ELECTRONIC SUPPLEMENTARY MATERIAL (ESI)

Development of biomimetic sensor for selective identification of cyanide

E.C. Figueira, L.C.S. Neres, M.R.S. Ruy, G.F. Troiano and M.D.P.T. Sotomayor*.

S1. Voltammetric responses to eight different ions used the proposed sensor.



Figure S1: Results obtained to evaluate the interference of the sodium and acetate in the sensor response. Measurements were conducted with a supporting electrolyte of NaOH (pH 12.0), amplitude: 10 mV, frequency: 20 Hz and step potential: 10 mV. The concentrations corresponding to successive addition of sodium acetate.



Figure S2: Results obtained to evaluate the interference of the borate in the sensor response. Measurements were conducted with a supporting electrolyte of NaOH (pH 12.0), amplitude: 10 mV, frequency: 20 Hz and step potential: 10 mV. The concentrations corresponding to successive addition of sodium borate.



Figure S3: Results obtained to evaluate the interference of the Ca²⁺ and Cl⁻ in the sensor response. Measurements were conducted with a supporting electrolyte of NaOH (pH 12.0), amplitude: 10 mV, frequency: 20 Hz and step potential: 10 mV. The concentrations corresponding to successive addition of calcium chloride.



Figure S4: Results obtained to evaluate the interference of the Pb²⁺ and NO₃⁻ in the sensor response. Measurements were conducted with a supporting electrolyte of NaOH (pH 12.0), amplitude: 10 mV, frequency: 20 Hz and step potential: 10 mV. The concentrations corresponding to successive addition of calcium chloride.



Figure S5: Results obtained to evaluate the interference of the NH₄⁺ and MoO₄²⁻ in the sensor response. Measurements were conducted with a supporting electrolyte of NaOH (pH 12.0), amplitude: 10 mV, frequency: 20 Hz and step potential: 10 mV. The concentrations corresponding to successive addition of calcium chloride.