Support Information

Synthesis, Mechanistic Investigation, and Application of Photoluminescent Sulfur and Nitrogen Co-Doped Carbon Dots

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Figure S2. UV absorption spectrum of S,N-CDs



Figure S3.Time-dependent fluorescence intensity of S,N-CDs($\lambda_{em} = 440 \text{ nm}$) in presence of H_2O_2



Figure S4 Effect of ionic strengths on the fluorescence intensity of sulfur-doped carbon dots



Figure S5. High-resolution O1s XPS spectra for S,N-CDs





Figure S7. Lifetime spectra for S,N-CDs



Figure S8. Raman spectra for S,N-CDs

im citrate					
Precursor	$S_{2p}(At.$	C_{1s}	Na _{1s}	O_{1s}	N_{1s}
Ratio	%)	(At. %)	(At. %)	(At. %)	(At. %)
0.05	0.885	46.346	18.791	33.261	0.717
0.075	0.82	47.413	17.78	33.245	0.742
0.1	0.789	42.34	21.276	34.816	0.78
1	5.335	37.758	17.761	33.283	5.863
2	12.892	16.432	23.809	42.774	4.093
5	14.467	15.884	20.942	43.669	5.039

Table S1 Atomic percent for S,N-CDs at different molar ratio of the sulfamide and sodium citrate