

Supporting Materials for

**Transparent anhydride-cured epoxy nanocomposites reinforced with
polyaniline stabilized nanosilica**

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S. 1 Chemical Structure Characterization of the Functionalized NanoSilica

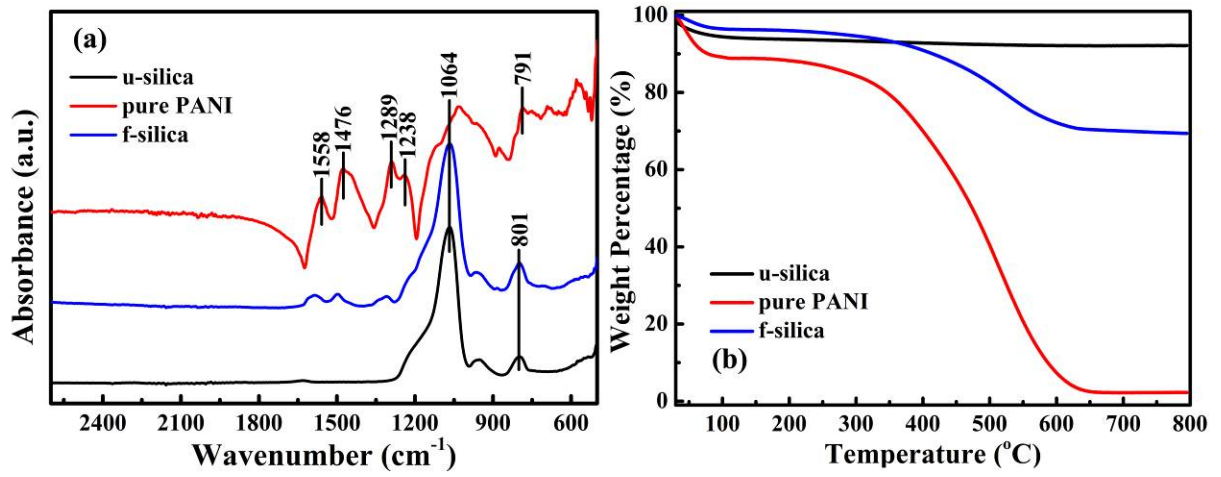


Fig. S1. (a) FT-IR spectra, and (b) TGA (air condition) of u-silica, pure PANI, and f-silica.

S. 2 Viscosity of DER 331 Resin NanoSuspension with PANI Nanoparticles

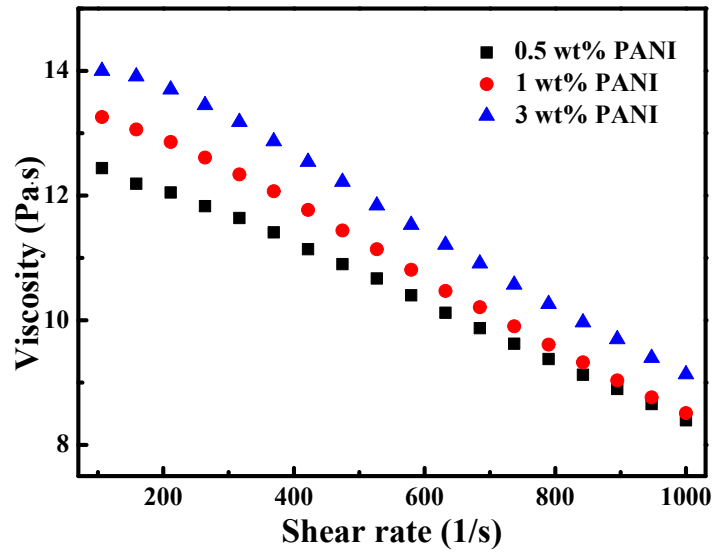


Fig. S2. Viscosity vs. shear rate of the epoxy resin nanosuspensions filled with different loadings of PANI nanoparticles at 25 °C.

S. 3 Temperature Dependent Viscosity of DER 331 Resin Monomer

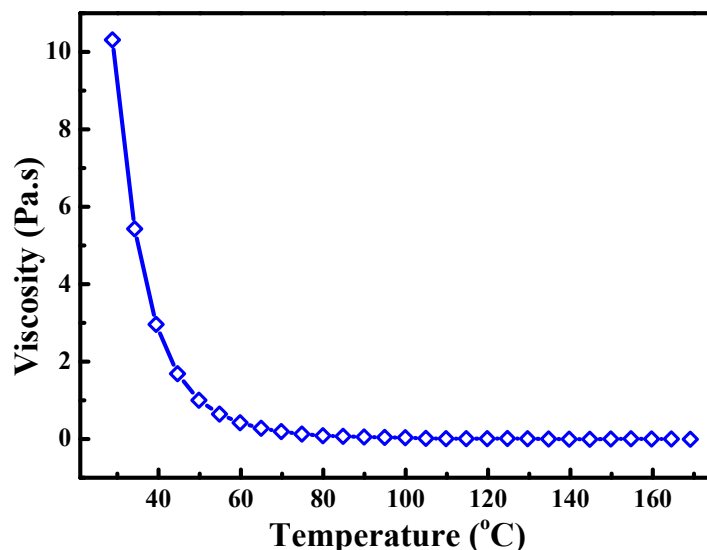


Fig. S3 Viscosity vs. Temperature of DER 331 resin monomer.

S. 4 Comparison of Curing Process

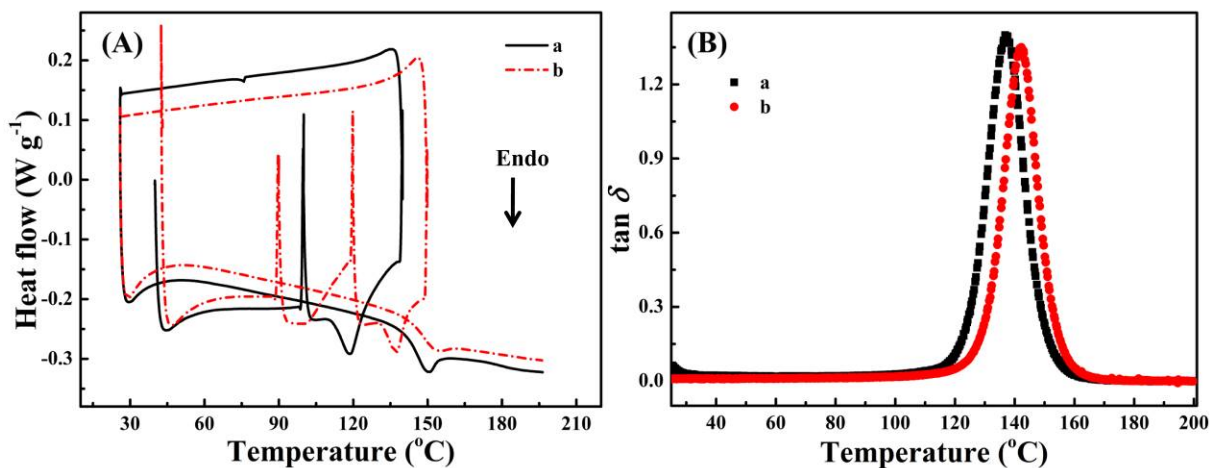


Fig. S4. For curing: (A) DSC curves; and (B) $\tan \delta$ vs. temperature of (a) at 100 °C ~ 3 h, and 140 °C ~ 6 h; and (b) at 90 °C ~ 2 h, 120 °C ~ 2 h, and 150 °C ~ 5 h.

The DSC measurements of (a) process were carried out under a nitrogen flow rate of approximately 20 mL min⁻¹ at a heating rate of 10 °C min⁻¹ from 25 to 100 °C; and an isothermal process at 100 °C was continued for 180 min. Then the temperature was increased to 140 °C at a

heating rate of $10\text{ }^{\circ}\text{C min}^{-1}$ and an isothermal process was conducted at $140\text{ }^{\circ}\text{C}$ for 360 min. After that, the temperature was decreased to $25\text{ }^{\circ}\text{C}$ at a rate of $10\text{ }^{\circ}\text{C min}^{-1}$ and then the temperature was increased from 25 to $200\text{ }^{\circ}\text{C}$ at a heating rate of $10\text{ }^{\circ}\text{C min}^{-1}$.

The DSC tests of **(b)** process were obtained under a nitrogen flow rate of approximately 20 mL min^{-1} at a heating rate of $10\text{ }^{\circ}\text{C min}^{-1}$ from 25 to $90\text{ }^{\circ}\text{C}$, and an isothermal process at $90\text{ }^{\circ}\text{C}$ was continued for 120 min. Then the temperature was increased to $120\text{ }^{\circ}\text{C}$ at a heating rate of $10\text{ }^{\circ}\text{C min}^{-1}$ and an isothermal process was conducted at $120\text{ }^{\circ}\text{C}$ for 120 min. After that, the temperature was increased to $150\text{ }^{\circ}\text{C}$ at a heating rate of $10\text{ }^{\circ}\text{C min}^{-1}$ and an isothermal process was conducted at $150\text{ }^{\circ}\text{C}$ for 240 min. Finally, the temperature was decreased to $25\text{ }^{\circ}\text{C}$ at a rate of $10\text{ }^{\circ}\text{C min}^{-1}$ and then the temperature was increased to $200\text{ }^{\circ}\text{C}$ at a heating rate of $10\text{ }^{\circ}\text{C min}^{-1}$.

S. 5 Thermogravimetric Analysis (TGA)

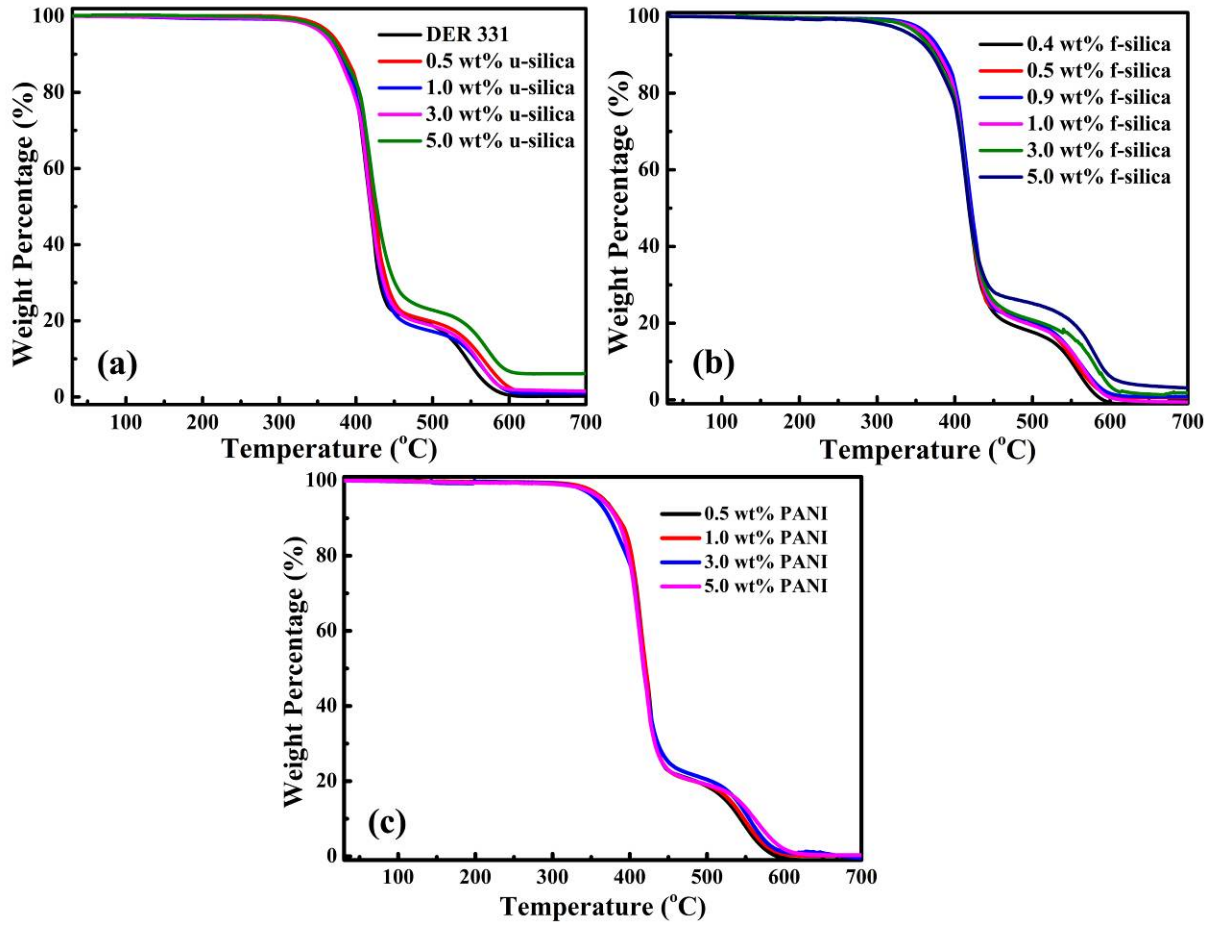


Fig. S5. TGA curves of (a) cured DER 331 resin and its nanocomposites with different u-silica loadings and (b) nanocomposites filled with different f-silica loadings (c) nanocomposites with different loadings of PANI nanoparticles.

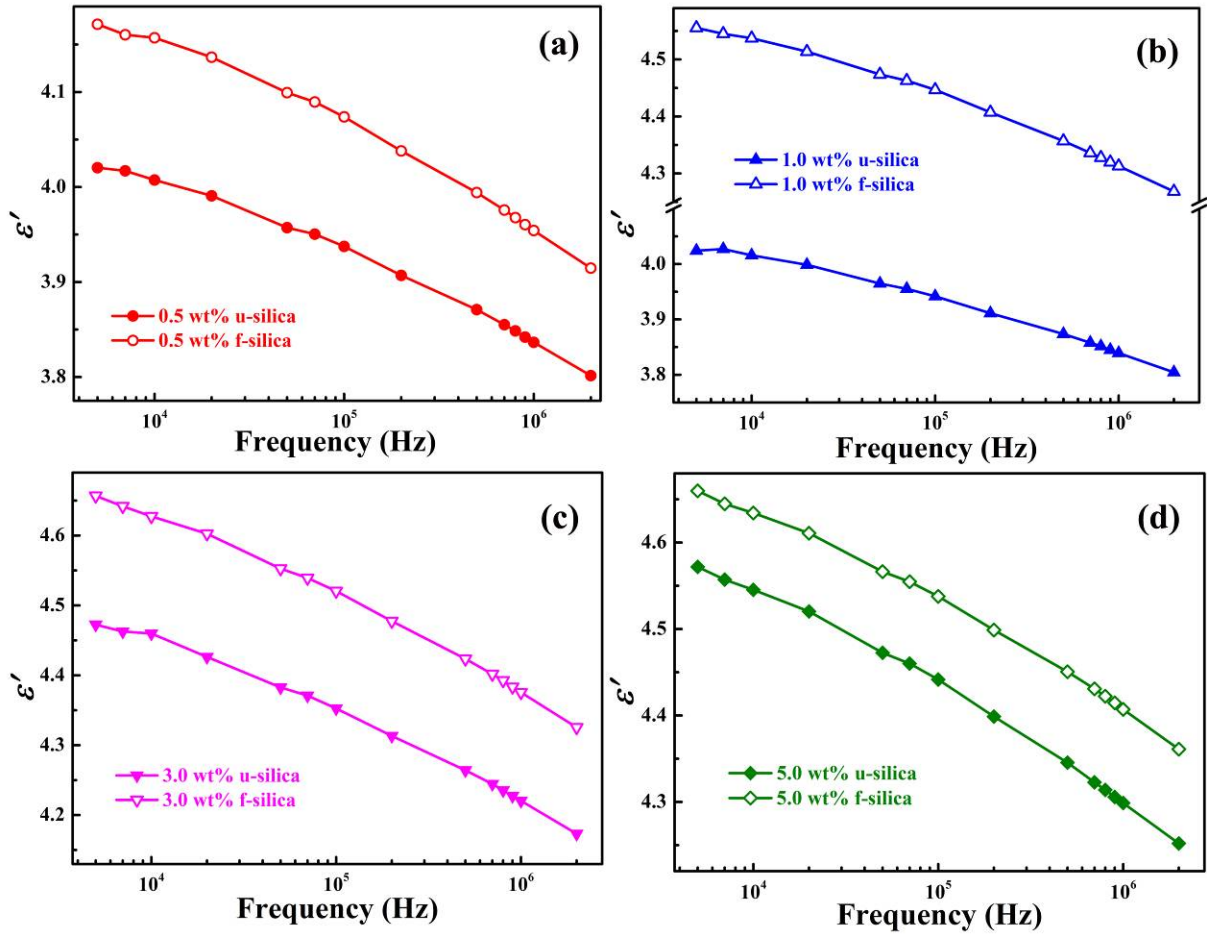


Fig. S6. Effect of surface treatment on the real permittivity (ϵ') for the cured DER 331 nanocomposites filled with different silica loadings.

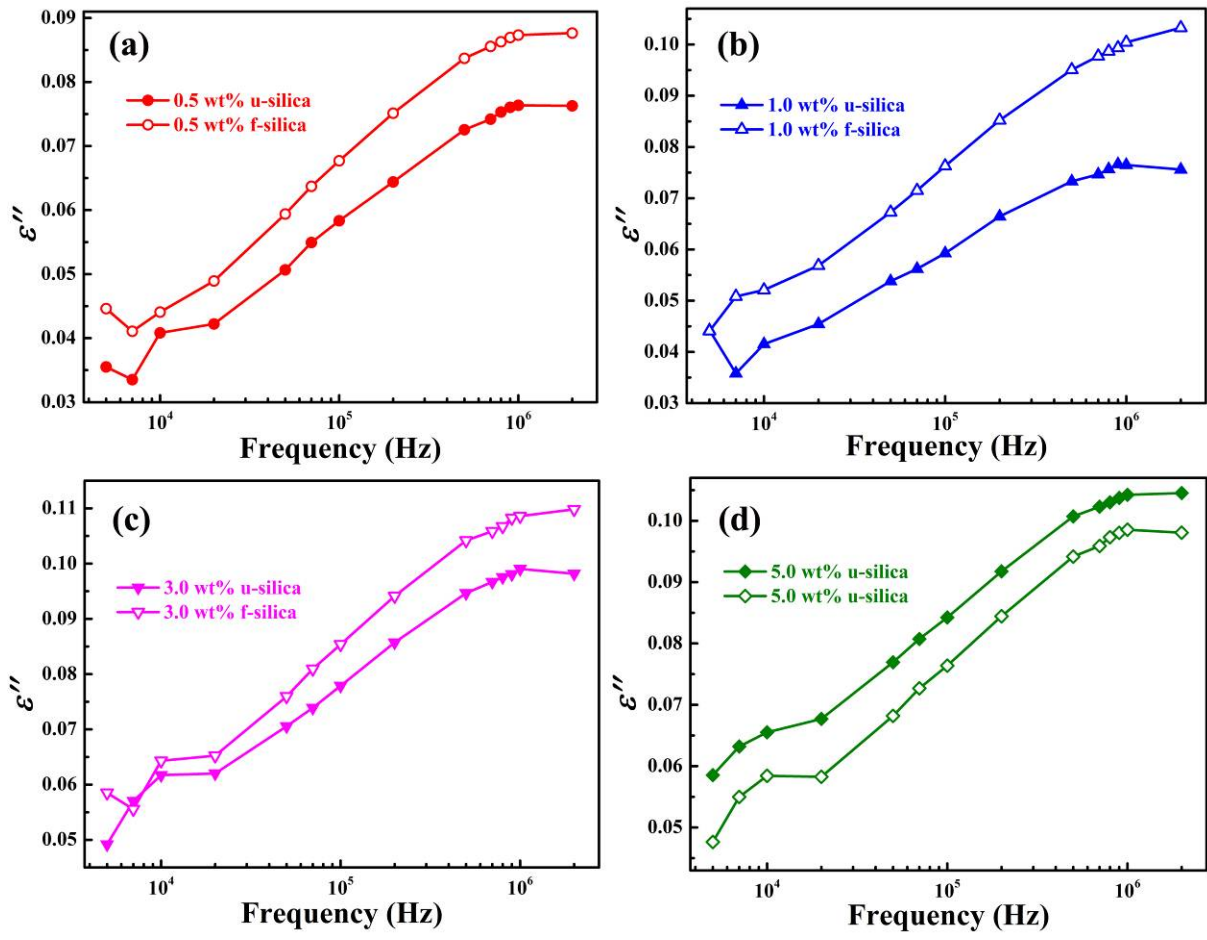


Fig. S7. Effect of surface treatment on the imaginary permittivity (ϵ'') for the cured DER 331 nanocomposites filled with different silica loadings.

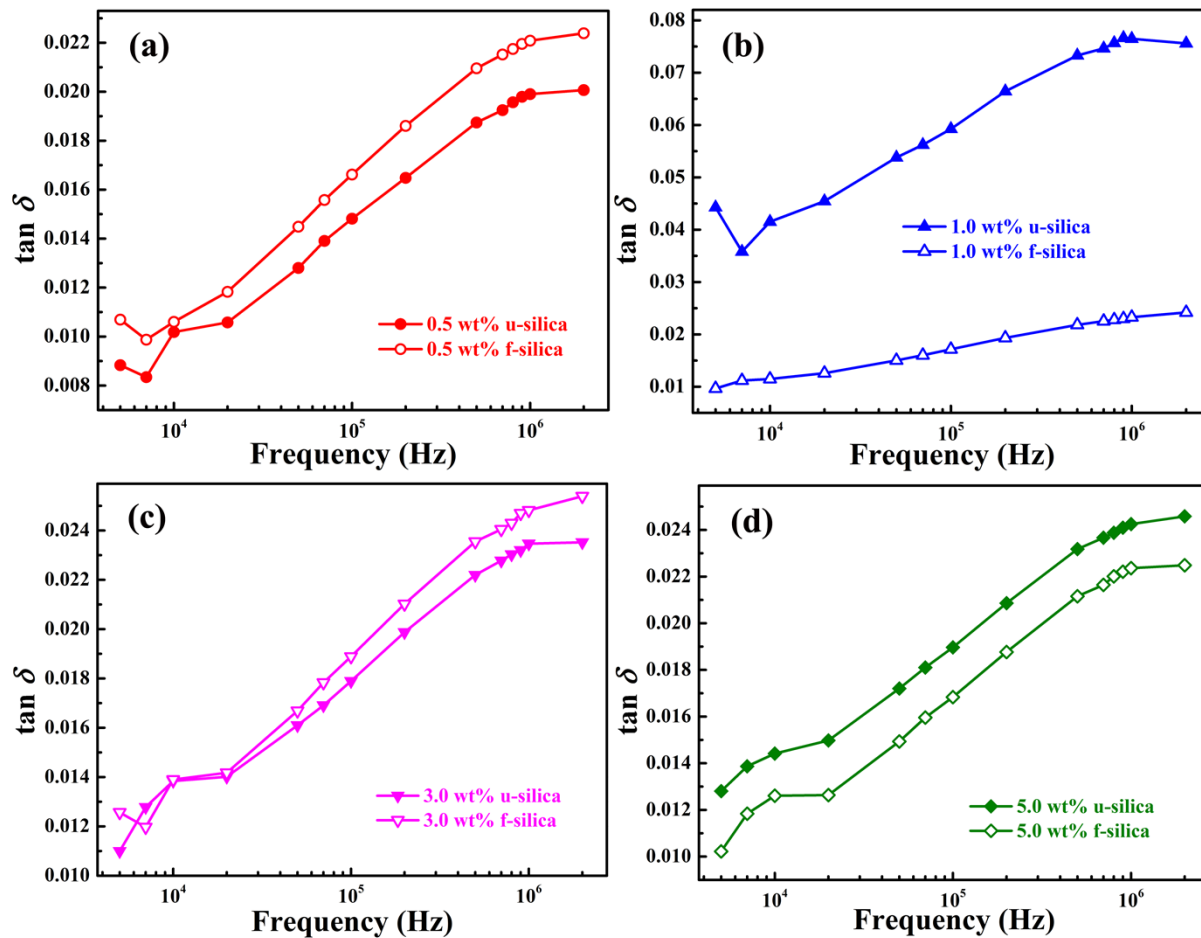


Fig. S8. Effect of surface treatment on the $\tan \delta$ for the cured DER 331 nanocomposites filled with different silica loadings.

S.6 Transparent Tests for DER 331 Nanocomposites Filled with PANI Nanoparticles

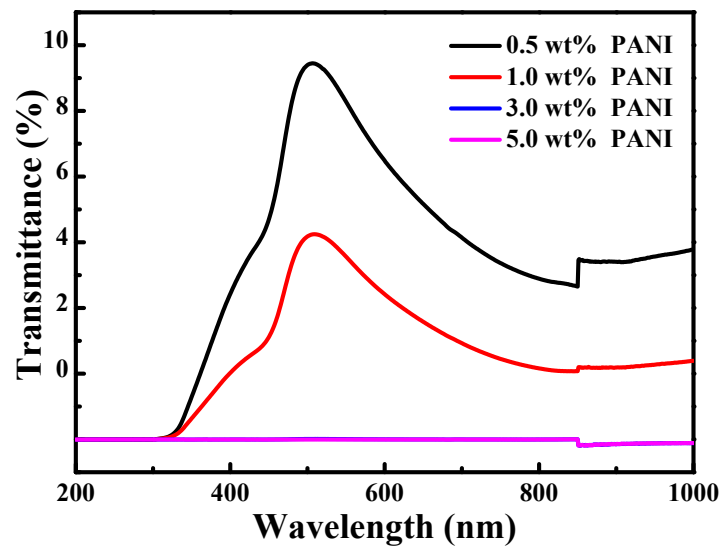


Fig. S6. Transmittance curves of the cured DER 331 resin filled with different PANI nanoparticles.