

Electronic Supplementary Information

**Sub-nanometer sized copper clusters: one-step synthesis and
electrochemical detection of glucose**

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Supplementary Results



Fig. S1 Digital photographs of the as-synthesized Cu nanocluster (left: solution; right: solid)

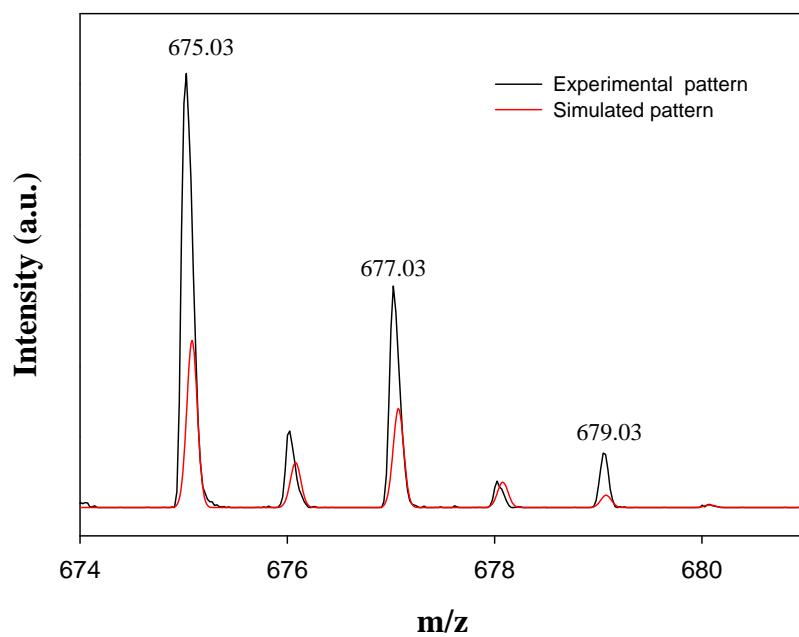


Fig. S2 The experimental (black curve) and simulated (red curve) isotopic patterns of $[\text{Cu}(\text{SG})_2\text{-H}]$

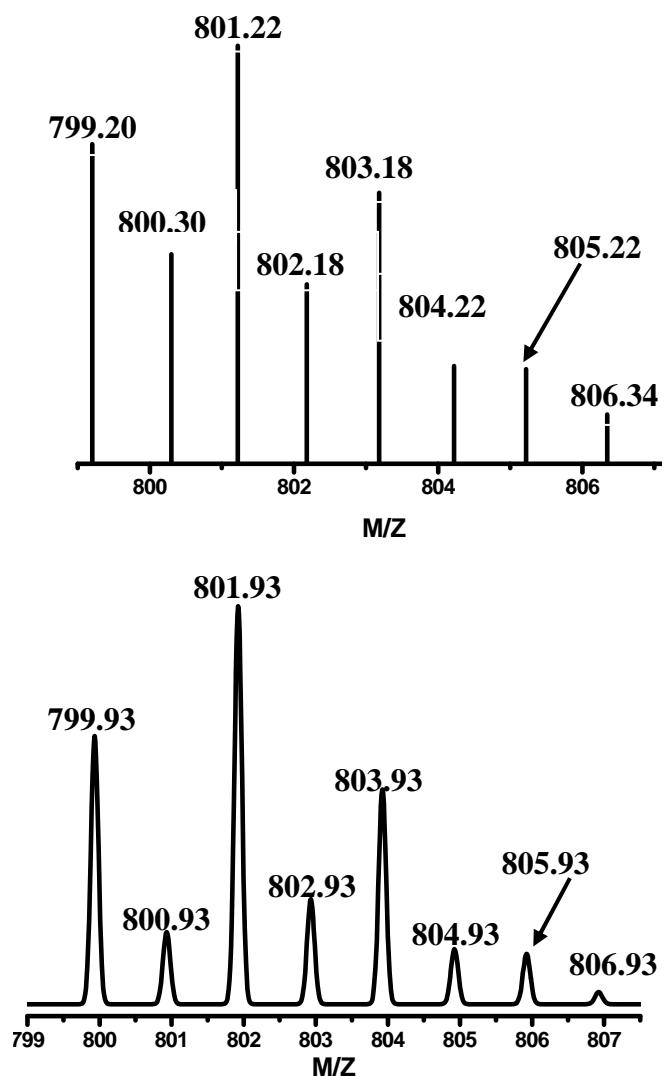


Fig. S3 The experimental (top) and simulated (bottom) isotopic patterns of $[\text{Cu}_3(\text{SG})_2]^+$.

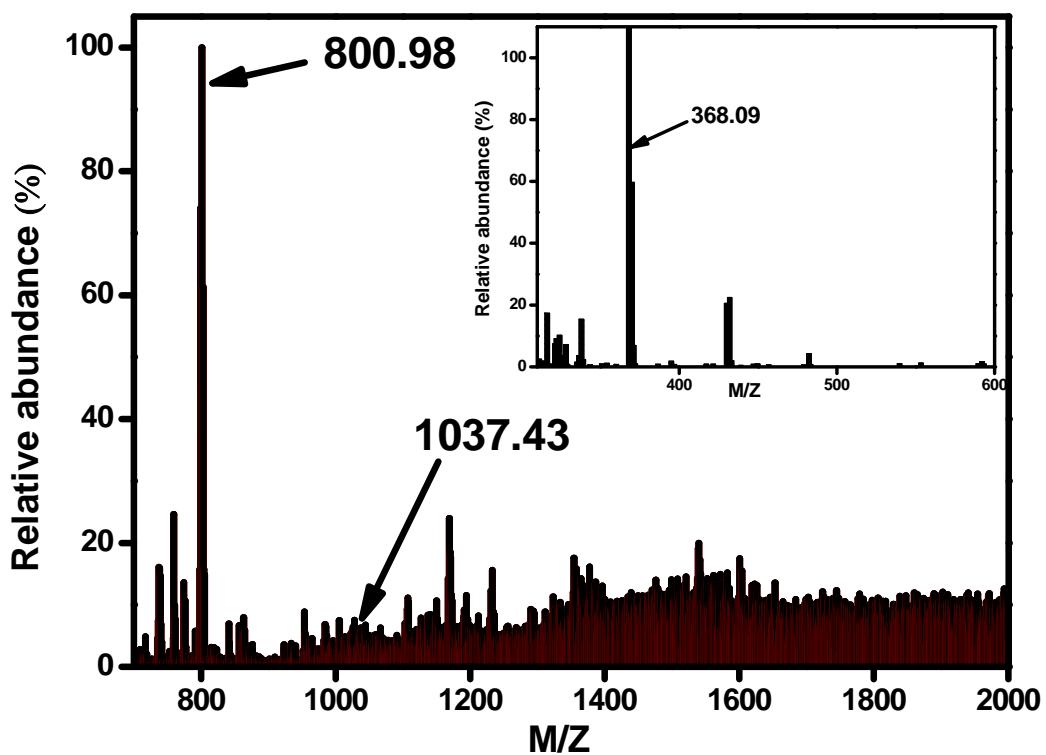


Fig. S4 The negative-mode ESI-MS of the as-synthesized Cu nanoclusters. Inset displays the lower mass region of the Cu clusters.

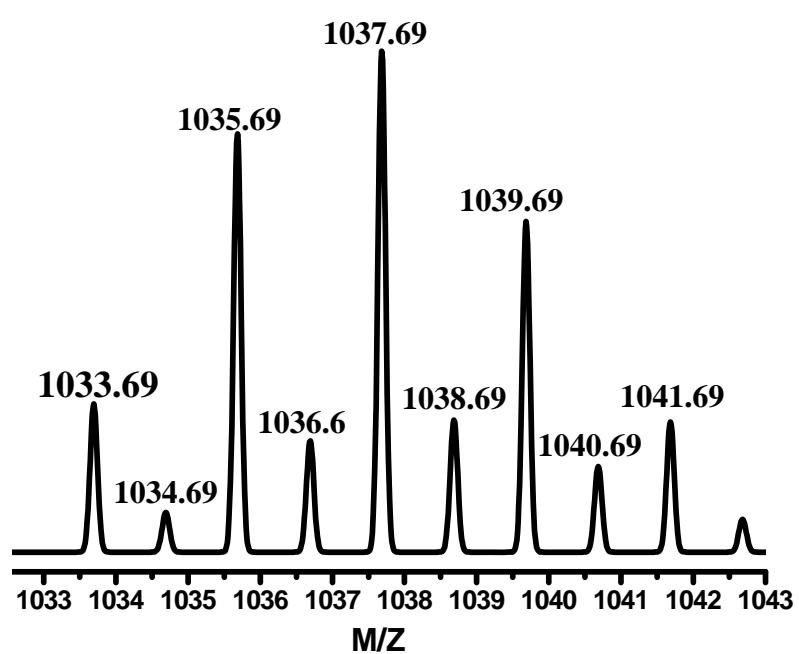
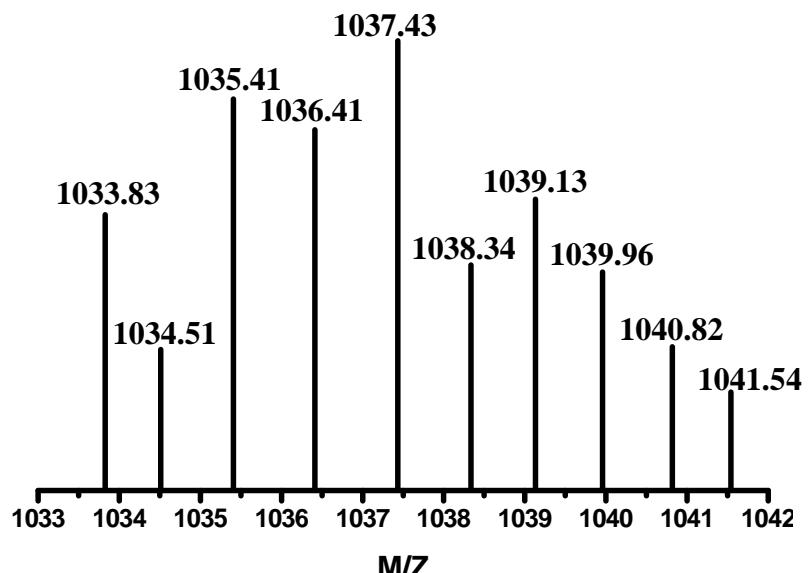


Fig. S5 The experimental (top) and simulated (bottom) isotopic patterns of [Cu₆(SG)₂+2Na]⁻.

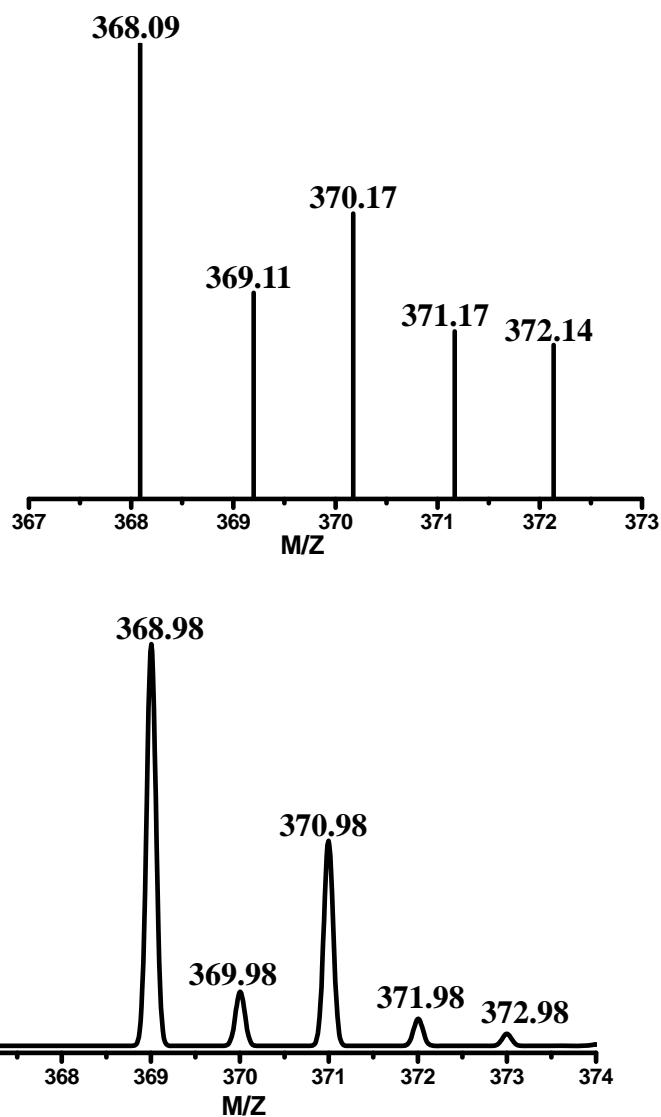


Fig. S6 The experimental (upper curve) and simulated (down curve) isotopic patterns of $[\text{Cu}(\text{SG})\text{-H}]^-$.

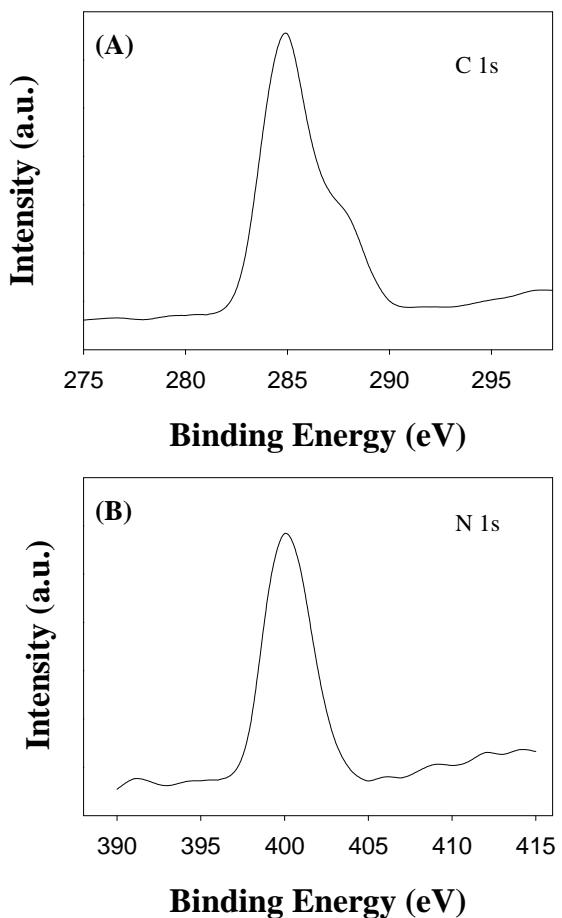


Fig. S7 XPS spectra of C 1s (A) and N 1s (B) of the as-prepared Cu nanoclusters.

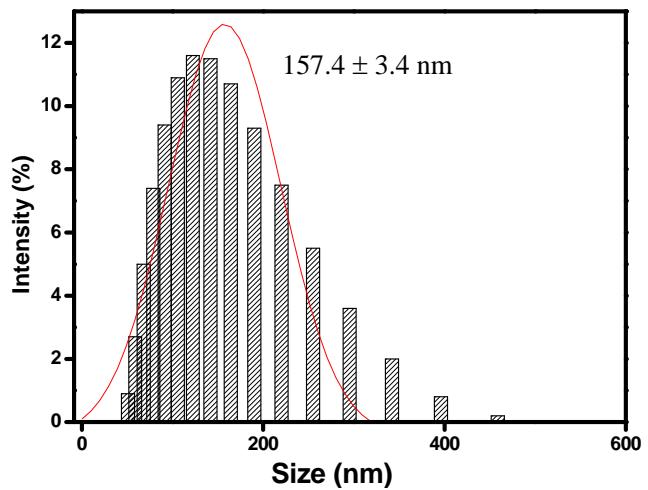


Fig. S8 Dynamic light scattering (DLS) measurement for the as-prepared Cu nanoparticles.

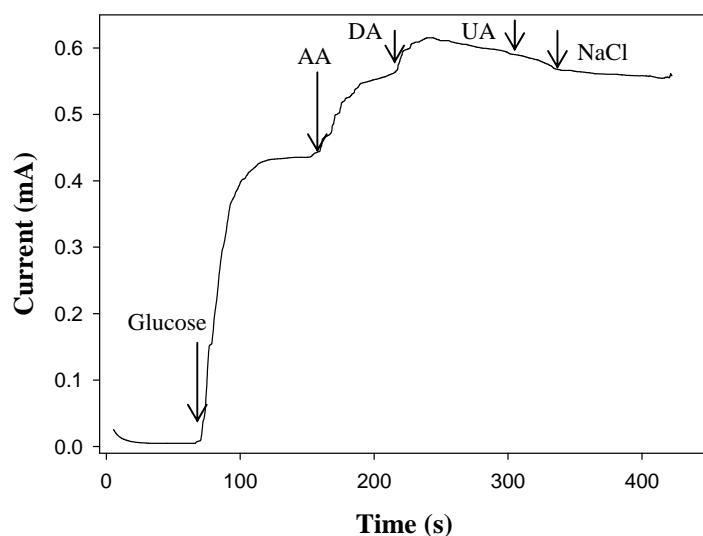


Fig. S9 Amperometric responses of the Cu NCs/TiO₂/ITO to the addition of different analytes in 0.1 M KOH.

20 mM glucose, 10 mM ascorbic acid (AA), 10 mM dopamine (DA), 10 mM uric acid (UA) and 10 mM NaCl at 0.5 V.

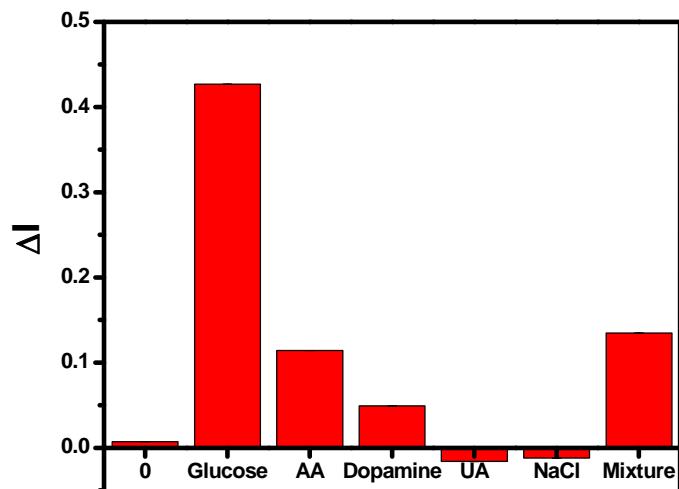


Fig. S10 The bar diagram from amperometric reponses of the Cu NCs/TiO₂/ITO to the addition of different analytes and their mixture in 0.1 M KOH. 20 mM glucose, 10 mM ascorbic acid (AA), 10 mM dopamine (DA), 10 mM uric acid (UA) and 10 mM NaCl at 0.5 V.