### **Supporting Information**

## Photoresponsive Real Time Monitoring Silicon Quantum Dots for Regulated Delivery of Anticancer Drug

Amrita Paul,<sup>a</sup> Avijit Jana,<sup>b</sup> S. Karthik,<sup>a</sup> Manoranjan Bera,<sup>a</sup> Yanli Zhao<sup>\*b</sup> and N. D. Pradeep Singh<sup>\*a</sup>

a Department of Chemistry, Indian Institute of Technology Kharagpur, 721302 India,

b Division of Chemistry and Biological Chemistry, School of Physical and Mathematical Sciences, Nanyang Technological University, 21 Nanyang Link, Singapore 637371

No	Contents	Page No.
1	Scheme for synthesis of <i>o</i> -nitrobenzyl chlorambucil conjugate.	S2
2	<sup>1</sup> H and <sup>13</sup> C NMR spectra	S2
3	EDX pattern of ONBCbl-SiQDs showing the presence of constituent elements	S7
4	Loading of <i>o</i> -nitrobenzyl chlorambucil on SiQDs by absorption spectra	S8
5	Measurement of fluorescence quantum yields <sup>1</sup>	S8
6	HPLC profile of chlorambucil release from ONBCbl-SiQDs	S9

#### **Supporting Information**



### 1. Synthesis of phototrigger *o*-nitrobenzyl chlorambucil conjugate:

Scheme S1: Synthesis of *o*-nitrobenzyl chlorambucil conjugate.

# 2.1. <sup>1</sup>H NMR (CDCl<sub>3</sub>, 200 MHz) and <sup>13</sup>C NMR (50 MHz, CDCl<sub>3</sub>) of 4-(3bromopropoxy)5-methoxy-nitrobenzaldehyde (1)





2.2. <sup>1</sup>H NMR (200 MHz, CDCl<sub>3</sub>) and <sup>13</sup>C NMR (50 MHz, CDCl<sub>3</sub>) spectra of 4-(3bromopropoxy)-5-methoxy-2-nitrobenzaldehyde (2)





2.3. <sup>1</sup>H NMR (CDCl<sub>3</sub>, 200 MHz) <sup>13</sup>C NMR (50 MHz, CDCl<sub>3</sub>) spectra of 4-(3bromopropoxy)-5-methoxy-2-nitrophenyl)methanol (3)





2.4. <sup>1</sup>H NMR (CDCl<sub>3</sub>, 200 MHz) and <sup>13</sup>C NMR (CDCl<sub>3</sub>, 50 MHz) spectra of 1-(bromoethyl)-4-(3-bromopropoxy)-5-methoxy-2-nitrobenzene (4)





2.5. <sup>1</sup>H NMR (CDCl<sub>3</sub>, 200 MHz) and <sup>13</sup>C NMR (CDCl<sub>3</sub>, 50MHz) spectra of 4-(3bromopropoxy)-5-methoxy-2-nitrobenzyl chlorambucil (ONBCbl) (5)





3. Energy Dispersion X-Ray (EDX) pattern of ONBCbl-SiQDs



Figure S1: EDX pattern of ONBCbl-SiQDs showing the presence of constituent elements.

#### 4. Loading of *o*-nitrobenzyl chlorambucil on SiQDs by absorption spectra.



Figure S2: (a) UV-vis absorption spectra of silicon quantum dots and *o*-nitrobenzyl chlorambucil loaded silicon quantum dots, (b) standard UV-vis absorption spectra of *o*-

nitrobenzyl, (c) calibration curve for the concentration of ONBCbl-SiQDs in the reaction mixture.

```
The concentration of ONBCbl-SiQDs in the reaction mixture after 24 h obtained from the absorption spectra (x)= 0.114 \text{ mg/ml}
```

```
Concentration of ONBCbl-SiQDs at 0 min obtained from absorption spectra (y) = 0.005 mg/ml
```

ONBCbl loaded on SiQDs

= x-y
= (0.114 - 0.005) mg/ml
= 0.109 mg/ml
= 0.109 mg of ONBCbl in 5 mg of SiQDs
= 0.0218 mg in 1mg of SiQDs
~ 22 μg/mg of ONBCbl was loaded on the SiQDs.

The amount of drug loaded on the nanoparticles was measured with UV–vis spectroscopy by dissolving drug-loaded nanoparticles in methanol. UV absorbance was monitored at a wavelength of 410 nm. The amount of the drug loaded was determined by calibration with a series of standards of known concentrations of drugs in the same solvent.

#### 5. Measurement of fluorescence quantum yields<sup>1</sup>

The fluorescence quantum yield (QY) of the silicon quantum dots (SiQDs) was determined by the reference point method.<sup>1</sup> Quinine sulfate in 0.1 M H<sub>2</sub>SO<sub>4</sub> (literature quantum yield: 54 %) was used as a standard sample to calculate the QY of SiQDs and ONBCbl-SiQDs, which were dissolved in acetonitrile. The absorbance values of the solutions at the excitation wavelength were measured by UV–Vis spectrophotometer. Photoluminescence (PL) emission spectra of all the sample solutions were recorded by Hitachi F-7000 fluorescence spectrophotometer at an excitation wavelength of 410 nm.

$$\frac{\Phi_{\rm S}}{\Phi_{\rm R}} = \frac{A_{\rm S}}{A_{\rm R}} \frac{({\rm Abs})_{\rm R}}{({\rm Abs})_{\rm S}} \frac{\eta_{\rm s}^2}{\eta_{\rm R}^2}$$

Where  $\Phi$  represents quantum yield, Abs represents absorbance, A represents the area under the fluorescence curve, and  $\eta$  is the refractive index of the medium. The subscripts S and R denote the corresponding parameters for the sample and reference, respectively.

	Intergrated	Abs.	Refractive	Fluorescence
Sample	Emission		Index of	Quantum Yield
	Intensity (I)		Solvent $(\eta)$	(QY)
Quinine sulfate	35605.59	0.058	1.33	54 % (known)
SiQDs	4268.27	0.035	1.327	18.8 %
ONBCbl-SiQDs	2345.49	0.043	1.327	4.2 %

Table S1: Quantum yield of the fluorescent SiQDs and ONBCbl-SiQDs

Fluorescence quantum yield (excitation wavelength 410 nm, error limit within  $\pm$  5 %).

6. HPLC profile of chlorambucil release from ONBCbl-SiQDs:



**Figure S3**: HPLC profile of chlorambucil release from ONBCbl-SiQDs upon irradiating with visible light ( $\geq$ 410 nm). ) at regular intervals (0-30 min, time interval of 5 min). The *x*-axes was offset by 30 s, to facilitate better visualization.

### **References:**

a) S. Mandal, C. Ghatak, V. G. Rao, S. Ghosh, N. Sarkar. J. Phys. Chem. C 2012, 116, 5585–5597. S. Zhu; b) Q. Meng; L. Wang; J. Zhang; Y. Song; H. Jin; K. Zhang; H. Sun; H. Wang; B.Yang. Angew. Chem. Int. Ed. 2013, 52, 3953–3957.