## Supplementary Information

## Fluorescence array-based sensing of nitroaromatics using conjugated polyelectrolytes

Jiatao Wu, ${ }^{\text {a,b }}$ Chunyan Tan, ${ }^{* a, b}$ Zhifang Chen, ${ }^{\text {a,b }}$ Yu Zong Chen, ${ }^{\mathrm{c}}$ Ying Tan, ${ }^{* a, \mathrm{~b}}$ and Yuyang Jiang ${ }^{\mathrm{b}, \mathrm{d}}$<br>${ }^{a}$ Department of Chemistry, Tsinghua University, Beijing 100084 P. R. China<br>${ }^{b}$ The Ministry-Province Jointly Constructed Base for State Key Lab- Shenzhen Key Lab of Chemical Biology, The Graduate School at Shenzhen, Tsinghua University, Shenzhen 518055 P. R. China<br>${ }^{c}$ Shenzhen Kivita Innovative Drug Discovery Institute, Shenzhen, 518055 China<br>${ }^{d}$ Department of Pharmacology and Pharmaceutical Sciences, School of Medicine, Tsinghua University, Beijing 100084 P. R. China

## Table of contents

S1 Experimental details

S4 Conjugated polyelectrolytes synthesis

S7 Training matrix

## 1. Experimental details

### 1.1. Materials

Bis(triphenylphosphine) palladium chloride $\left(\mathrm{PdCl}_{2}\left(\mathrm{PPh}_{3}\right)_{2}\right)$, tetrakis(triphenylphosphine) palladium $\left(\operatorname{Pd}\left(\mathrm{PPh}_{3}\right)_{4}\right)$, and cuprous iodide (CuI) were purchased from Aladdin Chemical Co. (Shanghai, China) and used as received. Boron tribromide $\left(\mathrm{BBr}_{3}\right)$, tetrabutylammonium fluoride (TBAF), 1,6-dibromohexane, and 1,4-dimethoxybenzene were purchased from JK Chemical (Beijing, China) and used without further purification. 2,5-dihydroxybenzoic acid (DHB) and silver trifluoroacetate were purchased from Sigma-Aldrich (Shanghai, China) and used as received. 4nitrobenzoic acid (C-1), 3-nitrobenzoic acid (C-2), 4-nitrophenol (C-3), 2-methyl-4,6-dinitrophenol (C4), 2,4-dinitrophenol (C-5), picric acid (C-6), 4-nitrobenzenesulfonic acid (C-7), 2,4dinitrobenzenesulfonic acid (C-8), 2,4,6-trinitrobenzensulfonic acid (C-9) were purchased from ENERGY CHEMICAL (Shanghai, China) and used without further purification. The water used in all experiments was prepared in a SG water purification system and displayed a resistivity of $\geq 18.2$ $\mathrm{M} \Omega \cdot \mathrm{cm}^{-1}$.

### 1.2. General methods

${ }^{1} \mathrm{H}$ and ${ }^{1} \mathrm{C}$ NMR spectra were recorded on a Bruker 400 MHz spectrometer, and chemical shifts were reported in parts per million using tetramethylsilane (TMS) as internal reference. The MALDI experiments were conducted on a 4700 proteomics analyzer MALDI-TOF/TOF-MS (Applied Biosystems, Framingham, MA, USA), equipped with a 337 nm nitrogen laser. Fluorescence spectra were recorded on a SPEX Fluorolog-3-TCSPC spectrometer with 1 cm path length cuvette, while absorption spectra were recorded on a Beckman DU 800 spectrophotometer. Nitroaromatics (NACs) arrays were measured on 96-well plates ( $300 \mu \mathrm{~L}$ Corning) using Tecan M1000 Pro plate reader.

### 1.3. Mass spectrometry

For all of the MALDI experiments, DHB was used as the matrix (M) and silver trifluoroacetate was used as cationization reagent (CR). All MALDI mass spectra were collected using linear ion mode.

The polymer samples were prepared at a concentration of $5 \mathrm{mg} / \mathrm{mL}$, matrix solutions were prepared at a concentration of $10 \mathrm{mg} / \mathrm{mL}$ and cationization reagent solutions were prepared at a concentration of 0.1 $\mathrm{mol} / \mathrm{L}$. These solutions were mixed in the given volume mixing ratio $\left(\mathrm{V}_{(\text {polymer })}: \mathrm{V}_{(\mathrm{M})}: \mathrm{V}_{(\mathrm{CR})}=5: 5: 1\right) .1 \mu \mathrm{~L}$ of the solution was hand-spotted on a stainless steel target and allowed to dry by air. All spectra data were taken from signal averaging of 300 laser shots and further processed using Data Explorer 4.5 (Applied Biosystems).

### 1.4. Sensor array detection

Conjugated polyelectrolytes (CPEs) were used at a concentration of $10-100 \mu \mathrm{M}$ in ultrapure water. Two different experiments were performed, the first in which the concentration of NACs was known, 1 $\mu \mathrm{M}$ for each nitroaromatic, and second, the solutions were normalized such that $\mathrm{A}_{300}=0.05$ and the concentrations varied. The samples were measured in 96 well polypropylene plates with an excitation at 420 nm . Six wells of a standard 96 well microtiterplate were filled with a stock solutions of CPEs, which the ultraviolet absorption of CPEs were fixed at $\mathrm{A}_{420}=0.1$. The emission spectrum was measured at 500 nm with the first well loaded with only CPEs in water for a measurement of $\mathrm{I}_{0}$. The percent quenching was calculated with the equation $\left.\left(\mathrm{I}_{0}-\mathrm{I}\right) / \mathrm{I}_{0} \times 100\right) \%$ quenching, where $\mathrm{I}_{0}$ is the fluorescence intensity without added NACs. Fluorescence data collection was carried out in six replicates and the data were subjected to linear discriminant analysis (LDA) using R software. Confidence ellipse was fitted by MATLAB 7.0 with $95 \%$ confidence interval.

### 1.5. Unknown samples discrimination

Unknown samples were identified according to their canonical scores or their placement in this 2D space. The array response to each unknown $\left(\mathrm{A}_{300}=0.05\right)$ was compared with the classification data, and unknowns were identified according to their placement in this 2-D space figure with a $95 \%$ confidence interval.

### 1.6. Data analysis method

Fluorescence intensity of each CPE-NAC solution was collected through plate reader. A training
matrix (six CPEs $\times$ nine NACs $\times$ six replicates) was generated using $\mathrm{I}_{0} / \mathrm{I}$ values for further computational analysis and were processed through LDA using R software (i3863.0.3). In the analysis, all of the variables were used in the complete mode, and the tolerance was set to 0.001 . The raw fluorescence intensity patterns were transformed into canonical score patterns in which the within-class variance to between-class variance was minimized according to preassigned grouping. Canonical scores were calculated by LDA using R software for identification of the nine NACs.

## 2. CPEs synthesis


monomer 1

monomer 2
i $\mathrm{K}_{2} \mathrm{CO}_{3}$, Acetone, $70^{\circ} \mathrm{C}$; ii $\mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{OH}$, Acetone, $\mathrm{H}_{2} \mathrm{O}, \mathrm{Et}_{3} \mathrm{~N}$, reflux;
iii 1)Trimethylsilylacetylene, $\mathrm{Pd}\left(\mathrm{PPh}_{3}\right)_{4} \mathrm{Cl}_{2}$, $\mathrm{Cul}, \mathrm{THF} / \mathrm{Et}_{3} \mathrm{~N}, 50^{\circ} \mathrm{C}$; 2) TBAF,THF,RT

## Scheme S1 Synthesis of monomer 1 and monomer 2

1,4-bis((6-bromohexyl)oxy)-2,5-diiodobenzene. A solution of $\operatorname{Br}\left(\mathrm{CH}_{2}\right)_{6} \mathrm{Br}(1.464 \mathrm{~g}, 6.0 \mathrm{mmol})$ and potassium carbonate $(1.38 \mathrm{~g}, 10.0 \mathrm{mmol})$ in 150 mL acetone fitted with a condenser were degassed with argon for 10 minutes. Then 362 mg of 2,5-diiodohydroquinone ( 1.0 mmol ) dissolve in 50 mL acetone was added slowly under argon. Then the reaction mixture was stirred at reflux for 20 hr . After cooling to room temperature, remove potassium carbonate through filtration. The reaction solvent was removed in vacuo. Then the remainders were partitioned between water and $\mathrm{CH}_{2} \mathrm{Cl}_{2}$. The organic solution was dried by adding anhydrous sodium sulfate, and then the crude product was purified by purified by
column chromatograph with a mixture of $\mathrm{CH}_{2} \mathrm{Cl}_{2} /$ hexane $(\mathrm{v} / \mathrm{v}=25 / 1)$ and white solid was collected and dried(yield: $344 \mathrm{mg} 50 \%) .{ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}, \delta_{\mathrm{ppm}}\right): 7.17(\mathrm{~s}, 2 \mathrm{H}), 3.94(\mathrm{t}, 4 \mathrm{H}), 3.43(\mathrm{t}, 4 \mathrm{H}), 1.91$ $(\mathrm{m}, 4 \mathrm{H}), 1.81(\mathrm{~m}, 12 \mathrm{H}), 1.53(\mathrm{t}, 8 \mathrm{H})$.

1,4-bis((6-bromohexyl)oxy)-2,5-diethynylbenzene. 4-bis((6-bromohexyl)oxy)-2,5-diiodobenzene ( $688 \mathrm{mg}, 1 \mathrm{mmol}$ ) was dissolved in 25 ml of dry THF/Et $\mathrm{E}_{3} \mathrm{~N}(3 / 1, \mathrm{v} / \mathrm{v}$ ) in a Schlenk flask and degassed with argon for 15 minutes. Then 21 mg of $\mathrm{Pd}\left(\mathrm{PPh}_{3}\right)_{2} \mathrm{Cl}_{2}(30 \mu \mathrm{~mol})$ and 12 mg of $\mathrm{CuI}(63 \mu \mathrm{~mol})$ were added, followed with the addition of 0.7 ml of trimethylsilylacetylene ( 5 mmol ). The reaction was stirred at $50^{\circ} \mathrm{C}$ for 22 hr . After filtration through a bed of celite, the solvent was removed in vacuo. The crude product was used for next step without further purification. The crude product obtained as was dissolved in 10 ml of THF. Tetrabutylammonium fluoride ( 5 ml of a 1 M solution in THF) was added and the resulting mixture was stirred at room temperature for 2 hr . Then the solution was diluted with 20 ml of ethyl ether, poured into a seperatory funnel and washed with water ( $30 \mathrm{ml} \times 1$ ). The organic layer was collected and the solvent was removed in vacuo. The crude product was purified by column chromatograph. A slight yellow solid was obtained (yield: $242 \mathrm{mg}, 50 \%$ ). ${ }^{1} \mathrm{H}$ NMR ( $\mathrm{CDCl}_{3}, \delta_{\text {ppm }}$ ): $6.95(\mathrm{~s}, 2 \mathrm{H}), 3.68(\mathrm{t}, 4 \mathrm{H}), 3.42(\mathrm{t}, 4 \mathrm{H}), 3.33(\mathrm{~s}, 2 \mathrm{H}), 1.89(\mathrm{~m}, 4 \mathrm{H}), 1.81(\mathrm{~m}, 4 \mathrm{H}), 1.52(\mathrm{~m}, 8 \mathrm{H})$.

Monomer 1. 1,4-bis((6-bromohexyl)oxy)-2,5-diiodobenzene ( $688 \mathrm{mg}, 1 \mathrm{mmol}$ ) was suspended in $40 \%$ trimethylamine in water $(20 \mathrm{~mL})$, ethanol $(30 \mathrm{~mL})$ and acetone ( 30 mL ). The suspension was heated to $130^{\circ} \mathrm{C}$ for 24 hours. The solution was concentrated to yield a solid, which was recrystallized from ethanol/ether (yield: $700 \mathrm{mg}, 96.4 \%) .{ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CH}_{3} \mathrm{DO}, \delta_{\mathrm{ppm}}\right): 7.01(\mathrm{~s}, 2 \mathrm{H}), 3.98(\mathrm{t}, 4 \mathrm{H}), 3.32(\mathrm{~m}$, $12 \mathrm{H}), 3.17(\mathrm{~m}, 4 \mathrm{H}), 1.91(\mathrm{~m}, 4 \mathrm{H}), 1.81(\mathrm{~m}, 4 \mathrm{H}), 1.73(\mathrm{~m}, 4 \mathrm{H}), 1.58(\mathrm{~m}, 4 \mathrm{H}) 1.39(\mathrm{t}, 18 \mathrm{H})$.

Monomer 2. 1,4-bis((6-bromohexyl)oxy)-2,5-diethynylbenzene ( $484 \mathrm{mg}, 1 \mathrm{mmol}$ ) was suspended in $40 \%$ trimethylamine in water $(20 \mathrm{~mL})$, ethanol $(30 \mathrm{~mL})$ and acetone $(30 \mathrm{~mL})$. The suspension was heated to $130^{\circ} \mathrm{C}$ for 24 hours. The solution was concentrated to yield a solid, which was recrystallized from ethanol/ether (yield: $485 \mathrm{mg}, 92.1 \%) .{ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CH}_{3} \mathrm{DO}, \delta_{\mathrm{ppm}}\right)$ : $7.01(\mathrm{~s}, 2 \mathrm{H}), 3.98(\mathrm{t}, 4 \mathrm{H}), 3.76(\mathrm{~s}$, $2 \mathrm{H}), 3.32(\mathrm{~m}, 12 \mathrm{H}), 3.20(\mathrm{~m}, 4 \mathrm{H}), 1.81(\mathrm{~m}, 4 \mathrm{H}), 1.70(\mathrm{~m}, 4 \mathrm{H}), 1.63(\mathrm{~m}, 4 \mathrm{H}), 1.48(\mathrm{~m}, 4 \mathrm{H}) 1.29(\mathrm{t}, 18 \mathrm{H})$.

CPEs synthesis. The backbone of P1 and P2 is synthesized via Sonogashira protocol. ${ }^{1,2}$ P3 was provided by Prof. Kirk S. Schanze in University of Florida. ${ }^{3}$ The synthesis of P4, P5 and P6 were referred to the method previously reported. ${ }^{4}$ The backbone of P1 and P2 is synthesized as follow. A solution of monomeric building blocks of diacetylenes $(0.2 \mathrm{mmol})$ and monomeric building blocks of bis(iodobenzenes) ( 0.2 mmol ) in 20 mL of dry $\mathrm{THF} / \mathrm{Et}_{3} \mathrm{~N}(\mathrm{~V} / \mathrm{V}=2 / 1)$ fitted with a condenser were degassed for 30 minutes. Then 12 mg of $\mathrm{Pd}\left(\mathrm{PPh}_{3}\right)_{4}$ and 5.3 mg of CuI were added under argon protection. The reaction mixture was stirred at $50{ }^{\circ} \mathrm{C}$ for 24 hr . The obtained reaction solution was poured into 300 mL of ethyl ether, and the precipitation were further purified by two repeated cycles of dissolution in THF and precipitation into ethyl ether and light yellow solid was collected and dried.


Fig. S1. ${ }^{1} \mathrm{H}$ NMR characterization of P 1 and P 2

## 3. Training matrix

Table S1. Training matrix of fluorescence response patterns obtained by six-CPE sensor array against NACs at $1 \mu \mathrm{M}$ concentration.

|  | P1 | P2 | P3 | P4 | P5 | P6 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| C-1 | 16.14149 | 0.410266 | 60.75646 | -0.19711 | 12.2807 | 24.69136 |
| C-1 | 15.51699 | 9.178514 | 60.8473 | 2.1682 | 30.07519 | 4.938272 |
| C-1 | 14.07732 | -7.12718 | 60.07928 | -9.13272 | 30.07519 | 25.92593 |
| C-1 | 13.06192 | 1.059059 | 57.61004 | -6.11038 | 22.80702 | -17.284 |
| C-1 | 15.99847 | 3.186719 | 59.39384 | 6.30749 | 15.03759 | -9.87654 |
| C-1 | 19.30686 | 3.659002 | 54.11677 | 5.978975 | 31.82957 | -17.284 |
| C-2 | 41.73802 | 23.89904 | 78.79274 | -1.89107 | 34.87179 | 34.14634 |
| C-2 | 41.95335 | 29.12499 | 83.75178 | 4.387292 | 51.79487 | 29.26829 |
| C-2 | 35.52422 | 15.73546 | 85.2208 | -4.23601 | 52.05128 | 39.02439 |
| C-2 | 40.44604 | 22.47997 | 84.20584 | 0.226929 | 38.20513 | 13.41463 |
| C-2 | 37.85696 | 19.96125 | 83.2265 | 2.950076 | 34.10256 | 23.17073 |
| C-2 | 40.29736 | 23.16594 | 83.99217 | 1.2548 | 51.28205 | 27.80488 |
| C-3 | 0.325309 | 1.244795 | 14.74688 | -2.77008 | 16.06218 | 1.538462 |
| C-3 | 3.948751 | 3.528411 | 16.90163 | -1.45429 | 1.036269 | -16.9231 |
| C-3 | 7.542165 | 4.858281 | 7.505607 | -5.60942 | -12.9534 | -60 |
| C-3 | 7.276913 | 1.741817 | 9.836591 | -6.30194 | 4.663212 | -18.4615 |
| C-3 | 2.342225 | 3.680652 | 11.13425 | -9.34903 | -4.40415 | -46.1538 |
| C-3 | -1.55147 | 3.250795 | 7.777956 | -9.8338 | 15.54404 | -15.3846 |
| C-4 | 40.68354 | 45.48809 | 82.25708 | 10.52632 | 35.43046 | 43.36283 |
| C-4 | 38.77676 | 48.15489 | 81.87633 | 16.06648 | 23.84106 | 16.81416 |
| C-4 | 36.50758 | 42.46703 | 81.60219 | 16.34349 | 18.21192 | 53.09735 |
| C-4 | 35.7326 | 45.02067 | 80.52087 | 29.08587 | 9.933775 | 28.17 |
| C-4 | 37.90143 | 45.16335 | 80.96253 | 24.37673 | 18.87417 | 22.12389 |
| C-4 | 40.20963 | 45.24208 | 77.15504 | 32.68698 | 6.291391 | 33.62832 |
| C-5 | 34.87454 | 37.12966 | 75.70617 | 12.38671 | 43.40528 | -4.34783 |
| C-5 | 37.83457 | 36.87534 | 75.73567 | 17.46224 | 36.21103 | -18.8406 |
| C-5 | 41.65428 | 41.04183 | 71.31794 | 15.34743 | 53.23741 | -34.7826 |
| C-5 | 39.79554 | 38.67249 | 71.06719 | 16.97885 | 39.08873 | -17.3913 |
| C-5 | 36.99814 | 40.85534 | 75.6693 | 17.82477 | 51.31894 | -14.4928 |
| C-5 | 33.88011 | 39.06243 | 74.84328 | 18.67069 | 23.98082 | -10.1449 |
| C-6 | 50.10756 | 44.73672 | 69.45859 | 33.83743 | 33.50384 | 28.07018 |
| C-6 | 47.44168 | 49.13181 | 75.35846 | 39.69754 | 50.12788 | 45.02924 |
| C-6 | 51.73577 | 43.62148 | 76.00094 | 34.73535 | 36.82864 | 36.54971 |
| C-6 | 50.42393 | 47.05661 | 75.71104 | 38.32703 | 29.92327 | 20.46784 |
| C-6 | 50.99338 | 47.59305 | 73.85411 | 42.91115 | 44.75703 | 28.07018 |
| C-6 | 51.24225 | 47.56953 | 75.14691 | 40.59546 | 46.03581 | 28.07018 |
| C-7 | 17.87052 | 11.52639 | 64.1996 | 3.148528 | 39.38619 | 16.4557 |
| C-7 | 20.14988 | 13.53305 | 64.3992 | 7.665982 | 31.71355 | 17.72152 |
| C-7 | 20.86803 | 12.52496 | 58.38723 | 4.791239 | 35.54987 | 29.11392 |
| C-7 | 16.26769 | 13.12411 | 56.84631 | 2.464066 | 36.31714 | 16.4557 |
| C-7 | 11.69858 | 15.84403 | 61.58084 | 3.627652 | 39.8977 | 26.58228 |


| C-7 | 14.10283 | 12.30147 | 61.89222 | 10.13005 | 25.57545 | 13.92405 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| C-8 | 32.37006 | 19.34727 | 67.00985 | 31.27985 | 37.80161 | -11.6402 |
| C-8 | 35.596 | 21.2372 | 67.61721 | 32.85578 | 30.8311 | -31.746 |
| C-8 | 38.21938 | 24.12116 | 63.45456 | 33.09456 | 15.81769 | -9.52381 |
| C-8 | 36.61666 | 23.75427 | 60.95104 | 33.33333 | 38.3378 | -20.6349 |
| C-8 | 33.53828 | 24.73976 | 66.11362 | 30.22923 | 31.36729 | -41.2698 |
| C-8 | 28.87359 | 21.86433 | 65.34331 | 30.99331 | 19.57105 | -22.963 |
| C-9 | 76.10861 | 66.21073 | 93.45402 | 40.78254 | 70.78086 | 30 |
| C-9 | 76.67056 | 65.16686 | 94.36342 | 43.34086 | 70.27708 | -13.3333 |
| C-9 | 77.47733 | 66.56039 | 92.13281 | 39.50339 | 69.01763 | 28.33333 |
| C-9 | 77.00996 | 65.88163 | 91.75532 | 45.89917 | 70.27708 | 33.33333 |
| C-9 | 75.91387 | 64.90461 | 93.80577 | 42.51317 | 75.81864 | 26.66667 |
| C-9 | 75.29071 | 65.37769 | 93.96877 | 46.87735 | 64.98741 | -6.66667 |

Table S2. Training matrix of fluorescence response patterns obtained by six-PPE sensor array $\left(A_{420}=0.1\right)$ against nine NACs $\left(A_{300}=0.05\right)$

|  | P1 | P2 | P3 | P4 | P5 | P6 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| C-1 | 46.91355 | 20.14826 | 88.68053 | 16.99127 | 74.9643 | 47.43548 |
| C-1 | 47.60584 | 20.47008 | 91.69224 | 1.61182 | 73.78886 | 46.84744 |
| C-1 | 48.81734 | 19.77943 | 93.1299 | 9.301545 | 80.18236 | 48.6769 |
| C-1 | 44.75976 | 18.78141 | 93.49358 | 3.626595 | 71.92134 | 51.06174 |
| C-1 | 46.54817 | 19.43591 | 92.85146 | 4.231028 | 75.85411 | 41.62039 |
| C-1 | 49.69369 | 24.65015 | 92.7094 | 11.98791 | 77.47995 | 47.04345 |
| C-2 | 72.96395 | 48.38693 | 95.59267 | 39.06916 | 80.50327 | 59.09456 |
| C-2 | 73.95948 | 47.94212 | 95.12436 | 35.39823 | 77.0307 | 59.25529 |
| C-2 | 73.9827 | 49.24587 | 92.72037 | 26.58145 | 78.75189 | 55.42459 |
| C-2 | 73.41383 | 48.98512 | 92.39255 | 42.74009 | 79.39607 | 50.12055 |
| C-2 | 73.28612 | 47.70694 | 93.93277 | 35.2999 | 76.69854 | 54.43343 |
| C-2 | 72.51988 | 46.77642 | 93.75585 | 33.56277 | 80.5536 | 53.79052 |
| C-3 | 44.80615 | 36.78269 | 77.87731 | 12.14393 | 48.13953 | 18.4257 |
| C-3 | 42.16322 | 37.83284 | 82.18941 | 12.27511 | 45.41004 | 18.95517 |
| C-3 | 41.0801 | 35.77581 | 82.89368 | 9.557721 | 48.31089 | 15.81363 |
| C-3 | 40.59134 | 35.28024 | 83.76475 | 13.71814 | 44.23501 | 16.3431 |
| C-3 | 40.75426 | 36.39725 | 81.36159 | 12.25637 | 46.29131 | 17.3844 |
| C-3 | 42.03047 | 40.22812 | 81.76932 | 13.56822 | 15.49572 | 11.01306 |
| C-4 | 93.10798 | 86.32544 | 98.71495 | 69.65075 | 90.744 | 67.98277 |
| C-4 | 93.28552 | 86.5391 | 98.7466 | 73.38638 | 88.45372 | 64.71644 |
| C-4 | 93.38849 | 86.64831 | 98.41109 | 70.55703 | 87.46443 | 61.37832 |
| C-4 | 93.37429 | 87.01391 | 98.39843 | 66.62246 | 88.44017 | 54.66619 |
| C-4 | 93.44175 | 86.65306 | 98.59467 | 73.34218 | 86.32606 | 60.58866 |
| C-4 | 93.40269 | 87.30355 | 98.63265 | 68.08134 | 90.82532 | 59.58363 |
| C-5 | 91.31916 | 85.31034 | 98.20271 | 67.5162 | 92.36682 | 63.44302 |
| C-5 | 91.22377 | 85.31969 | 98.6256 | 62.03024 | 90.67055 | 62.80096 |
| C-5 | 90.76043 | 85.03459 | 98.71018 | 66.69546 | 93.37397 | 62.43981 |


| C-5 | 90.83197 | 85.09534 | 98.72427 | 65.78834 | 91.28015 | 66.53291 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| C-5 | 90.9955 | 85.69359 | 98.67494 | 64.44924 | 91.41267 | 54.69502 |
| C-5 | 91.48269 | 86.47878 | 98.57626 | 63.84449 | 69.53353 | 55.25682 |
| C-6 | 94.94334 | 93.52408 | 99.37436 | 84.4883 | 95.48328 | 80.55774 |
| C-6 | 94.76766 | 93.17597 | 99.35981 | 84.71776 | 93.79774 | 76.9835 |
| C-6 | 94.60577 | 93.10443 | 99.35254 | 85.45204 | 95.36476 | 80.63629 |
| C-6 | 94.6471 | 92.70386 | 99.33799 | 85.86508 | 94.46932 | 80.71485 |
| C-6 | 94.778 | 93.29041 | 99.36709 | 86.00275 | 95.27258 | 79.02592 |
| C-6 | 94.98123 | 93.83882 | 99.35254 | 85.91097 | 88.72794 | 77.76905 |
| C-7 | 71.25449 | 55.64494 | 95.45531 | 37.83098 | 85.70931 | 73.80451 |
| C-7 | 71.84945 | 55.53899 | 95.00252 | 36.27131 | 80.43882 | 73.59398 |
| C-7 | 72.12306 | 57.42652 | 91.59819 | 27.20348 | 81.42559 | 71.12782 |
| C-7 | 71.67446 | 57.98768 | 91.26279 | 37.21436 | 83.07407 | 66.16541 |
| C-7 | 71.24177 | 58.70973 | 93.1634 | 25.2811 | 78.46529 | 68.66165 |
| C-7 | 70.92997 | 62.66138 | 92.64911 | 27.5662 | 84.32784 | 64.27068 |
| C-8 | 87.09509 | 78.67633 | 98.29909 | 43.51554 | 90.70468 | 89.39176 |
| C-8 | 87.29429 | 78.20638 | 98.18312 | 65.05895 | 91.73621 | 88.38282 |
| C-8 | 87.29429 | 77.8226 | 96.59819 | 61.1647 | 91.87529 | 87.66215 |
| C-8 | 86.87138 | 78.41394 | 96.3552 | 58.30654 | 89.01252 | 84.11646 |
| C-8 | 87.11961 | 78.03015 | 97.38237 | 66.3094 | 91.73621 | 85.47132 |
| C-8 | 86.66605 | 77.86959 | 97.28297 | 56.66309 | 91.5044 | 84.3759 |
| C-9 | 98.97053 | 99.41635 | 99.83441 | 93.35562 | 96.49972 | 91.96694 |
| C-9 | 98.94878 | 99.4307 | 99.83441 | 94.19139 | 96.40288 | 92.49587 |
| C-9 | 99.02128 | 99.44984 | 99.84714 | 94.31676 | 96.48589 | 91.63636 |
| C-9 | 99.05028 | 99.46419 | 99.84077 | 93.85708 | 96.52739 | 90.87603 |
| C-9 | 99.05753 | 99.44506 | 99.82804 | 94.44212 | 95.91865 | 91.33884 |
| C-9 | 99.1844 | 99.47376 | 99.84077 | 94.5257 | 96.59657 | 90.90909 |

Table S3. Training matrix of fluorescence response patterns obtained by six-PPE sensor array $\left(\mathrm{A}_{420}=\mathbf{0 . 1}\right)$ against three unknown samples ( $\mathrm{A}_{300}=\mathbf{0 . 0 5}$ ).

|  | P1 | P2 | P3 | P4 | P5 | P6 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| A | 98.80454 | 99.40763 | 99.81507 | 96.16766 | 97.31016 | 91.76265 |
| B | 42.05178 | 30.76349 | 74.59006 | 7.225549 | 46.76185 | -2.75742 |
| C | 95.18986 | 92.38701 | 99.38355 | 83.47305 | 93.87544 | 73.71728 |
| C-1 | 46.91355 | 20.14826 | 88.68053 | 16.99127 | 74.9643 | 47.43548 |
| C-1 | 47.60584 | 20.47008 | 91.69224 | 1.61182 | 73.78886 | 46.84744 |
| C-1 | 48.81734 | 19.77943 | 93.1299 | 9.301545 | 80.18236 | 48.6769 |
| C-1 | 44.75976 | 18.78141 | 93.49358 | 3.626595 | 71.92134 | 51.06174 |
| C-1 | 46.54817 | 19.43591 | 92.85146 | 4.231028 | 75.85411 | 41.62039 |
| C-1 | 49.69369 | 24.65015 | 92.7094 | 11.98791 | 77.47995 | 47.04345 |
| C-2 | 72.96395 | 48.38693 | 95.59267 | 39.06916 | 80.50327 | 59.09456 |
| C-2 | 73.95948 | 47.94212 | 95.12436 | 35.39823 | 77.0307 | 59.25529 |
| C-2 | 73.9827 | 49.24587 | 92.72037 | 26.58145 | 78.75189 | 55.42459 |


| C-2 | 73.41383 | 48.98512 | 92.39255 | 42.74009 | 79.39607 | 50.12055 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| C-2 | 73.28612 | 47.70694 | 93.93277 | 35.2999 | 76.69854 | 54.43343 |
| C-2 | 72.51988 | 46.77642 | 93.75585 | 33.56277 | 80.5536 | 53.79052 |
| C-3 | 44.80615 | 36.78269 | 77.87731 | 12.14393 | 48.13953 | 18.4257 |
| C-3 | 42.16322 | 37.83284 | 82.18941 | 12.27511 | 45.41004 | 18.95517 |
| C-3 | 41.0801 | 35.77581 | 82.89368 | 9.557721 | 48.31089 | 15.81363 |
| C-3 | 40.59134 | 35.28024 | 83.76475 | 13.71814 | 44.23501 | 16.3431 |
| C-3 | 40.75426 | 36.39725 | 81.36159 | 12.25637 | 46.29131 | 17.3844 |
| C-3 | 42.03047 | 40.22812 | 81.76932 | 13.56822 | 15.49572 | 11.01306 |
| C-4 | 93.10798 | 86.32544 | 98.71495 | 69.65075 | 90.744 | 67.98277 |
| C-4 | 93.28552 | 86.5391 | 98.7466 | 73.38638 | 88.45372 | 64.71644 |
| C-4 | 93.38849 | 86.64831 | 98.41109 | 70.55703 | 87.46443 | 61.37832 |
| C-4 | 93.37429 | 87.01391 | 98.39843 | 66.62246 | 88.44017 | 54.66619 |
| C-4 | 93.44175 | 86.65306 | 98.59467 | 73.34218 | 86.32606 | 60.58866 |
| C-4 | 93.40269 | 87.30355 | 98.63265 | 68.08134 | 90.82532 | 59.58363 |
| C-5 | 91.31916 | 85.31034 | 98.20271 | 67.5162 | 92.36682 | 63.44302 |
| C-5 | 91.22377 | 85.31969 | 98.6256 | 62.03024 | 90.67055 | 62.80096 |
| C-5 | 90.76043 | 85.03459 | 98.71018 | 66.69546 | 93.37397 | 62.43981 |
| C-5 | 90.83197 | 85.09534 | 98.72427 | 65.78834 | 91.28015 | 66.53291 |
| C-5 | 90.9955 | 85.69359 | 98.67494 | 64.44924 | 91.41267 | 54.69502 |
| C-5 | 91.48269 | 86.47878 | 98.57626 | 63.84449 | 69.53353 | 55.25682 |
| C-6 | 94.94334 | 93.52408 | 99.37436 | 84.4883 | 95.48328 | 80.55774 |
| C-6 | 94.76766 | 93.17597 | 99.35981 | 84.71776 | 93.79774 | 76.9835 |
| C-6 | 94.60577 | 93.10443 | 99.35254 | 85.45204 | 95.36476 | 80.63629 |
| C-6 | 94.6471 | 92.70386 | 99.33799 | 85.86508 | 94.46932 | 80.71485 |
| C-6 | 94.778 | 93.29041 | 99.36709 | 86.00275 | 95.27258 | 79.02592 |
| C-6 | 94.98123 | 93.83882 | 99.35254 | 85.91097 | 88.72794 | 77.76905 |
| C-7 | 71.25449 | 55.64494 | 95.45531 | 37.83098 | 85.70931 | 73.80451 |
| C-7 | 71.84945 | 55.53899 | 95.00252 | 36.27131 | 80.43882 | 73.59398 |
| C-7 | 72.12306 | 57.42652 | 91.59819 | 27.20348 | 81.42559 | 71.12782 |
| C-7 | 71.67446 | 57.98768 | 91.26279 | 37.21436 | 83.07407 | 66.16541 |
| C-7 | 71.24177 | 58.70973 | 93.1634 | 25.2811 | 78.46529 | 68.66165 |
| C-7 | 70.92997 | 62.66138 | 92.64911 | 27.5662 | 84.32784 | 64.27068 |
| C-8 | 87.09509 | 78.67633 | 98.29909 | 43.51554 | 90.70468 | 89.39176 |
| C-8 | 87.29429 | 78.20638 | 98.18312 | 65.05895 | 91.73621 | 88.38282 |
| C-8 | 87.29429 | 77.8226 | 96.59819 | 61.1647 | 91.87529 | 87.66215 |
| C-8 | 86.87138 | 78.41394 | 96.3552 | 58.30654 | 89.01252 | 84.11646 |
| C-8 | 87.11961 | 78.03015 | 97.38237 | 66.3094 | 91.73621 | 85.47132 |
| C-8 | 86.66605 | 77.86959 | 97.28297 | 56.66309 | 91.5044 | 84.3759 |
| C-9 | 98.97053 | 99.41635 | 99.83441 | 93.35562 | 96.49972 | 91.96694 |
| C-9 | 98.94878 | 99.4307 | 99.83441 | 94.19139 | 96.40288 | 92.49587 |


| C-9 | 99.02128 | 99.44984 | 99.84714 | 94.31676 | 96.48589 | 91.63636 |
| :--- | :--- | :--- | :--- | :--- | :--- | :--- |
| C-9 | 99.05028 | 99.46419 | 99.84077 | 93.85708 | 96.52739 | 90.87603 |
| C-9 | 99.05753 | 99.44506 | 99.82804 | 94.44212 | 91.33884 |  |
| C-9 | 99.1844 | 99.47376 | 99.84077 | 94.5257 | 96.59657 |  |

## References

1 X. Zhao, M. R. Pinto, L. M. Hardison, J. Mwaura, J. Muller, H. Jiang, D. Witker, V. D. Kleiman, J. R. Reynolds and K. S. Schanze, Macromolecules, 2006, 39, 6355-6366.
2 Y. Wu, Y. Tan, J. Wu, S. Chen, Y. Z. Chen, X. Zhou, Y. Jiang and C. Tan, ACS Appl. Mater. Interfaces, 2015, 7, 6882-6888.
3 L. K. Ista, D. Dascier, E. Ji, A. Parthasarathy, T. S. Corbitt, K. S. Schanze and D. G. Whitten, ACS Appl. Mater. Interfaces, 2011, 3, 2932-2937.
4 Z. Chen, P. Wu, R. Cong, N. Xu, Y. Tan, C. Tan and Y. Jiang, ACS Appl. Mater. Interfaces, 2015.

