

Supporting Information

Highly Conductive, Optically Transparent, Low Work-Function Hydrogen-doped Boron-doped ZnO Electrodes for Efficient ITO-Free Polymer Solar Cells

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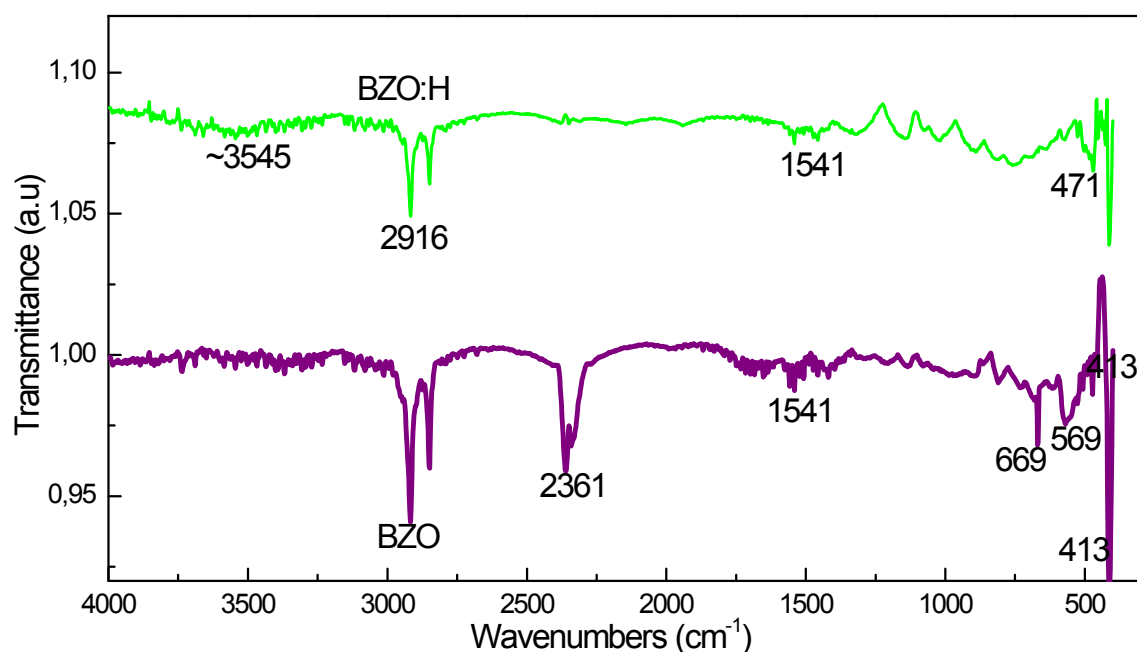
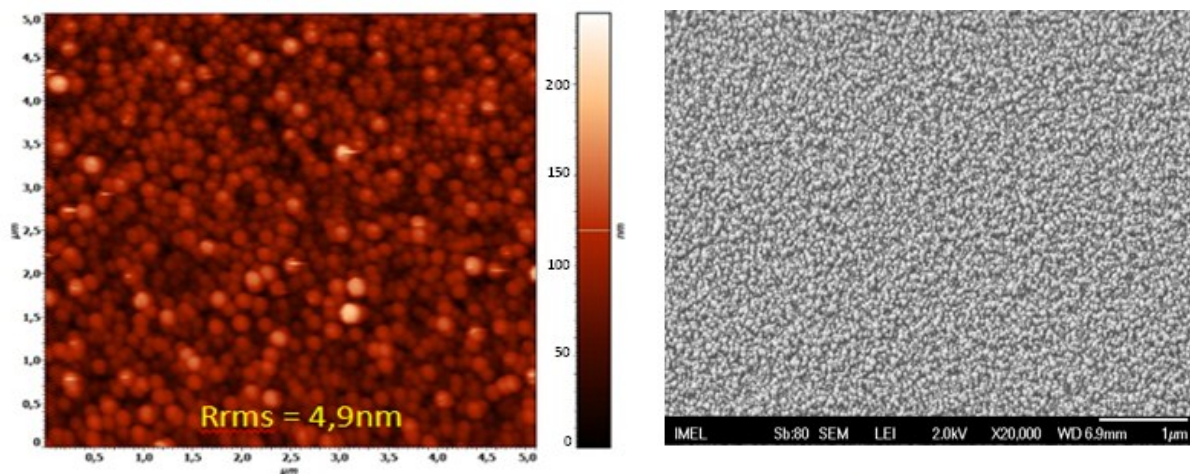


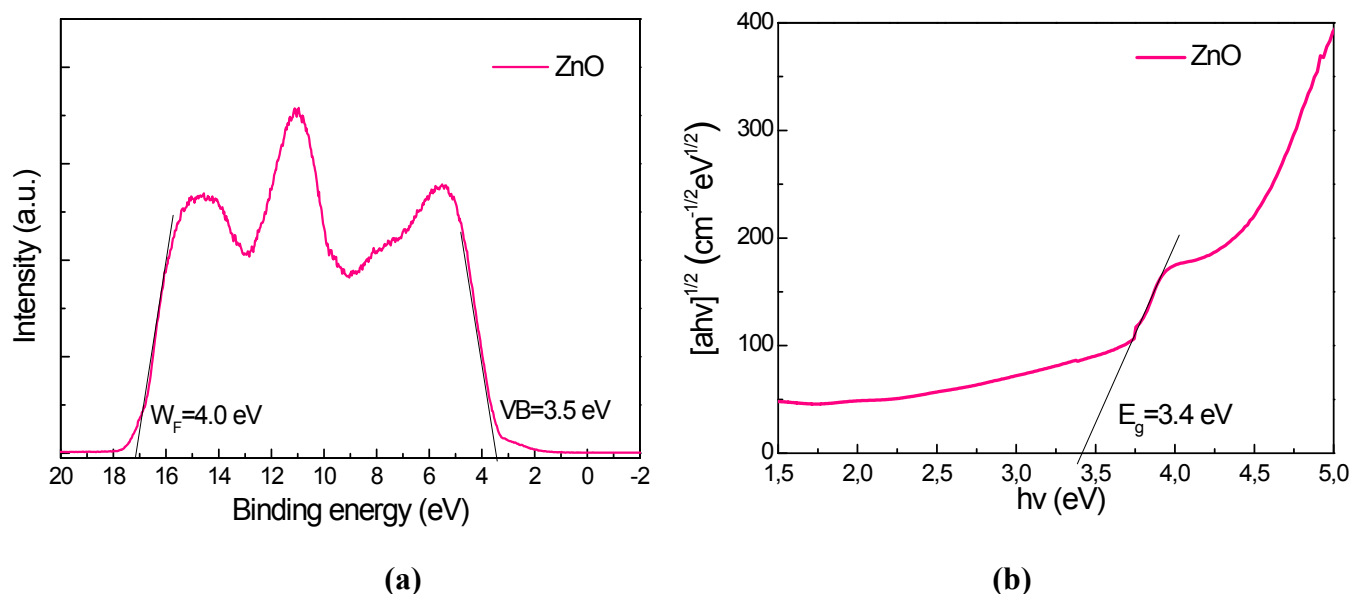
Figure S1. FTIR study of as-deposited and H-annealed BZO films (2% B content). The as-deposited BZO film presents a sharp peak at 413 cm⁻¹ (slightly blue shifted in comparison to the band at 408 cm⁻¹ reported in literature) attributed to the characteristic Zn–O stretching band of the hexagonal wurtzite ZnO structure.^{1,2} It also appears a broad band at 569 cm⁻¹ ascribed to other Zn–O stretching modes of wurtzite structure. The large band at 2361 cm⁻¹ is ascribed to CO₂ from the atmosphere adsorbed by the metallic cations. Upon H₂ annealing the BZO film presents a decrease in the intensity of the characteristic wurtzite Zn–O bond at 413 cm⁻¹ (which is also blue shifted) indicating decrease in the concentration of those structures. Also, a large red shift of the band attributed to other Zn–O stretching modes of wurtzite structure (569→ 471 cm⁻¹) is shown indicating considerable weakening of those bonds. A very significant result is also the appearance of a broad band at ~ 3545cm⁻¹ attributed to stretching of O–H bonds.³ Thus, upon H₂ thermal treatment the ZnO film is hydrogenated, in the sense that it contains -OH groups within its structure.



(a)

(b)

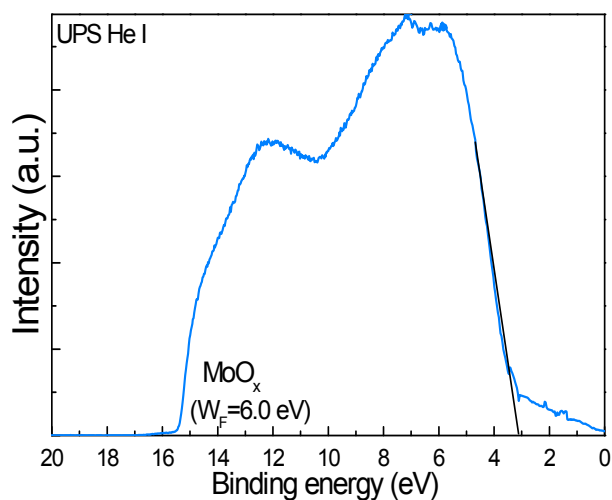
Figure S2 (a) Atomic Force Microscopy (AFM) 2-D and (b) SEM topographies of ZnO films (40 nm thick) derived via a sol-gel method deposited on BZO:H substrates. These films exhibit a short and dense rod-like morphology which is beneficial for the device operation, as it was previously demonstrated by our group.⁶



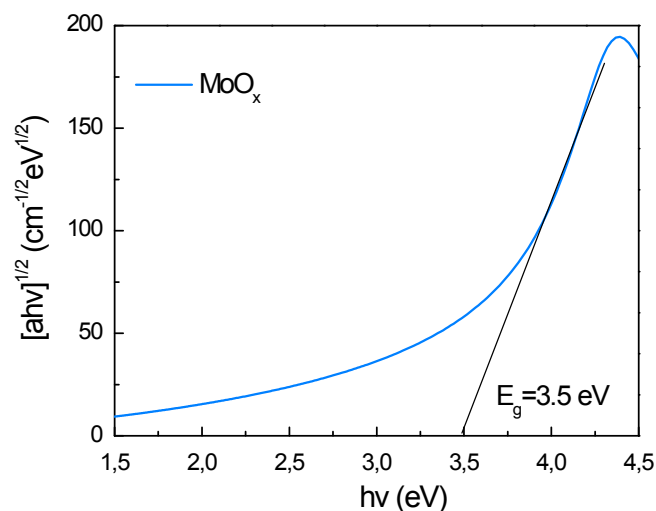
(a)

(b)

Figure S3. (a) UPS spectrum and (b) tauc plot (as derived from UV-vis absorption spectrum) of 40 nm sol-gel derived ZnO films. According to these spectra the ionization energy of ZnO is -7.5 eV, the energy bandgap is 3.4 eV and the conduction band minimum is placed at -4.1 eV.

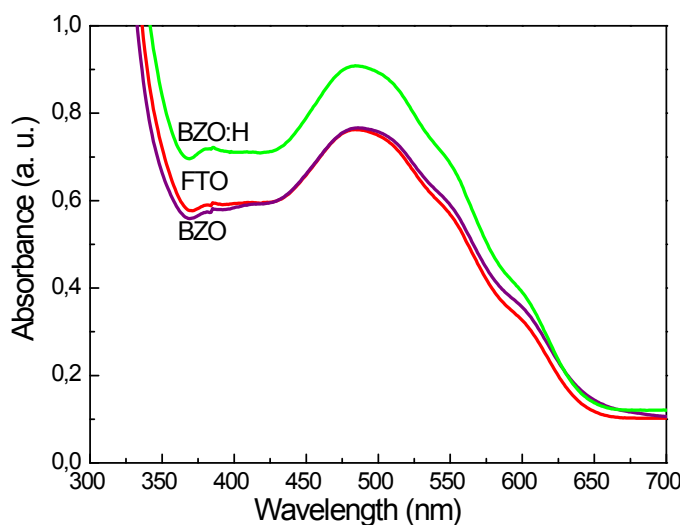


(a)

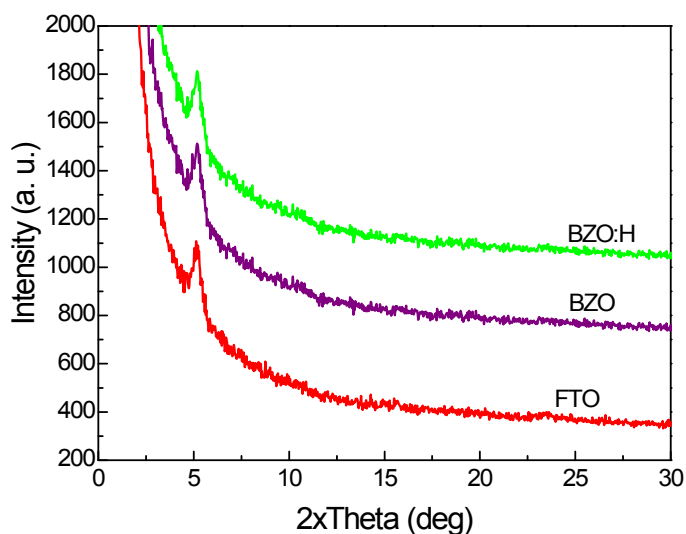


(b)

Figure S4 (a) UPS spectrum and (b) Tauc plot, as derived from absorption measurements, of a MoO_x film with a thickness of 20nm for the estimation of work function, ionization energy and optical energy band gap.

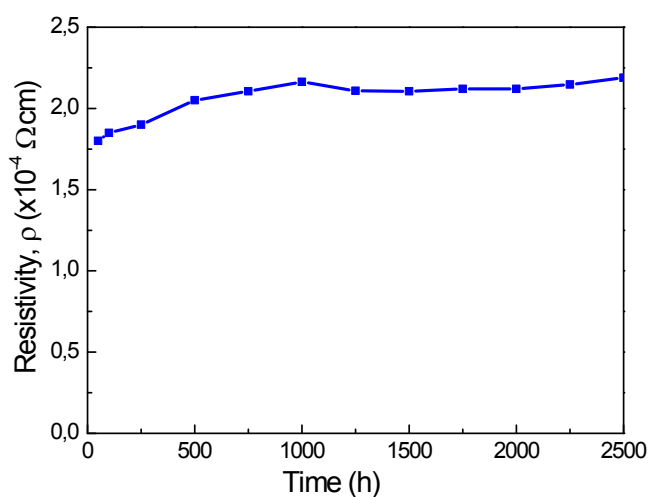


(a)

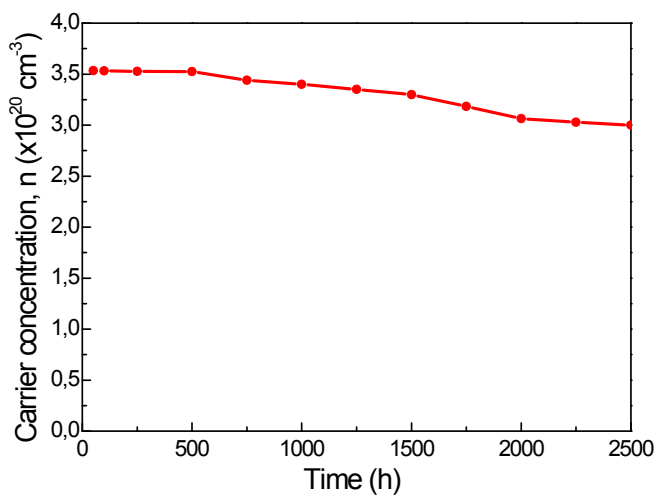


(b)

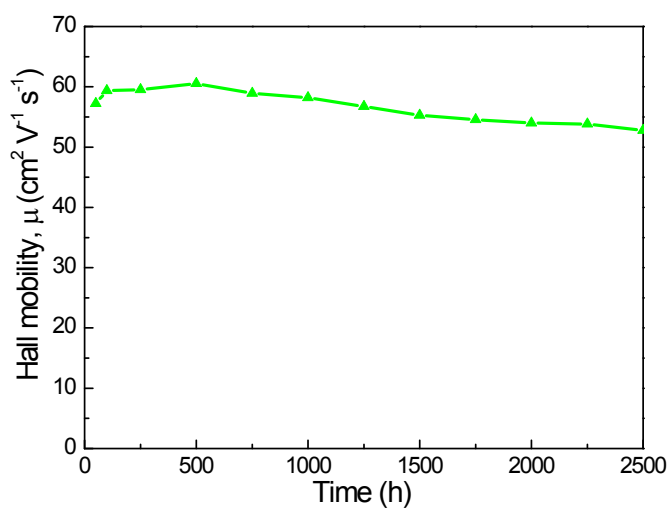
Figure S5. (a) UV/vis absorption spectra and (b) XRD diffraction patterns of 100 nm P3HT:PC₇₁BM films deposited on different TCO layers. Although the XRD patterns of P3HT:PC₇₁BM films deposited on different TCO substrates exhibited great similarities, the difference of the absorption spectrum of the film deposited on BZO:H substrate is evident. This may be attributed to the better optical transparency of the BZO:H substrate when compared to the BZO and FTO ones.



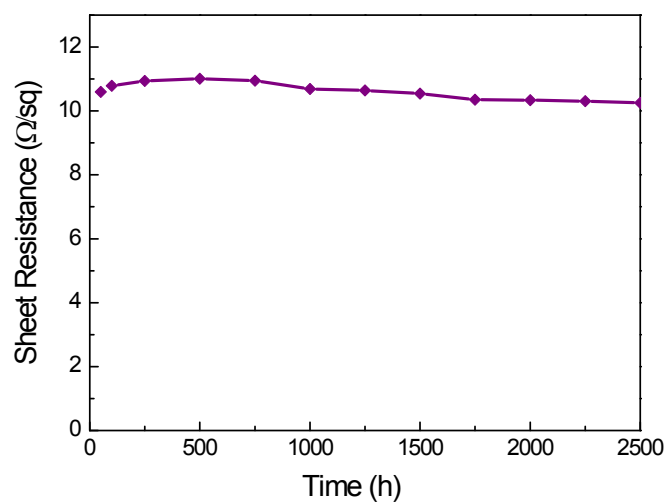
(a)



(b)



(c)



(d)

Figure S6. Stability study of BZO:H films. Variation with time of: (a) resistivity, (b) carrier concentration, (c) Hall mobility and (d) sheet resistance of BZO:H films.

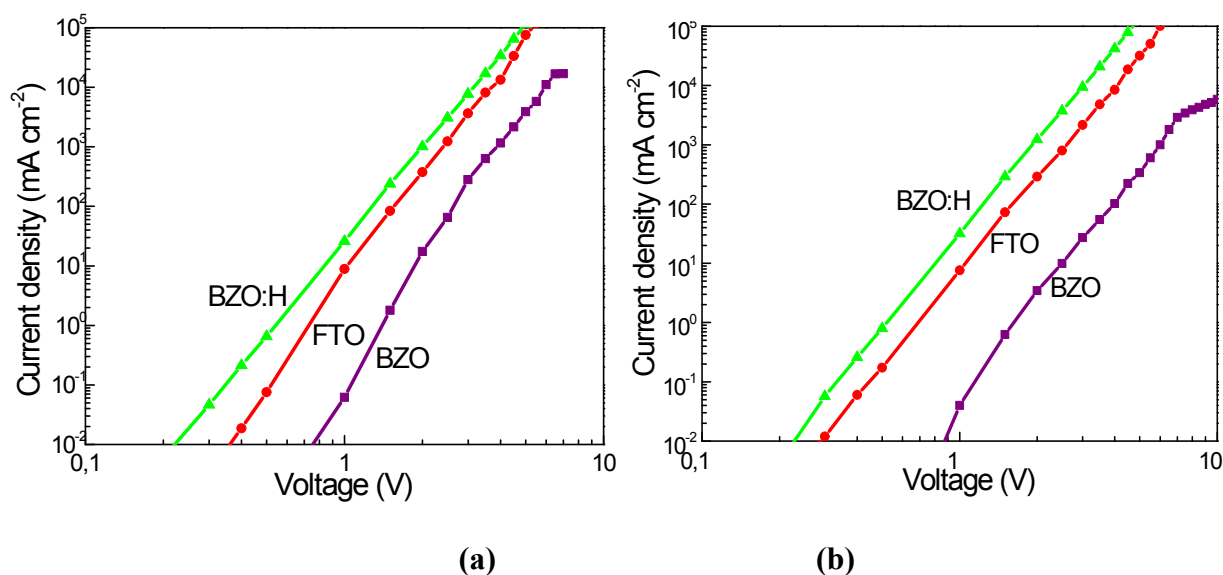


Figure S7 Electron only (a) PCDTBT:PC₇₁BM and (b) PTB7:PC₇₁BM based devices with the structure: BZO or BZO:H or FTO cathode electrode/ZnO/active layer/MoO_x/Al.

References

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