

## Supporting Information

### **Hexagonal Magnetite Nanoprisms: Preparation, Characterization and Cellular Uptake**

*Hongwang Wang\**, *Tej B. Shrestha*, *Matthew T. Basel*, *Marla Pyle*, *Yubisela Toledo*,  
*Andrew Konecny*, *Prem Thapa*, *Myles Ikenberry*, *Keith L. Hohn*, *Viktor Chikan*,  
*Deryl L. Troyer*, *Stefan H. Bossmann\**

#### **Experimental Materials**

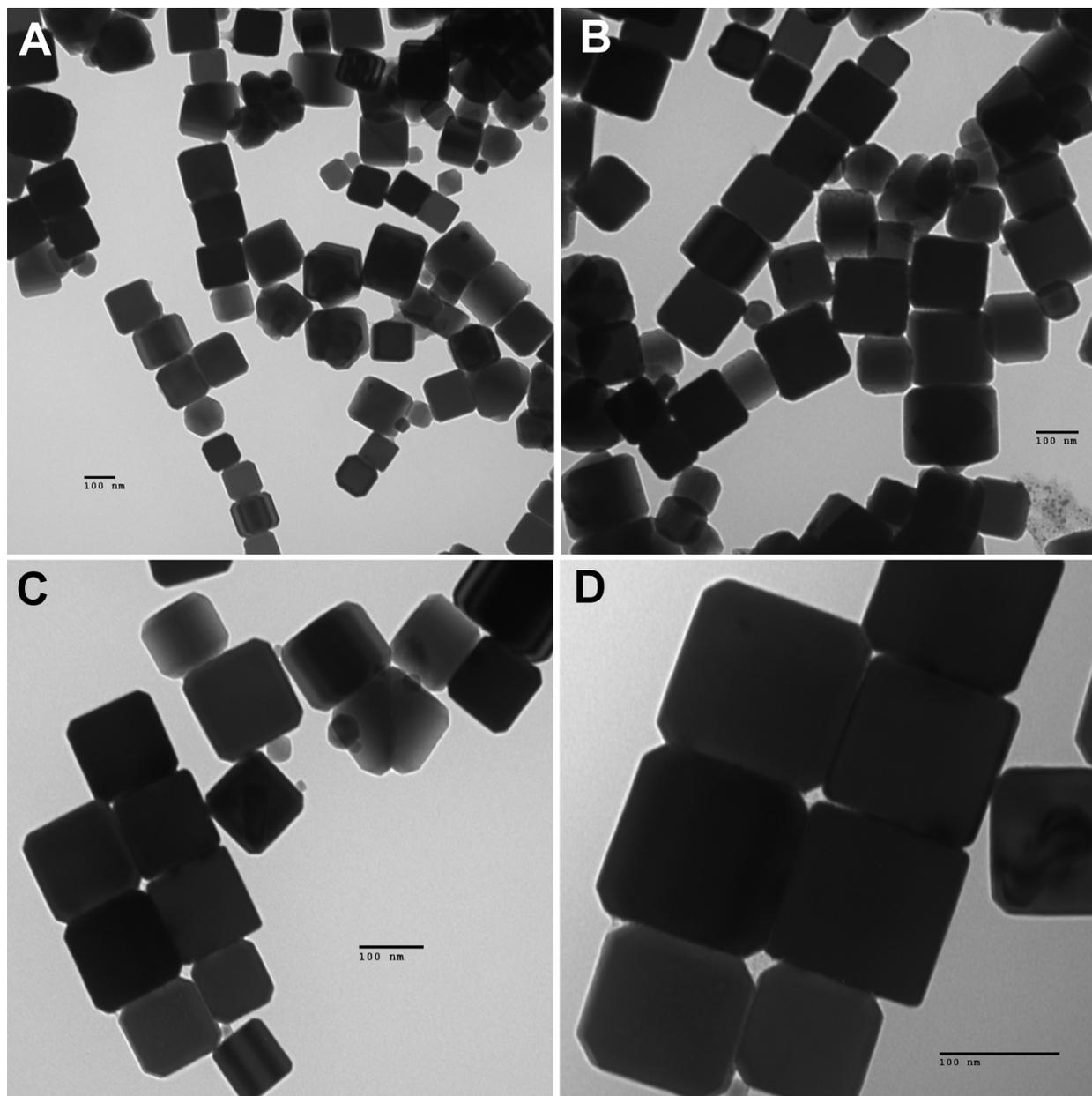
Dopamine hydrochloride, Boc anhydride, benzyl bromide, trifluoroacetic acid, succinic acid anhydride, tetraethylene glycol, EDC, DMAP, iron(III)acetylacetonate ( $\text{Fe}(\text{acac})_3$ ), oleic acid, stearic acid, benzyl ether, fetal bovine serum (FBS), neocuproine, ascorbic acid, ammonium acetate, and concentrated hydrochloric acid (HCl) were purchased from Sigma-Aldrich (St. Louis, MO). RAW264.7 mouse monocyte/macrophage (Mo/Ma) cells were purchased from ATCC (Manassas, VA). RPMI were purchased from Invitrogen (Carlsbad, CA). Thiazolyl blue and sodium dodecyl sulfate were purchased from Fisher Scientific (Pittsburgh, PA). Ferrozine was purchased from Hach (Loveland, CO).

#### **Characterization of the Hexagonal $\text{Fe}_3\text{O}_4$ Nanoplatelets**

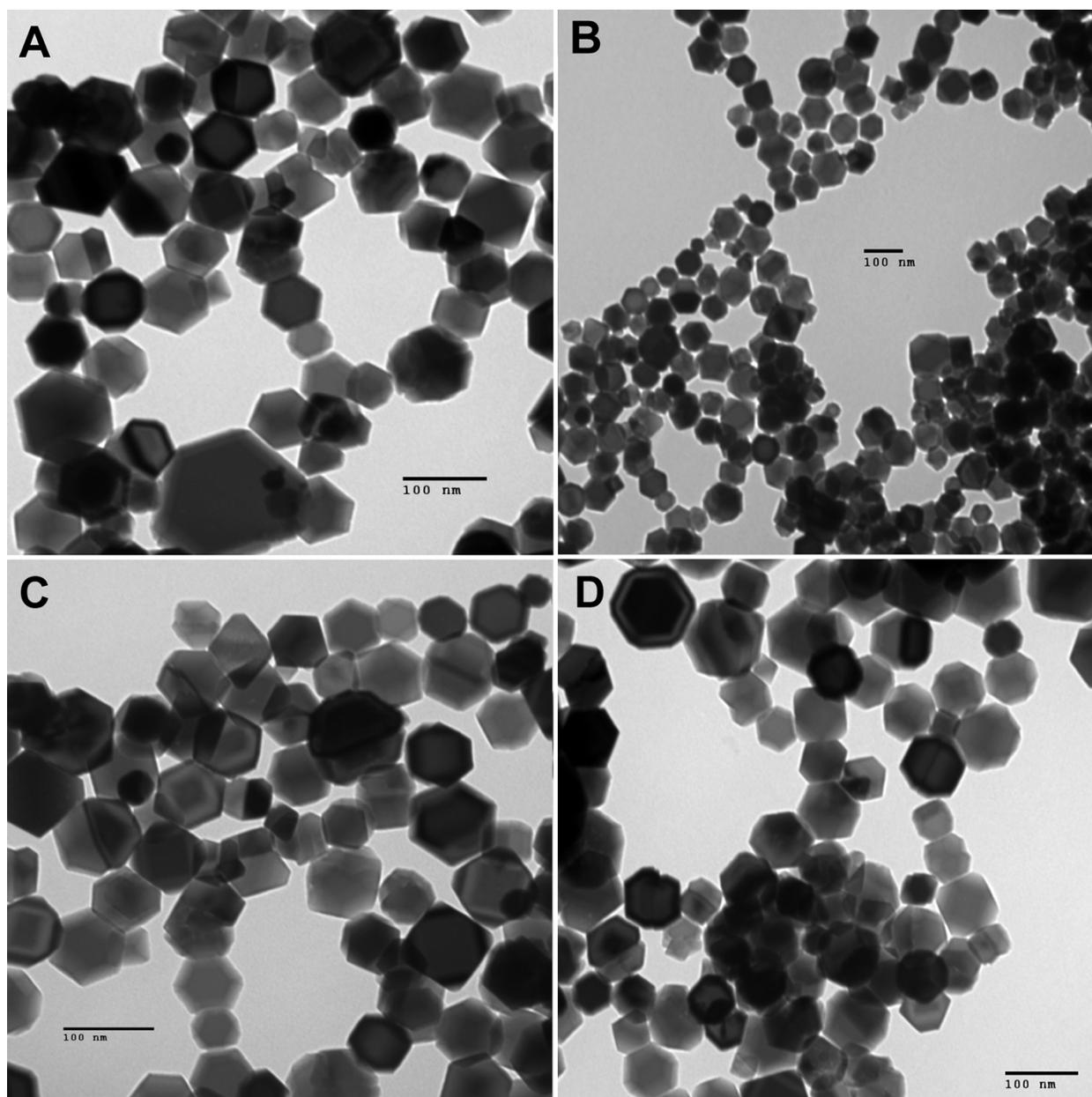
The morphology of the hexagonal  $\text{Fe}_3\text{O}_4$  nanoplatelets is characterized by transmission electron microscopy (TEM). The TEM samples were prepared by immersing carbon-coated 200-mesh copper grids into a solution of nanoplatelets, followed by washing the grids with dropwise chloroform and drying overnight in a desiccator. The dried grids were analyzed with a Philips CM100 microscope operated at 100 kV. High-resolution TEM was recorded on FEI Tecnai F20XT, 200 kV; FEI, Hillsboro, OR. Powder X-ray diffraction (XRD) patterns were obtained on a Bruker D8 X-ray diffractometer with  $\text{Cu K}\alpha$  radiation. X-ray photoelectron spectroscopy (XPS) data was recorded with a Perkin–Elmer PHI 5400 electron spectrometer using acrochromatic  $\text{AlK}\alpha$  radiation (1486.6 eV). Analysis was carried out under vacuum less than  $5 \times 10^{-9}$  Torr and heated to 120 °C to remove any adsorbed molecules on the surface. The XPS binding energies were measured with a precision of 0.025 eV. The analyzer pass energy was set to 17.9 eV, the contact time was 50 ms, and the area scanned was 4 mm<sup>2</sup>.

**Preparation of Hexagonal Fe<sub>3</sub>O<sub>4</sub> Nanoplatelets**

0.71 g of Fe(acac)<sub>3</sub> was added to a mixture of 1.27 g of oleic acid and 0.5 g of stearic acid in 10.4 g of benzyl ether. After degassing at room temperature for 1 hour, the reaction mixture was heated to 290 °C at the rate of 20 °C/min with vigorous stirring. The reaction mixture was maintained at 290 °C for 30 min, and then cooled to room temperature naturally. The resulted mixture was diluted with 10 mL hexane and 30 mL toluene. The nanoparticles were collected by centrifugation and further washed with chloroform. In the absence of stearic acid, under otherwise the same conditions, only Fe<sub>3</sub>O<sub>4</sub> nanocubes were obtained (Figure S1)



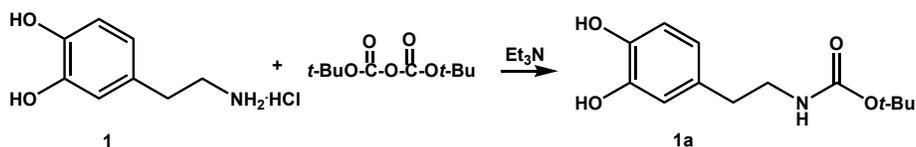
*Figure S1: TEM of Fe<sub>3</sub>O<sub>4</sub> nanocubes*



*Figure S2: TEM of Fe<sub>3</sub>O<sub>4</sub> HMNPs*

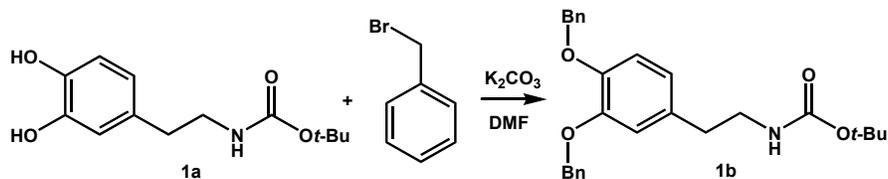
## Synthesis of Dopamine Based Water Soluble Ligand 4

### *N*-tert-Butoxycarbonyl-3,4-dihydroxyphenylethylamine (**1a**)



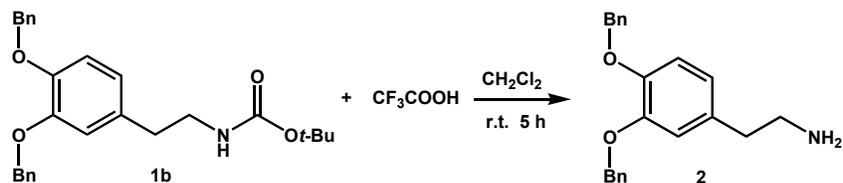
A solution of dopamine hydrochloride **1** (310 mg, 1.63 mmol) in methanol (8 mL) was stirred under N<sub>2</sub> for 5 minutes. TEA (1.8 mmol) was added followed by Boc-anhydride (393 mg, 1.8 mmol). The mixture was stirred under N<sub>2</sub> for 12 hours and the solvent was removed under reduced pressure. The remaining residue was dissolved in 40 mL CH<sub>2</sub>Cl<sub>2</sub> and washed with 1 N HCl (3×5 mL) and brine (5 mL). The organic layer was dried over anhydrous Na<sub>2</sub>SO<sub>4</sub>. After filtration, the organic phase was kept at -5 °C for 3 hours. A white precipitate came out as product **1a** and collected by filtration. (98% yield). <sup>1</sup>H NMR (DMSO-d<sub>6</sub>) δ: 1.73 (s, 9H); 2.48 (t, 2H); 3.02 (q, 2H); 6.40 (d, 1H); 6.54 (s, 1H); 6.61 (d, 1H); 6.83 (t, 1H); 6.85 (s, 1H); 6.76 (s, 1H).

### *N*-tert-Butoxycarbonyl-3,4-dibenzyloxyphenylethylamine (**1b**)



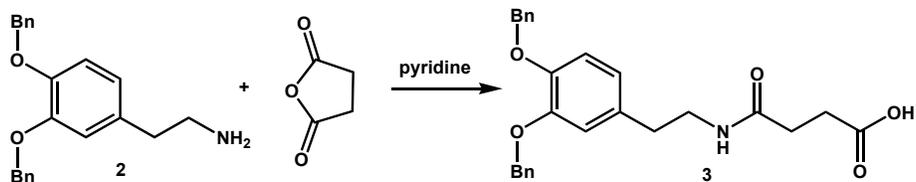
3.47 g of compound **1a** was dissolved in 100 mL DMF. 12.6 g K<sub>2</sub>CO<sub>3</sub> was then added and the system was protected under N<sub>2</sub>. 4.69 g (2 eq.) of benzyl bromide was added drop wise. The mixture was stirred at room temperature for 24 hours without light. The solid was removed by filtering through a short pad of celite and the filter-cake was washed with ether (3×100 mL). The combined filtrate and washing solution were washed with ice water (3×50 mL) and brine (15 mL). The organic layer was dried over anhydrous Na<sub>2</sub>SO<sub>4</sub> and concentrated to 150 mL. After resting at -5 °C for 5 hours, a white precipitate occurred as product **1b** and was collected by vacuum filtration. (93% yield). <sup>1</sup>H NMR (CDCl<sub>3</sub>) δ: 1.45 (s, 9H); 2.70 (t, 2H); 3.31 (q, 2H); 4.49 (s, 1H); 5.15 (d, 4H); 6.71 (d, 1H); 6.80 (s, 1H); 6.88 (d, 1H); 7.32 (t, 2H); 7.37 (t, 4H); 7.45 (d, 4H).

2-(3,4-Bis-benzyloxy-phenyl)-ethylamine (**2**).



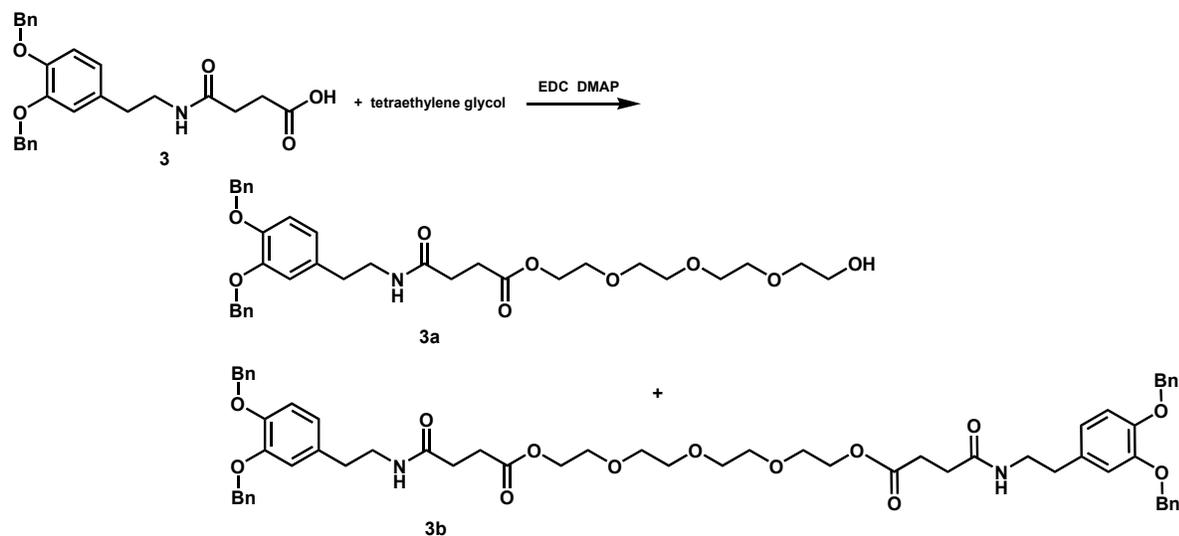
4.3g of compound **1b** was dissolved in 150 mL 5% TFA  $\text{CH}_2\text{Cl}_2$  solution and stirred at room temperature for 5 hours. The solvent was removed in vacuum and a clear oil was obtained as the product **2**. (100% yield).  $^1\text{H}$  NMR ( $\text{CDCl}_3$ )  $\delta$ : 2.79 (t, 2H); 3.08 (m, 2H); 5.11 (s, 4H); 6.68 (d, 1H); 6.75 (s, 1H); 6.90 (d, 1H); 7.32 (t, 2H); 7.35 (t, 4H); 7.42 (d, 4H).  $^{13}\text{C}$  NMR ( $\text{CDCl}_3$ )  $\delta$ : 32.90; 41.85; 71.50; 72.00; 115.60; 116.25; 122.30; 127.60; 127.85; 128.35; 128.45; 128.63; 128.85; 136.70; 136.85; 148.45; 149.00; 160.88; 161.20; 161.58; 161.90.

*N*-[2-(3,4-Bis-benzyloxy-phenyl)-ethyl]-succinamic acid (**3**)



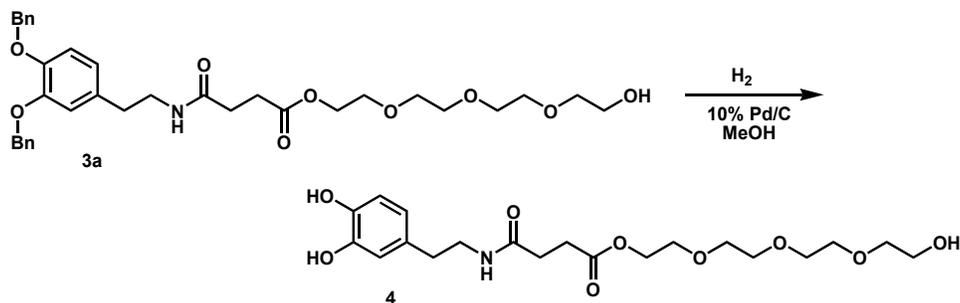
1.43 g of compound **2** and 0.43 g of succinic anhydride (1/1 ratio) were dissolved in 6 mL pyridine. The solution was stirred at room temperature for 5 hours. The solvent was removed by co-evaporation with toluene (toluene 5×5 ml). A white solid was obtained and washed with  $\text{CH}_2\text{Cl}_2$  3 times. After drying in vacuum, 1.4 g of product **3** was obtained. (89% yield).  $^1\text{H}$  NMR ( $\text{DMSO-d}_6$ )  $\delta$ : 2.29 (t, 2H); 2.42 (t, 2H); 2.60 (t, 2H); 3.21 (q, 2H); 5.09 (d, 4H); 6.71 (d, 1H); 6.94 (s, 1H); 6.96 (d, 1H); 7.32 (t, 2H); 7.38 (d, 4H); 7.45 (t, 4H); 7.90 (t, 1H); 12.08 (s, 1H). MS-ESI+:  $m/z$  434.2. Molecular weight calculated as 433.5.

2-(2-(2-(2-hydroxy-ethoxy)ethoxy)ethoxy)ethyl-N-[2-(3,4-Bis-benzyloxy-phenyl)-ethyl]-succinamic acid ester (**3a**)



0.964 g of compound **3** and 0.426g of EDC (1/1 ratio) were dissolved in 100 mL CH<sub>2</sub>Cl<sub>2</sub> and stirred at room temperature for 10 minutes. 0.433 g of tetraethylene glycol was added followed by 5 mg DMAP. After stirring for 12 hours at room temperature, the organic phase was washed with 10% H<sub>3</sub>PO<sub>4</sub> solution (3×10 mL), water (3×10 mL) and brine (10 mL). The organic phase was dried over anhydrous MgSO<sub>4</sub>. After removing the solvent in vacuum, the residue was loaded on a column and eluted with 1/1 acetone/methylene chloride. 0.77 g of product **3a** was obtained. (79% yield). 0.21 g of side product **3b** was isolated. <sup>1</sup>H NMR for **3a** (CDCl<sub>3</sub>) δ: 2.39 (t, 2H); 2.57 (t, 1H); 2.70 (q, 4H); 3.44 (q, 2H); 3.60 (t, 2H); 3.65 (broad 12H); 4.24 (t, 2H); 5.15 (d, 4H); 5.74 (t, 1H); 6.71 (d, 1H); 6.81 (s, 1H); 6.89 (d, 1H); 7.31 (t, 2H); 7.37 (t, 4H); 7.46 (d, 4H). MS-ESI+: m/z 610.4. Molecular weight calculated as 609.3. <sup>1</sup>H NMR for **3b** (CDCl<sub>3</sub>) δ: 2.37 (t, 4H); 2.67 (m, 8H); 3.42 (q, 4H); 3.63 (s, 8H); 3.67 (t, 4H); 4.22 (t, 4H); 5.15 (d, 8H); 5.70 (t, 2H); 6.70 (d, 2H); 6.80 (s, 2H); 6.88 (d, 2H); 7.31 (t, 4H); 7.36 (t, 8H); 7.45 (d, 8H). MS-ESI+: m/z 1024.7. Molecular weight calculated as 1024.2.

2-(2-(2-(2-hydroxy-ethoxy)ethoxy)ethoxy)ethyl-N-[2-(3,4-dihydroxy-phenyl)-ethyl]-succinamic acid ester (**4**)



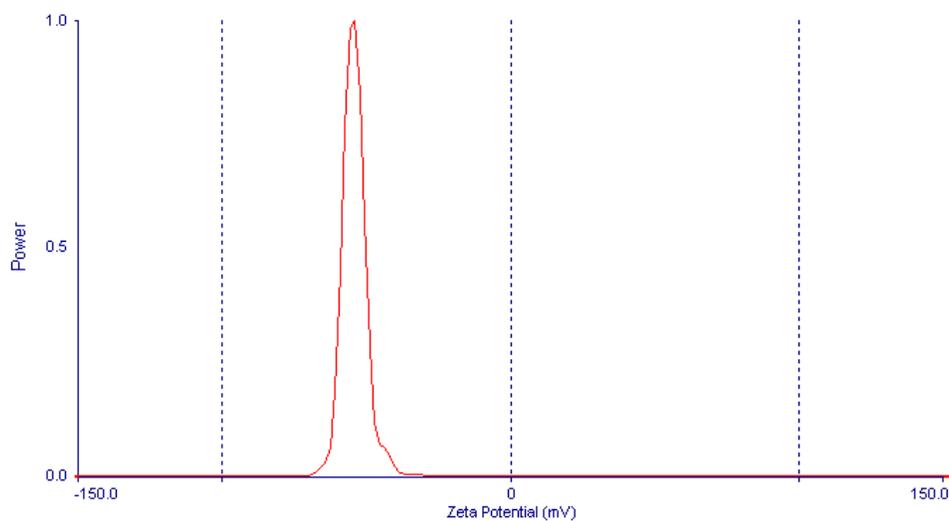
0.34 g of compound **3a** was dissolved in 50 mL methanol. 77 mg of Pd/C was added under N<sub>2</sub>. After evacuating three times, 1 atm. H<sub>2</sub> was applied and the mixture was stirred for 24 hours at room temperature. The catalyst was removed by filtering through a short pad of celite. After removing the solvent in vacuum, 0.23 g of product **4** was obtained. (100% yield). <sup>1</sup>H NMR (DMSO-*d*<sub>6</sub>) δ: 2.33 (t, 2H); 2.48 (q, 2H); 3.15 (broad multiplet, 4H); 3.41 (t, 2H); 3.49 (t, 2H); 3.51 (broad multiplet, 8H); 3.59 (t, 2H); 4.11 (t, 2H); 6.41 (d, 1H); 6.55 (s, 1H); 6.61 (d, 1H). MS-ESI<sup>+</sup>: *m/z* 430.4. Molecular weight calculated as 429.4.

#### Surface Capping of Hexagonal Fe<sub>3</sub>O<sub>4</sub> Nanoplatelets With Ligand **4**

20 mg of hexagonal Fe<sub>3</sub>O<sub>4</sub> nanoplatelets was dispersed in 10 mL chloroform by sonication; 35 mg of ligand **4** in 10 mL of methanol was added. The mixture was sonicated for 5 min, and then was further shaken at room temperature for 12 hours. The nanoplatelets were precipitated by centrifugation, and the free ligand **4** was removed by 6 cycles of methanol washing.

## Zeta Potential Measurements

Zeta potential measurements were performed using ZetaPALS Zeta Potential Analyzer purchased by Brookhaven Instruments Corporation.<sup>18</sup> The HMNPs were characterized in 1 X PBS at 298K. Their concentration was 0.1mg/mL.



*Figure S5: Zeta potential of HMNPs in 1 X PBS*

## Reference

- 18 A. S. Perera, H. Wang, M. T. Basel, M. R. Pokhrel, P. S. Gamage, M. Kalita, S. Wendel, B. Sears, D. Welideniya, Y. Liu, C. Turro, D. L. Troyer and S. H. Bossmann, *Langmuir*, 2013, **29**, 308.