG=C01 [X(C12H6O4)]\@

Solubility study of BBQ derivatives

The solubilities of BBQ derivatives were measured for the non-substituted BBQ and R₂-BBQ (R = F, Cl, Br, Me, *n*Bu, and OMe) using tetraglyme as the electrolyte solution. The obtained solubilities and the related battery performance are summarized in the following table.

Solubility of the BBQs in tetraglyme and the related battery performance.

Sample	Solubility in tetraglyme (g L ⁻¹)	Capacity retention ^a (%)
BBQ	4.0-5.0	67
F ₂ -BBQ	18-19	12
Cl ₂ -BBQ	9.0-10	68
Br ₂ -BBQ	3.0-4.0	>99 ^b
Me ₂ -BBQ	7.0-8.0	46
<i>n</i> Bu ₂ -BBQ	>23	37
(OMe) ₂ -BBQ	<1.0	60

^a Ratio of the 20th capacity to the 1st value. ^b The capacity was rapidly decreased at around 25th cycle.

Experimental method for solubility measurement

R₂-BBQ (0.5 mg) was added to tetraglyme (0.5 mL), and the mixture was stirred at 20 °C for 5 min. The addition of 0.5 mg R₂-BBQ and stirring of the mixture were repeated, until undissolved R₂-BBQ remained after the stirring.

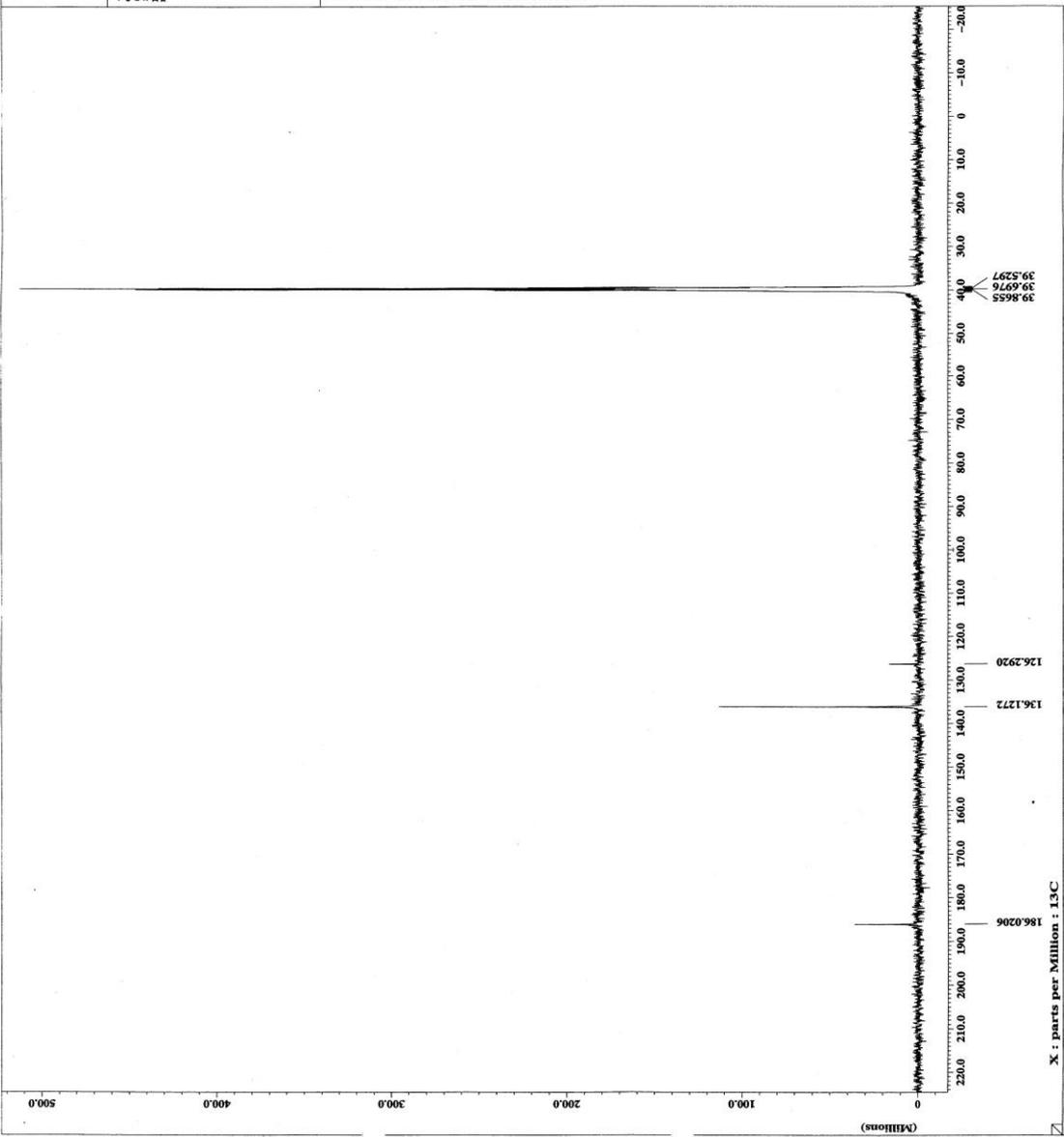
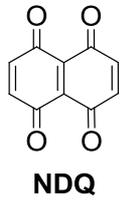
References

- 1 M. J. Frisch, G. W. Trucks, H. B. Schlegel, G. E. Scuseria, M. A. Robb, J. R. Cheeseman, G. Scalmani, V. Barone, B. Mennucci, G. A. Petersson, H. Nakatsuji, M. Caricato, X. Li, H. P. Hratchian, A. F. Izmaylov, J. Bloino, G. Zheng, J. L. Sonnenberg, M. Hada, M. Ehara, K. Toyota, R. Fukuda, J. Hasegawa, M. Ishida, T. Nakajima, Y. Honda, O. Kitao, H. Nakai, T. Vreven, J. A. Montgomery, Jr., J. E. Peralta, F. Ogliaro, M. Bearpark, J. J. Heyd, E. Brothers, K. N. Kudin, V. N. Staroverov, T. Keith, R. Kobayashi, J. Normand, K. Raghavachari, A. Rendell, J. C. Burant, S. S. Iyengar, J. Tomasi, M. Cossi, N. Rega, J. M. Millam, M. Klene, J. E. Knox, J. B. Cross, V. Bakken, C. Adamo, J. Jaramillo, R. Gomperts, R. E. Stratmann, O. Yazyev, A. J. Austin, R. Cammi, C. Pomelli, J. W. Ochterski, R. L. Martin, K. Morokuma, V. G. Zakrzewski, G. A. Voth, P. Salvador, J. J. Dannenberg, S. Dapprich, A. D. Daniels, O. Farkas, J. B. Foresman, J. V. Ortiz, J. Cioslowski and D. J. Fox, Gaussian 09, Revision C.01, Gaussian, Inc., Wallingford CT, 2010.
- 2 W. J. Hehre, L. Radom, P. V. R. Schleyer and P. A. Pople, *Ab initio Molecular Orbital Theory*, Wiley, New York, 1986.
- 3 R. Dennington, T. Keith and J. Millam, Semichem Inc., Shawnee Mission, KS, 2009.



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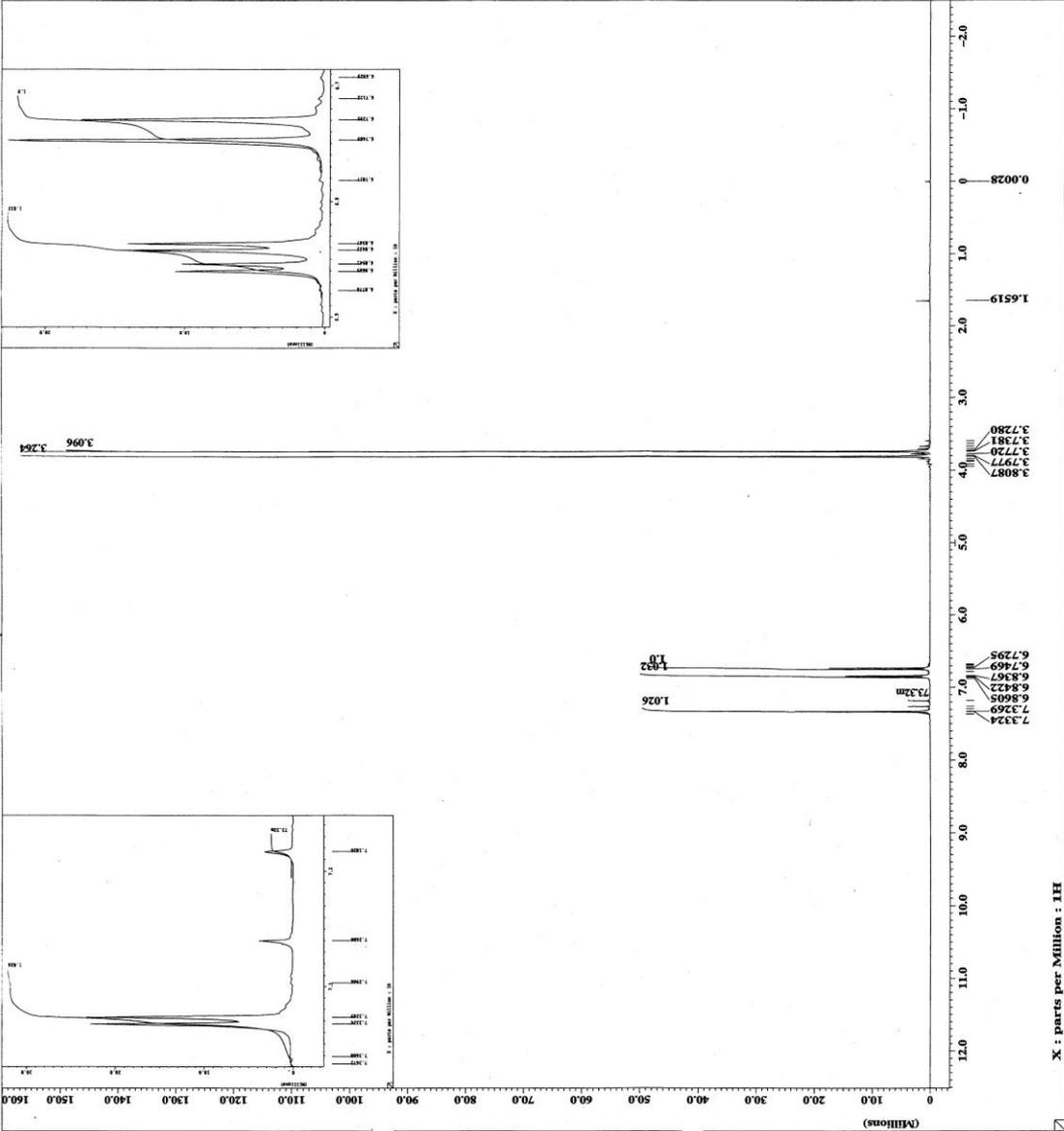
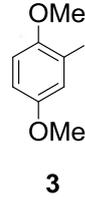
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X_resolution = 1.44454088 (Hz)
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Solvent = DMSO-D6
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PPM

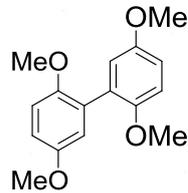
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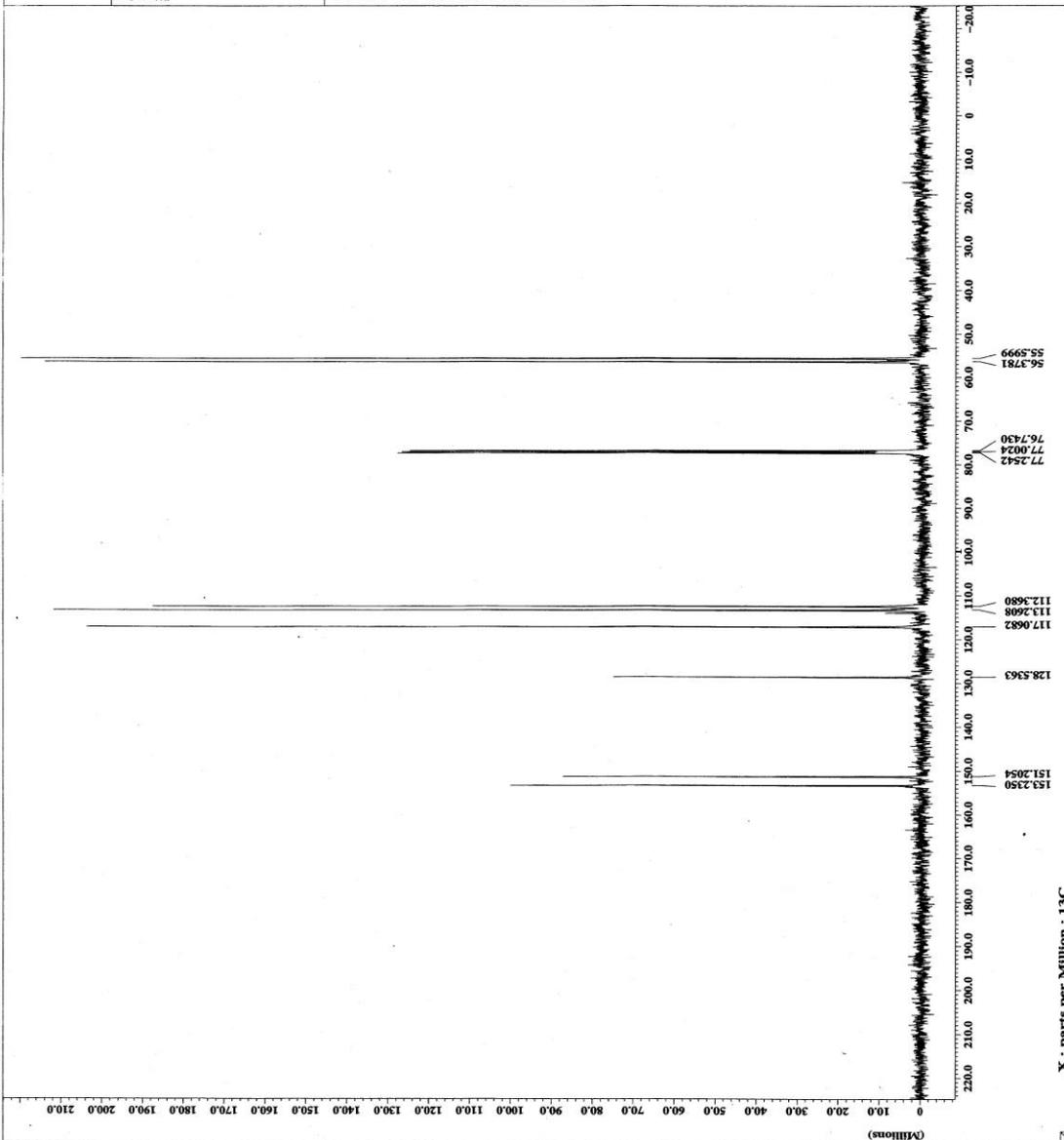


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SFC 1
SamplePhase

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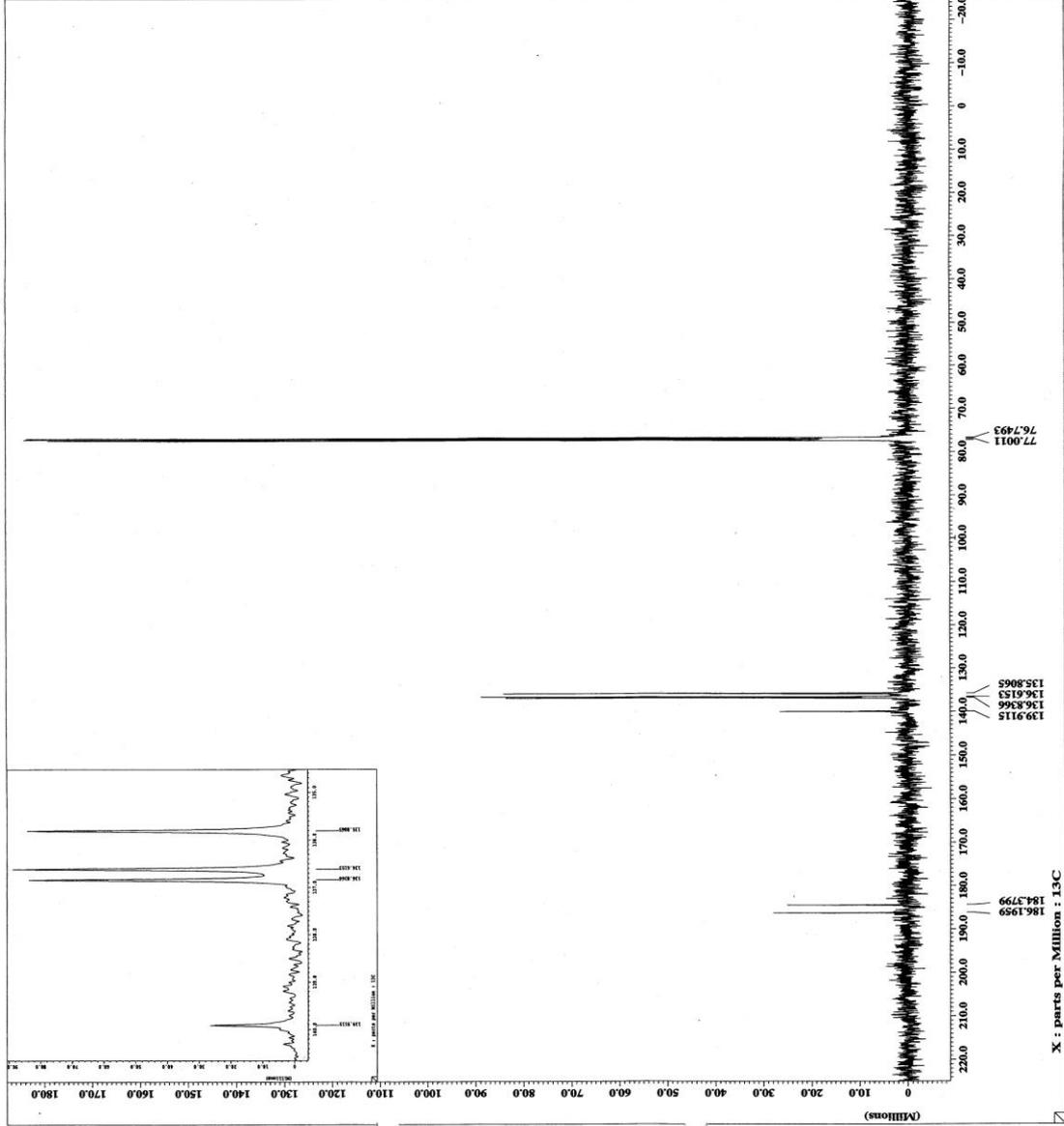
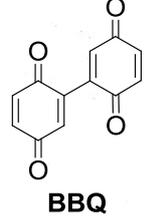
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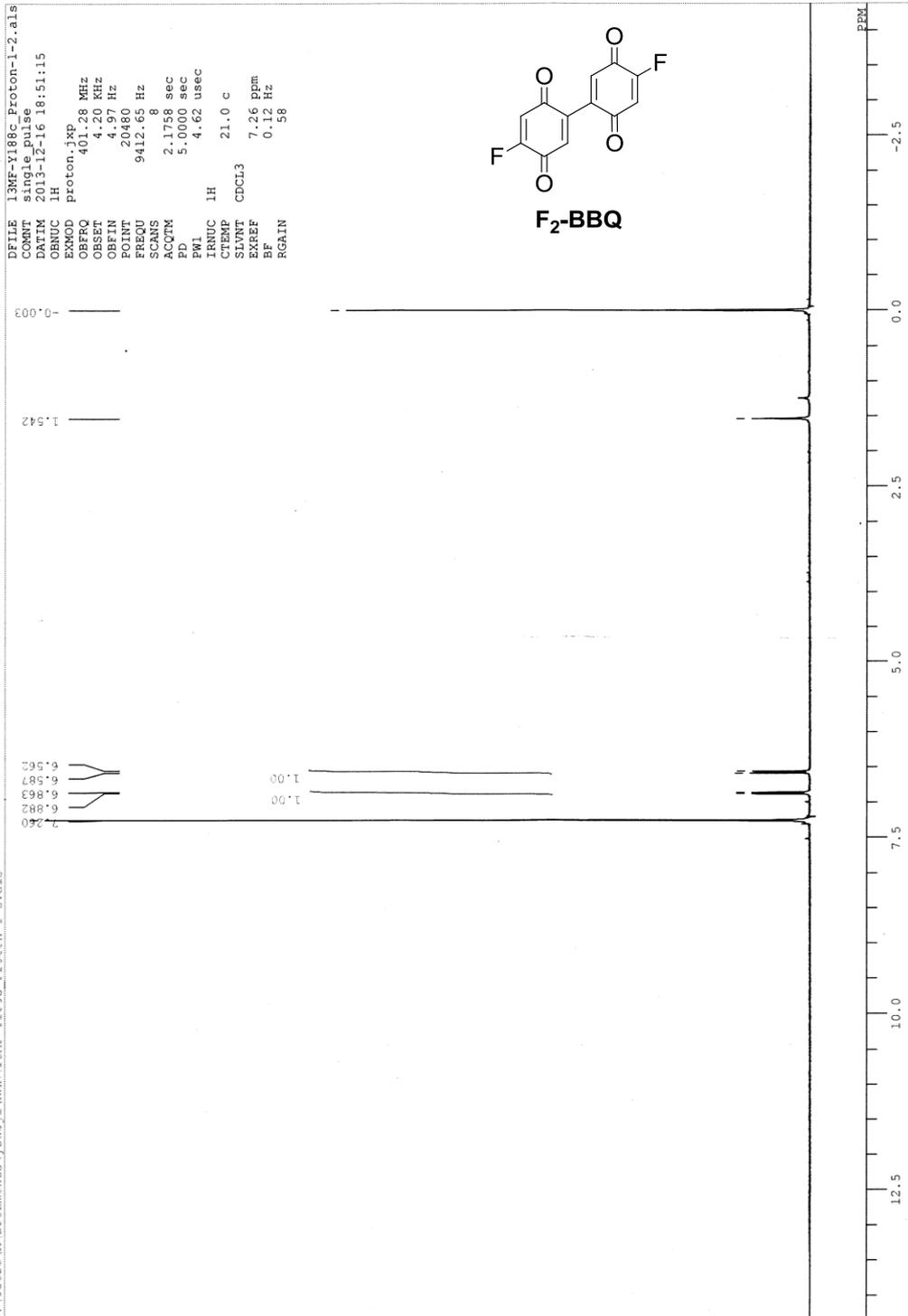


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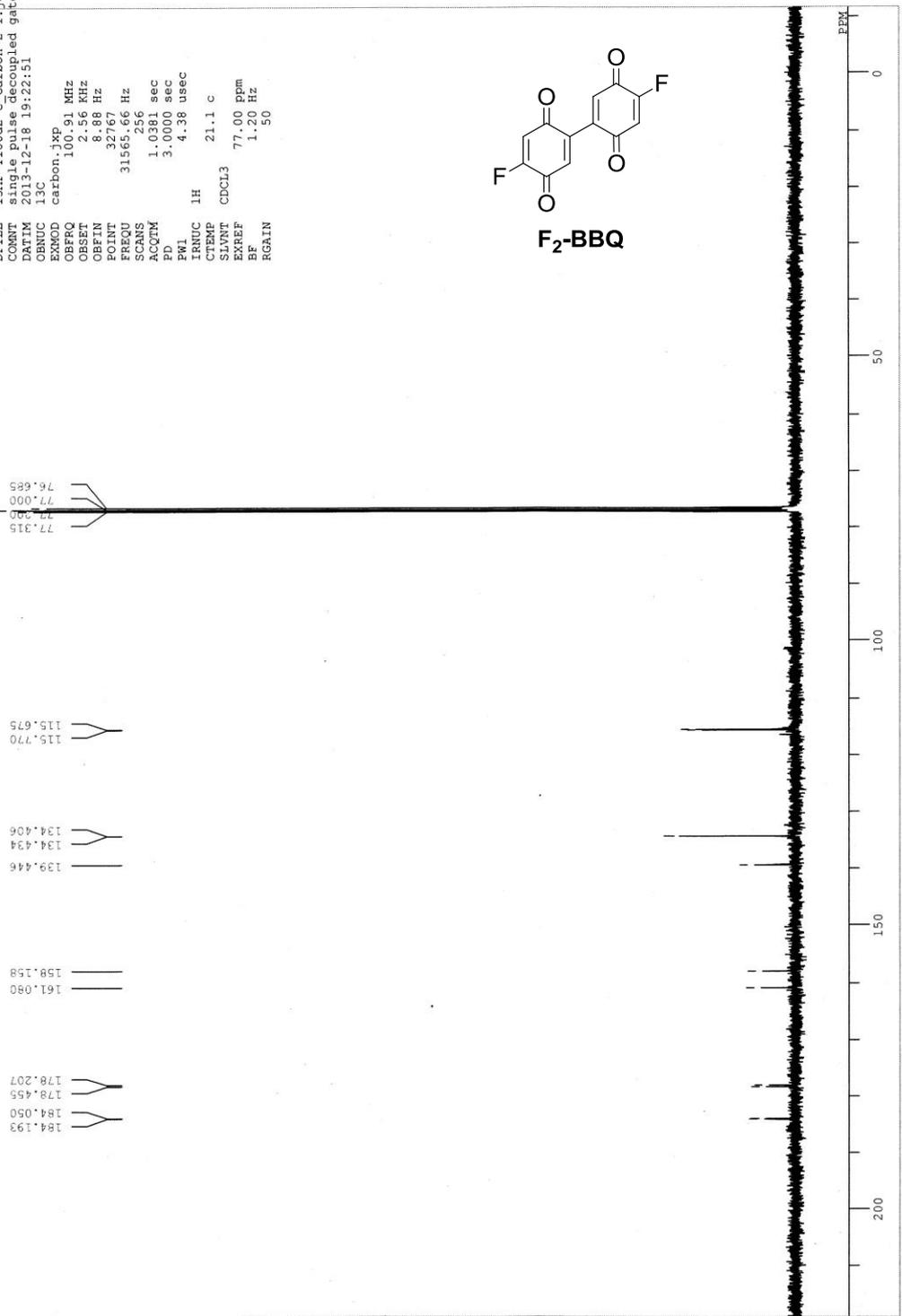
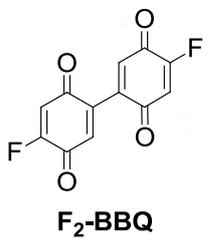
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Nuc98 = 13C
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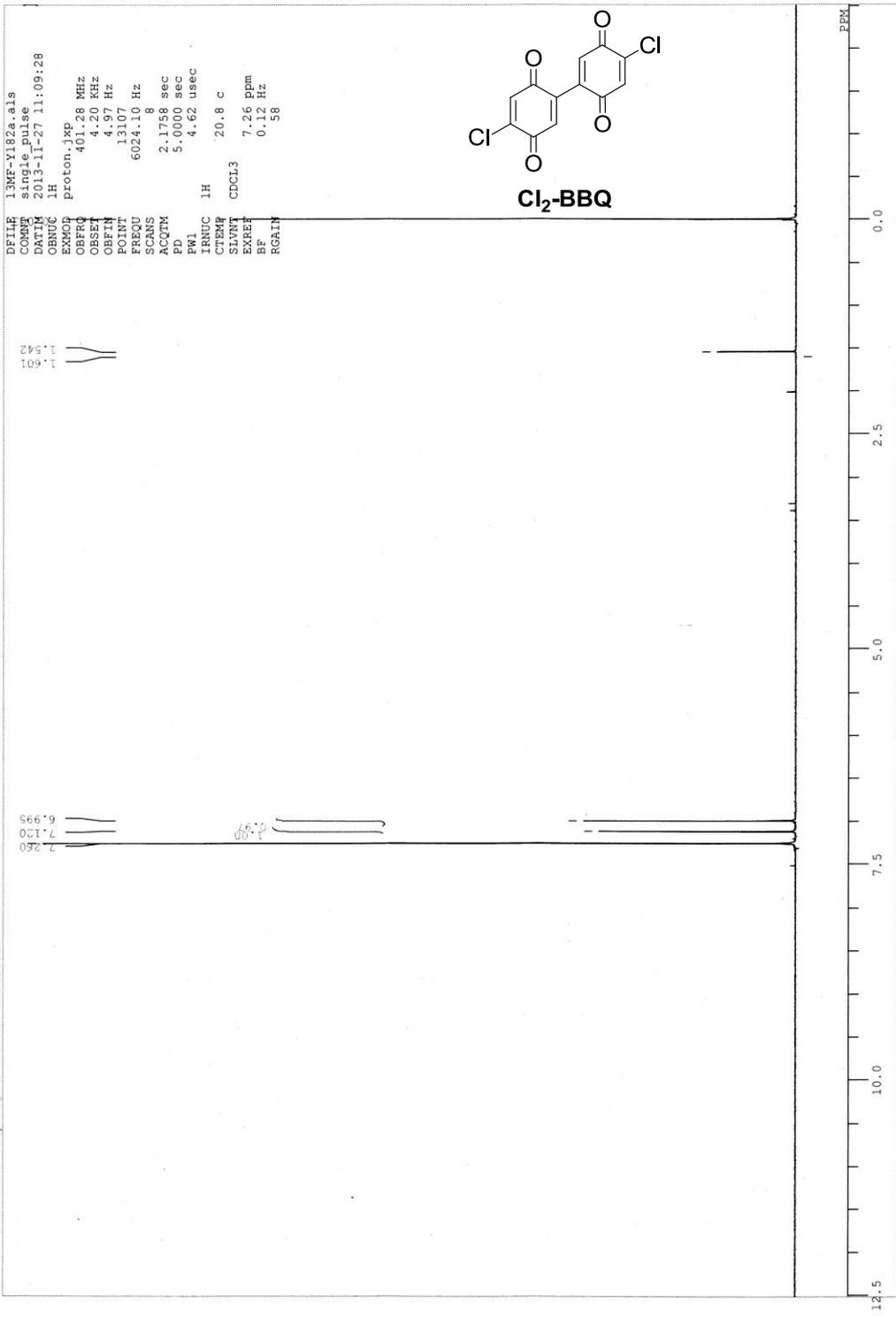
C:\User\A\Documents\yokoji\NMR\13MF-YI86c-Proton-1-2.als

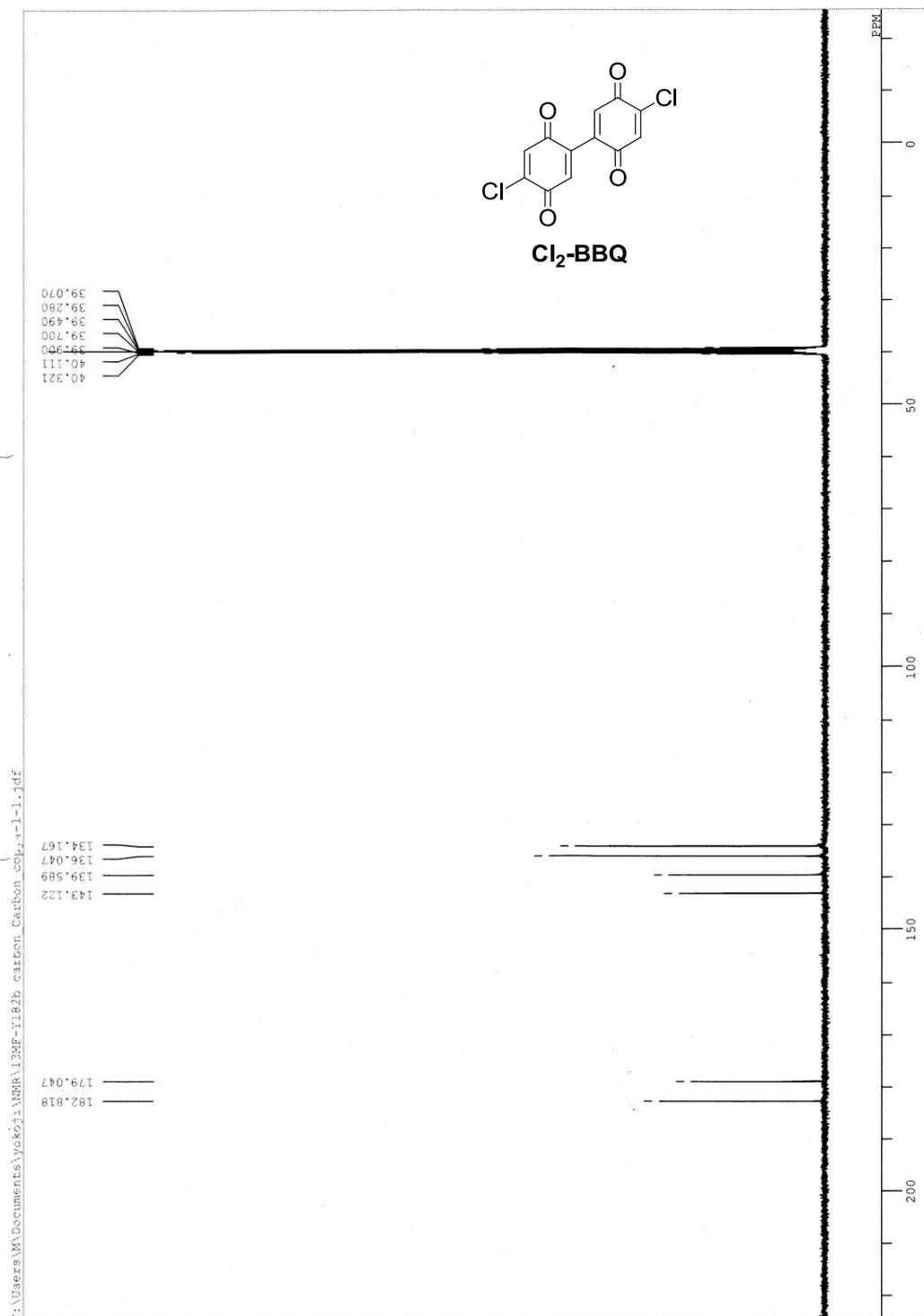


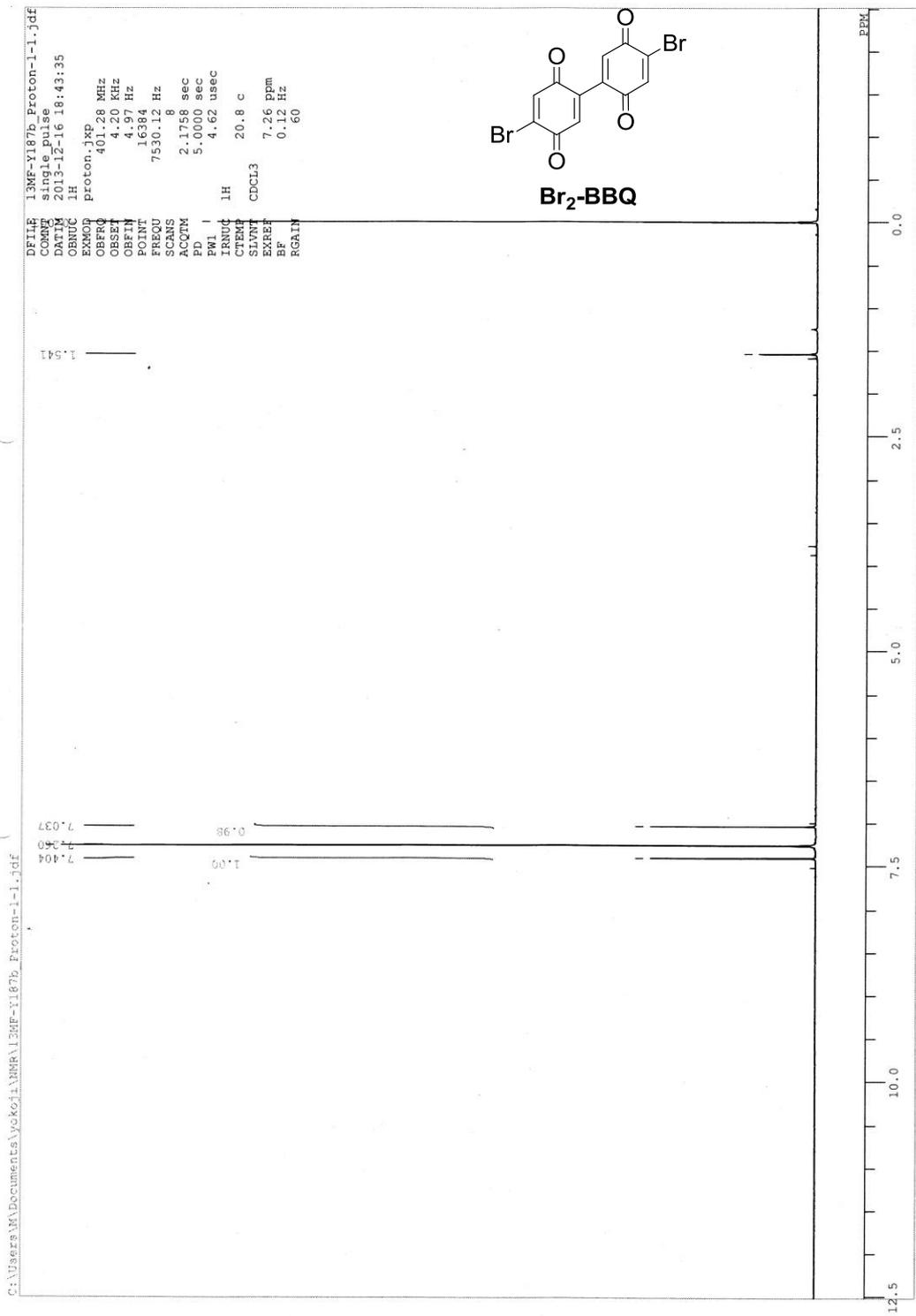
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 DFILE 13MF-VI88d2_c_Carbon-2-1.fdi
 COMNT single pulse decoupled gated NOE
 DATIM 2013-12-18 19:22:51
 ORNUC 13C
 EXMOD carbon_jxp
 ORFRO 100.91 MHz
 ORSET 2.56 KHz
 ORFIN 8.88 Hz
 POINT 32767
 FREOU 31565.66 Hz
 SCANS 256
 ACQTM 1.0381 sec
 PD 3.0000 sec
 PW1 4.38 usec
 IRNUC 1H
 CTEMP 21.1 C
 SLVNT CDCL3
 EXREF 77.00 ppm
 BF 1.20 Hz
 RGAIN 50



C:\Users\VA\Documents\yoko\1\NMR\13MF-Y182a.als

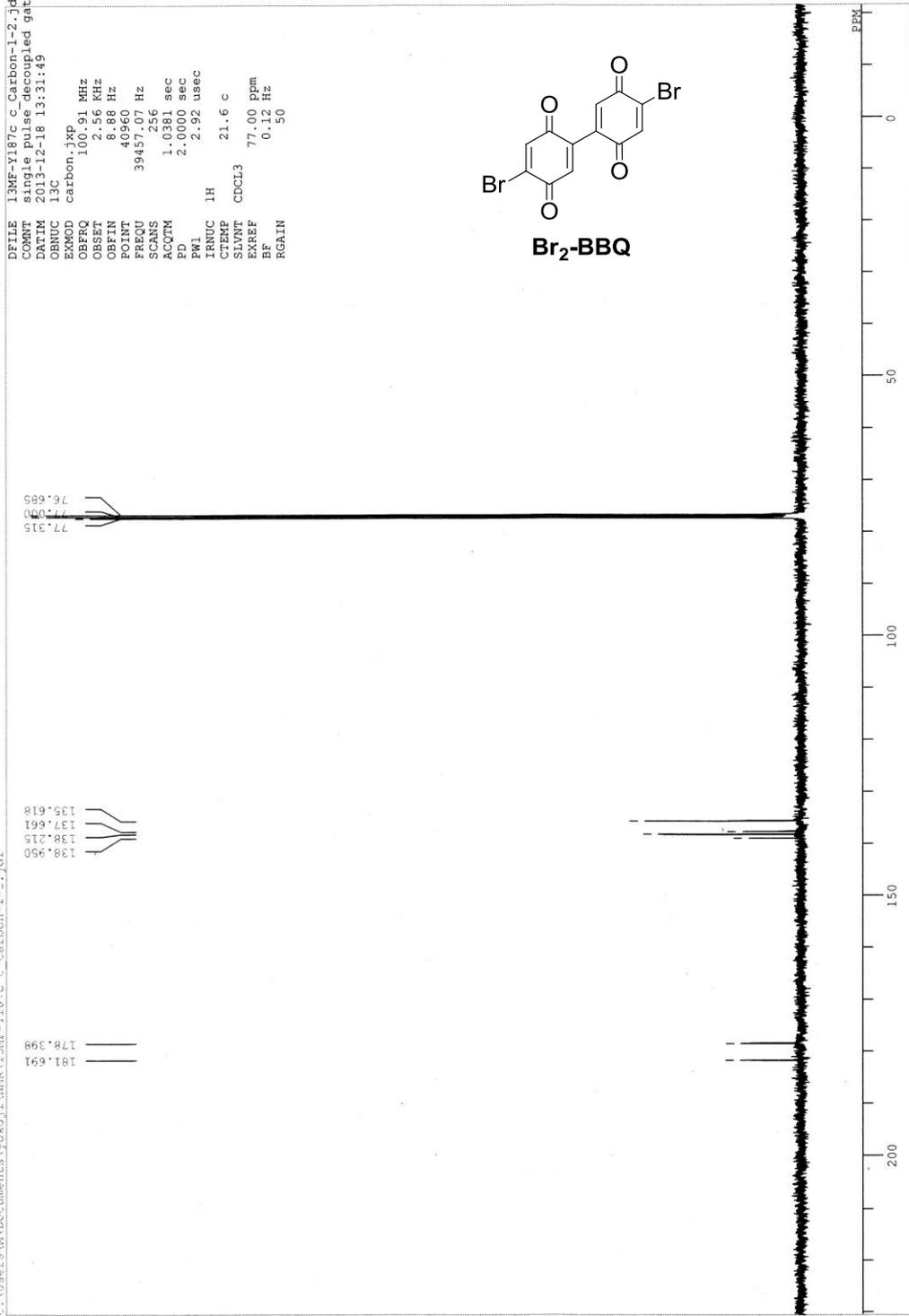
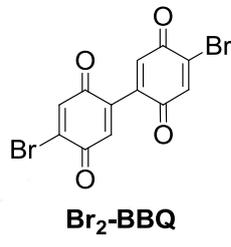






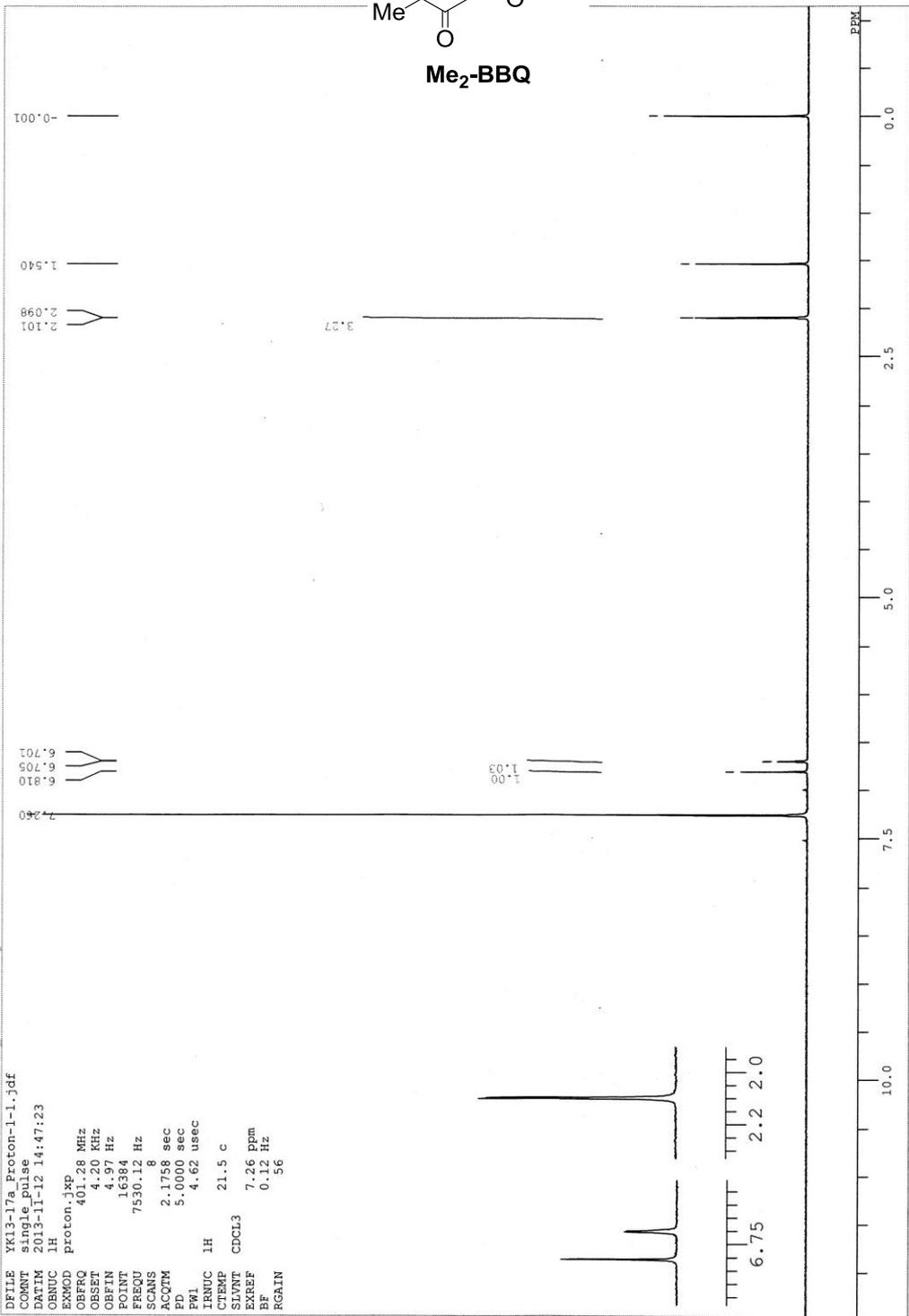
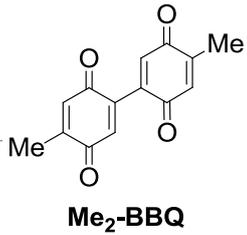
D:\Users\W\Documents\work\1\NMR\13C\13C-12-18\13C-12-18-13:31:49
 L3MF-V187c c_Carbon-1-2_1.jdf
 single pulse decoupled gated NOE

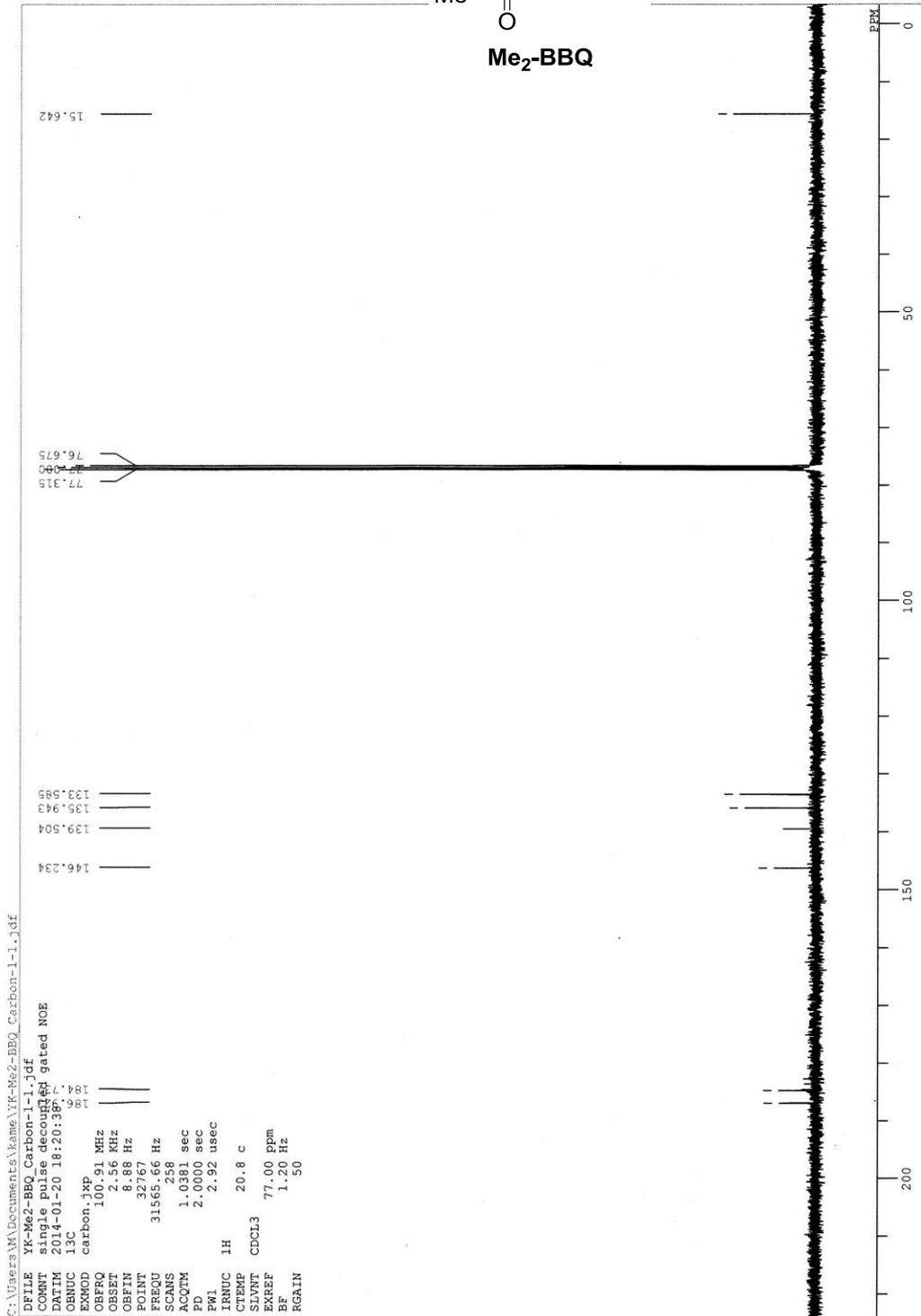
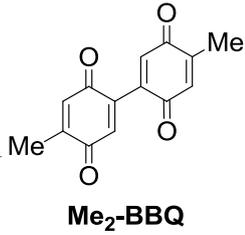
DATE_ 2013-12-18 13:31:49
 COM1 13C
 EXMOD carbon_1xp
 OBFO 100.91 MHz
 OBSE 2.56 KHz
 OFIN 8.88 Hz
 POINT 40960
 FREQ 39457.07 Hz
 SCANS 256
 ACQTM 1.0381 sec
 PD 2.0000 sec
 PW1 2.92 usec
 IRNUC 1H
 CTEMP 21.6 c
 SLVNT CDCl3
 EXREF 77.00 ppm
 BF 0.12 Hz
 RGAIN 50

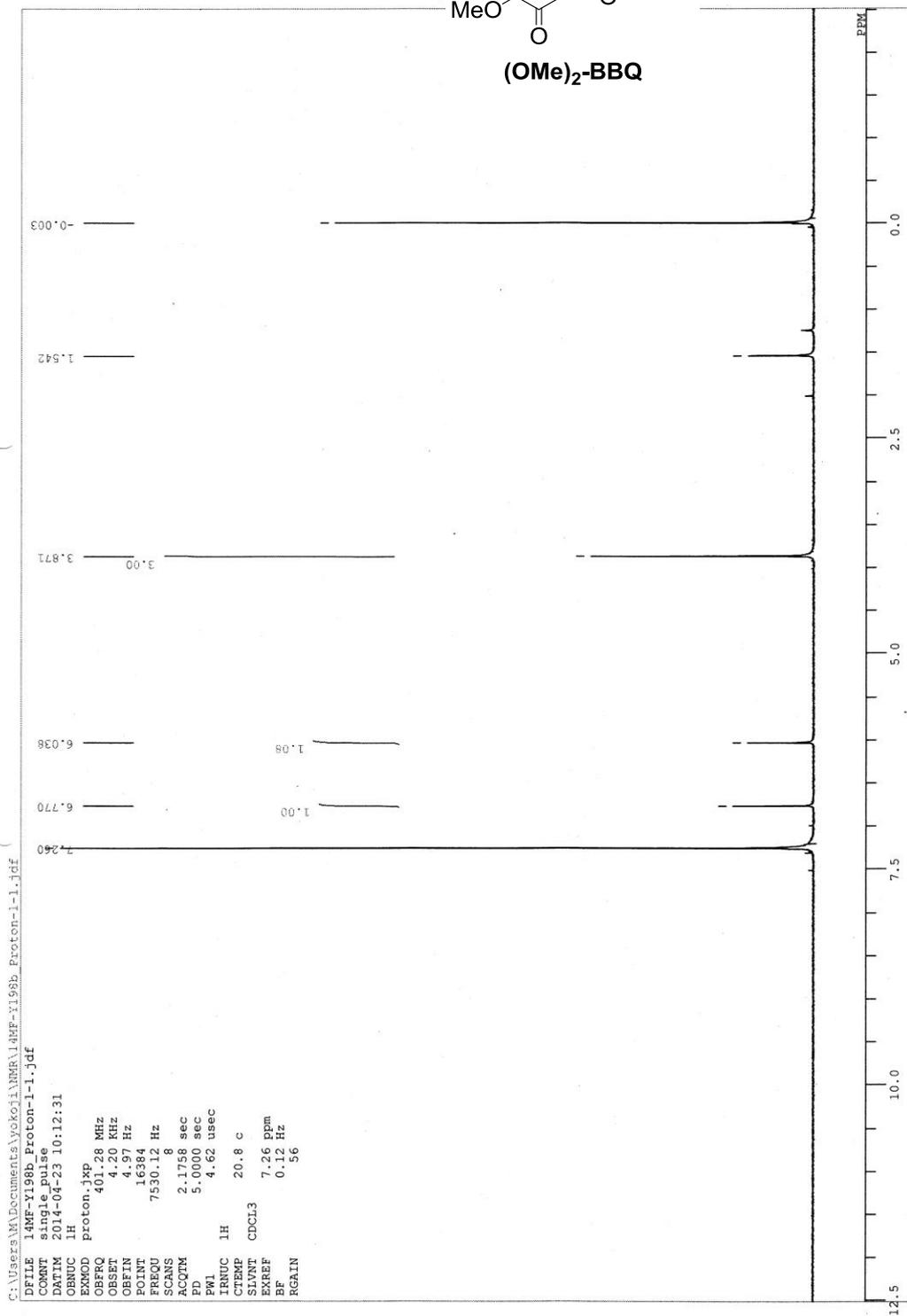
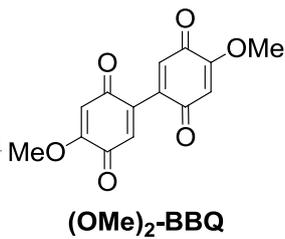


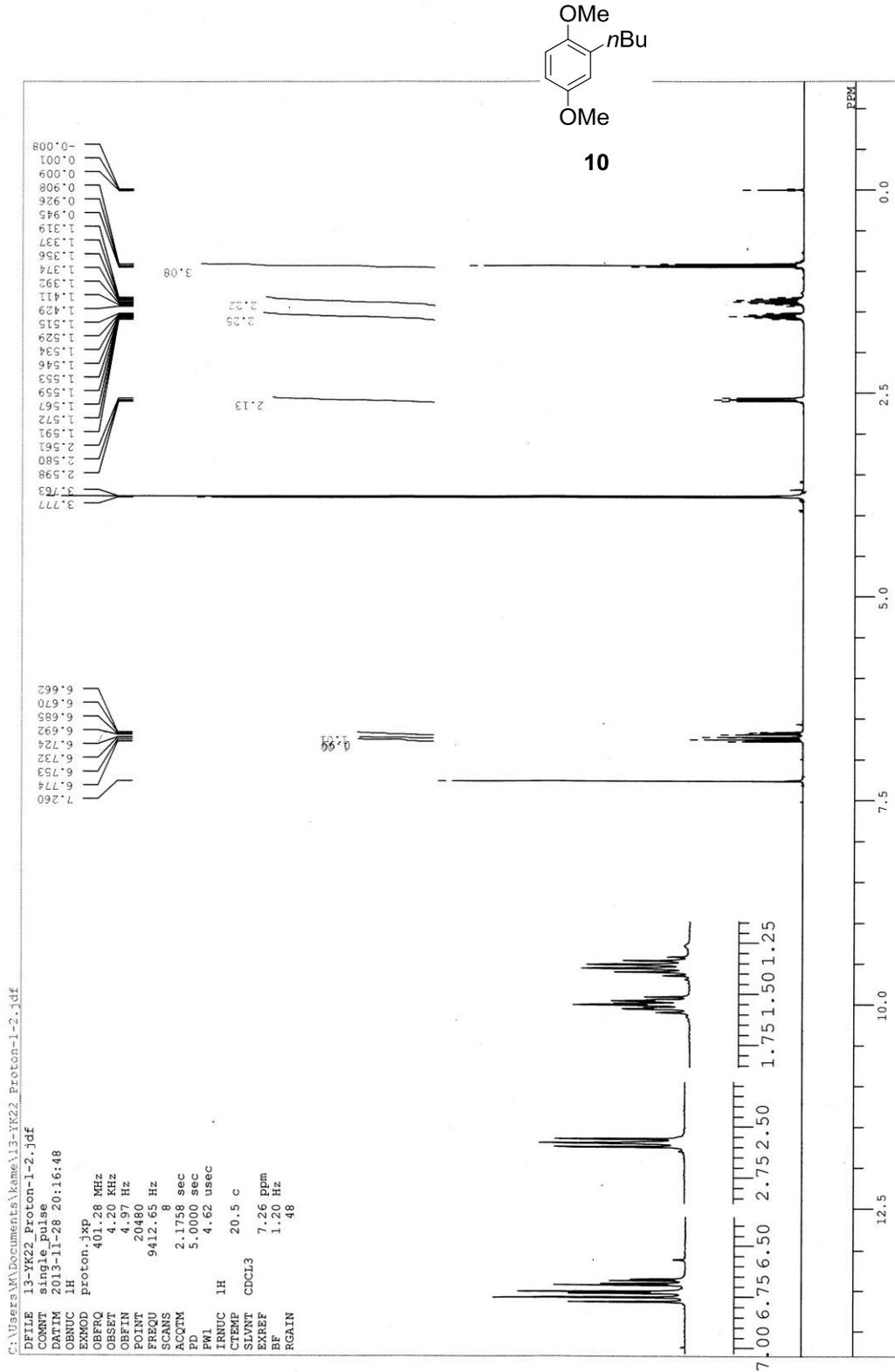
C:\Users\A\Documents\kame\YK13-17a_Proton-1-1.jbf

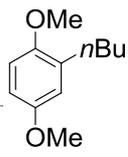
DFILE YK13-17a_Proton-1-1.jbf
 COMMENT single_pulse
 DATIM 2013-11-12 14:47:23
 OBNUC 1H
 EXMOD proton.jxp
 OBFREQ 401.28 MHz
 OBSSET 4.20 KHz
 OBFIN 4.97 Hz
 POINT 16384
 FREQU 7530.12 Hz
 SCANS 2.1758 sec
 SCOTM 5.0000 sec
 PFI 4.62 usec
 INUC 1H
 TEMPE 21.5 C
 SOLVT CDCL3
 SLOTT 7.26 ppm
 EXREF BF
 RGAIN 56



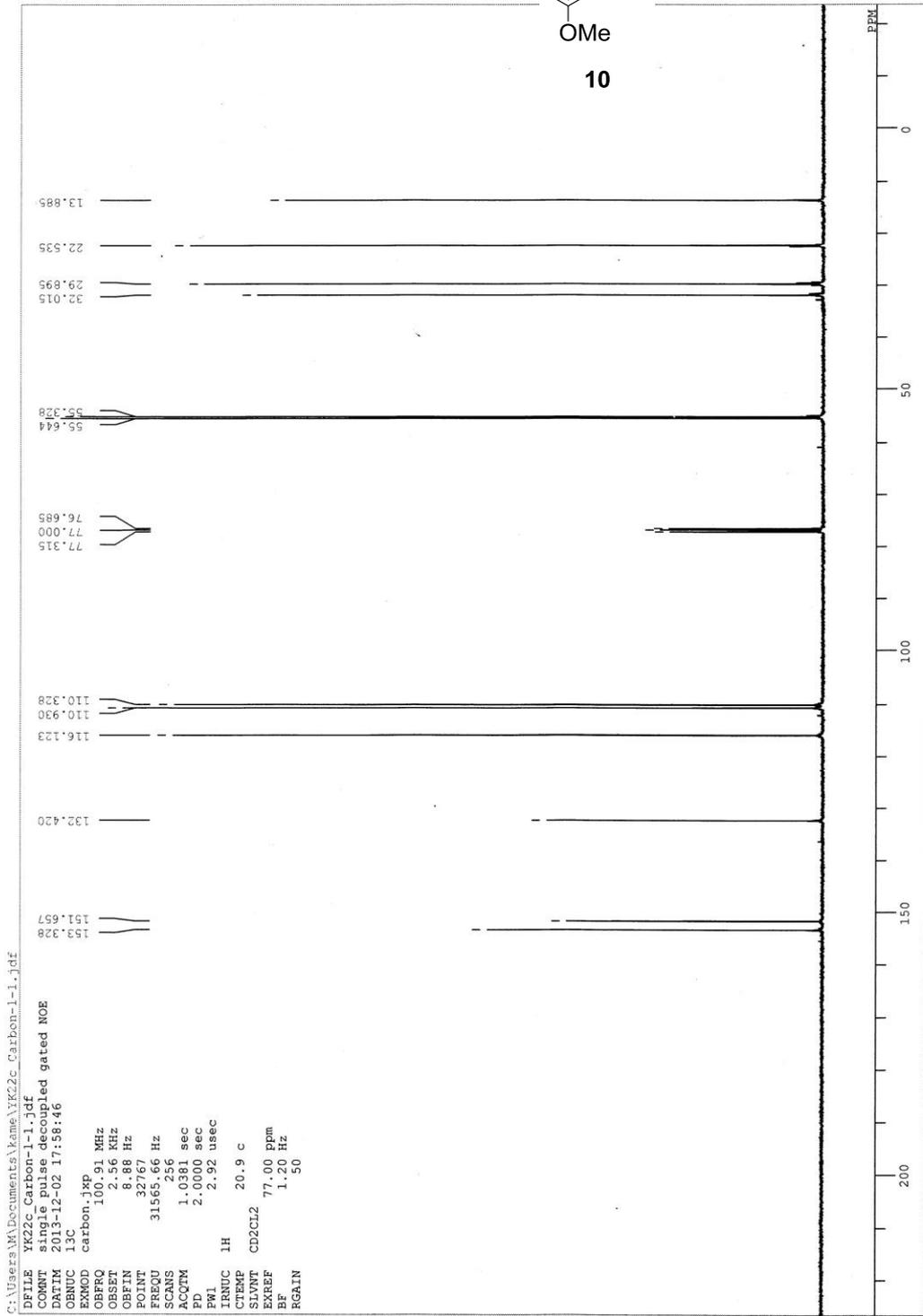








10



C:\Users\A\Documents\kame\YK22c_Carbon-1-1-1.jdf

```

DEFILE YK22c_Carbon-1-1-1.jdf
COMNT sing15 pulse decoupled gated NOE
DATIM 2013-12-02 17:58:46
ORNIC 13C
EXMOD carbon_1xp
ORPRO 100.91 MHz
ORSET 2.56 KHz
ORFIN 8.88 Hz
POINT 32767
FREQU 31565.66 Hz
SCANS 256
ACQTM 1.0381 sec
PD 2.0000 sec
PW 2.92 usec
IRNIC 1H
CTEMP 20.9 C
SIVNT CD2CL2
EXREF 77.00 ppm
BF 1.20 Hz
RGAIN 50
  
```

C:\Users\W\Documents\kame\13-YK24a Proton-1-2.jdf

DTITLE 13-YK24a Proton-1-2.jdf

COMNT single pulse

DAYIM 2013-12-04 13:24:14

ORNUC 1H

EXMOD proton.jxp

OBFRQ 401.28 MHz

OBSEI 4.20 KHz

OBFIN 4.57 Hz

PRGFI 20483

PRGFI 9412.65 Hz

SCANS

ACQTM 2.1758 sec

PC 5.0000 sec

PR 4.62 usec

IRNUC 1H

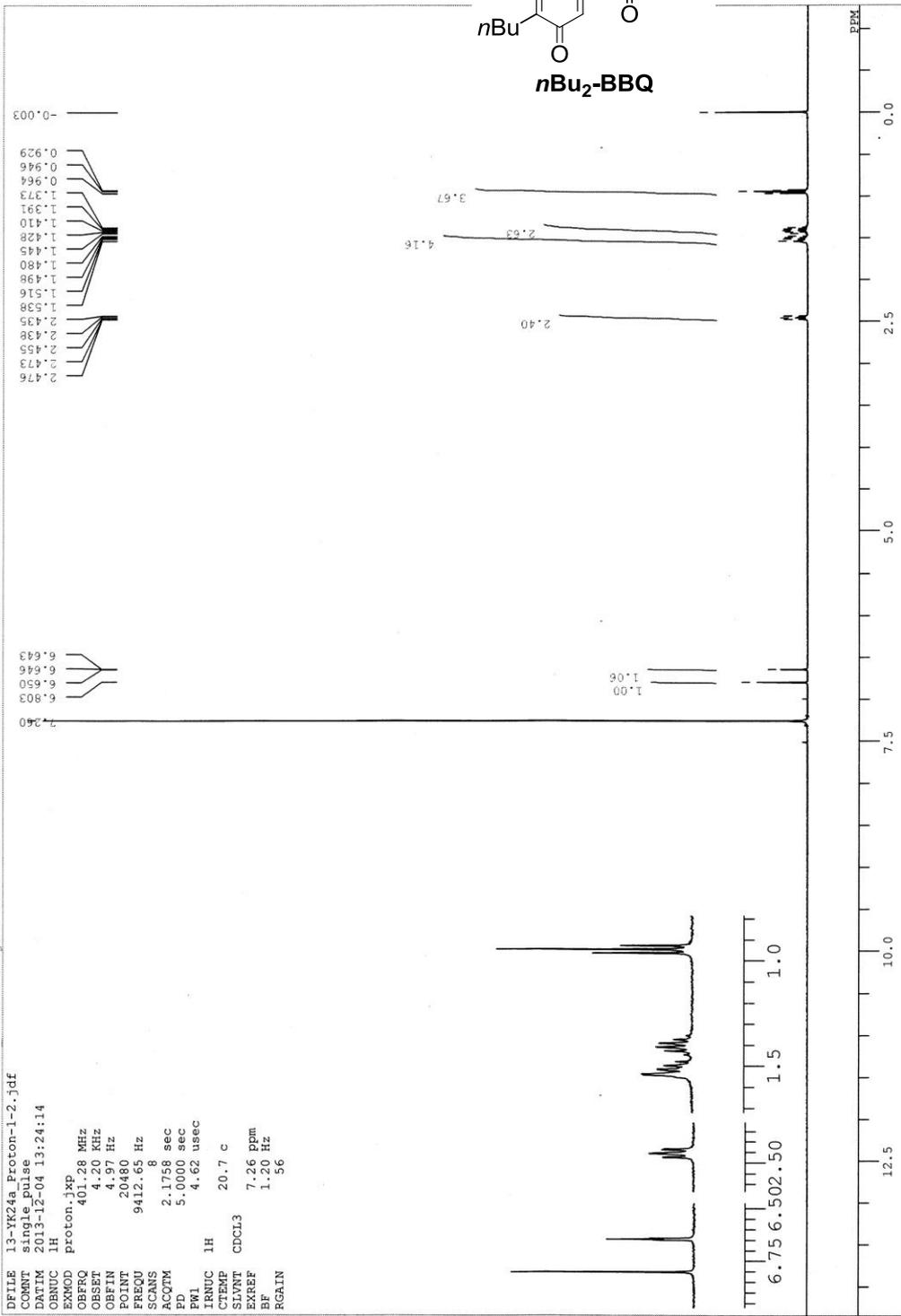
CTEMP 20.7 C

SOLVT CDCL3

EXREF 7.26 ppm

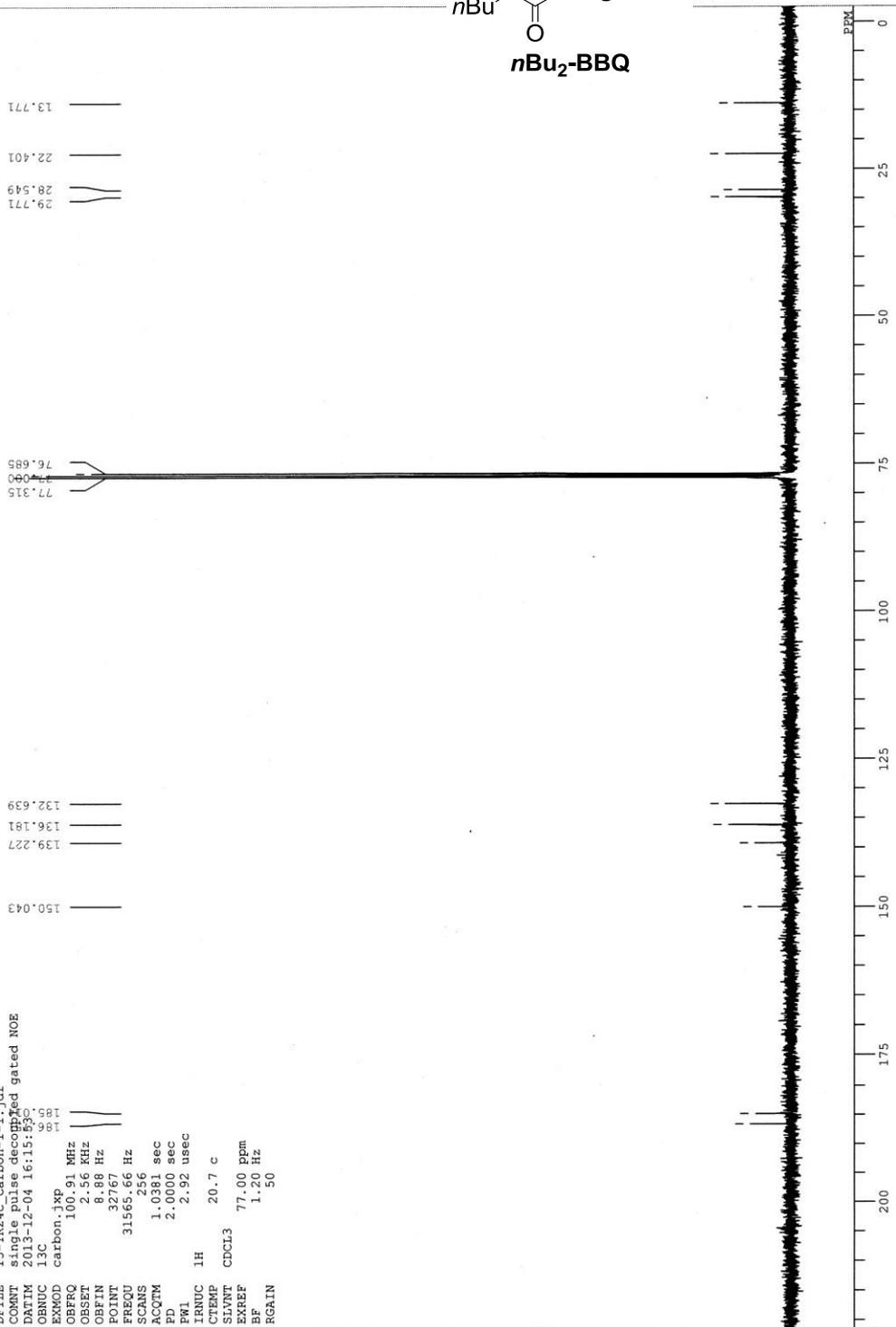
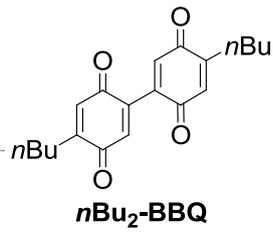
BF 1.20 Hz

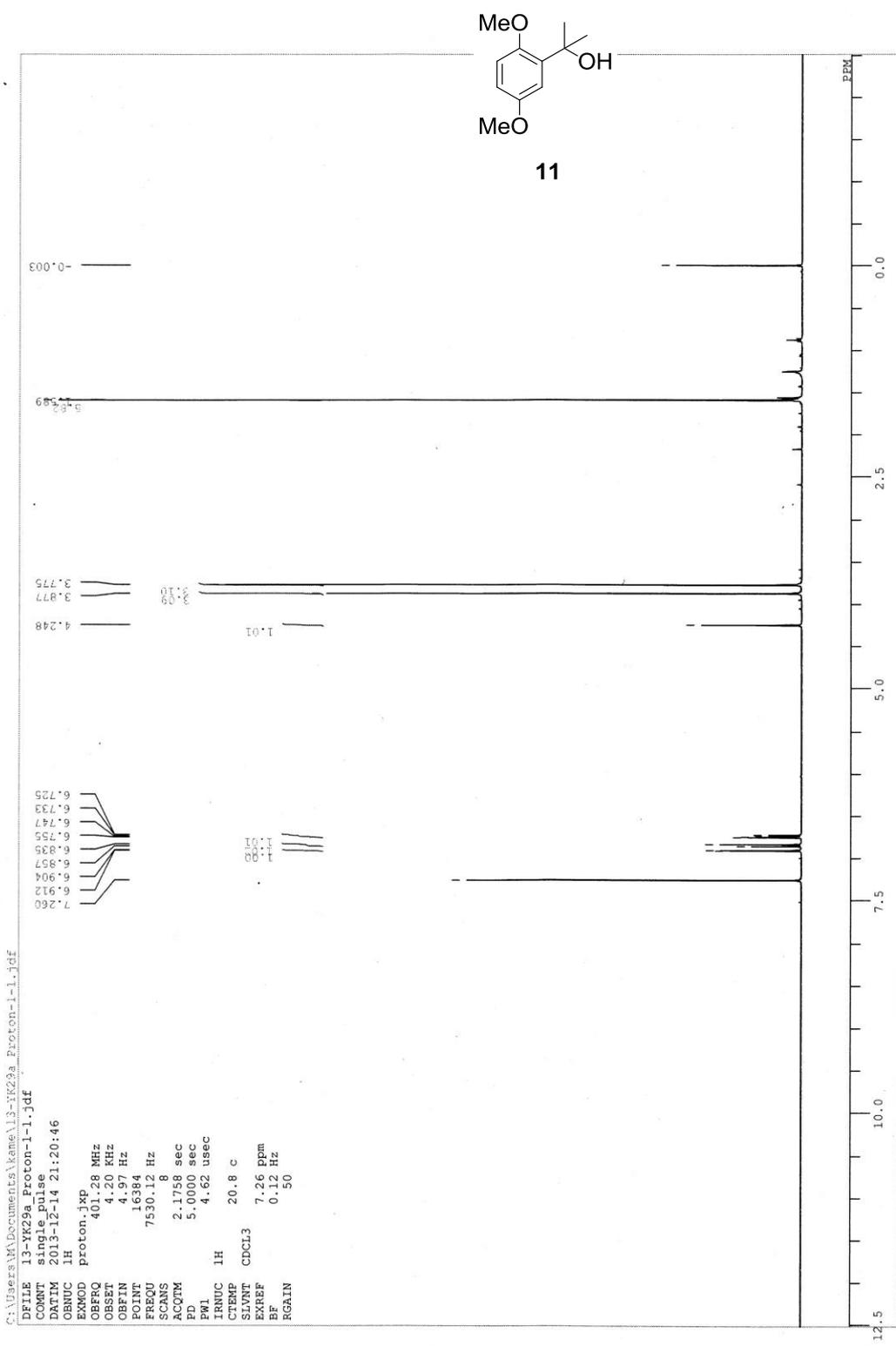
RGAIN 56

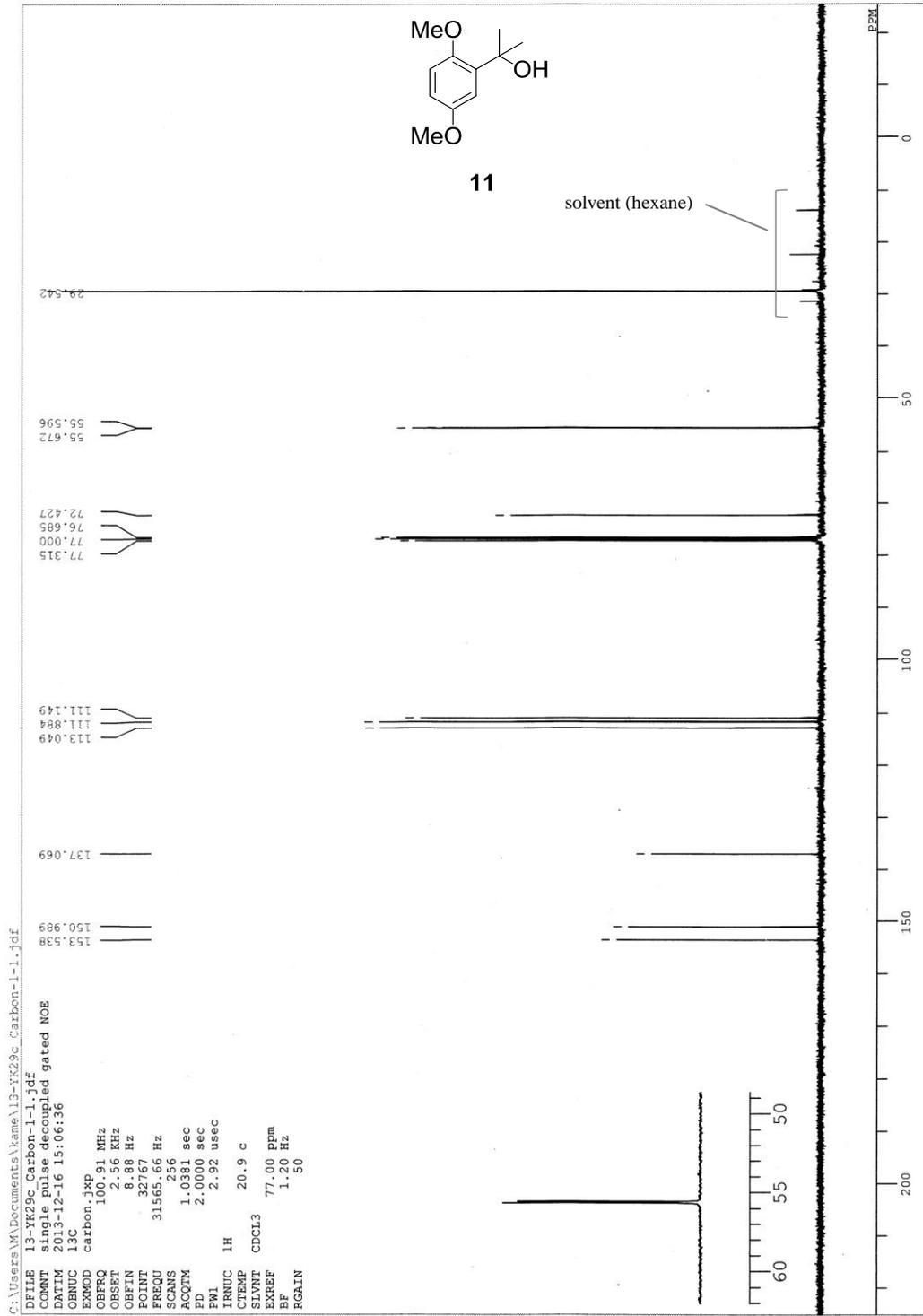


C:\Users\RAM\Documents\kame\13-YK24c_Carbon-1-1-1.jdf

DFIL 13-YK24c_Carbon-1-1-1.jdf
 COM1 single pulse decoupled gated NOE
 DATE 2013-12-04 16:15:55
 OBNUC 13C
 EXMOD carbon_1XP
 OBSFRQ 100.61 MHz
 OBSSET 4.86 kHz
 OFFSET 32.69 Hz
 FREQ1 31565.66 Hz
 SCANS 56
 ACQTM 1.0381 sec
 PD 2.0000 sec
 PUL 2.92 usec
 INUC 1H
 TEMPC 20.7 C
 STUNT CDCL3
 EXREF 77.00 ppm
 BF 1.20 Hz
 RGAIN 50







C:\Users\MA\Documents\kama\13-YK31b_1-Proton-1-1.jdf

FILE 13-YK31b_1-Proton-1-1.jdf

COM1 single pulse

DATE 2013-12-20 12:13:58

ORNUC 1H

EXMOD proton.jxp

OBFRQ 401.28 MHz

OBSEI 4.20 KHz

ROBIN 14.84 Hz

RGAIN 7530.12 Hz

SCANS 16

ACQTM 2.1758 sec

PC 5.0000 sec

PW 4.62 usec

IRNUC 1H

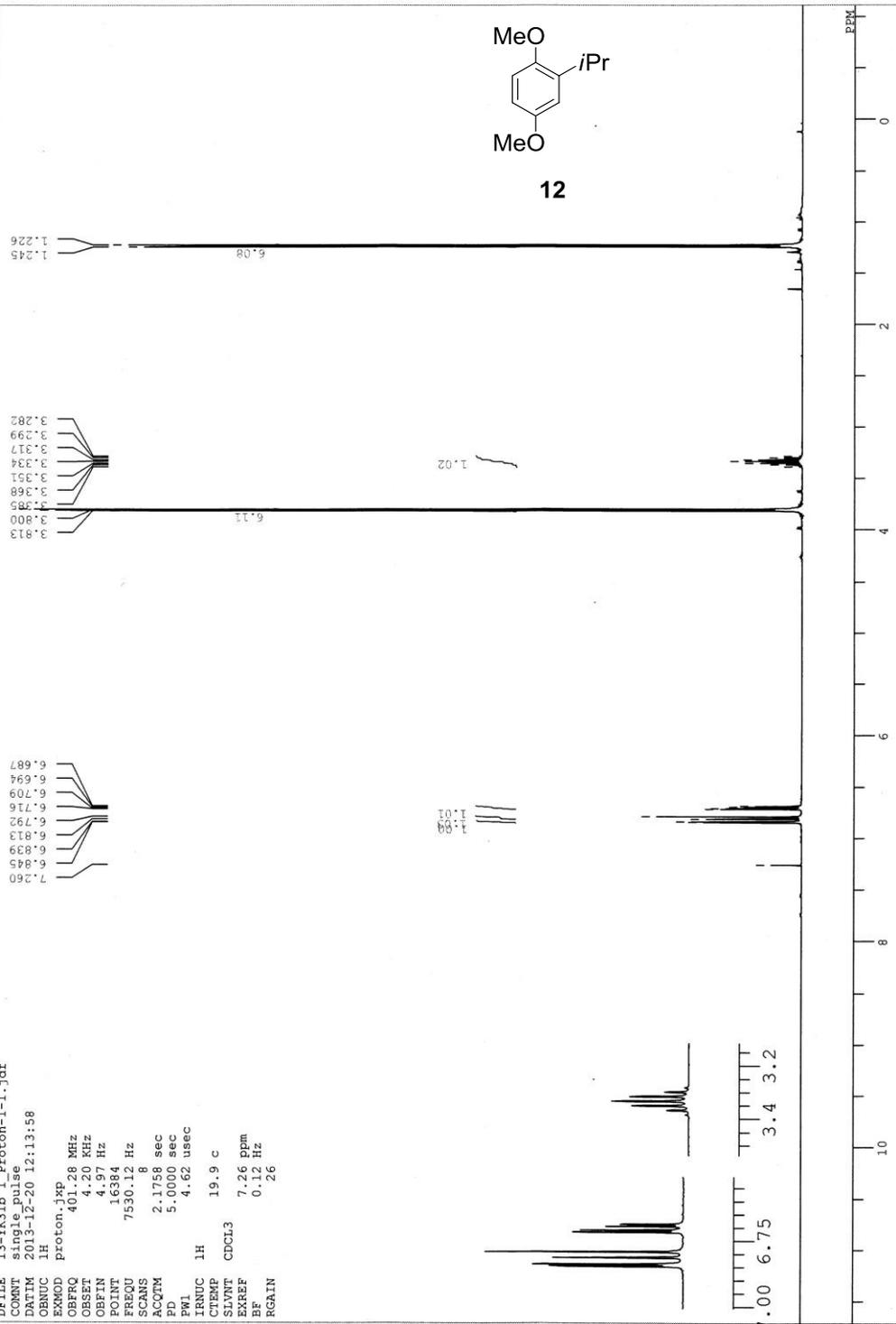
CTEMP 19.9 C

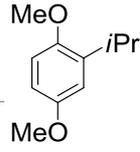
SOLNT CDCl3

EXREF 7.26 ppm

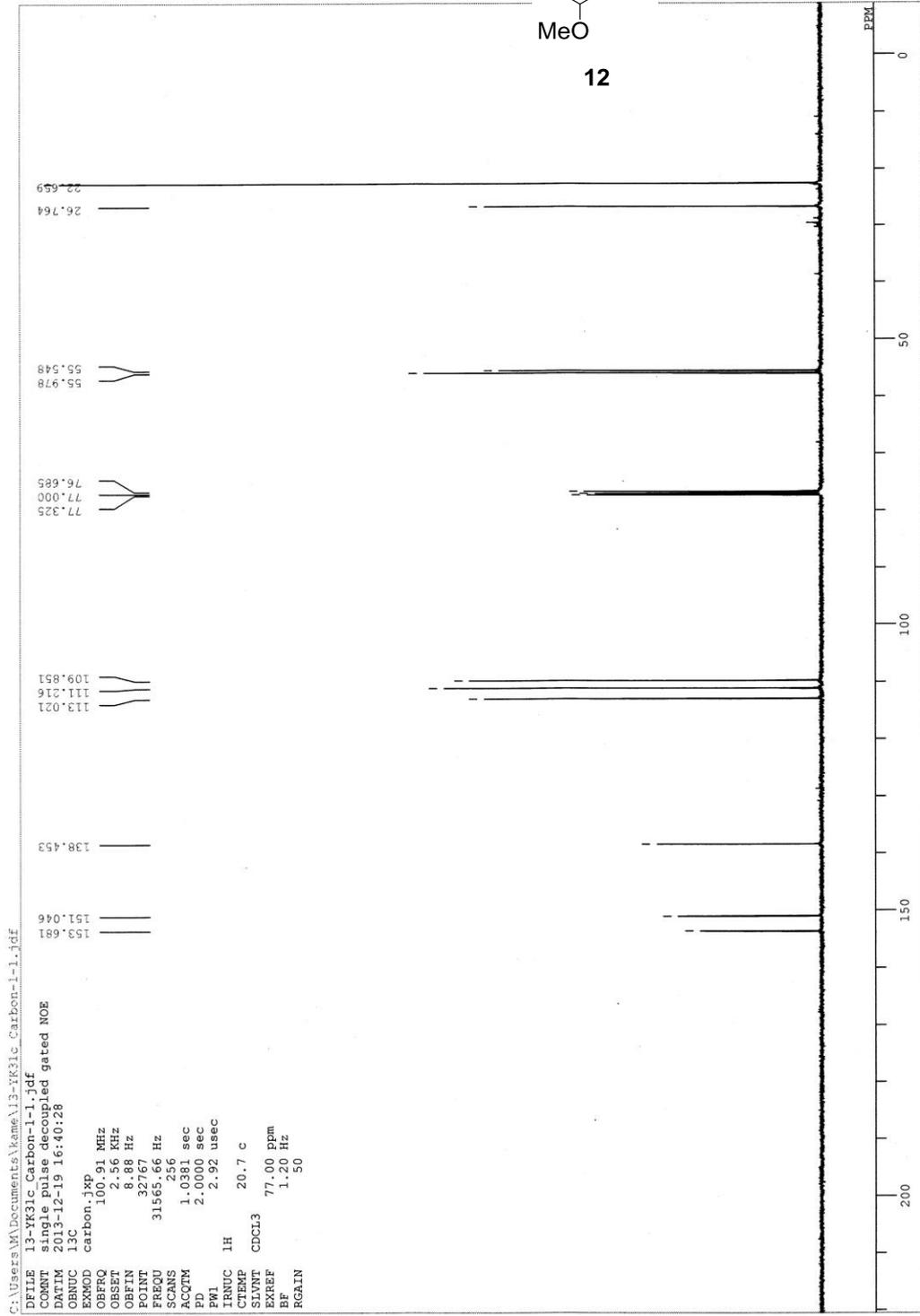
BF 0.12 Hz

RGAIN 26



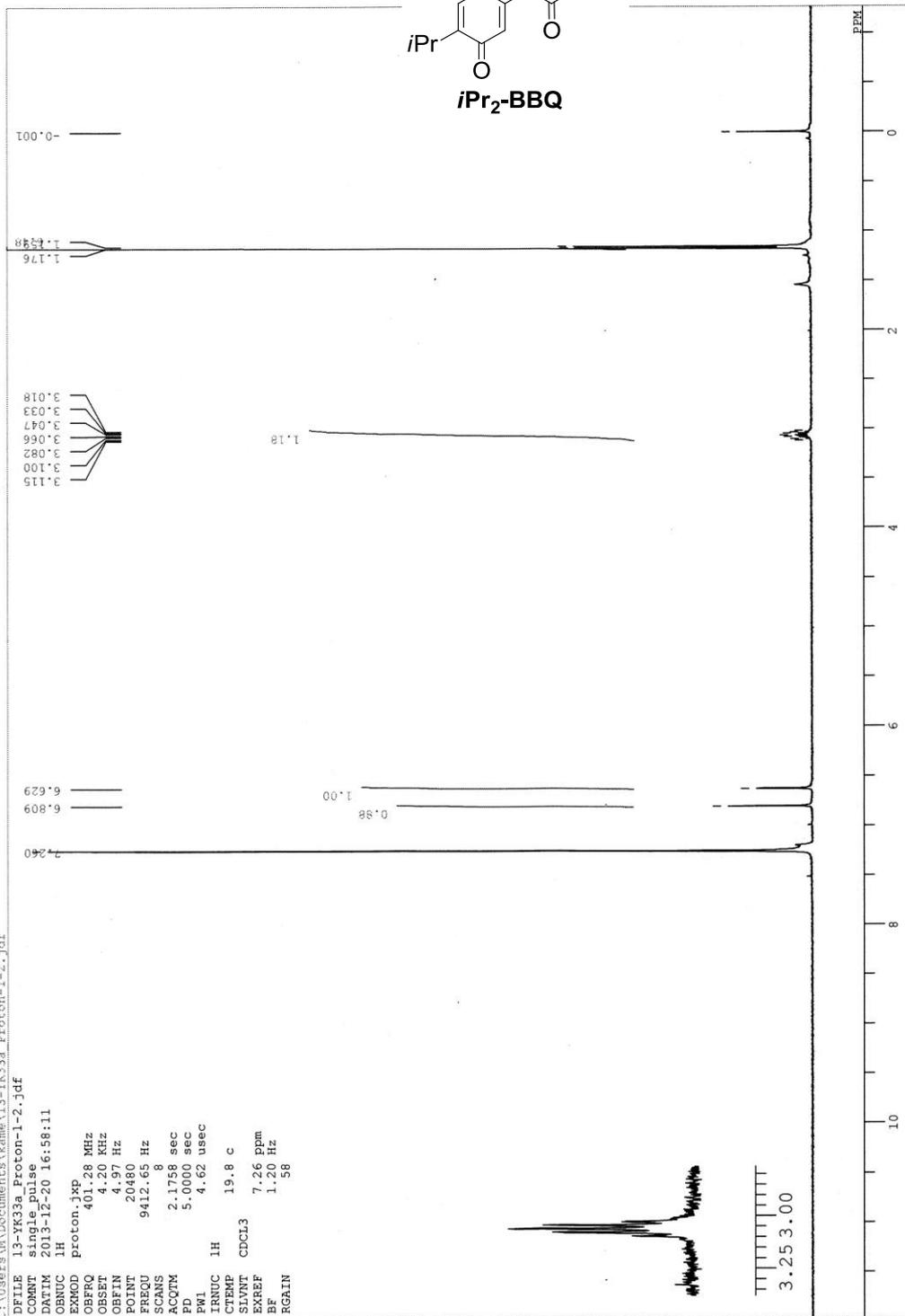
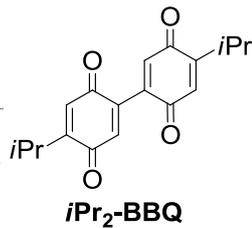


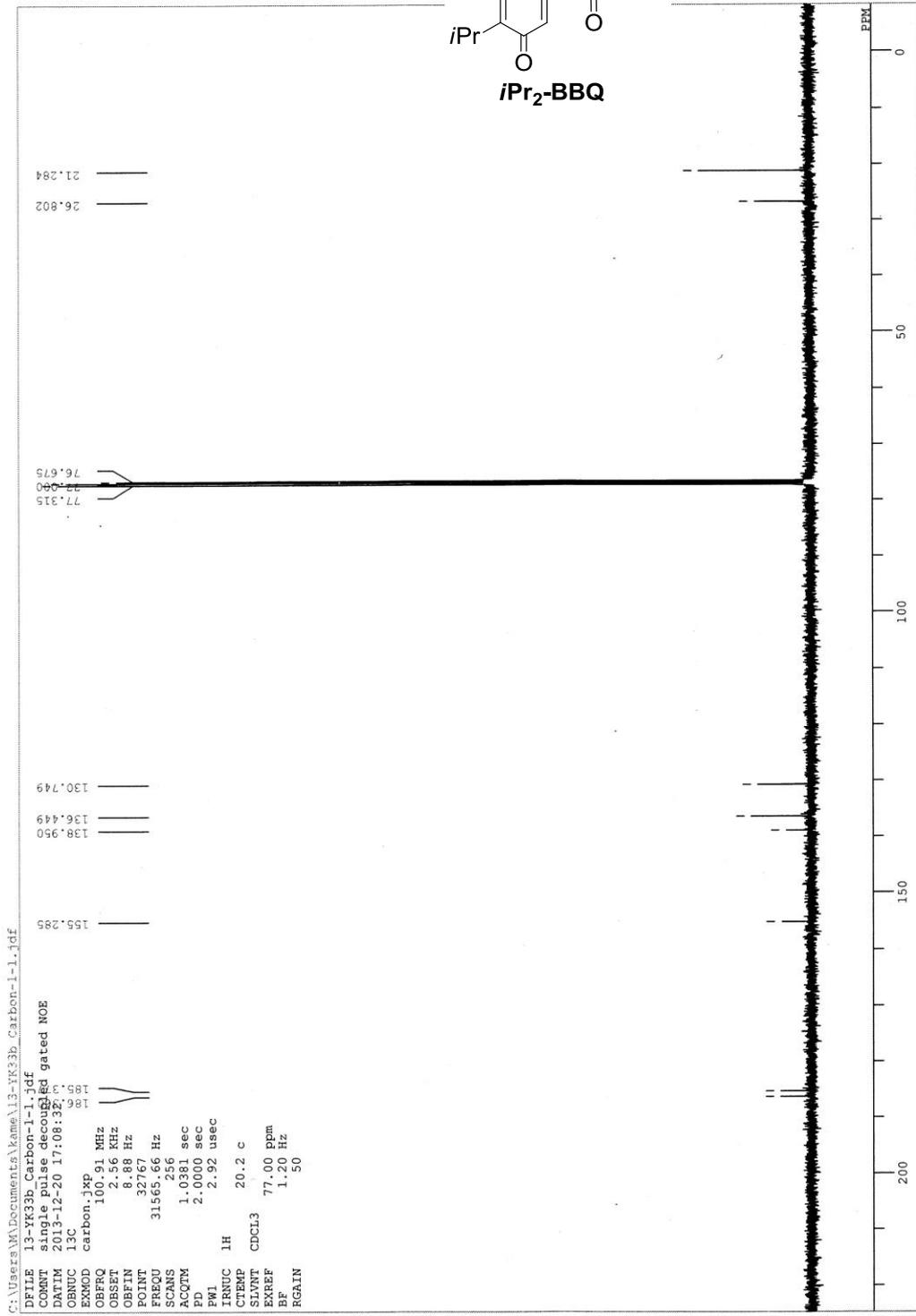
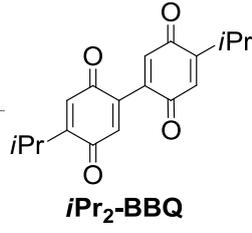
12

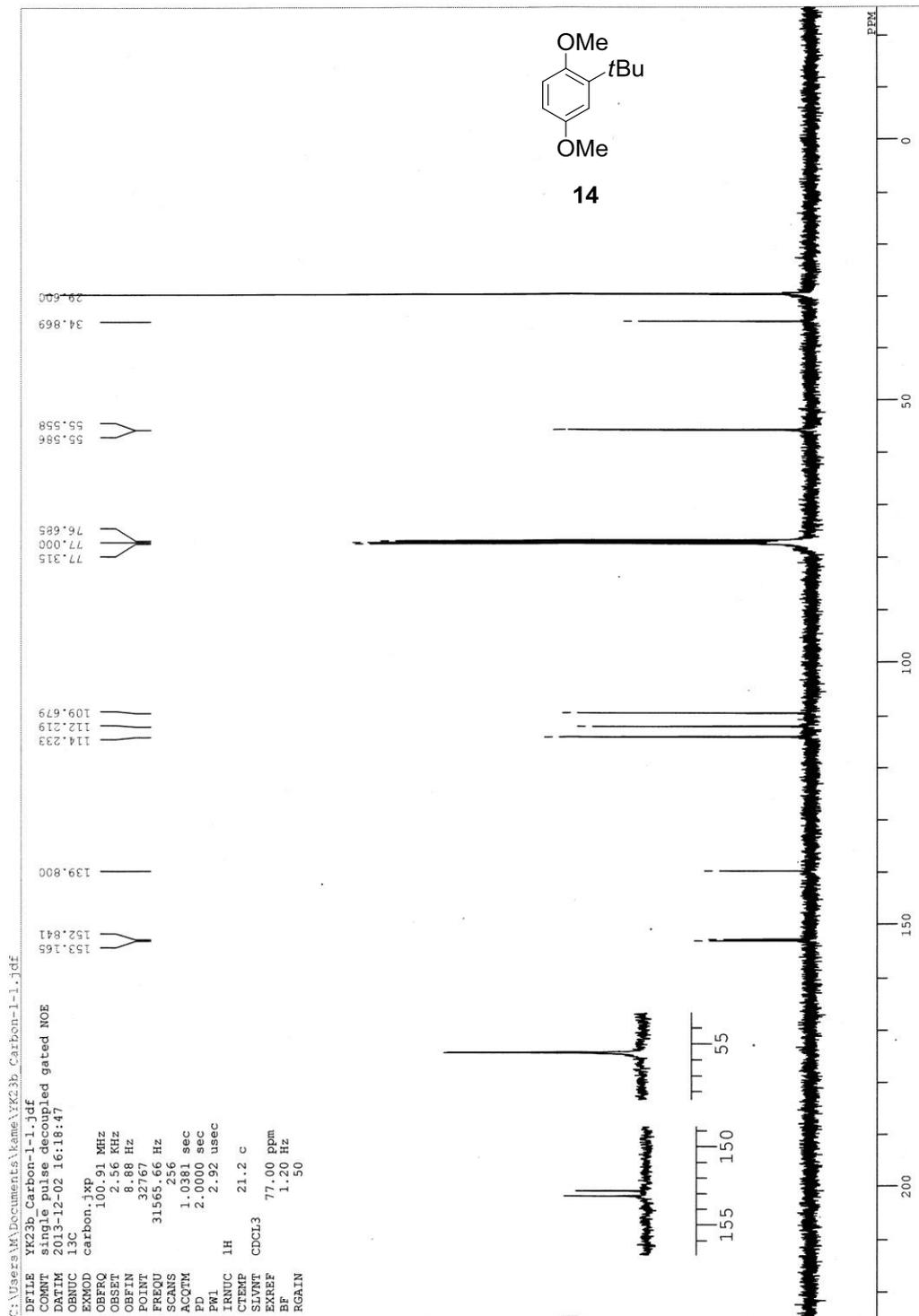


C:\Users\VA\Documents\kama\13-YK33a_Proton-1-2.jdf

DFILE 13-YK33a_Proton-1-2.jdf
 COM1 single_pulse
 DATIM 2013-12-20 16:58:11
 OBNUC 1H
 EXMOD proton_1xp
 OBFRQ 401.28 MHz
 OBSET 4.20 KHz
 OBFIN 4.97 Hz
 POINT 20480
 FRFQ 9412.63 Hz
 SCANS 2.1758 sec
 ACQIM 5.0000 sec
 FID 4.62 usec
 INUC 1H 19.8 C
 CTEMP CDCL3 7.26 ppm
 SUMP EXREF BF 1.20 Hz
 RGAIN 58







C:\Users\A\Documents\kama\13-YK25a_Proton-1-4.jdf

DFILE 13-YK25a_Proton-1-4.jdf
COMNT single pulse
DATIM 2013-12-03 22:32:13
ORNUC 1H
EXMOD proton.jxp
OBFRQ 401.28 MHz
OBSET 4.20 KHz
OBFIN 4.97 Hz
POINT 16400
FREQU 7530.12 Hz
SCANS 8
ACQTM 2.1758 sec
PD 5.0000 sec
PW 4.62 usec
IRNUC 1H 20.3 c
CTEMP
SLVNT CDCl3 7.26 ppm
EXREF
BF 1.20 Hz
RGAIN 56

