

Sr₂(OH)₃NO₃: the First Nitrate as a Deep UV Nonlinear Optical Material with Large SHG Responses

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Table S1. Crystal Data and Structure Refinement for $\text{Sr}_2(\text{OH})_3\text{NO}_3$.^a

Formula	$\text{Sr}_2(\text{OH})_3\text{NO}_3$
Formula Mass (amu)	288.27
Crystal System	Hexagonal
Space Group	<i>P</i> E rro!2 <i>m</i>
<i>a</i> (\AA)	6.6037 (2)
<i>c</i> (\AA)	3.55850(10)
$\alpha(^{\circ})$	90
$\gamma(^{\circ})$	120
<i>V</i> (\AA^3)	134.392(7)
<i>Z</i>	1
ρ (calcd) (g/cm ³)	3.562
Temperature (K)	293(2)
λ (\AA)	1.54184
<i>F</i> (000)	134
μ (mm ⁻¹)	26.04
Final R indices (I>2 σ (I)) <i>a</i> <i>R</i> ₁ /w <i>R</i> ₂	0.040/0.104
GOF on <i>F</i> ²	1.24
Absolute Structure Parameter	0.1(2)

^a $\text{R}_1(F) = \sum ||F_o - |F_c|| / \sum |F_o|$. $w\text{R}_2(F_o^2) = [\sum w(F_o^2 - F_c^2)^2 / \sum w(F_o^2)^2]^{1/2}$.

Table S2. Selected Bond lengths (Å) and angles (deg) for $\text{Sr}_2(\text{OH})_3\text{NO}_3$.

$\text{Sr1}-\text{O}2^{\text{i}}$	2.6259 (13)	$\text{Sr1}-\text{Sr1}^{\text{vii}}$	3.8127 (1)
$\text{Sr1}-\text{O}2^{\text{ii}}$	2.6259 (13)	$\text{Sr1}-\text{H}2$	2.8421
$\text{Sr1}-\text{O}2^{\text{iii}}$	2.6259 (13)	$\text{O}1-\text{N}1$	1.254 (14)
$\text{Sr1}-\text{O}2^{\text{iv}}$	2.6259 (13)	$\text{O}1-\text{Sr1}^{\text{viii}}$	2.798 (10)
$\text{Sr1}-\text{O}2^{\text{v}}$	2.6259 (13)	$\text{N}1-\text{O}1^{\text{ix}}$	1.254 (14)
$\text{Sr1}-\text{O}2$	2.6259 (13)	$\text{N}1-\text{O}1^{\text{x}}$	1.254 (14)
$\text{Sr1}-\text{O}1$	2.798 (10)	$\text{O}2-\text{Sr1}^{\text{xi}}$	2.6259 (13)
$\text{Sr1}-\text{O}1^{\text{ii}}$	2.798 (10)	$\text{O}2-\text{Sr1}^{\text{vi}}$	2.6259 (13)
$\text{Sr1}-\text{O}1^{\text{iii}}$	2.798 (10)	$\text{O}2-\text{Sr1}^{\text{xii}}$	2.6259 (13)
$\text{Sr1}-\text{Sr1}^{\text{v}}$	3.5585	$\text{O}2-\text{H}2$	0.8193
$\text{O}2^{\text{i}}-\text{Sr1}-\text{O}2^{\text{ii}}$	136.849 (19)	$\text{O}2^{\text{iii}}-\text{Sr1}-\text{Sr1}^{\text{vi}}$	47.35 (3)
$\text{O}2^{\text{i}}-\text{Sr1}-\text{O}2^{\text{iii}}$	85.31 (5)	$\text{O}2^{\text{iv}}-\text{Sr1}-\text{Sr1}^{\text{vi}}$	132.65 (3)
$\text{O}2^{\text{ii}}-\text{Sr1}-\text{O}2^{\text{iii}}$	79.12 (4)	$\text{O}2^{\text{v}}-\text{Sr1}-\text{Sr1}^{\text{vi}}$	132.65 (3)
$\text{O}2^{\text{i}}-\text{Sr1}-\text{O}2^{\text{iv}}$	79.12 (4)	$\text{O}2-\text{Sr1}-\text{Sr1}^{\text{vi}}$	47.35 (3)
$\text{O}2^{\text{ii}}-\text{Sr1}-\text{O}2^{\text{iv}}$	85.31 (5)	$\text{O}1-\text{Sr1}-\text{Sr1}^{\text{vi}}$	90.0
$\text{O}2^{\text{iii}}-\text{Sr1}-\text{O}2^{\text{iv}}$	136.849 (19)	$\text{O}1^{\text{ii}}-\text{Sr1}-\text{Sr1}^{\text{vi}}$	90.000 (1)
$\text{O}2^{\text{i}}-\text{Sr1}-\text{O}2^{\text{v}}$	79.12 (4)	$\text{O}1^{\text{iii}}-\text{Sr1}-\text{Sr1}^{\text{vi}}$	90.000 (1)
$\text{O}2^{\text{ii}}-\text{Sr1}-\text{O}2^{\text{v}}$	136.849 (19)	$\text{Sr1}^{\text{v}}-\text{Sr1}-\text{Sr1}^{\text{vi}}$	180.0
$\text{O}2^{\text{iii}}-\text{Sr1}-\text{O}2^{\text{v}}$	136.85 (2)	$\text{O}2^{\text{i}}-\text{Sr1}-\text{Sr1}^{\text{vii}}$	43.45 (3)
$\text{O}2^{\text{iv}}-\text{Sr1}-\text{O}2^{\text{v}}$	79.12 (4)	$\text{O}2^{\text{ii}}-\text{Sr1}-\text{Sr1}^{\text{vii}}$	117.7 (2)
$\text{O}2^{\text{i}}-\text{Sr1}-\text{O}2$	136.85 (2)	$\text{O}2^{\text{iii}}-\text{Sr1}-\text{Sr1}^{\text{vii}}$	43.45 (3)
$\text{O}2^{\text{ii}}-\text{Sr1}-\text{O}2$	79.12 (4)	$\text{O}2^{\text{iv}}-\text{Sr1}-\text{Sr1}^{\text{vii}}$	117.7 (2)
$\text{O}2^{\text{v}}-\text{Sr1}-\text{O}2$	85.31 (5)	$\text{O}1-\text{Sr1}-\text{Sr1}^{\text{vii}}$	72.9 (2)
$\text{O}2^{\text{i}}-\text{Sr1}-\text{O}1$	71.0 (3)	$\text{O}1^{\text{ii}}-\text{Sr1}-\text{Sr1}^{\text{vii}}$	47.1 (2)
$\text{O}2^{\text{ii}}-\text{Sr1}-\text{O}1$	65.9 (3)	$\text{O}1^{\text{iii}}-\text{Sr1}-\text{Sr1}^{\text{vii}}$	167.1 (2)
$\text{O}2^{\text{iii}}-\text{Sr1}-\text{O}1$	71.0 (3)	$\text{Sr1}^{\text{v}}-\text{Sr1}-\text{Sr1}^{\text{vii}}$	90.0
$\text{O}2^{\text{iv}}-\text{Sr1}-\text{O}1$	65.9 (3)	$\text{Sr1}^{\text{vi}}-\text{Sr1}-\text{Sr1}^{\text{vii}}$	90.0
$\text{O}2^{\text{v}}-\text{Sr1}-\text{O}1$	137.21 (5)	$\text{O}2^{\text{i}}-\text{Sr1}-\text{H}2$	120.7
$\text{O}2-\text{Sr1}-\text{O}1$	137.21 (5)	$\text{O}2^{\text{ii}}-\text{Sr1}-\text{H}2$	91.4

02 ⁱ —Sr1—01 ⁱⁱ	65. 9 (3)	02 ⁱⁱⁱ —Sr1—H2	70. 2
02 ⁱⁱ —Sr1—01 ⁱⁱ	137. 21 (5)	02 ^{iv} —Sr1—H2	150. 7
02 ⁱⁱⁱ —Sr1—01 ⁱⁱ	65. 9 (3)	02 ^v —Sr1—H2	83. 7
02 ^{iv} —Sr1—01 ⁱⁱ	137. 21 (5)	02—Sr1—H2	16. 6
02 ^v —Sr1—01 ⁱⁱ	71. 0 (3)	01—Sr1—H2	138. 0
02—Sr1—01 ⁱⁱ	71. 0 (3)	01 ⁱⁱ —Sr1—H2	54. 8
01—Sr1—01 ⁱⁱ	120. 0	01 ⁱⁱⁱ —Sr1—H2	80. 4
02 ⁱ —Sr1—01 ⁱⁱⁱ	137. 21 (5)	Sr1 ^v —Sr1—H2	128. 8
02 ⁱⁱ —Sr1—01 ⁱⁱⁱ	71. 0 (3)	Sr1 ^{vi} —Sr1—H2	51. 2
02 ⁱⁱⁱ —Sr1—01 ⁱⁱⁱ	137. 21 (5)	Sr1 ^{vii} —Sr1—H2	89. 5
02 ^{iv} —Sr1—01 ⁱⁱⁱ	71. 0 (3)	N1—01—Sr1	137. 1 (2)
02 ^v —Sr1—01 ⁱⁱⁱ	65. 9 (3)	N1—01—Sr1 ^{viii}	137. 1 (2)
02—Sr1—01 ⁱⁱⁱ	65. 9 (3)	Sr1—01—Sr1 ^{viii}	85. 9 (4)
01—Sr1—01 ⁱⁱⁱ	120. 0	01 ^{ix} —N1—01	120. 000 (1)
01 ⁱⁱ —Sr1—01 ⁱⁱⁱ	120. 0	01 ^{ix} —N1—01 ^x	120. 0
02 ⁱ —Sr1—Sr1 ^v	47. 35 (3)	01—N1—01 ^x	120. 0
02 ⁱⁱ —Sr1—Sr1 ^v	132. 65 (3)	Sr1 ^{xi} —02—Sr1 ^{vi}	93. 10 (6)
01—Sr1—Sr1 ^v	90. 0	Sr1—02—Sr1 ^{xii}	93. 10 (6)
01 ⁱⁱ —Sr1—Sr1 ^v	90. 000 (1)	Sr1 ^{xi} —02—H2	96. 7
02 ⁱ —Sr1—Sr1 ^{vi}	132. 65 (3)	Sr1—02—H2	96. 8
02 ⁱⁱ —Sr1—Sr1 ^{vi}	47. 35 (3)	Sr1 ^{xii} —02—H2	96. 7

Symmetry transformations used to generate equivalent atoms:

- (i) $-y+1, x-y, z-1$; (ii) $-x+y+1, -x+1, z$; (iii) $-y+1, x-y, z$;
- (iv) $-x+y+1, -x+1, z-1$; (v) $x, y, z-1$; (vi) $x, y, z+1$; (vii) $y, x-1, -z$;
- (viii) $y+1, x, -z$; (ix) $-y+1, x-y-1, z$; (x) $-x+y+2, -x+1, z$; (xi) $y, x, -z+1$; (xii) $y, x, -z$.

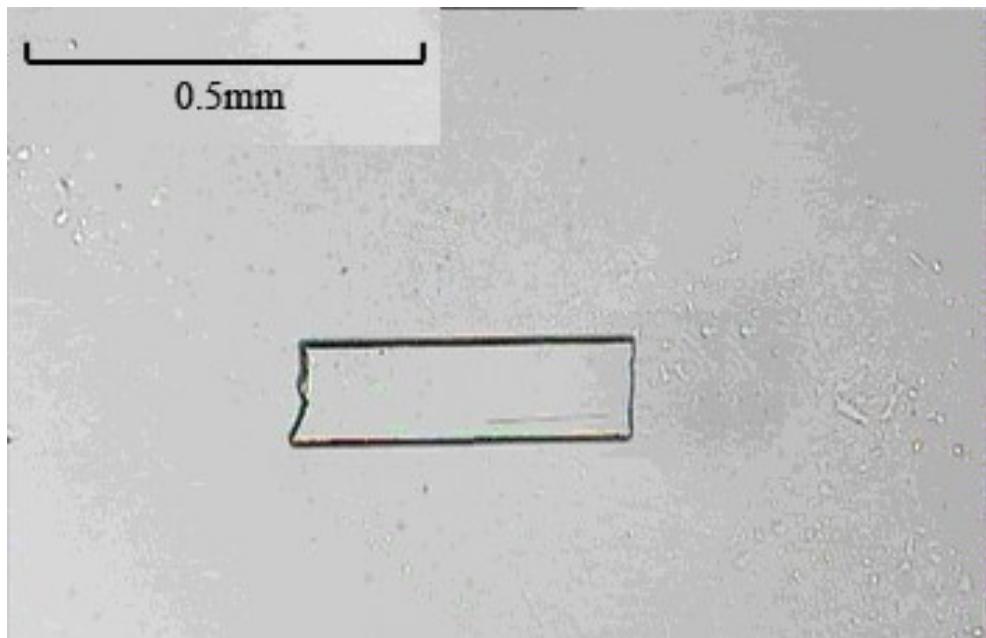


Figure S1. Photograph of $\text{Sr}_2(\text{OH})_3\text{NO}_3$ crystal.

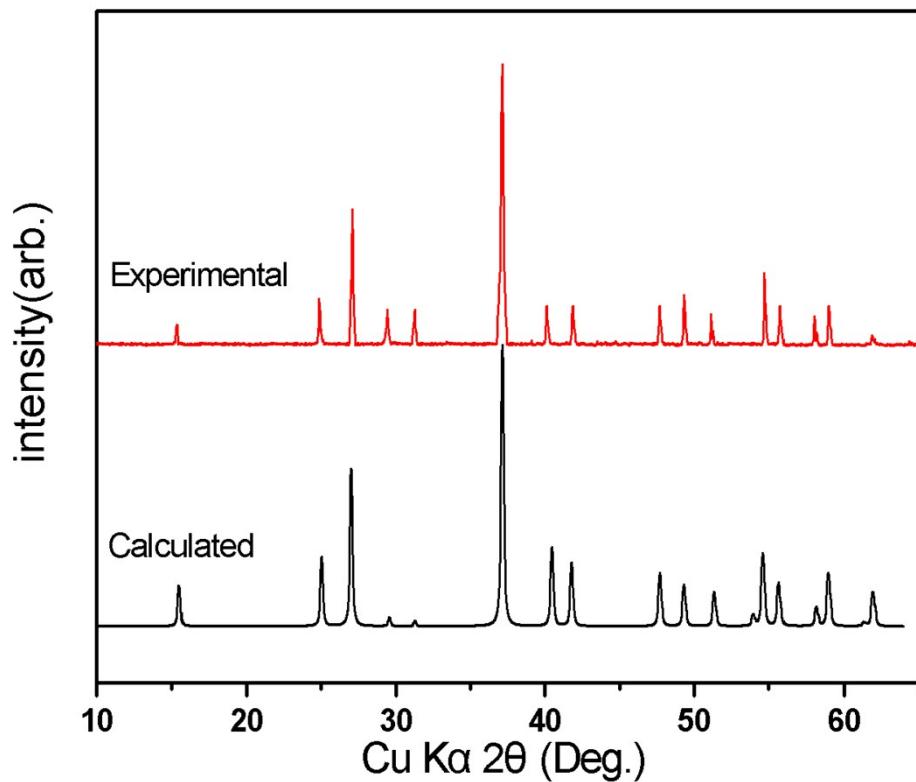


Figure S2. Experimental and calculated XRD patterns for $\text{Sr}_2(\text{OH})_3\text{NO}_3$. The black curve is the calculated one, the red is the experimental one.

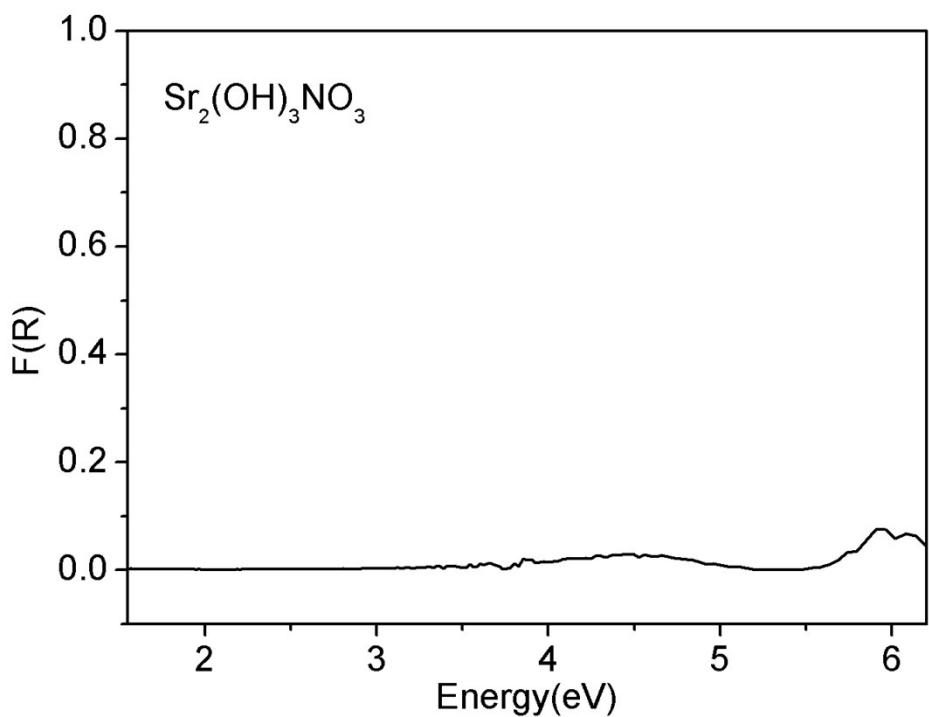


Figure S3. UV absorption spectra and Optical diffuse reflectance spectra of $\text{Sr}_2(\text{OH})_3\text{NO}_3$.

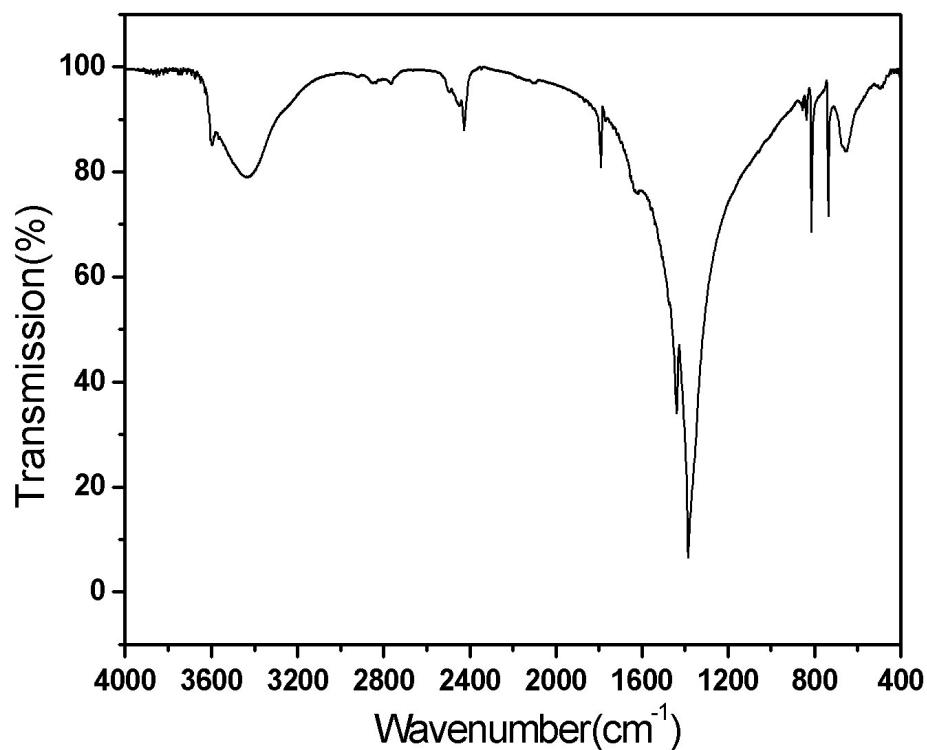


Figure S4. IR spectrum of $\text{Sr}_2(\text{OH})_3\text{NO}_3$.

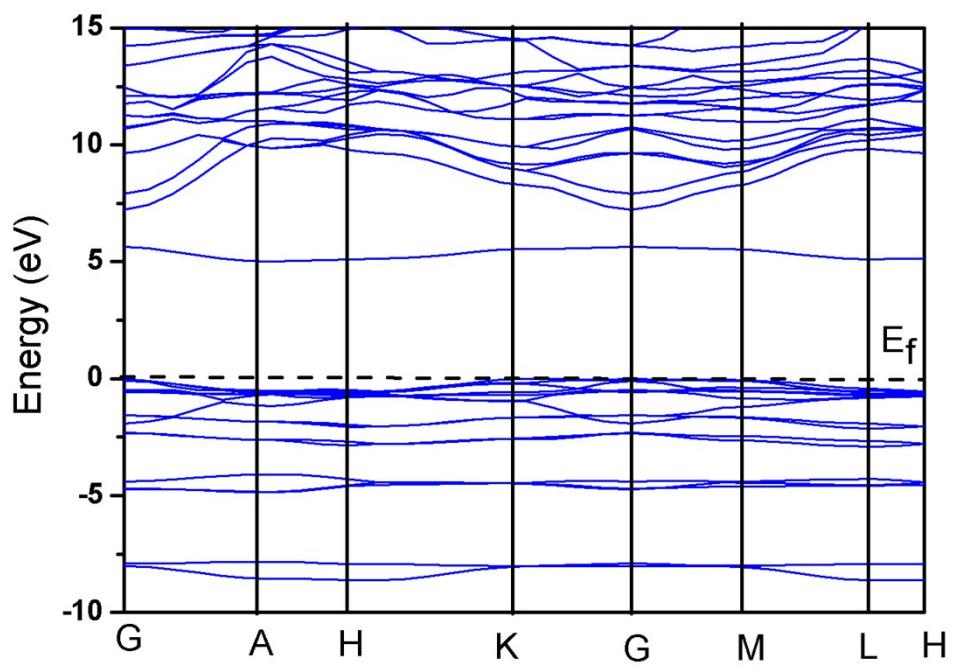


Figure S5. Calculated band structure of $\text{Sr}_2(\text{OH})_3\text{NO}_3$ (the Fermi level is set at 0 eV).