Electronic Supplementary Material (ESI) for Analytical Methods.

Supporting Information

Electrochemical sensor based on fullerene nanorods for detection of paraben an endocrine disruptor

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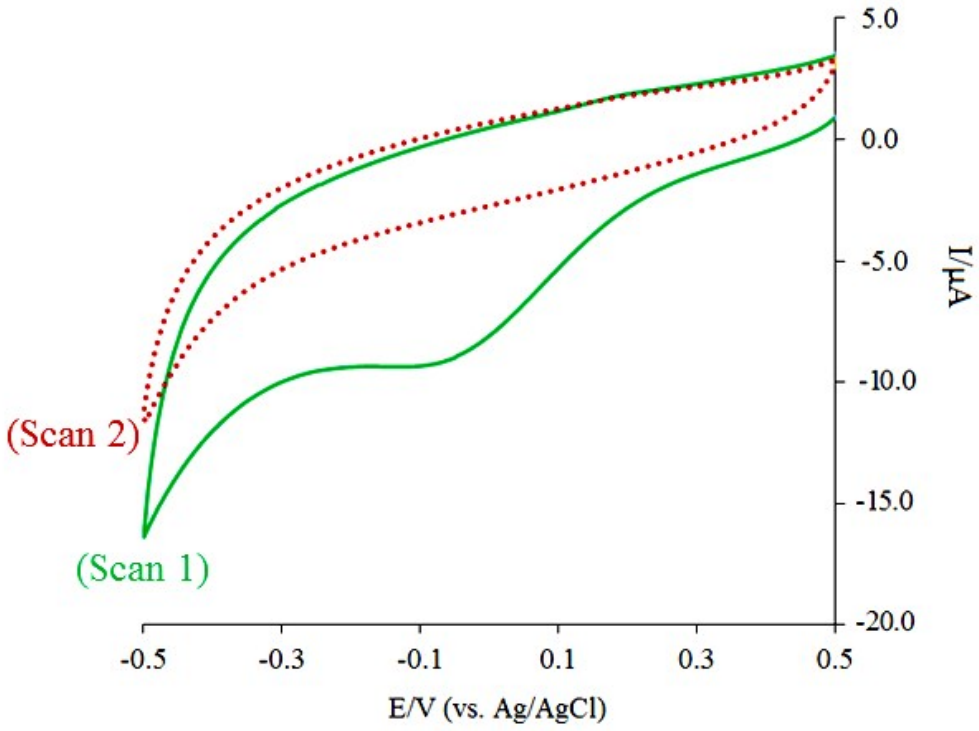

Figure S1: Electrochemical reduction of nitrophenyl diazonium salt ($Cl^-N_2^+$–Ph–NO_2) at GCE (scan rate 100 mVs^{-1}). The reduction peak observed at -0.07 V corresponds to electrochemical reduction of nitrophenyl diazonium salt.

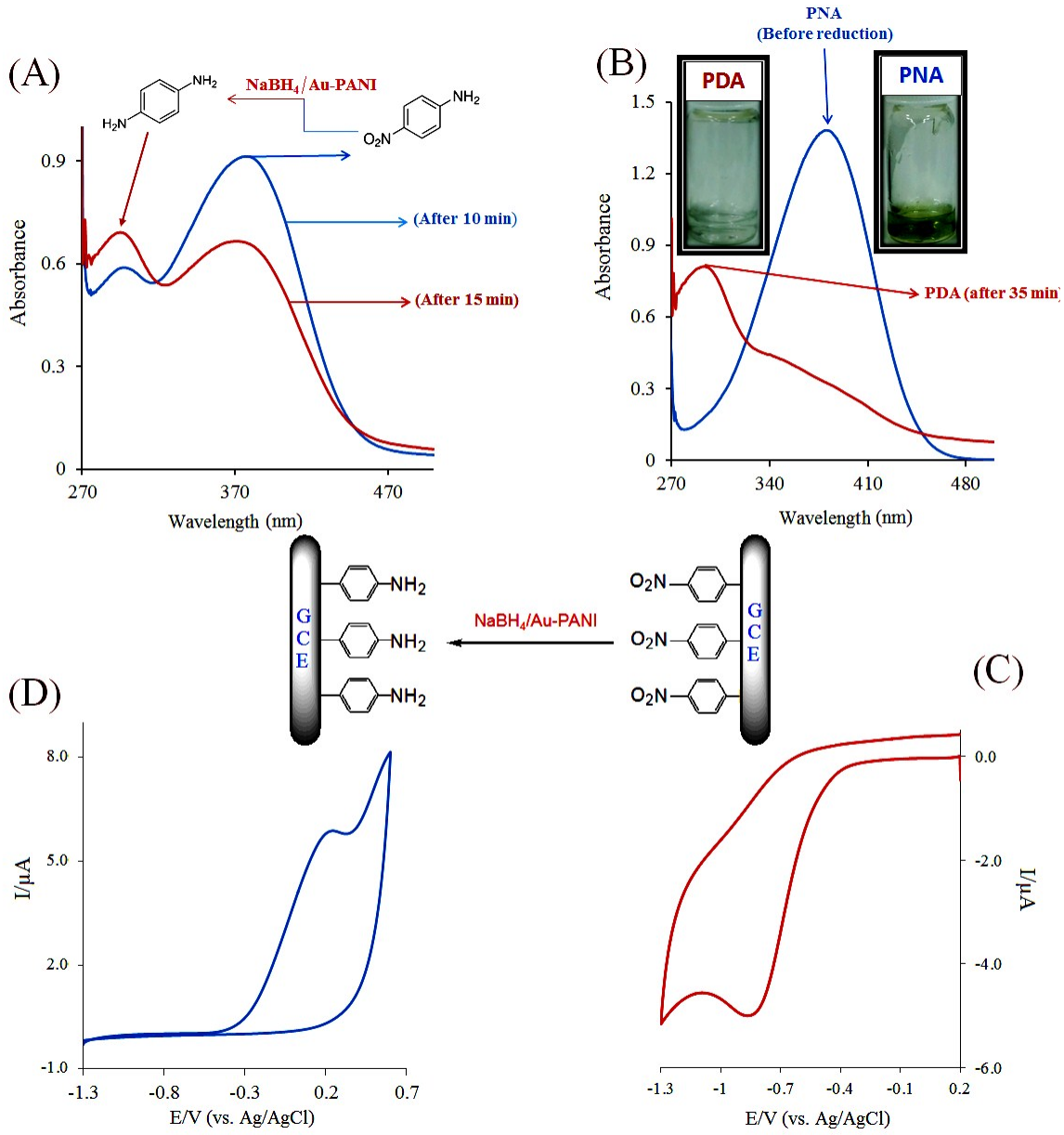

Figure S2: UV-vis spectrum of *p*-nitroaniline (PNA) reduction to *p*-phenylenediamine (PDA) after addition of $NaBH_4$/Au–PANI system observed before (A) and after (B) reduction process. Cyclic voltammograms (CV) are confirming the presence of nitrophenyl group at GCE–Ph–NO_2 (C) and phenylamine group formed at GCE–Ph–NH_2 (D) after reduction with $NaBH_4$/Au–PANI system.

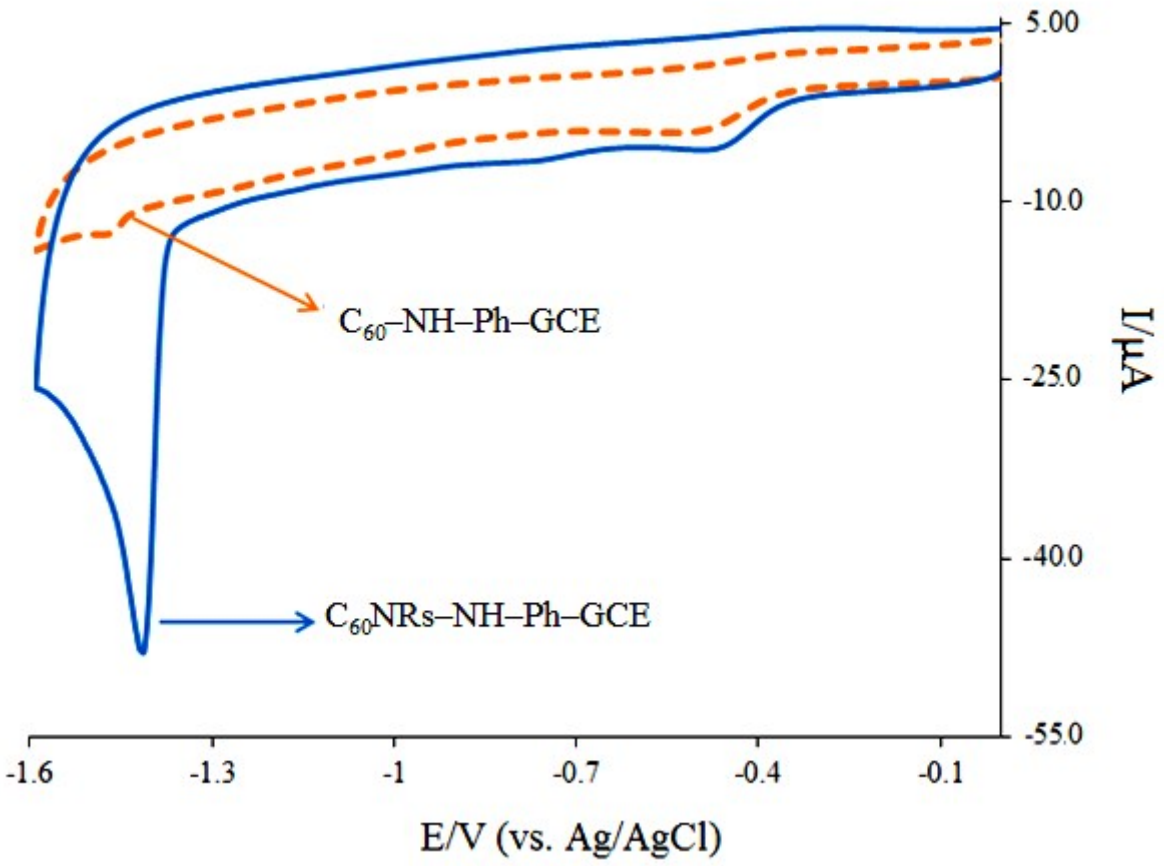

Figure S3: Electrochemical reduction (ER) of C_{60}NRs–NH–Ph–GCE and C_{60}–NH–Ph–GCE in 1.0 M KOH at a scan rate of 10 mVs^{-1}.