

# Novel structurally tuned DAMN receptor for “*in-situ*” diagnosis of bicarbonate in environmental waters

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## General information

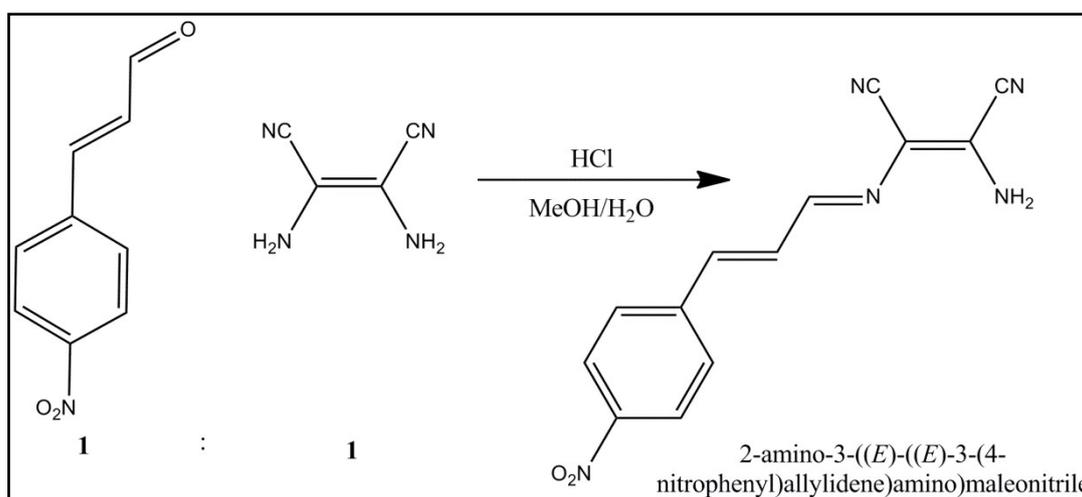
All solvents were purchased from commercial sources and used without further purification. Starting materials, 4-nitrocinamaldehyde, 98%, (predominantly *trans*) and 1, 2-diaminomaleonitrile were purchased from Sigma-Aldrich. Mass spectra were recorded on a Bruker HR-MS spectrometer using CH<sub>3</sub>CN as solvent. <sup>13</sup>CMR and <sup>1</sup>H nmr spectra were recorded using a Bruker instrument operating at 500MHz. Single-Crystal XRD data was collected on a Bruker APEX II diffractometer (Mo KR,  $\lambda = 0.71069 \text{ \AA}$ ). Absorption measurements were carried out using an Agilent spectrophotometer (Product no: G9821A, Serial no: MY1321007, Cary Win Uv software. For absorption titration experiments, 3ml volumes of receptor in DMSO were used in a quartz cuvette (Hellma) at  $25 \pm 3 \text{ }^\circ\text{C}$ , followed by addition of stock solutions of the appropriate anions in the form of Sodium or potassium salts dissolved in Milli-Q water (resistivity=18.2 M $\Omega$ cm at 25  $^\circ\text{C}$ ). Graph plotting and liner-curve fitting was done in Origin Pro 8 and Excel 2007. The synthesis of receptor molecule was carried under ambient conditions (298 K).

Single-crystals were obtained through solvent evaporation method. Receptor was dissolved in minimal amount of DMSO and sample vial was left under ambient conditions via perforated cap. After 3-4 weeks, needle shaped crystals obtained were suitable for X-ray diffraction, were collected by filtration. Single-crystal analysis revealed that receptor molecule crystallizes in the P21/c space group.

## Synthesis and characterization of receptor

Receptor was synthesised by the below mentioned procedure:

It involves dropwise addition of equimolar methanolic solution of diaminomaleonitrile (50 mg, 0.46 mM) to a stirring solution of 4-nitrocinnamaldehyde (81.94 mg) in H<sub>2</sub>O, containing 1-2 drops of concentrated HCl. A yellow-colored solid got precipitated immediately. Later is filtered, washed several times with absolute EtOH and ultimately dried under vacuum and characterised by standard spectroscopic techniques, like NMR and HR-MS and Single-crystal XRD.



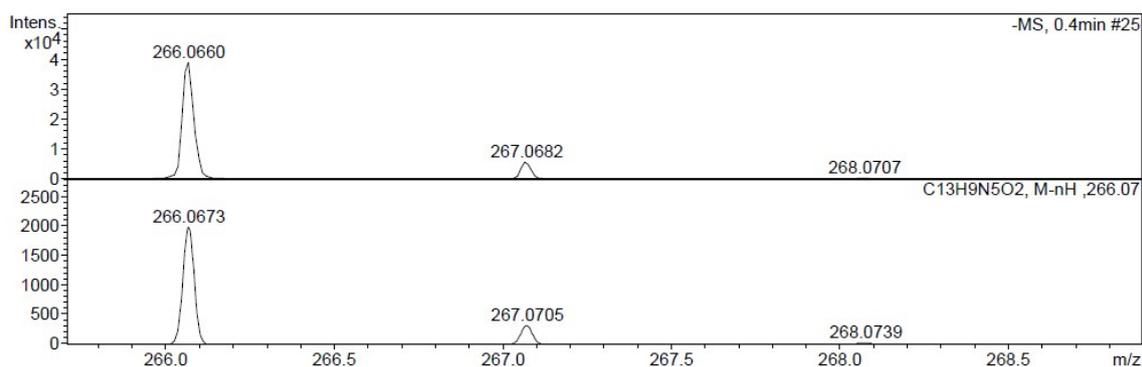
**Fig. 1** Synthetic Scheme of receptor.

Characterisation of receptor (NMR and HR-MS):

(Yellow colour, 95% yield); (Ratio of *trans* and *cis* isomers: 6:1); <sup>1</sup>H NMR- (500MHz; *d*<sub>6</sub>-DMSO): δ 8.26, 8.22 (two diastereomers, d, *J* = 9, 9 Hz, 2H), δ 8.12, 8.06 (two diastereomers, d, *J* = 9, 4 Hz, 1H), 7.98 (s, 2H), δ 7.85, 7.93 (two diastereomers, d, *J* = 3.5, 9 Hz, 2H), δ 7.93, 7.57 (two diastereomers, d, *J* = 16, 16 Hz, 1H), δ 7.29, 7.20 (two diastereomers, dd, *J* = 8.8, 16 Hz, dd, *J* = 8.8, 16 Hz, 1H): <sup>13</sup>C NMR- (500 MHz; *d*<sub>6</sub>-DMSO): δ 156.40, 147.79, 142.44, 141.27, 131.59, 128.93, 127.68, 124.64, 114.73, 114.07, 103.73. MS (HR-MS, negative mode) found 267.064 for C<sub>13</sub>H<sub>9</sub>N<sub>5</sub>O<sub>2</sub>

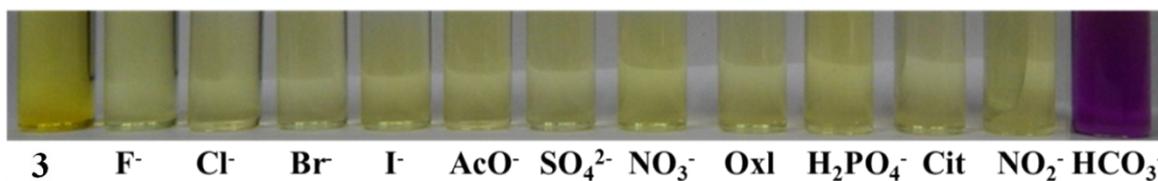


**Fig. 3**  $^{13}\text{C}$  nmr spectra of receptor.



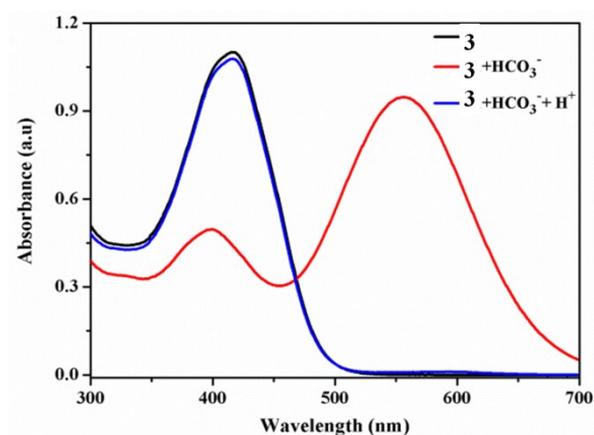
**Fig. 4** HR-MS of receptor.

### Selectivity of receptor and Naked-eye changes



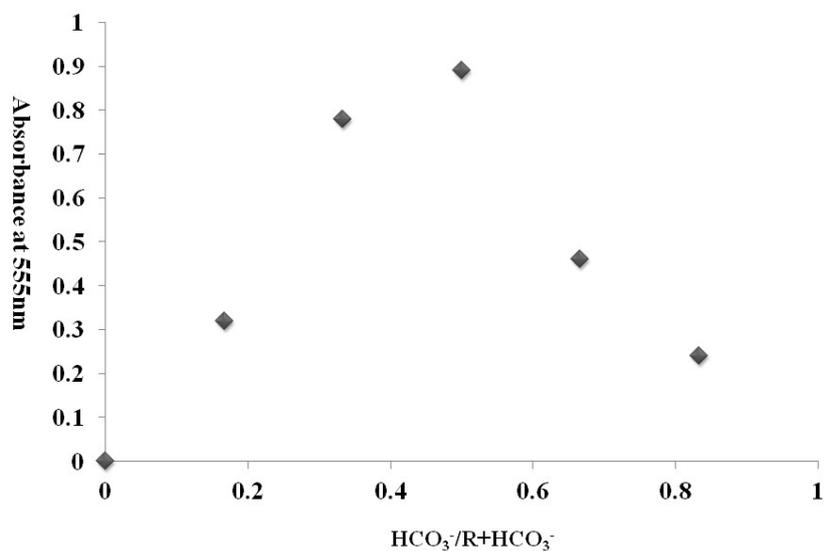
**Fig. 5** Diagram representing exclusive recognition response of receptor **3** ( $16\ \mu\text{M}$ ) with various anions. Observations were made in presence of 10 equivalents of  $\text{HCO}_3^-$ , while rest have been added with more than 200 equivalents.

### Reversal of receptor carbonate interaction



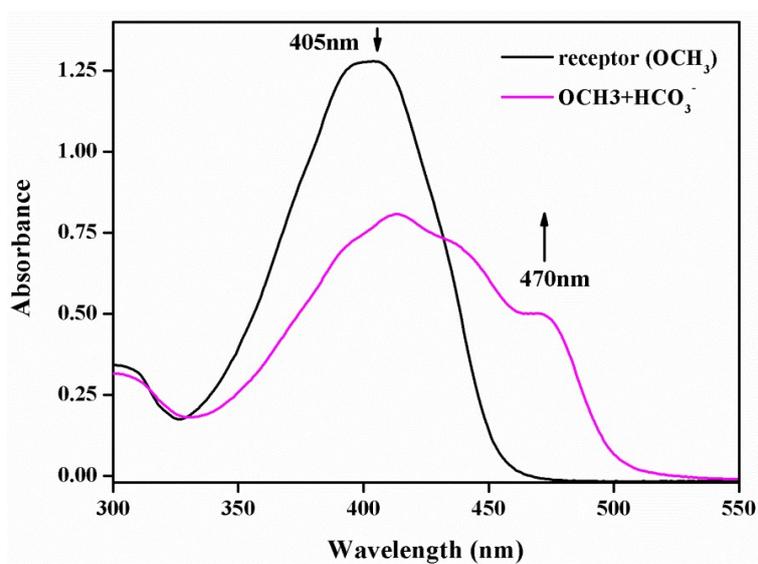
**Fig. 6** Recyclable nature of receptor (**3**), driven by  $H^+$  sources. Observations were made with 25 $\mu$ M of **3** with 60 $\mu$ L of 10 mM  $HCO_3^-$ . The reversibility was obtained upon addition of 20  $\mu$ L of 1 mM  $H_2SO_4$ .

**Jobs Plot**

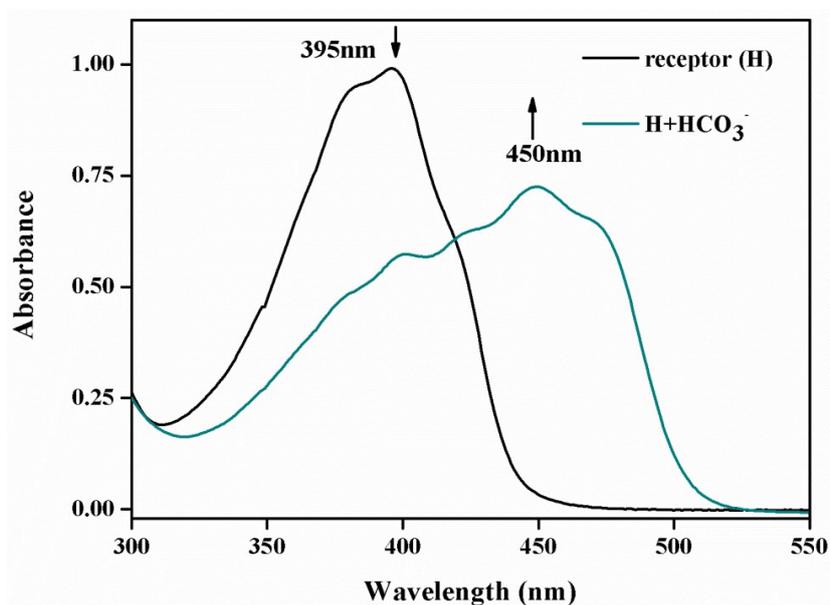


**Fig. 7** Jobs plot for determination of stoichiometry of receptor-bicarbonate interaction (R = receptor **3**).

**Interaction of various receptor derivatives with bicarbonate**



**Fig.8** Absorption changes of -OCH<sub>3</sub> (1) receptor with bicarbonate anion.

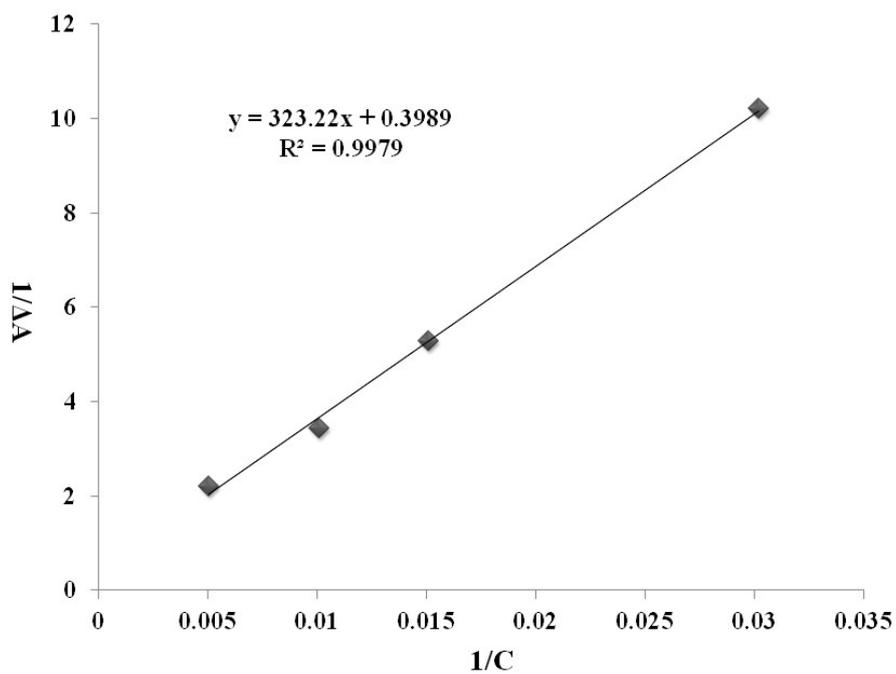


**Fig. 9** Absorption changes of -H (2) receptor with bicarbonate anion.

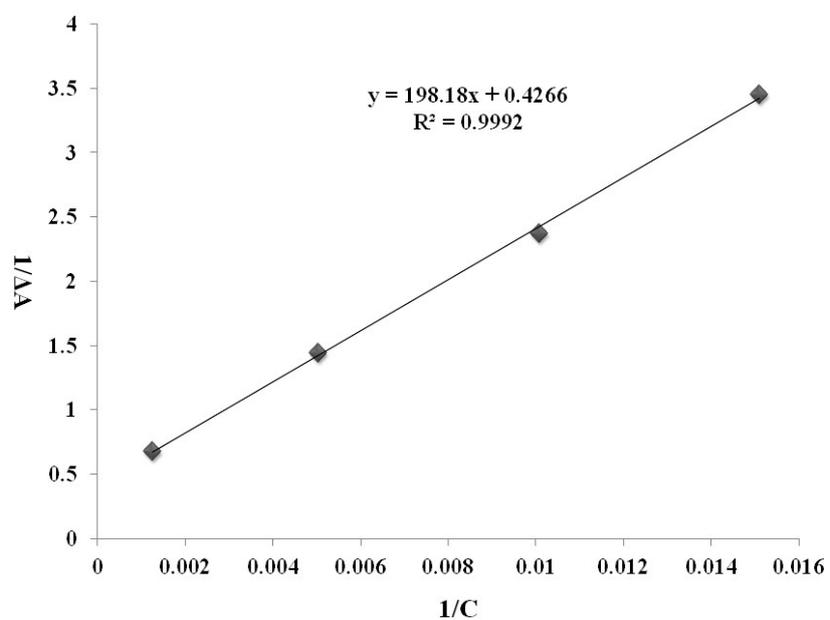
### Binding constant calculations

For the 1:1 interaction, binding constants were obtained with the help of Benesi-Hilderbrand Plots below. Here variation of reciprocal of change in emerging absorption signal after bicarbonate interaction ( $1/\Delta A$ ) was fitted with respect to reciprocal of HCO<sub>3</sub><sup>-</sup> concentration in  $\mu\text{M}$  ( $1/C$ ). Binding constant (K) can be calculated by the below relation:

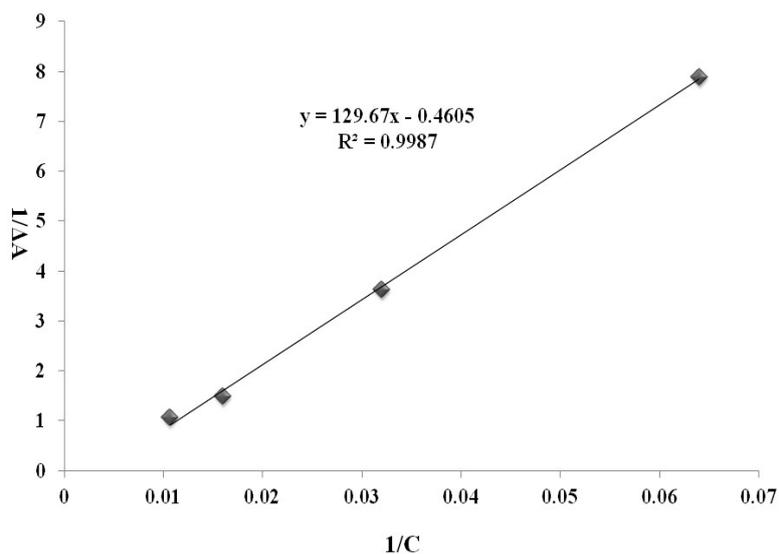
$$K = (\text{Intercept/Slope}) * 1000000 \text{ M}^{-1}$$



**Fig. 10** Inverse of absorption changes of  $-\text{OCH}_3$  (1,  $28\mu\text{M}$ ) receptor with inverse of  $\text{HCO}_3^-$  concentrations.



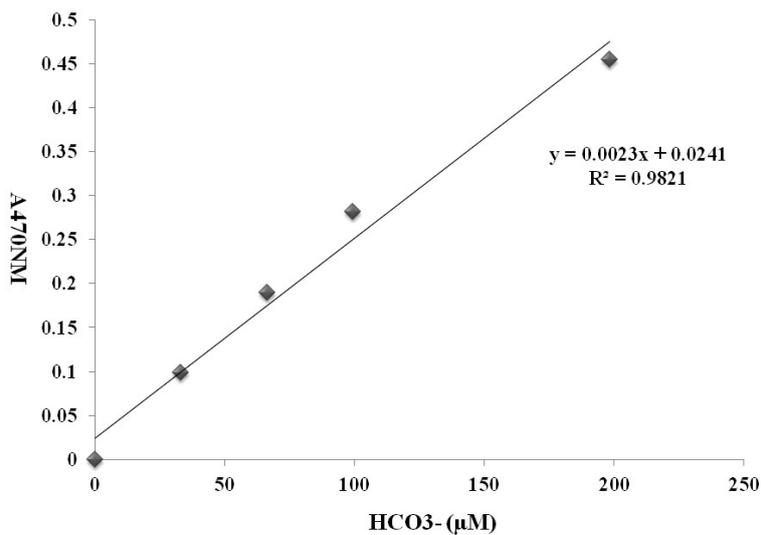
**Fig. 11** Inverse of absorption changes of  $-\text{H}$  (2,  $20\mu\text{M}$ ) receptor with inverse of  $\text{HCO}_3^-$  concentrations.



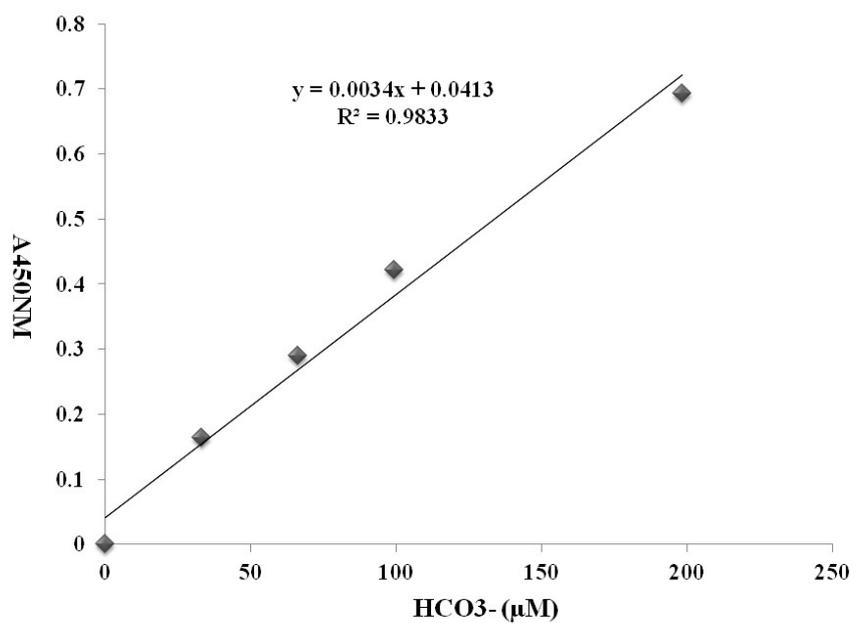
**Fig. 12** Inverse of absorption changes of  $-\text{NO}_2$  receptor (**3**,  $30 \mu\text{M}$ ) with inverse of  $\text{HCO}_3^-$  concentrations.

### Sensitivity of various receptors

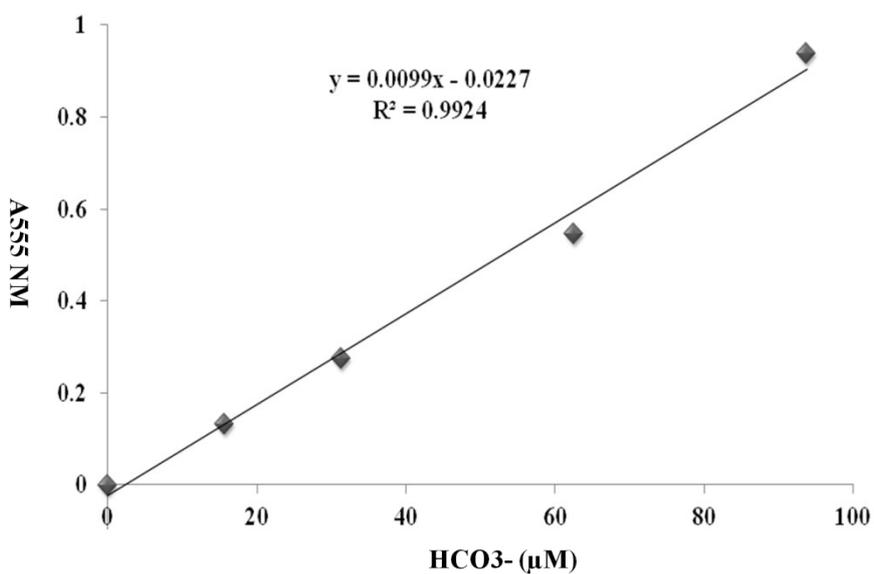
Sensitivity of receptor is presented by the slope of below curve ( $\theta$ ), presenting variation of absorbance of emerging signal with bicarbonate anion concentrations.



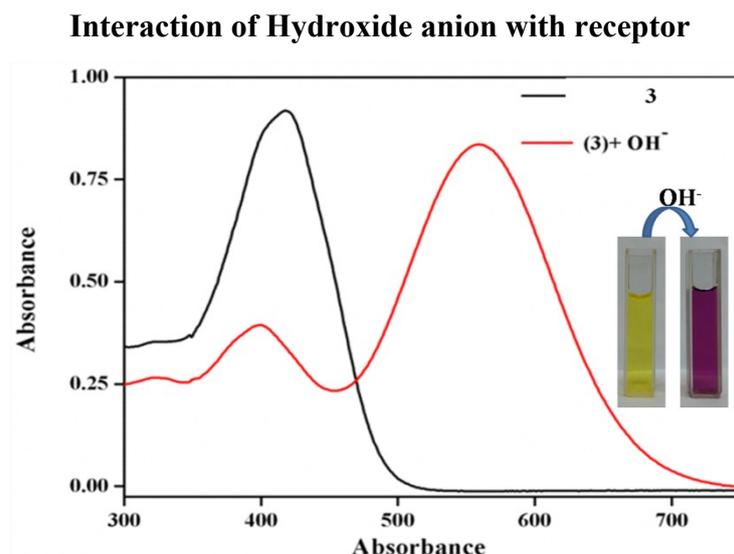
**Fig.13** Variation of absorbance at 470nm of  $-\text{OCH}_3$  receptor (**1**,  $28 \mu\text{M}$ ) with bicarbonate.



**Fig14** Variation of absorbance at 450 nm of  $-H$  receptor (**2**, 20  $\mu M$ ) with bicarbonate anion concentrations.

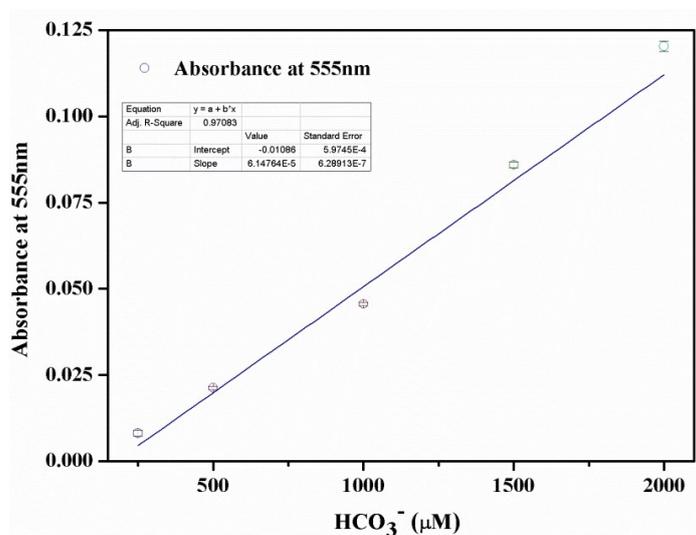


**Fig. 15** Variation of absorbance at 555nm of  $-NO_2$  receptor (**3**, 30  $\mu M$ ) with bicarbonate anion concentrations.



**Fig. 16** Interaction of hydroxide anion ( $\text{OH}^-$ ) with  $-\text{NO}_2$  receptor (**3**).

### Limit of Detection



**Fig. 17** Calibration curve for determination of bicarbonate anion in water samples. Here receptor concentration used is  $170 \mu\text{M}$ . All the experiments were carried out at  $298 \text{ K}$ .

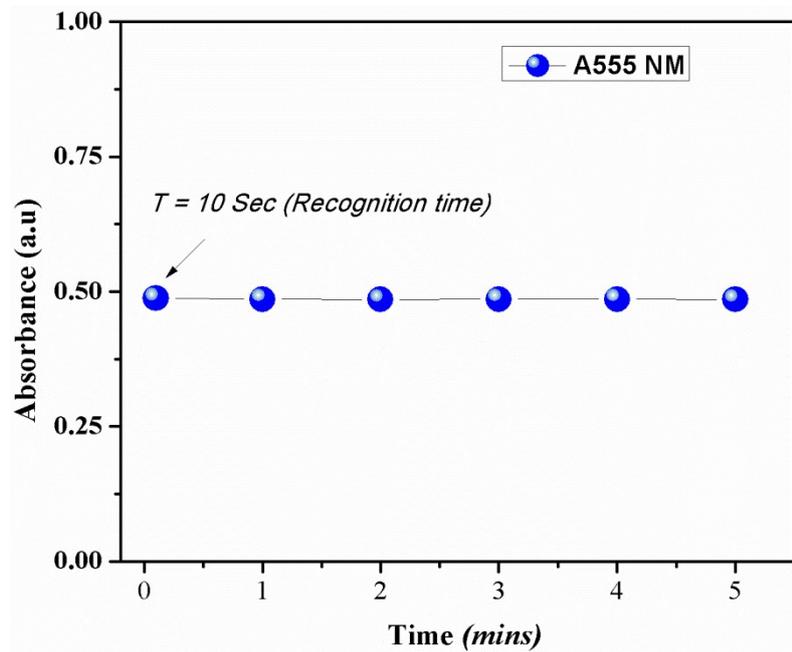
Limit of detection (LOD) =  $3\sigma/m$

Here “ $\sigma$ ” refers to the standard deviation of blank measurements ( $n=3$ ).

“ $m$ ” refers to the slope of the calibration curve in the above figure.

Hence LOD =  $(3 \times 0.000834 / 0.00006147) = 40.71 \mu\text{M}$

### Response time of receptor



**Fig. 18** Sensor response of receptor (3) in presence of  $\text{HCO}_3^-$ . The studies were conducted at room temperature, with receptor concentration of  $(20 \mu\text{M})$  with  $80\text{-}100 \mu\text{M}$  of  $\text{HCO}_3^-$ .