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Supporting Information for

Modulation of the Ligand-Based Fluorescence of 3d Metal Complexes upon Spin State Change

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Table S1: Crystallographic data.

Compound	**[NiL1]**	**[CuL1(EtOH)]·2 $CHCl_3$**
formula	$C_{26}H_{24}N_4NiO_6$	$C_{30}H_{32}N_4Cl_6CuO_{67}$
CCDC number	1055605	1055606
M_r / g mol^{-1}	547.20	836.83
crystal system	Triclinic	Triclinic
space group	*P-1*	*P-1*
a / Å	4.084(1)	8.651(1)
b / Å	12.918(3)	11.798(2)
c / Å	22.854(6)	18.199(3)
α / °	101.906(19)	71.833(11)
β / °	91.82(2)	86.734(12)
γ / °	94.00(2)	88.122(12)
V / Å³	1175.6(5)	1761.8(4)
Z	2	2
ρ / g cm^{-3}	1.546	1.577
μ / mm^{-1}	0.877	1.126
crystal size	0.06 × 0.14 × 0.29	0.084 × 0.103 × 0.115
T / K	133(2)	133(2)
λ (MoK_α) / Å	0.71073	0.71073
θ-range / °	1.7–27.8	1.179–28.0
reflns. collected	8045	7594
indep. reflns. (R_{int})	4136 (0.084)	3682 (0.173)
parameters	334	434
R (*F*) (all data)	0.0497 (0.1498)	0.1176 (0.1779)
wR2	0.0720	0.2963
GooF	0.680	0.988

Figure S1: Powder diffraction patterns of all synthesised complexes.

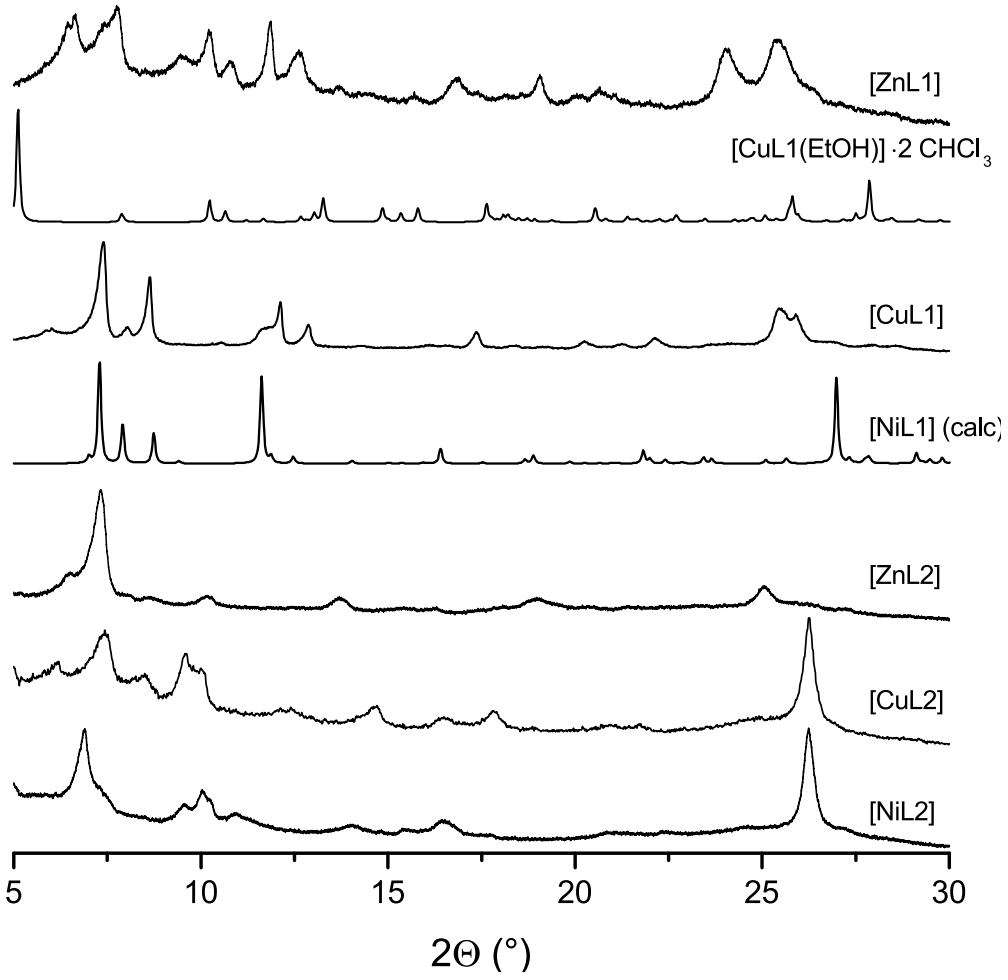

Figure S2: Intensity of characteristic absorption bands for the complex [NiL1] in trichloromethane and pyridine solutions, upon complexation and de-complexation of the pyridine molecules.

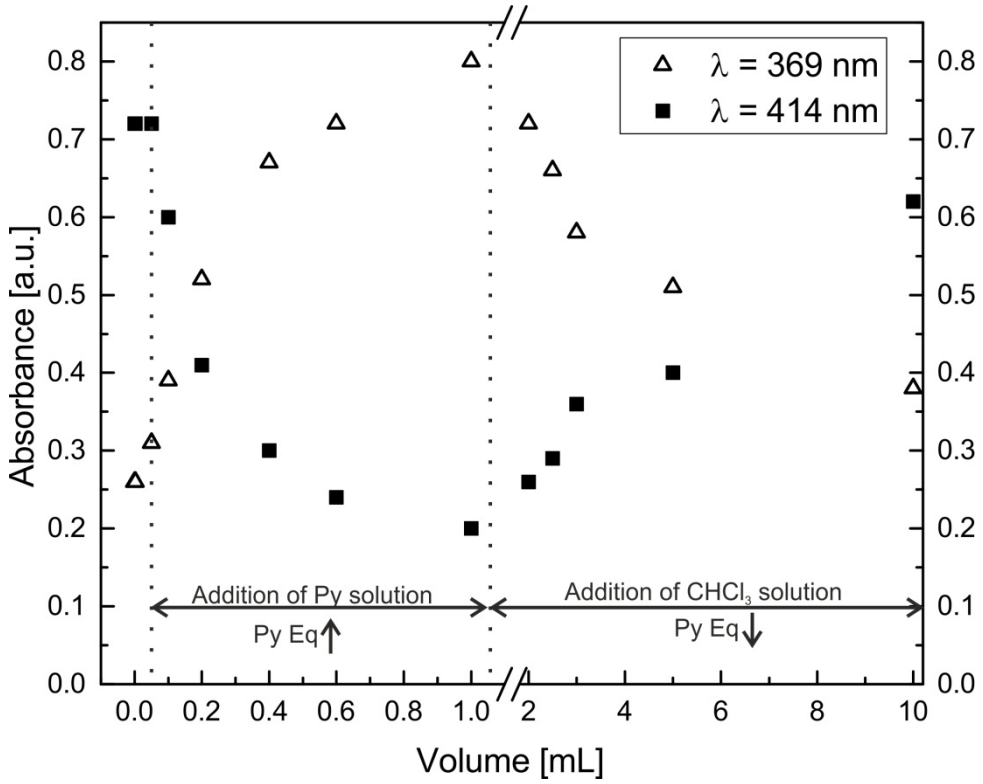

Figure S3: UV-Vis spectra of [NiL1], [NiL2], [ZnL1], and [ZnL2] in $CHCl_3/Et_3N$ and $CHCl_3$/HCOOH mixtures.