

Electronic Supplementary Information

Crystal Structure, Electronic Band Structure and High-Temperature Thermoelectric Properties of Te-Substituted Tetrahedrites $\text{Cu}_{12}\text{Sb}_{4-x}\text{Te}_x\text{S}_{13}$ ($0.5 \leq x \leq 2.0$)

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1. Table S1. Single-crystal data collection and structure refinement at 300 K.

Chemical formula	Cu ₁₂ Sb ₂ Te ₂ S ₁₃ (series S1)
Molar mass (g.mol ⁻¹)	1678.02
Symmetry	Cubic
Space group	<i>I</i> -43 <i>m</i>
<i>a</i> (Å)	10.3520(1)
<i>V</i> (Å ³)	1108,91(1)
<i>Z</i>	2
ρ (g.cm ⁻³)	5.025
Radiation	Mo K α
θ range	3 – 79.39°
Linear Abs. coeff. μ (mm ⁻¹)	17.24
Reflection collected	3465
Reflection unique	666
Number of parameters	18
Final <i>R</i> value (%)	2.79
<i>S</i> (Goodness of fit)	0.942

2. SEM images and X-ray mapping of the $x = 0.61$, 1.40 and 1.79 samples (series S1)

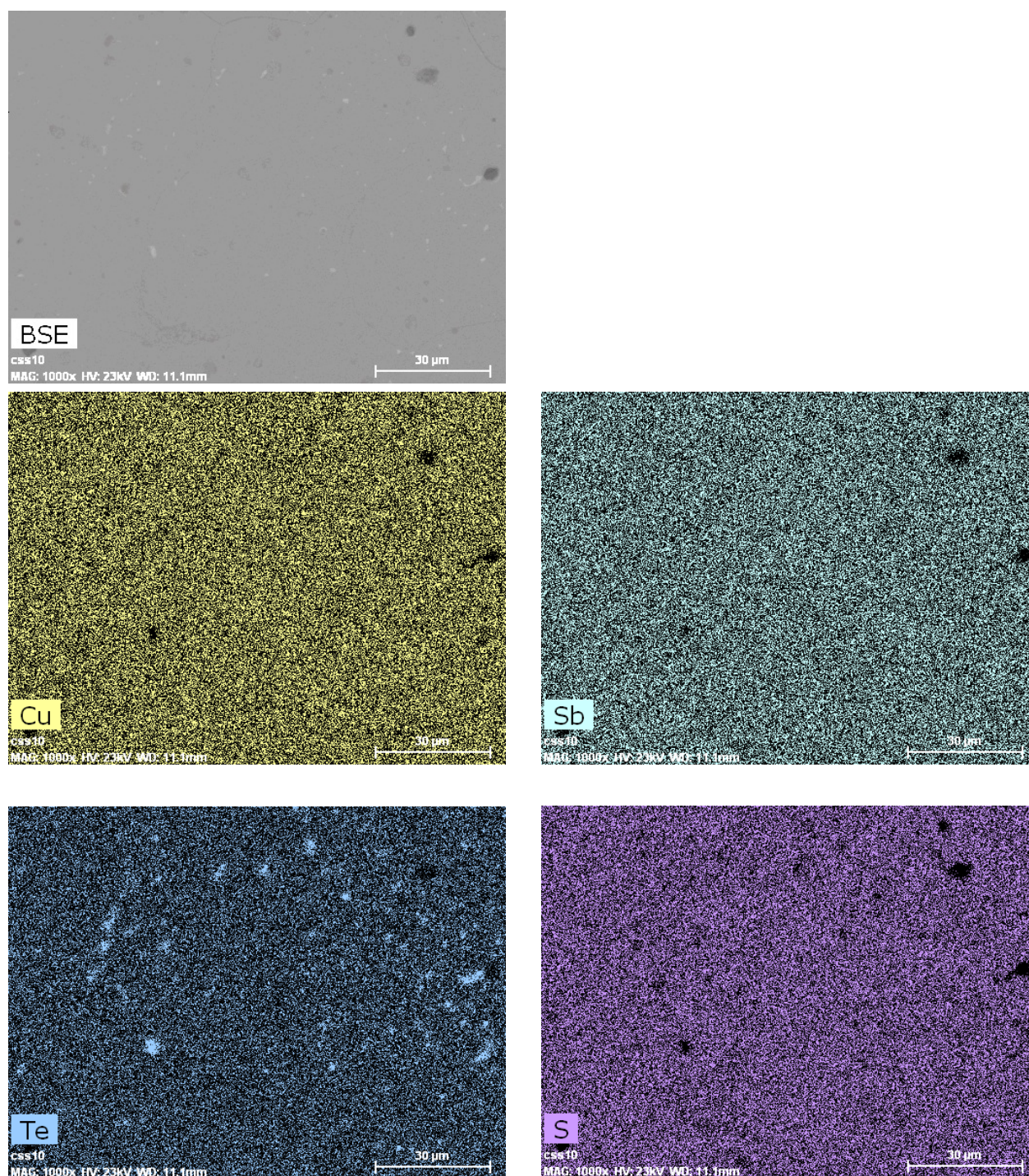


Figure S1. BSE images and X-ray mapping performed on the $x = 0.61$ sample.

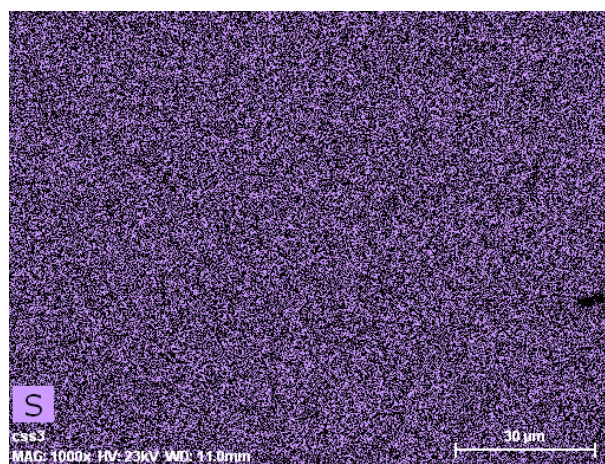
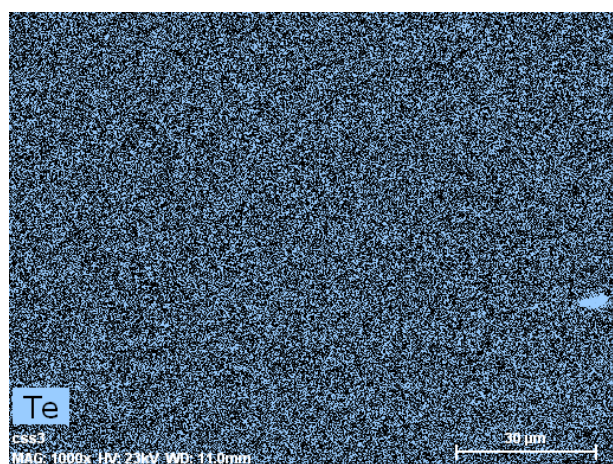
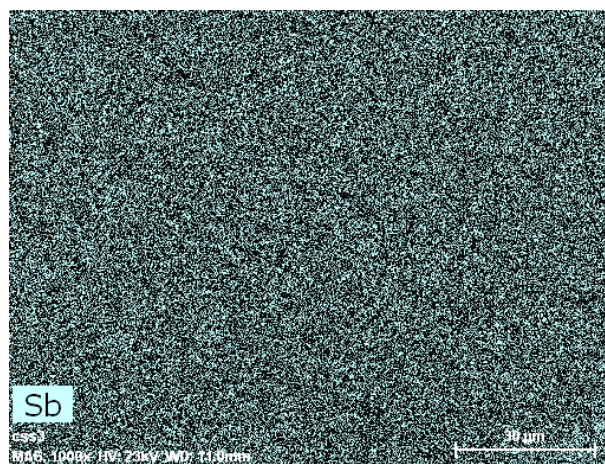
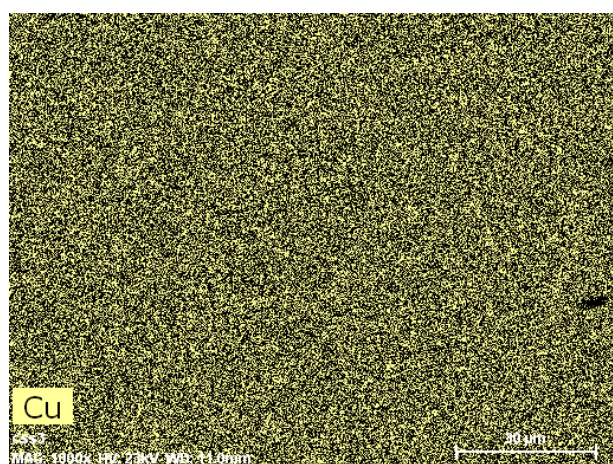
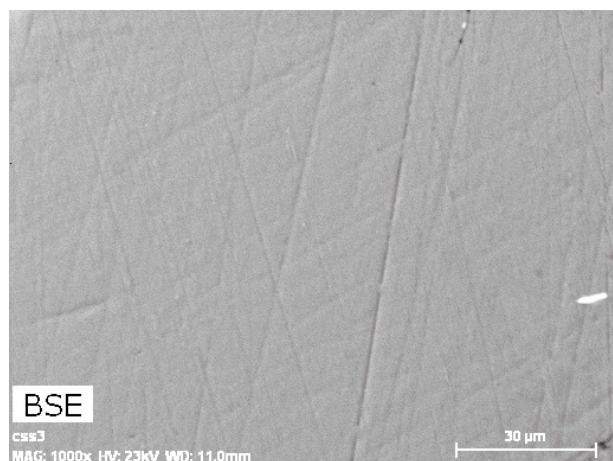


Figure S2. BSE images and X-ray mapping performed on the $x = 1.40$ sample.

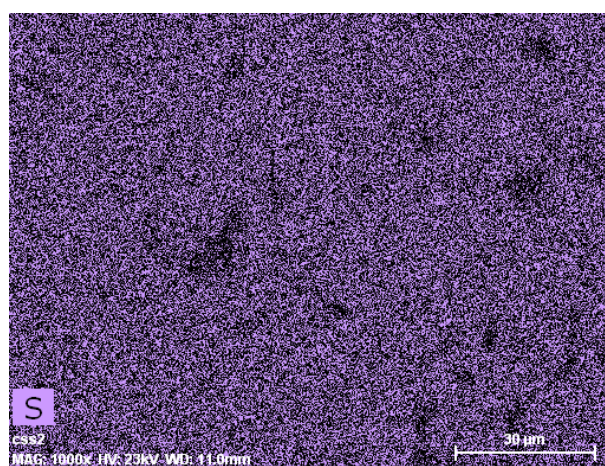
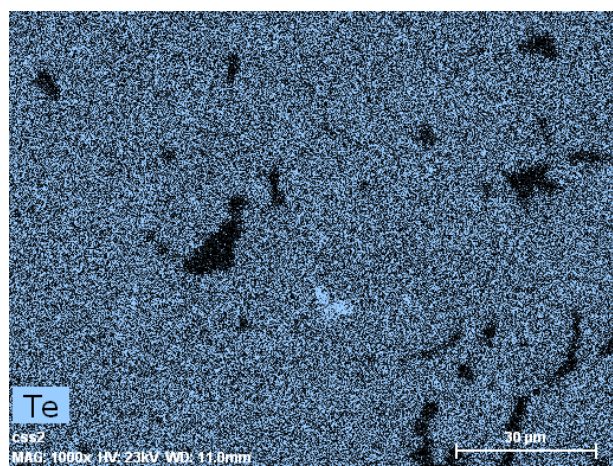
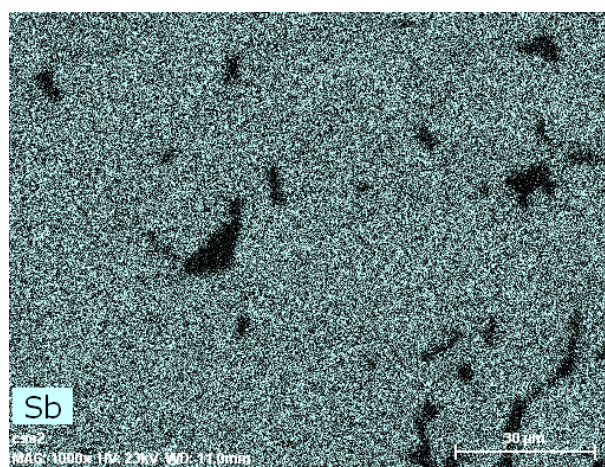
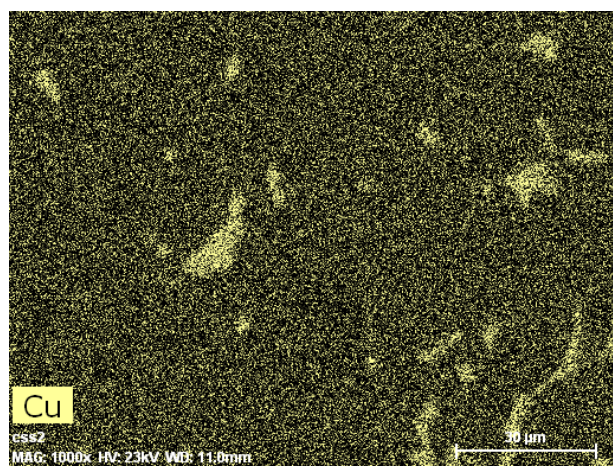
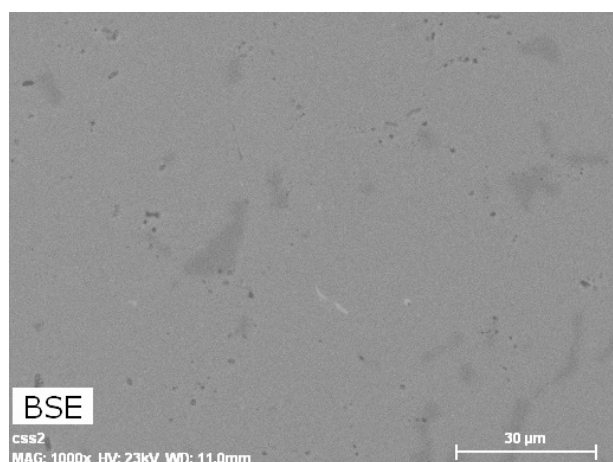


Figure S3. BSE images and X-ray mapping performed on the $x = 1.79$ sample.

2. Table S2. Atomic coordinates, site occupancies and equivalent atomic thermal displacement parameters ($\text{\AA}^2$).

Site	Atom	x	y	z	s.o.f	U_{11}	U_{22}	U_{33}	U_{23}	U_{13}	U_{12}
12d	Cu1	0.5	0	0.25	0.25	0.0192(3)	0.0192(3)	0.0256(5)	0	0	0
12e	Cu2	0	0	0.2168(3)	0.25	0.0737(1)	0.0737(1)	0.0185(1)	0	0	-0.052(1)
8c	Sb	0.26391(4)	0.26391(4)	0.26391(4)	0.1667	0.0124(2)	0.0124(2)	0.0124(2)	-0.0008(1)	-0.0008(1)	-0.0008(1)
24g	S1	0.8855(1)	0.8855(1)	0.3622(2)	0.5	0.0141(4)	0.0141(4)	0.0110(6)	0.001(4)	0.001(4)	-0.011(5)
2a	S2	0	0	0	0.04167	0.0174(1)	0.0174(1)	0.0174(1)	0	0	0

3. Table S3. Refined bond distances and bond angles from single-crystal XRD data at room temperature. Data obtained by Pfitzner *et al.* and Makovicky and Skinner on single-crystals of the ternary compound have been added for comparison.

Composition	$\text{Cu}_{12}\text{Sb}_2\text{Te}_2\text{S}_{13}$	$\text{Cu}_{12}\text{Sb}_4\text{S}_{13}$	$\text{Cu}_{12.3}\text{Sb}_4\text{S}_{13}$
Cu1 – S1 ($\times 4$)	2.33508	2.3133(8)	2.311(4)
Cu2 – S1 ($\times 2$)	2.25911	2.264(2)	2.246(7)
Cu2 – S2	2.16048	2.249(2)	2.250(6)
Cu2 – Cu2 ($\times 4$)	3.05538	3.180(3)	3.186(8)
Cu2 – Sb/Te ($\times 2$)	3.47533	3.4090(4)	3.386
Sb/Te – S1 ($\times 3$)	2.45991	2.438(1)	2.452(4)
S1 – Cu1 – S1 ($\times 2$)	112.55	106.37(4)	105.45(15)
S1 – Cu2 – S1	91.24	96.6(1)	96.04(25)
S2 – Cu2 – S1 ($\times 2$)	134.38	131.70(7)	131.98(10)
Sb/Te – Cu2 – Sb/Te	170.48	174.6(1)	175.2
S1 – Sb/Te – S1 ($\times 3$)	96.67	95.87(4)	95.70(10)

5. High-temperature PXRD patterns of the $x = 0.61$ (a) and 1.79 (b) samples (series S1)

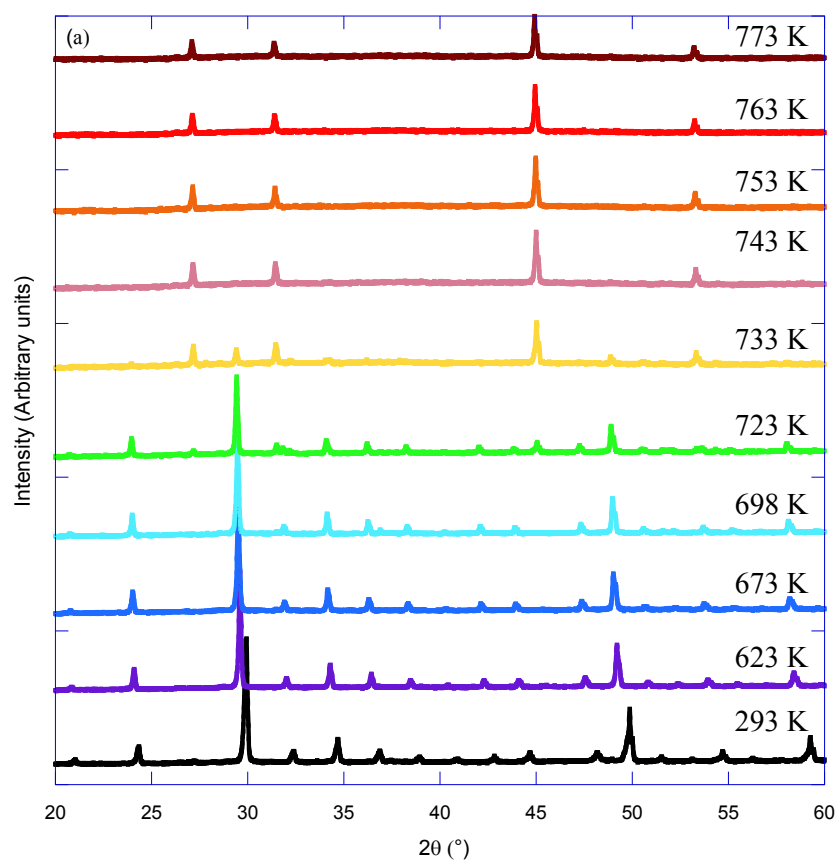


Figure S4. High-temperature PXRD spectra of the $x = 0.61$ sample.

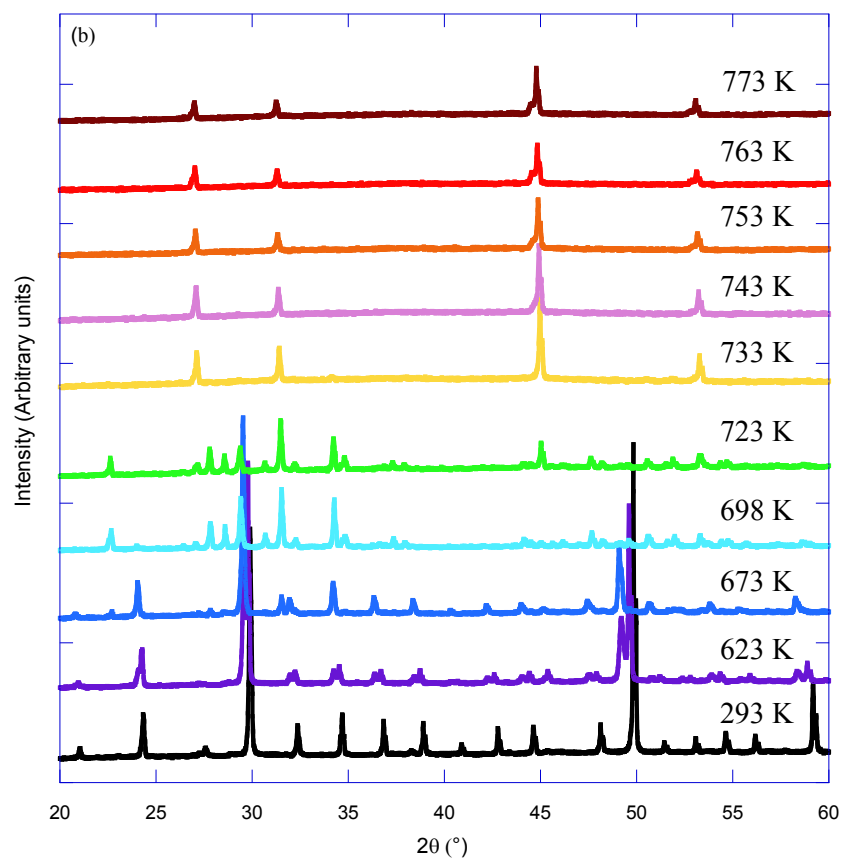


Figure S5. High-temperature PXRD spectra of the $x = 1.79$ sample.

6. Temperature dependence of the specific heat of the $x = 0.80$ sample (series S1)

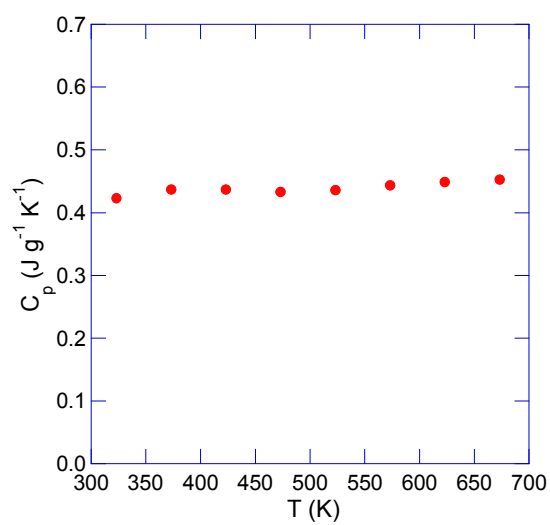


Figure S6. Temperature dependence of the specific heat C_p of the $x = 0.80$ sample.