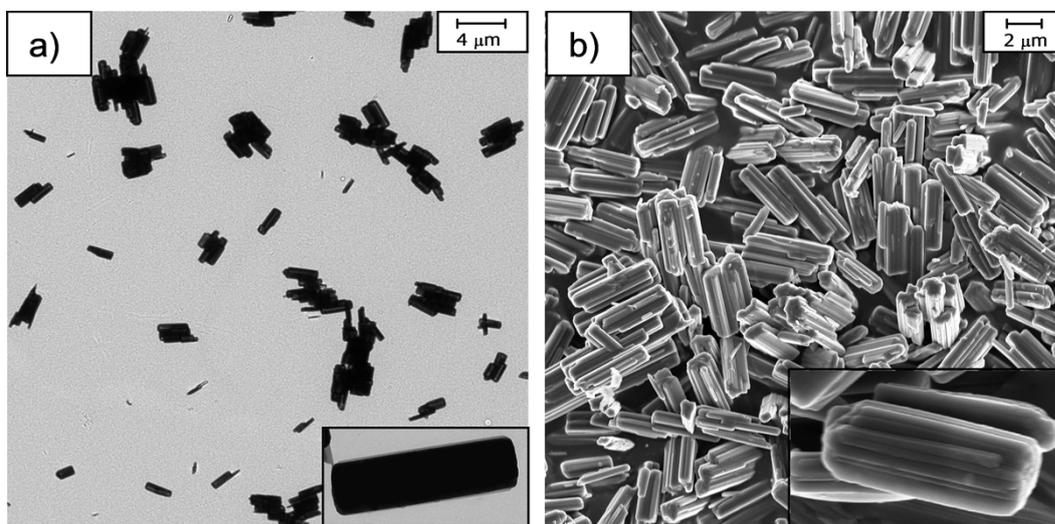


### Electronic Supporting Information

**Figure S1.-** HR-TEM (a) and HR-SEM (b) images of the bulk polymer  $\{[\text{Fe}(\text{HTrz})_2(\text{Trz})](\text{BF}_4)\}_n$  prepared in following a classical procedure.



**Table S1.-** Elemental analysis of samples described in this work

		C [%]	N[%]	H[%]
<b>1</b>	exptl	19.30	32.81	2.58
	calcd	19.41	33.95	2.85
Proposed Formula: $[\text{Fe}(\text{HTrz})_2(\text{Trz})](\text{BF}_4) \cdot (\text{H}_2\text{O})_{1.25}$				
<b>2</b>	exptl	19.18	33.31	2.61
	calcd	19.41	33.95	2.85
Proposed Formula: $[\text{Fe}(\text{HTrz})_2(\text{Trz})](\text{BF}_4) \cdot (\text{H}_2\text{O})_{1.25}$				
<b>3</b>	exptl	19.31	33.71	2.65
	calcd	19.41	33.95	2.85
Proposed Formula: $[\text{Fe}(\text{HTrz})_2(\text{Trz})](\text{BF}_4) \cdot (\text{H}_2\text{O})_{1.25}$				
<b>4</b>	exptl	19.59	34.43	2.52
	calcd	19.41	33.95	2.85
Proposed Formula: $[\text{Fe}(\text{HTrz})_2(\text{Trz})](\text{BF}_4) \cdot (\text{H}_2\text{O})_{1.25}$				
<b>1@SiO<sub>2</sub></b>	exptl	16.68	28.33	2.36
	calcd	16.43	28.75	2.25
Proposed Formula: $[\text{Fe}(\text{HTrz})_2(\text{Trz})](\text{BF}_4) \cdot (\text{SiO}_2)_{1.2} \cdot (\text{H}_2\text{O})_{0.9}$				
<b>2@SiO<sub>2</sub></b>	exptl	15.50	26.07	2.13
	calcd	15.11	26.44	2.16
Proposed Formula: $[\text{Fe}(\text{HTrz})_2(\text{Trz})](\text{BF}_4) \cdot (\text{SiO}_2)_{1.8} \cdot (\text{H}_2\text{O})_{1.1}$				
<b>3@SiO<sub>2</sub></b>	exptl	15.38	25.46	2.20
	calcd	14.82	25.91	2.19
Proposed Formula: $[\text{Fe}(\text{HTrz})_2(\text{Trz})](\text{BF}_4) \cdot (\text{SiO}_2)_{1.9} \cdot (\text{H}_2\text{O})_{1.3}$				
<b>4@SiO<sub>2</sub></b>	exptl	16.84	28.56	1.98
	calcd	16.45	28.78	2.11
Proposed Formula: $[\text{Fe}(\text{HTrz})_2(\text{Trz})](\text{BF}_4) \cdot (\text{SiO}_2)_{1.3} \cdot (\text{H}_2\text{O})_{0.6}$				

Figure S2.- Thermal variation of the  $\chi_M T$  product of 1-dansyl.

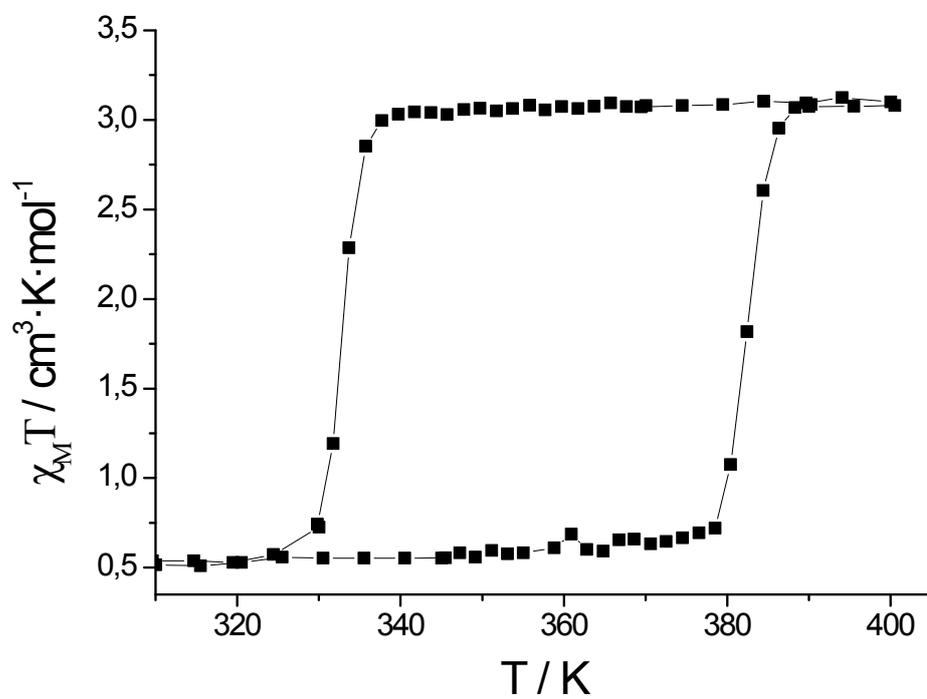
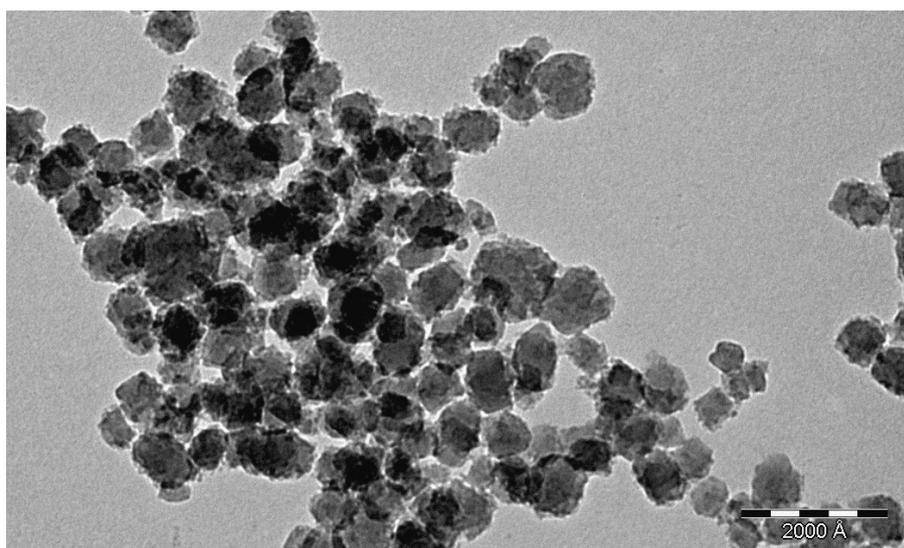


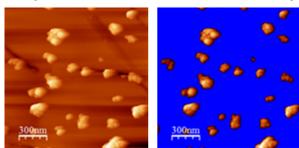
Figure S3.- HR-TEM image of 1-dansyl



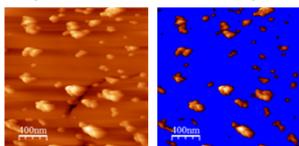
**Figure S4.-** AFM images of **1@SiO<sub>2</sub>-SH** nanoparticles deposited on a gold surface at two different immersion times.

immersion time: 6 hours

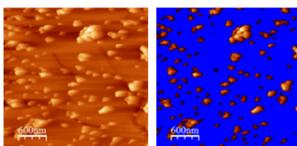
$$\text{Density} = 31 / 2,25 = 13,77 \text{ particles}/\mu^2$$



$$\text{Density} = 55 / 4 = 13,75 \text{ particles}/\mu^2$$

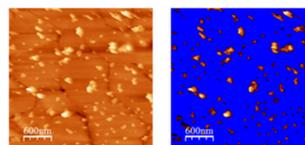


$$\text{Density} = 125 / 9 = 13,88 \text{ particles}/\mu^2$$

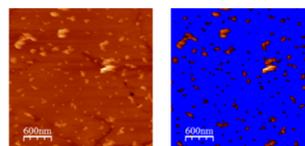


immersion time: 12 hours

$$\text{Density} = 153 / 9 = 17 \text{ particles}/\mu^2$$



$$\text{Density} = 169 / 9 = 18,77 \text{ particles}/\mu^2$$



$$\text{Density} = 59 / 2,89 = 20,41 \text{ particles}/\mu^2$$

