

Supplimentary Information

A facile electrochemical synthesis strategy of Cu₂O (cubes, sheets and flowers) microstructured materials for sensitive detection of 4-nitrophenol†

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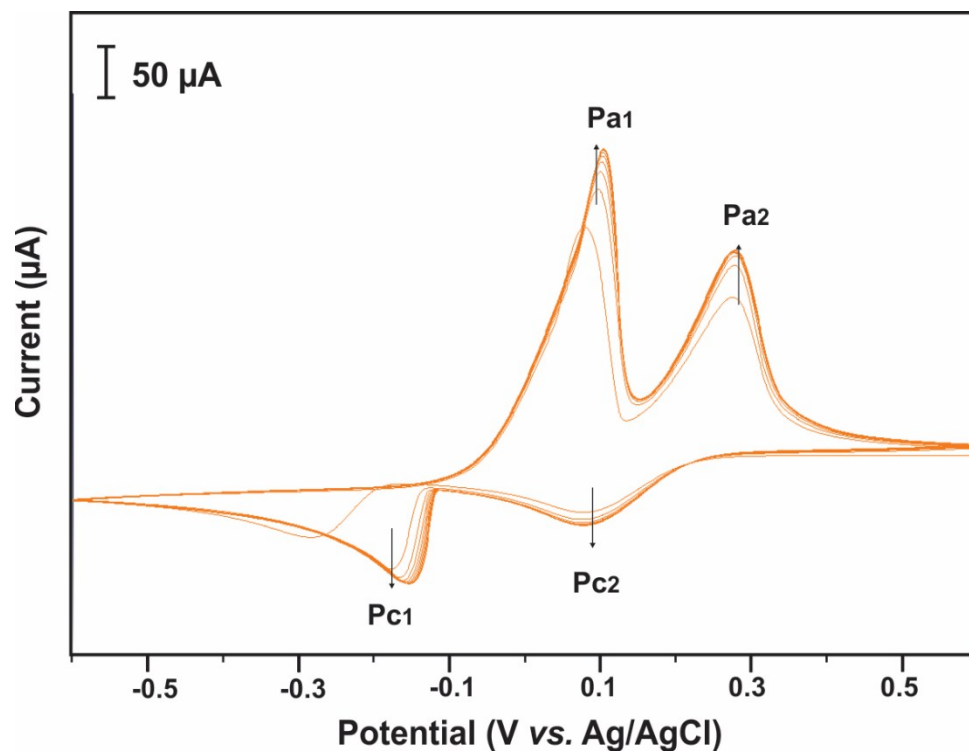


Fig. S1 Consecutive cyclic voltammograms for 10 cycles deposition of Cu_2O on the GCE using 10 mM KCl containing 5 mM of CuCl_2 at a scan rate of 50 mV s^{-1} .

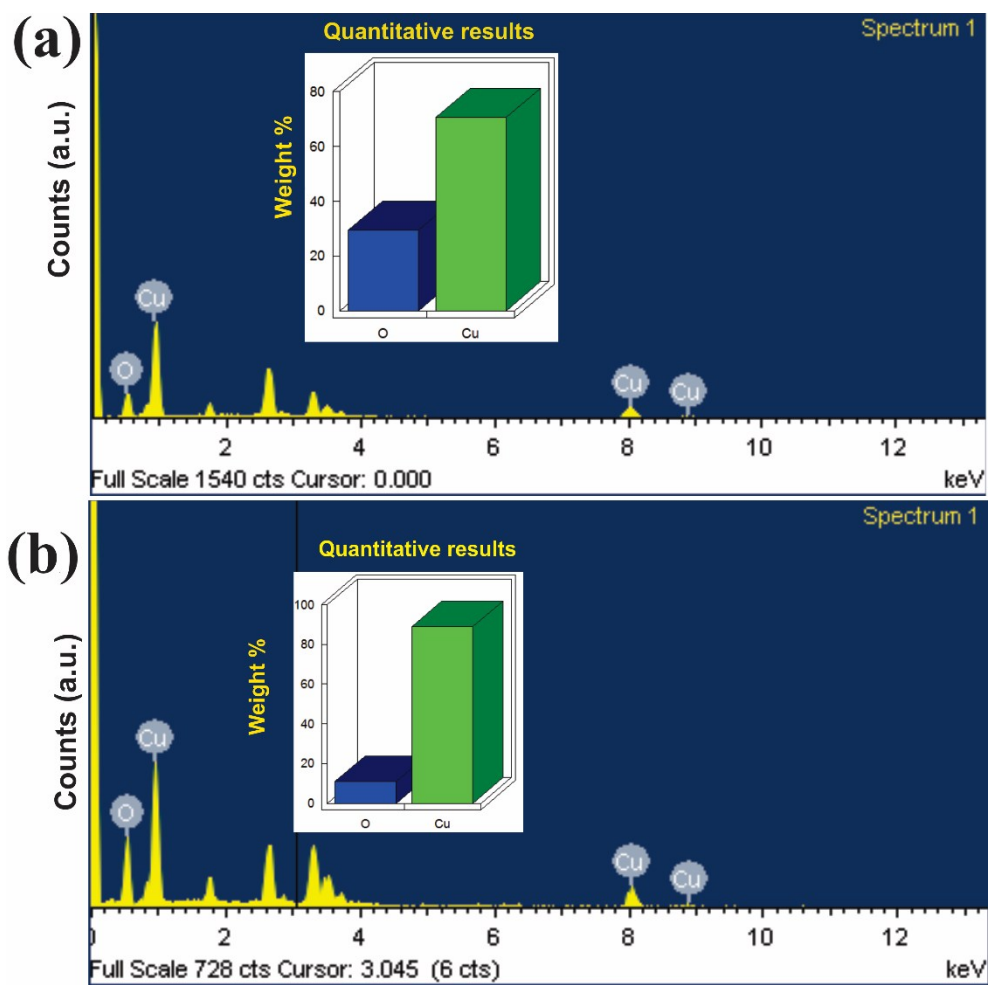


Fig. S2. The EDX spectra for Cu_2O -cubes (a) and Cu_2O -flowers (b). Inset: the corresponding bar diagram.

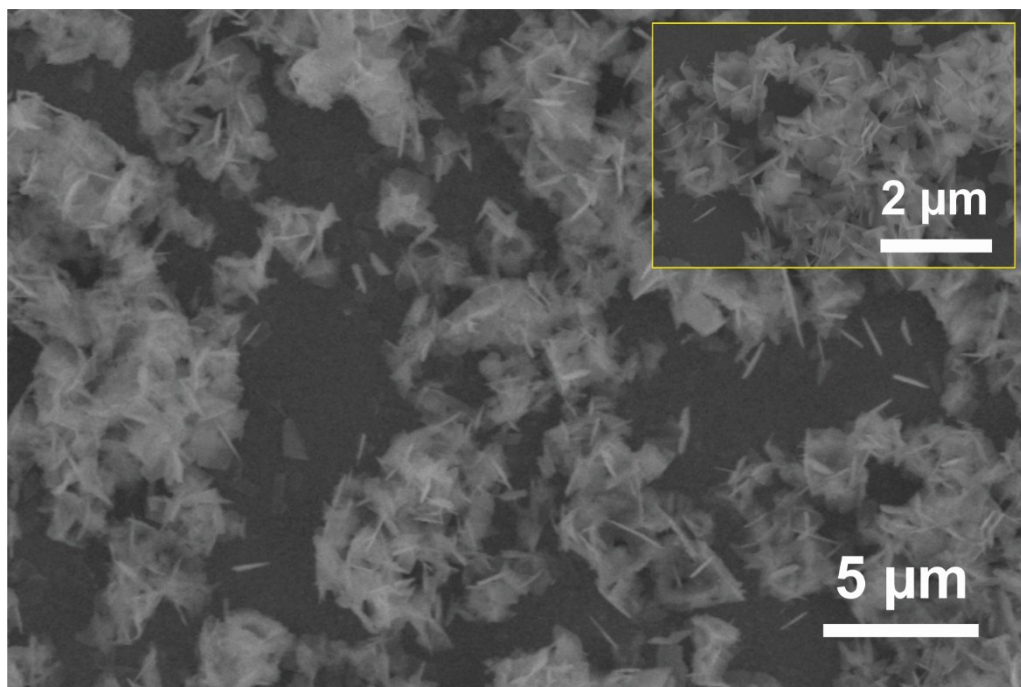


Fig. S3. SEM images of the electrochemically prepared Cu₂O-cubes at 20 cycles.

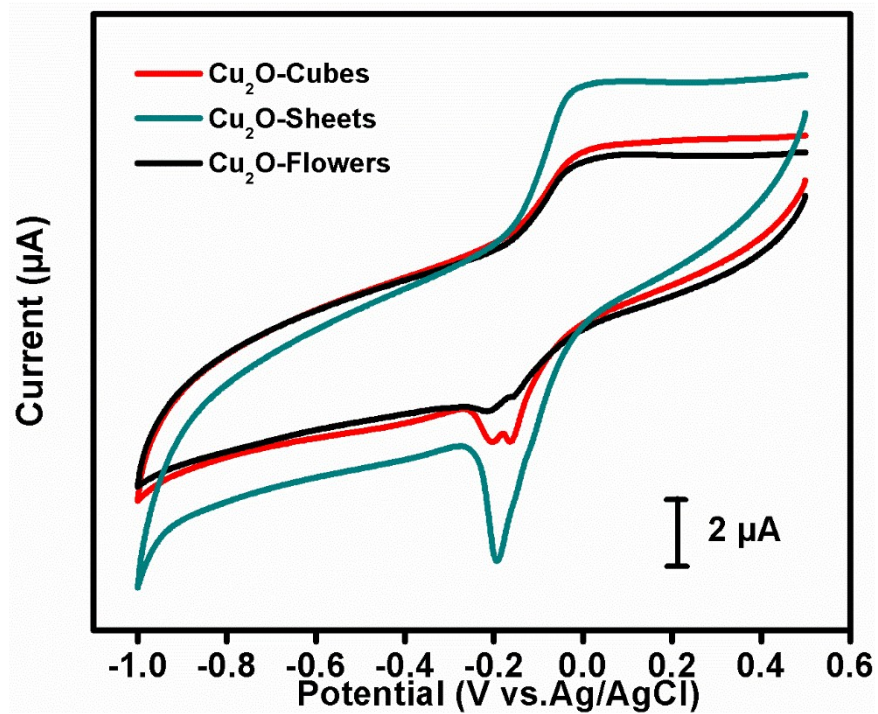


Fig. S4. (a) CV capacitive current profiles for different modified electrodes in absence of 4-NP concentrations in 0.05 M PBS solutions (pH 5.0).

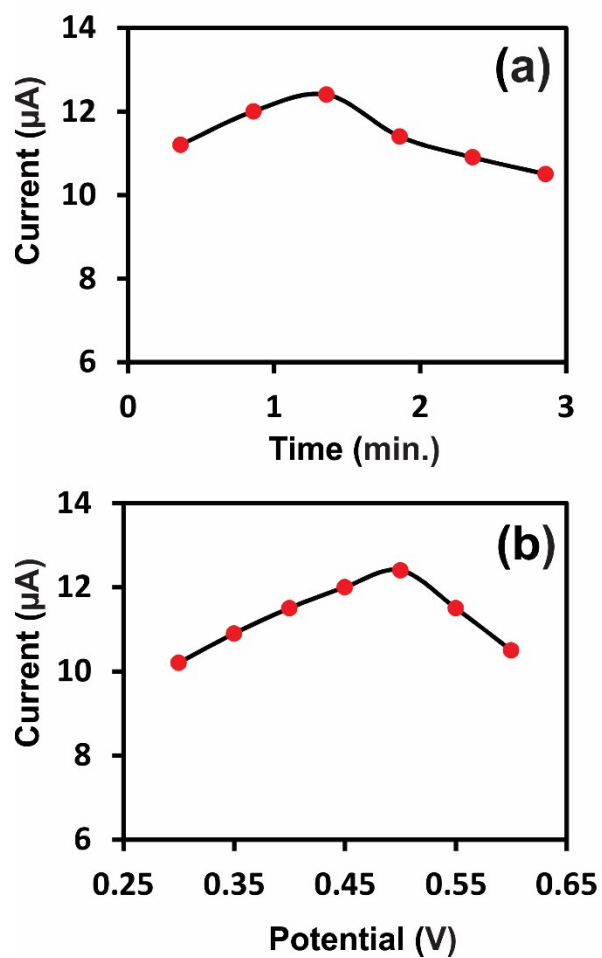


Fig. S4. The Cu₂O modified electrodes during the detection of 150 μ M of 4-NP, (a) Accumulated potential and (b) with variations of peak current.

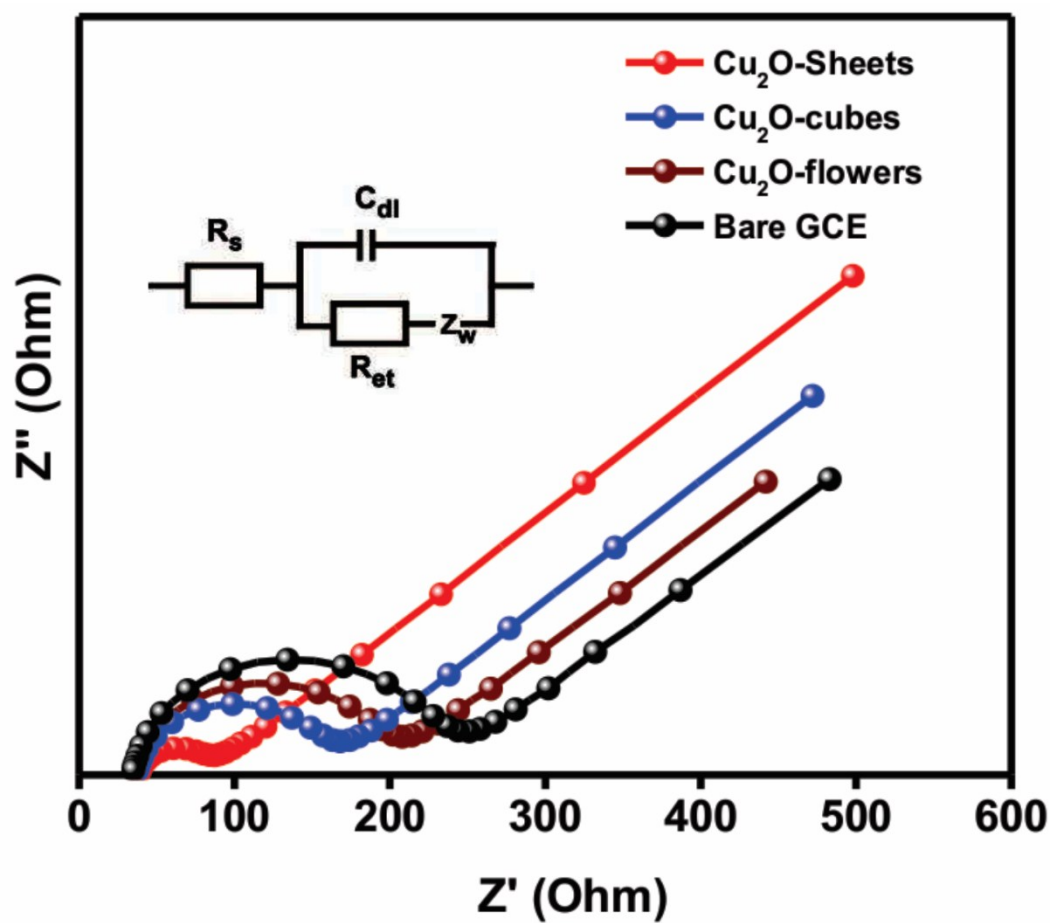


Fig. S5. (a) Nyquist plots at different electrodes by using 5 mM Fe (CN)₆^{3-/4-} containing in 0.1 M KCl as a supporting electrolyte solutions.

Table S1. Determination of 4-NP in various real samples using DPV.

Real samples	Analyte	Added (μM)	Found (μM)	Recovery (%)
Tap water	4-NP	50	49.1	98.2
		140	138.9	99.2
River water	4-NP	50	48.9	97.8
		140	139.5	99.5