

## Supporting information

# Exploiting Metal Organic Frameworks as efficient enzymes immobilization matrices for the building-up of sensitive electrochemical biosensor

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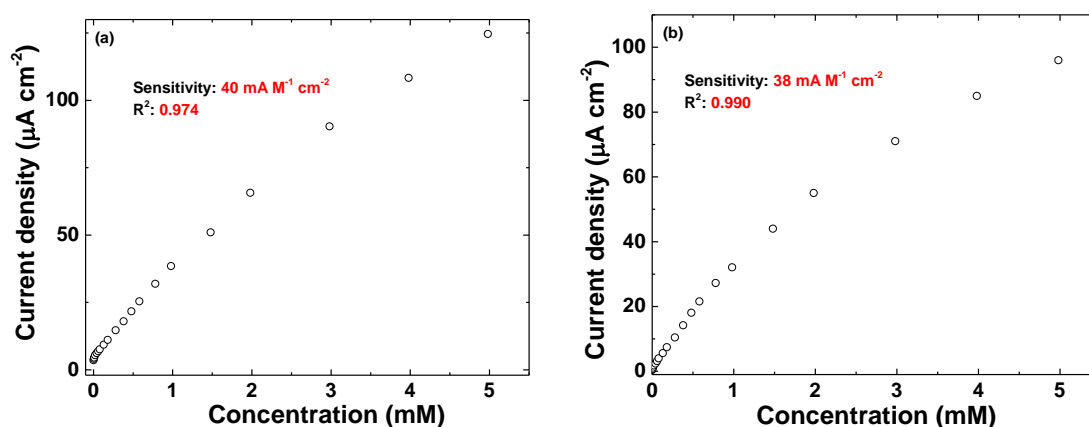


Figure SI 1. Current density versus glucose concentration of GOx-MIL-100(Fe)- PtNPs-CIE at pH=4 when the electrode was dried at (a) room temperature and (b)  $50^\circ\text{C}$ . Note that the sensitivity was obtained here without optimization.

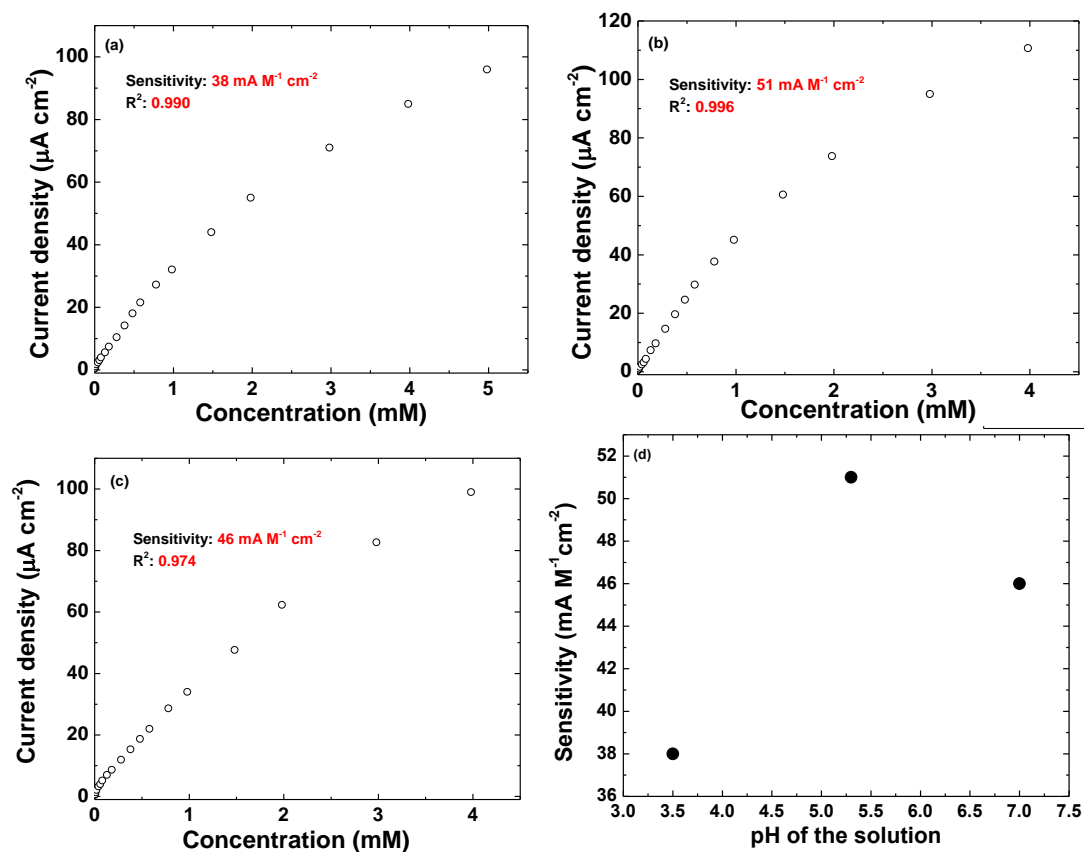


Figure SI 2. Current density versus glucose concentration for  $\text{GOx-MIL-100(Fe)-PtNPs-CIE}$ . For each electrode,  $\text{GOx}$  is solubilized at pH (a) 3.5, (b) 5.3 and (c) 7. Curve (d) shows the sensitivity versus pH of the solution in which  $\text{GOx}$  was dissolved. Note that the sensitivity was obtained here without optimization.

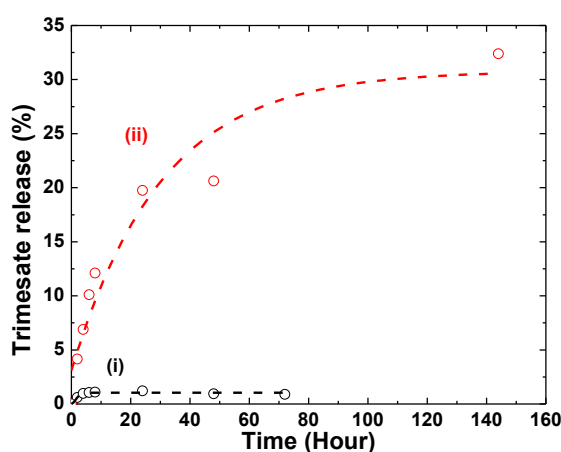


Figure SI 3. Release of trimesate measured by HPLC during  $\text{MIL-100(Fe)}$  incubation in acetate buffer of  $\text{pH}=5.3$  when (i) immobilized at the electrode surface and kept at 0.5 V vs.  $\text{Ag}^+/\text{Ag}$  and (ii) incubated in the form of powder.

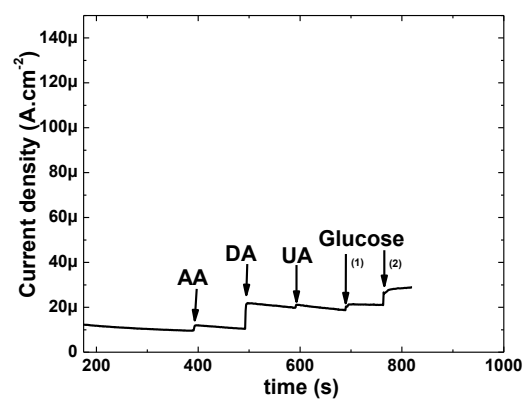


Figure SI 4. Amperometric current responses at a potential of 0.5 V vs. AgCl/Ag upon the addition of 0.1 mM ascorbic acid (AA), 0.1 mM dopamine (DA), 0.1 mM uric acid (UA), 0.04 mM (addition (1)) and 0.08 mM (addition (2)) glucose .

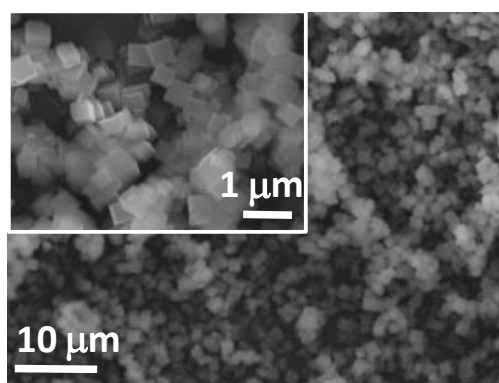
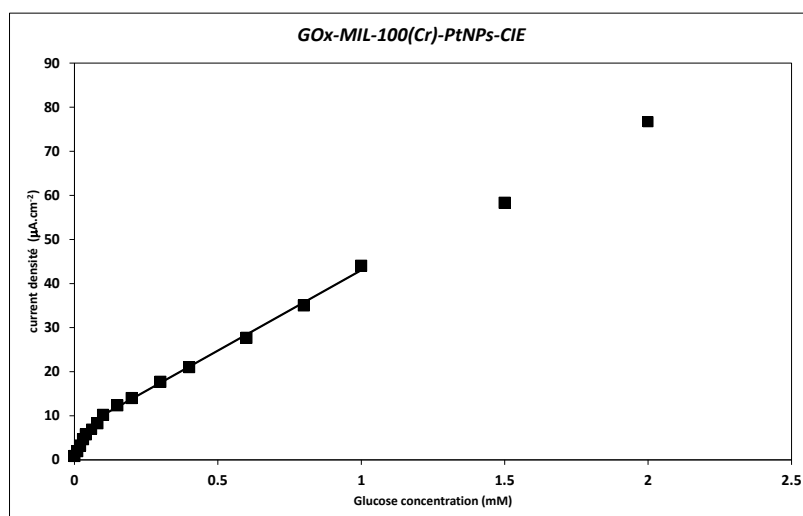


Figure SI5. SEM images of MIL-127(Fe) nanoparticles.



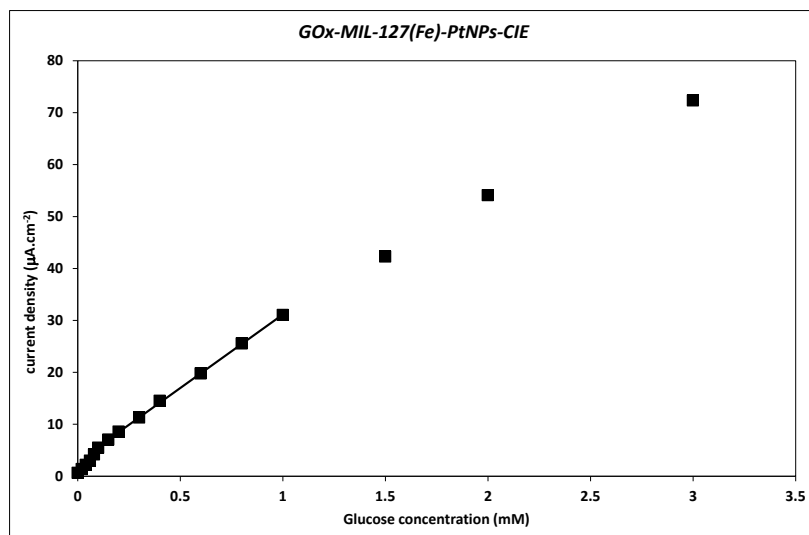
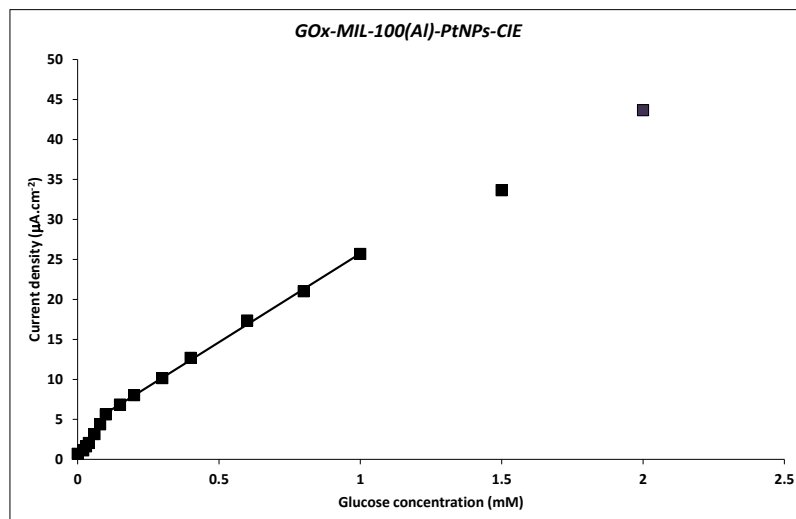


Figure S16. Glucose calibration curves of current vs glucose concentration of GOx-MOF-PtNp-CIE electrode

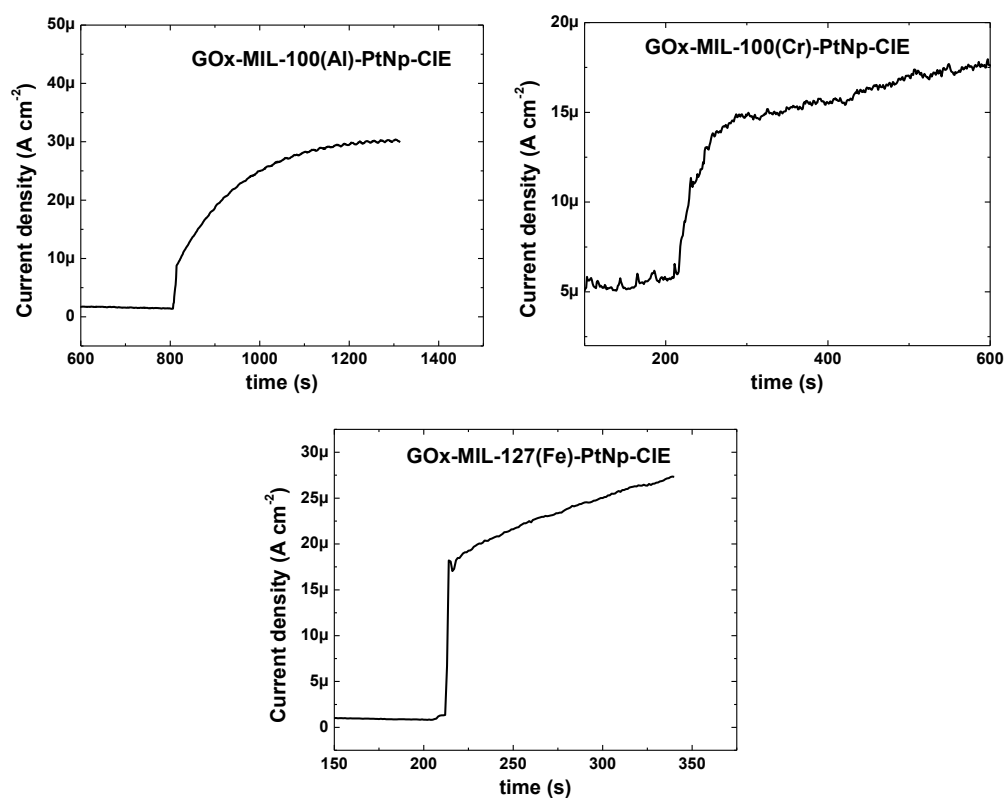


Figure SI7. Chronoamperometric responses of GOx-MIL-100(Al)-PtNp-CIE, GOx-MIL-100(Cr)-PtNp-CIE and GOx-MIL-127(Fe)-PtNp-CIE after adding respectively 1, 0.3 and 0.7mM of glucose. Response times were measured by the time taken to achieve 90% of the current density.