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## Supporting Information

## Amine Vapor Responsive Lanthanide Complex Entrapment: Control of Ligand-to-Metal and Metal-to-Metal Energy Transfer

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1. Spectroscopic and time resolved measurements



Figure S1. a) Schematic representation of ZL framework, b) Molecular structure of acac,



**Figure S2.** XRD pattern of a) nanozeolite (NZL) (black line) and b)  $Eu_4Tb_6(acac)_n@NZL$  (red line). c)  $Eu_4Tb_6(acac)_n@NZL$  upon exposure to  $Et_3N$ .



Figure S3. SEM images of (a) nanozeolite (NZL) and b) Eu<sub>4</sub>Tb<sub>6</sub>(acac)<sub>n</sub>@NZL.



**Figure S4.** Excitation spectra of  $Eu_4Tb_6(acac)_n@NZL$  monitored at 544 nm (black line) and monitored at 612 nm (red line), emission spectrum (blue line) of  $Eu_4Tb_6(acac)_n@NZL$  excitated at 310 nm.



**Figure S5.** Excitation spectra of  $Eu_4Tb_6(acac)_n@NZL$  before (black line) and after exposure to n-BuNH<sub>2</sub> (red line), Benzylamine (blue line), a) monitored at 612 nm, b) monitored at 544 nm; c) Emission spectra of  $Eu_4Tb_6(acac)_n@NZL$  before (black line) and after exposure to n-BuNH<sub>2</sub> (red line), Benzylamine (blue line) excited at 310 nm.



**Figure S6.** Excitation spectra of  $Eu_4Tb_6(acac)_n@NZL$  before (black line) and after exposure to N-methylaniline (orange line) a) monitored at 612 nm, b) monitored at 544 nm; c) Emission spectra of  $Eu_4Tb_6(acac)_n@NZL$  before (black line) and after exposure to N-methylaniline (orange line) excited at 310 nm.



**Figure S7.** FTIR spectra of (a)  $Eu_4Tb_6(acac)_n@NZL$  (black line), (b)  $Eu_4Tb_6(acac)_n@NZL$  upon exposure to  $Et_3N$  (red line).



**Figure S8.** The compared decay curves of  $Tb^{3+}(acac)_n@NZL$  (black line) and  $Eu_4Tb_6(acac)_n@NZL$  (red line) using an excitation of 310 nm and monitored at 544 nm upon exposure to a)Et<sub>3</sub>N, b) n-BuNH<sub>2</sub>, c) Benzylamine, d) t-BuNH<sub>2</sub> and e) En vapor.



**Figure S9.** The compared decay curves of  $Eu^{3+}(acac)_n @NZL$  (black line) and  $Eu_4Tb_6(acac)_n @NZL$  (red line) using an excitation of 310 nm and monitored at 612 nm upon exposure to a)Et<sub>3</sub>N, b) n-BuNH<sub>2</sub>, c) Benzylamine, d) t-BuNH<sub>2</sub> and e) En vapor.



**Figure S10.** Digital photographs of  $Eu_4Tb_6(acac)_n@NZL$  upon contact with the equilibrated vapor of  $Et_3N$  for varied periods of time under near UV irradiation at 302 nm.



**Figure S11.** Emission spectrum of  $Eu_4Tb_6(acac)_n@NZL$  upon contact with the equilibrated vapor of  $Et_3N$  for 30 s; Inset: Digital photographsof  $Eu_4Tb_6(acac)_n@NZL$  upon contact with the equilibrated vapor of  $Et_3N$  for 30 s under near UV irradiation at 302 nm.

Table S1	The ec	uilibrated	vapor	pressure	of	amine	solve	ents
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	Et <sub>3</sub> N	t-BuNH <sub>2</sub>	n-BuNH <sub>2</sub>	En	Benzyla mine	N-methy laniline	aniline
Equilibrated vapor pressure(KPa)	9.06	49.54	12.29	1.65	0.09	0.06	0.09