

Electronic Supplementary Material (ESI) for Journal of Materials Chemistry C.

This journal is © The Royal Society of Chemistry 2015

Facile synthesis of CuO micro-sheets over peeling off Cu foil in oxalic acid solution and their sensing properties towards n-Butanol

Chengjun Dong ^{a,b}, Lihong Wang ^a, Gang Chen ^a, Xuechun Xiao ^{a,b}, Igor Djerdj ^c and Yude

Wang ^{*a,b}

a School of Physical Science and Technology, Yunnan University, 650091 Kunming, People's Republic of China. Fax: +86-871-65153832; Tel: +86-871-65031124; E-mail: ydwang@ynu.edu.cn

b Yunnan Province Key Lab of Mico-Nano Materials and Technology, Yunnan University, 650091 Kunming, People's Republic of China

c RuđerBošković Institute, Bijeni čka 54, 10000 Zagreb, Croatia. Fax: +38514680114; Tel: +38514680113; E-mail: igor.djerdj@irb.hr.

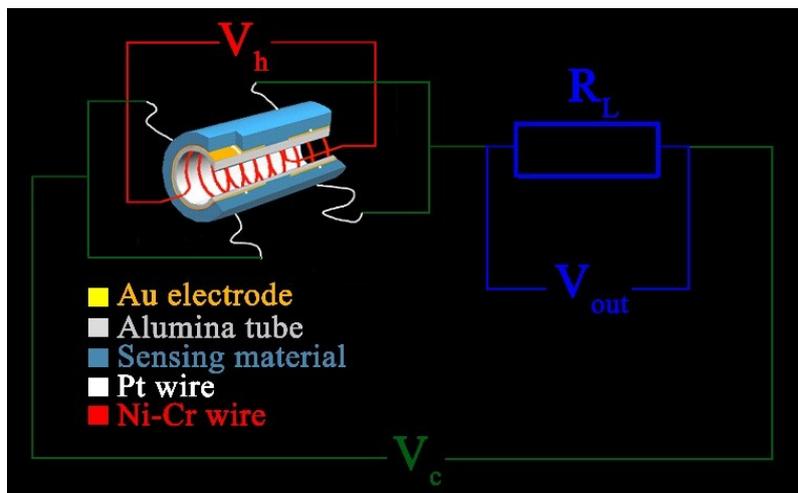


Fig. S1 Schematic structure along with the testing principle of the gas sensor.

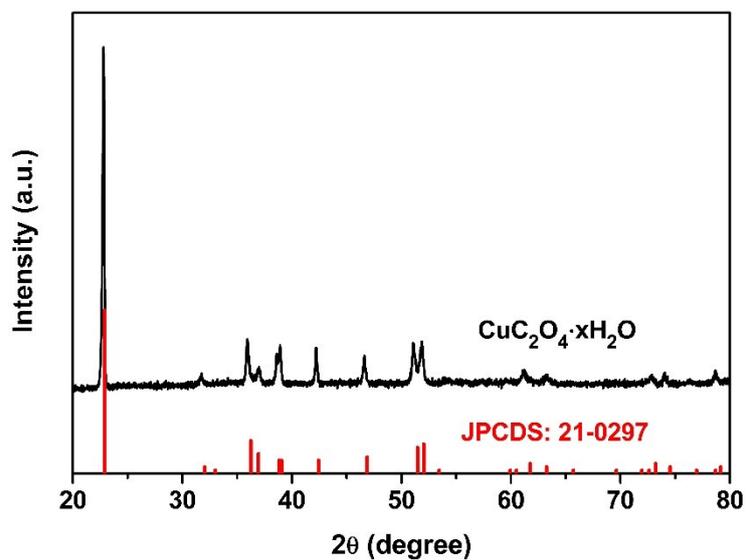


Fig. S2 XRD pattern of $\text{CuC}_2\text{O}_4 \cdot x\text{H}_2\text{O}$ precursors.

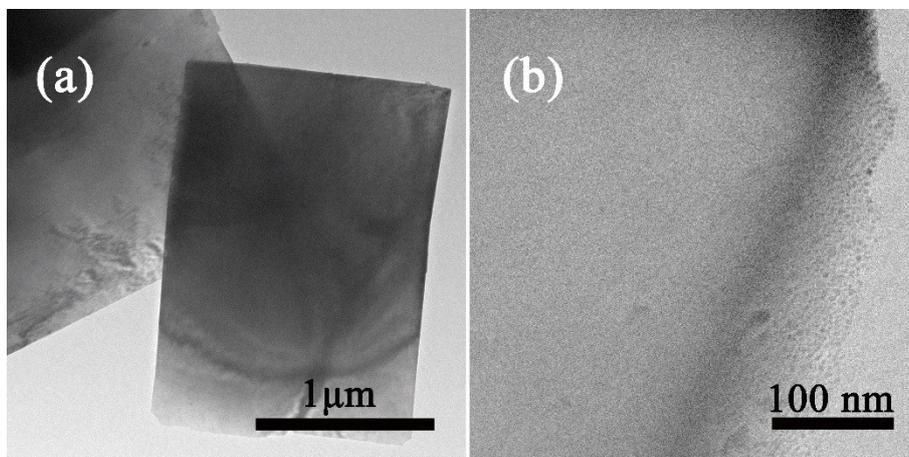


Fig.S3 Typical low- (a) and high-magnification (b) TEM images of $\text{CuC}_2\text{O}_4 \cdot x\text{H}_2\text{O}$ precursors.

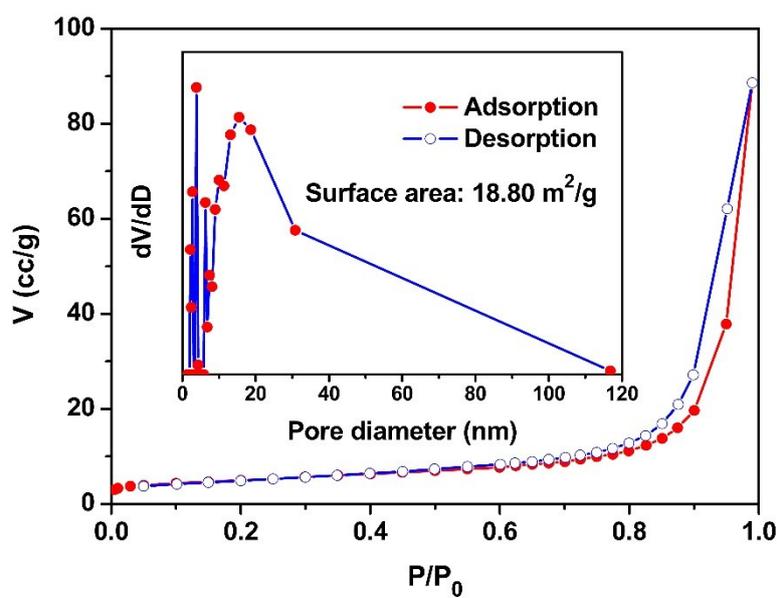


Fig.S4 Nitrogen adsorption-desorption isotherm of the CuO sheets (the inset displays the corresponding pore size distribution obtained from the desorption curve).

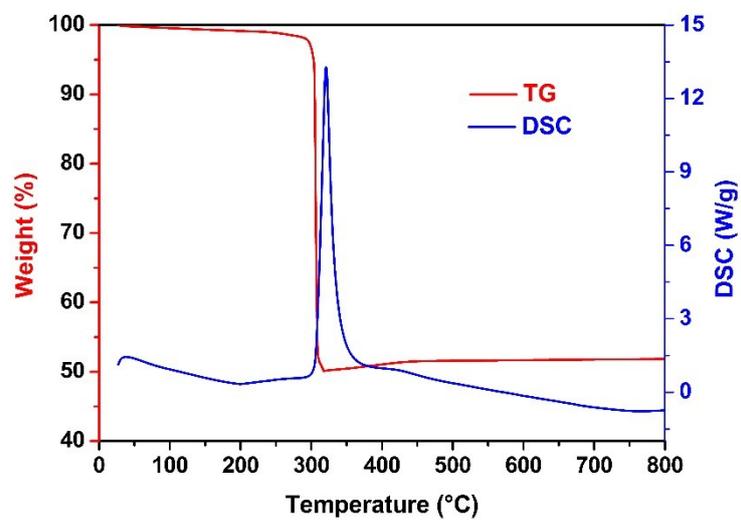


Fig.S5 TG and DSC curves of thermal conversion $\text{CuC}_2\text{O}_4 \cdot x\text{H}_2\text{O}$ precursor to CuO .

Table S1. Rietveld refinement parameters obtained from X-ray diffraction data of CuO using MAUD software package.

Lattice parameters

Symmetry: Monoclinic

Space group: C2/c

Cell parameters:

a: 4.6865 Å

b: 3.4312 Å

c: 5.1361 Å

α, β, γ (°): 90, 99.40, 90

Crystallite size (nm): 42.73

Atomic positions

Cu:

x: 0.0839

y: 0.2393

z: 0

O:

x: -0.1861

y: 0.4193

z: 0.2249

Reliability factors

Rwp (%): 6.60

Rp (%): 5.22
